



1 **Tolerance of soil bacterial community to tetracycline**
2 **antibiotics induced by As, Cd, Zn, Cu, Ni, Cr and Pb pollution**

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13
14 **Abstract.** The widespread use of both heavy metals and antibiotics in livestock farming and their subsequent
15 arrival on agricultural soils through manure/slurry spreading has become a problem of vital importance for human
16 health and the environment. In the current research, a laboratory experiment was carried out for 42 days to study
17 co-selection for tolerance of three tetracycline antibiotics (tetracycline, TC; oxytetracycline, OTC;
18 chlortetracycline, CTC) in soils polluted with heavy metals (As, Cd, Zn, Cu, Ni, Cr and Pb) at high concentration
19 levels (1000 mg kg⁻¹ of each one, separately). Pollution Induced Community Tolerance (PICT) of the bacterial
20 community was estimated using the leucine incorporation technique. The Log IC₅₀ (logarithm of the concentration
21 causing 50% inhibition in bacterial community growth) values obtained in uncontaminated soil samples for all the
22 heavy metals tested showed the following toxicity sequence: Cu>As>Cr≥Pb≥Cd>Zn>Ni. However, in polluted
23 soil samples the toxicity sequence was: Cu>Pb≥As≥Cd≥Cr≥Ni≥Zn. Moreover, at high metal concentrations the
24 bacterial communities show tolerance to the metal itself, this taking place for all the metals tested in the long term.
25 The bacterial communities of the soil polluted with heavy metals showed also long-term co-tolerance to TC, OTC,
26 and CTC. This kind of studies, focusing on the eventual increases of tolerance and co-tolerance of bacterial
27 communities in agricultural soil, favored by the presence of other pollutants, is of crucial importance, mostly
28 bearing in mind that the appearance of antibiotic resistance genes in soil bacteria could be transmitted to human
29 pathogens.



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32 **Keywords:** Chlortetracycline; heavy metals; trace elements; oxytetracycline; PICT; tetracycline

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34 1. Introduction

35 The accumulation of heavy metals in soils is a widespread problem all over the world. Some
36 anthropogenic activities such as mining, industrial production, agriculture, or livestock farming, are among the
37 main sources of release of heavy metals to the terrestrial environment (Abdu et al., 2017). The use of heavy
38 metals in livestock production is a common practice since commercial feeds are enriched with essential
39 elements (such as Cu and Zn) to prevent diseases and obtain an optimal growth rate (Sager, 2007; Wu et al.,
40 2013). Moreover, veterinary antibiotics are also widely used for the treatment of bacterial infections (Boxal et
41 al., 2003) and their consumption has been increasing in recent decades, with the estimate of world use for 2030
42 being 105,596 tons (Van Boeckel et al., 2015). The antibiotics most widely used in veterinary medicine in the
43 European Union are tetracyclines (TCs), and specifically tetracycline (TC), oxytetracycline (OTC) and
44 chlortetracycline (CTC) (European Medicines Agency, 2016). Both heavy metals and antibiotics for veterinary
45 use are poorly absorbed by the intestines of animals, therefore, a high percentage of them are expelled in feces
46 and urine (Kornegay et al., 1976; Sarmah et al., 2006). The presence of high levels of heavy metals and
47 veterinary antibiotics in farmland soils is due to the repeated applications of manure and slurries (as well as
48 sewage sludge) as organic fertilizers (Hamscher et al., 2002; Nicholson et al., 2003). Once in the soil, these
49 compounds can interact with soil microbial communities and modify their structure and function (Hattori,
50 1992; Thiele-Bruhn and Beck, 2005; Chien et al., 2008; Giller et al., 2009; Caban et al., 2018).

51 The increase of the concentration of any pollutant in soils may suppose a selection pressure for soil
52 bacterial communities, causing tolerance to that pollutant (Blanck, 2002). This effect may be useful to quantify
53 the harmful effects produced by pollutants on soil bacteria and is called pollution-induced community tolerance
54 (PICT). Agricultural soils, highly influenced by anthropogenic activities, deserve special attention since they
55 are recognized as the largest reservoirs of antibiotic-resistant genes, receiving antibiotics from veterinary use
56 (through repeated applications of manure and slurries, as indicated above), as well as heavy metals (Ji et al.,
57 2012). Therefore, the resistance of soil bacterial communities to antibiotics has become a crucial threat at a
58 world scale, and the study of whether bacterial communities generate co-tolerance to antibiotics in the presence
59 of heavy metals (as well as how and in which degree is developed) is of vital importance.

60 Several previous studies have focused on the co-tolerance among different heavy metals and
61 antibiotics (Sarma et al., 2010; Fernández-Calviño and Bååth, 2013; Song et al., 2017; Zhong et al., 2021),
62 however, until now, there were not studies evaluating the effect of a wide range of different heavy metals on
63 the tolerance shown by soil bacterial communities against each of the three tetracycline antibiotics most used



64 in livestock. Therefore, the objective of this study is to determine the eventual development of tolerance on
65 soil bacterial communities to heavy metals in agricultural soils contaminated individually with As, Cd, Zn, Cu,
66 Ni, Cr and Pb, and also the eventual generation of co-tolerance to the antibiotics tetracycline, oxytetracycline
67 and chlortetracycline, using the leucine incorporation technique as the endpoint. The results of this research
68 could be relevant in order to define appropriate management practices for wastes and fertilizers in agricultural
69 soils.

70

71 **2. Material and methods**

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73 **2.1. Chemicals**

74 Tetracycline hydrochloride (CAS 64-75-5; $\geq 95\%$ in purity), oxytetracycline hydrochloride (CAS
75 2058-46-0; $\geq 95\%$ in purity) and chlortetracycline hydrochloride (CAS 64-72-2; $\geq 97\%$ in purity), all three
76 supplied by Sigma–Aldrich (Steinheim, Germany), were used for soil spiking.

77 The heavy metals added to the soil samples for determining bacterial community tolerance were
78 arsenic (as $\text{Na}_2\text{HAsO}_4 \cdot 7\text{H}_2\text{O}$, CAS; 10048-95-0), cadmium (as $\text{Cd}(\text{NO}_3)_2 \cdot 4\text{H}_2\text{O}$, CAS 10022-68-1), zinc (as
79 $\text{Zn}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$, CAS; 10196-18-6), copper (as $\text{Cu}(\text{NO}_3)_2 \cdot 2.5\text{H}_2\text{O}$, CAS 19004-19-4), nickel (as $\text{Ni}(\text{NO}_3)_2$
80 $\cdot 6\text{H}_2\text{O}$, CAS 13478-00-7), chromium (as $\text{K}_2\text{Cr}_2\text{O}_7$, CAS 7778-60-9) and lead (as $\text{Pb}(\text{NO}_3)_2$, CAS 10099-74-8),
81 all of them supplied by Panreac (Barcelona, Spain).

82

83 **2.2. Soil samples**

84 An agricultural soil from Sarria (NW of Spain) was selected from a set of soils previously analyzed
85 by Conde-Cid et al. (2018). Moreover, this soil was previously used for measurement of the bacterial growth
86 after being polluted with antibiotics (tetracycline, oxytetracycline and chlortetracycline), with results shown in
87 Santás-Miguel et al. (2020a) and Santás-Miguel et al. (2020b).

88 The main characteristic of the soil studied are shown in Table S1 (Supplementary Material). Briefly,
89 its texture was silt loam, with pH in water of 6.0 and pH in KCl (0.1 M) of 5.2. The organic carbon (C) and
90 total nitrogen (N) contents were 1.8% and 0.2%, respectively and the effective cation exchange capacity
91 (eCEC) was $13.16 \text{ cmol}_c \text{ kg}^{-1}$.

92

93



94 **2.3. Experimental design**

95 An amount of 42 g of the air-dried soil was placed in polypropylene tubes (500 mL) and then
96 moistened up to 60-80% of water holding capacity and incubated at 22 °C in the dark for 1 week, this time
97 being adequate for the recovery and stabilization of the growth of the bacterial communities of the soil after
98 moisture adjustment (Meisner et al., 2013). After this time, the soil sample was distributed in 8 polypropylene
99 tubes (100 mL), with 7 of them being individually polluted with 1000 mg kg⁻¹ of one of the 7 heavy metals
100 (As, Cd, Zn, Cu, Ni, Cr, Pb), and the 8th tube acting as control (soil without metal). Then, the 8 microcosms
101 were distributed in 24 polypropylene tubes (50 mL) (8 microcosms x 3 replicates), placing 1.75 g (dry weight)
102 in each tube, and were incubated in the dark for 42 days.

103 The toxicity on the bacterial communities exerted individually by the heavy metals was measured by
104 estimating the bacterial growth on days 1 and 42 of incubation, following the protocol established by Bååth
105 (1994) and Bååth et al. (2001). The bacterial community tolerance was determined according to Bååth (1992),
106 and to Díaz-Raviña et al. (1994), with certain modifications indicated below, and using the leucine
107 incorporation method (Bååth, 1994; Bååth et al., 2001). Briefly, soil samples obtained from each microcosm
108 were mixed with distilled water (rate 1:20 (w/v)), using a multi-vortex shaker at maximum intensity for 3 min.
109 Then, the soil/water mixture was centrifuged at 1000 x g for 10 min to obtain the soil bacterial suspension
110 (supernatant). Aliquots of 1.50 mL of the supernatant were transferred to 2 mL centrifugation tubes, in which
111 150 µL of pollutant solutions (containing heavy metals or antibiotics) were subsequently added. As a result, a
112 total of 7 heavy metals (As, Cd, Zn, Cu, Ni, Cr and Pb) and 3 antibiotics (TC, OTC and CTC) were added to
113 the bacterial suspensions (a total of 10 pollutants studied), and these plus a control (only water added), were
114 used for each sample. The final concentrations of heavy metals added ranged between 10⁻² and 10⁻⁶ mol L⁻¹,
115 while for antibiotics ranged from 400 to 6x10⁻³ mg L⁻¹. The bacterial community growth was estimated after a
116 pre-incubation step of 24 h, where bacterial suspensions were added with the different concentrations of
117 antibiotics before the leucine incorporation assay (Berg et al., 2010; Fernández-Calviño and Bååth, 2013). The
118 [³H] leucine incorporation was then measured on each micro-centrifugation tube, as follows: a volume of 0.2
119 µL [³H] Leu (3.7 MBq mL⁻¹ and 0.574 TBq mmol⁻¹; Amersham) was added with non-labeled Leu to each tube,
120 resulting in 275 nM Leu in the bacterial suspensions. After 3 h of incubation, the bacterial growth was
121 terminated by adding 75 mL of 100% trichloroacetic acid. Washing was performed as described by Bååth et
122 al. (2001), and a subsequent measurement of radioactivity was carried out using a liquid scintillation counter
123 (Tri-Carb 2810 TR, PerkinElmer, USA).



124

125 **2.4. Data analyses**

126 The tolerance of the bacterial community to the seven heavy metals (As, Cd, Zn, Cu, Ni, Cr and Pb)
127 and the three tetracycline antibiotics (TC, OTC and CTC) was estimated as Log IC₅₀, the logarithm of the
128 concentration that resulted in 50% inhibition of bacterial community growth. Log IC₅₀ was calculated using a
129 logistic model, as follows:

$$130 Y = c/[1 + e^{b(X-a)}]$$

131 where Y is the measured level of Leu incorporation, X is the logarithm of the substance (heavy metals (As, Cd,
132 Zn, Cu, Ni, Cr, Pb) or antibiotics (TC, OTC or CTC)) added to the bacterial suspension, a is the log IC₅₀, c the
133 bacterial growth rate in absence of the toxic substance, and b is a slope parameter indicating the inhibition rate.
134 A higher value of log IC₅₀ indicates a higher community tolerance, while a lower value indicates that the
135 pollutants (heavy metals or antibiotics) are more toxic to the bacterial community.

136



137 3. Results and Discussion

138 3.1. Toxicity of heavy metals on soil bacteria

139 The toxicity of the heavy metals on soil bacterial communities was tested in the agricultural soil in the
140 short and long terms (1 and 42 days of incubation). The bacterial growth results obtained for the soil samples
141 polluted with 1000 mg kg⁻¹ of As, Cd, Zn, Cu, Ni, Cr, or Pb (separately) and for the unpolluted soil (0 mg kg⁻¹)
142 allowed to establish a toxicity index for these heavy metals (Table 1). On day 1, the toxicity sequence was the
143 following: Cr>Cu>Ni>Cd>Zn>As>Pb. After 42 days of incubation the sequence was similar, mainly affecting
144 to Ni and Cd: Cr>Cu>Cd>Ni>Zn>As>Pb.

145 The data obtained in this study differ from those observed by other authors, which was expected, since
146 the toxicity of heavy metals on soil microorganisms depends on the physico-chemical characteristics of the
147 soils, the concentrations of heavy metals added, and the incubation period (Hattori, 1992; Diaz-Raviña and
148 Bååth, 1996; Nannipieri et al., 1997; Ahmad et al., 2005). Hattori (1992) evaluated the influence of Cd, Cr, Cu,
149 Ni, Pb, and Zn on the microorganisms present in two soils after four weeks and obtained the toxicity sequence
150 Cd> Cu> Ni> Cr> Zn for one of them (a Gleysol), and the sequence Cu> Cd> Ni> Zn> Cr for the other soil
151 (an Andosol). Ahmad et al. (2005) measured the response of the microbial populations to Cu, Cd, Cr, Hg, Mn,
152 Ni, Pb and Zn at five incubation times, finding that the sensitivity of the bacterial populations depended on the
153 functional group to which they belonged, being the aerobic-heterotrophic bacteria more sensitive to Ni, Pb and
154 Cu, while nitrogen-fixing bacteria were more sensitive to Cd and Pb. Lastly, Bååth (1989) observed that the
155 relative toxicity of metals decreased in the order Cd> Cu> Zn> Pb, although differences were found among
156 different investigations. Analyzing data obtained in different studies, it can be observed that, in general, the
157 heavy metals that exert a higher effect on soil microorganisms are Cd and Cu. These results agree with those
158 obtained using as endpoints other organisms such as *Daphnia magna*, *Vibrio fischeri*, *Lemma minor* and
159 *Eisenia fetida*, which show that there is great variability in the toxicities of heavy metals on them, but also
160 show that Cd and Cu are the most toxic heavy metals (Neuhauser et al., 1985; Arambašić et al., 1995; Huynh
161 and Bulich, 1995; Dirilgen, 2001; Hsieh et al., 2004; Teodorovic et al., 2009).

162

163 3.2. Soil bacterial communities tolerance to heavy metals in the long-term

164 Figure 1 shows the dose-response curves obtained for each heavy metal after 42 days of incubation,
165 with data for each microcosm. These dose-response curves are sigmoidal, such as those reported by other
166 authors also for heavy metal (Díaz-Raviña et al., 1994; Cruz-Paredes et al., 2017; Song et al., 2017; Santás-



167 Miguel et al., 2020c). The inhibition curves show the absence of bacterial growth inhibition at low
168 concentrations of heavy metals, but at high doses inhibition takes place, and it increased with dose. As general
169 trend, it is observed that the inhibition curves obtained for all heavy metals tested in the soil shift to the right
170 with respect to the control (unpolluted soil). This shift to the right in the dose-response curves suggests the
171 existence of soil bacterial community tolerance to the high heavy metals concentrations (1000 mg kg^{-1}) added
172 on the soil used in this research.

173 In this study, the dose-response curves fitted well to the logistic model, with R^2 ranging between 0.966
174 and 0.991 (mean $R^2=0.983$) for the control sample (0 mg kg^{-1}), and R^2 between 0.861 and 0.987 (mean $R^2=$
175 0.950) for samples polluted with heavy metals (1000 mg kg^{-1} of each one, individually).

176 Table 2 shows the Log IC_{50} values obtained from each dose-response curve. The unpolluted soil
177 showed the lowest value of Log IC_{50} for Cu (-4.17 ± 0.05), and the highest value for Ni (2.56 ± 0.08), i.e., the
178 bacteria found in the bacterial suspension of unpolluted soils show high toxicity effects due to Cu and lower
179 toxicity for Ni. As regards the Log IC_{50} values obtained for all the heavy metals tested, the following toxicity
180 sequence was observed for the control (unpolluted soil): $\text{Cu} > \text{As} > \text{Cr} \geq \text{Pb} \geq \text{Cd} > \text{Zn} > \text{Ni}$. This toxicity sequence is
181 similar to that reported by Díaz-Raviña et al. (1994), who studied the bacterial community tolerance to Cu, Cd,
182 Zn, Ni and Pb using the thymidine incorporation technique, and observed that the most toxic heavy metal on
183 bacterial communities was Cu, while Ni showed the lowest toxicity. However, the order of the rest of heavy
184 metals differs from that observed in the current study, with the sequence being: $\text{Cu} > \text{Cd} > \text{Zn} > \text{Pb} > \text{Ni}$.

185 Regarding the samples polluted with 1000 mg kg^{-1} of each of the heavy metals separately, the lowest
186 Log IC_{50} value corresponded to Cu (-2.61 ± 0.12), and the highest to Zn (-1.63 ± 0.09) (Table 2), i.e., the bacterial
187 communities of these microcosms are more sensitive to Cu than to Zn. The toxicity sequence, in function of
188 the Log IC_{50} values, was: $\text{Cu} > \text{Pb} \geq \text{As} \geq \text{Cd} \geq \text{Cr} \geq \text{Ni} \geq \text{Zn}$.

189 The Log IC_{50} values obtained for each soil sample polluted with the high heavy metal concentrations
190 increased with respect to the control (unpolluted soil), indicating that the bacteria exposed to high levels of
191 heavy metals developed tolerance to these metals. The highest increases in Log IC_{50} ($\Delta \text{Log IC}_{50}$, Table 2)
192 corresponded to Cr and Cu (1.6 units), followed by As, Zn, Cd and Pb (1.4, 1.3, 1.2 and 1.1 units, respectively),
193 whereas it was 0.7 units for Ni. To note that the presence of high amounts of heavy metals in the soil causes
194 that bacterial communities increase their tolerance to heavy metals (Díaz-Raviña et al. 1994; Díaz-Raviña and
195 Bååth, 1996; Díaz-Raviña and Bååth, 2001; Fernández-Calviño and Bååth, 2013; Santás-Miguel et al., 2020c;



196 Zhong et al., 2021), although the magnitude of these increases seems to depend on the toxicity or initial pressure
197 exerted by the metal on the soil bacterial communities.

198 Figure S1 (Supplementary Material) shows the relation between the increase in tolerance to heavy
199 metals and the toxicity exerted by each metal on bacterial growth in the short term (1 day) and in the long term
200 (42 days). It can be observed that the higher is the bacterial growth inhibition on days 1 and 42, the greater is
201 the increase in tolerance generated by bacterial communities to heavy metals. This behavior was observed for
202 all the heavy metals studied except Ni. In fact, eliminating Ni the relationship becomes significant at day 1
203 ($r=0.789$, $P < 0.05$) and at day 42 ($r=0.730$; $P < 0.05$).

204 The results suggest that the increase in soil bacterial community tolerance after the addition of heavy
205 metals can be attributed to the immediate effect caused by the decay of sensitive species, and to a posterior
206 effect caused by different competitive capacities and posterior adaptation in bacteria remaining. This effect of
207 heavy metals in soil has been observed previously by Diaz-Raviña and Bååth (1996) for Zn, Cd, Ni, Cu and
208 Pb, and by Fernandez-Calviño and Bååth (2016) for Cu.

209

210 **3.3. Soil bacterial communities tolerance to tetracycline antibiotics in the long-term**

211 Figures 2-4 show dose-response curves obtained at 42 days of incubation for each heavy metal added
212 to the soil at 1000 mg kg^{-1} , after the exposure of bacterial suspension from each microcosm to concentrations
213 of tetracyclines antibiotics (TC, OTC and CTC) ranging from 400 to $6 \times 10^{-3} \text{ mg L}^{-1}$. The inhibition curves
214 obtained for these tetracycline antibiotics are sigmoidal. As general trend, the dose-response curves are shifted
215 to right respect to the control sample, and in some cases the tetracyclines antibiotics did not show an inhibitory
216 effect on bacterial communities.

217 These dose-response curves fitted well to the logistic model, with R^2 ranging between 0.925 and 0.982
218 (mean $R^2=0.956$) for TC samples, between 0.771 and 0.997 (mean $R^2= 0.952$) for OTC samples, and from
219 0.959 to 0.992 (mean $R^2=0.976$) for CTC samples.

220 Table 3 shows the Log IC_{50} values obtained from each dose-response curve, being 1.83 ± 0.06 for the
221 control sample after exposure of the bacterial suspension to different concentrations of TC. In general, the Log
222 IC_{50} values obtained for each microcosm polluted with 1000 mg kg^{-1} of each heavy metal and subsequently
223 exposed to TC, increased with respect to the control. The magnitude of these increase varies in function of the
224 heavy metal, being around 0.4 units for Pb, Zn and As, and 0.5 units to Ni. However, the bacterial communities
225 of soils contaminated with Cd, Cu and Cr do not show inhibition in their growth when exposed to TC, therefore,



226 their LogIC₅₀ cannot be determined due to the high community tolerance to TC. Consequently, heavy metals
227 that do not show inhibition have been assigned a Log IC₅₀ value of >2.6, since it is the maximum concentration
228 tested in the current study.

229 The Log IC₅₀ value obtained for the control soil in presence of OTC was 1.11±0.08. Regarding the
230 Log IC₅₀ values obtained for the samples polluted with heavy metals, it can be observed that after the exposure
231 of the bacterial suspension to OTC, there was an increase in these values with respect to the control, going
232 from 0.2 to 0.6 for Pb, Cd, Zn, As and Ni, and being 1.6 for Cu. Regarding Cr, the exposure of the soil bacterial
233 suspension to OTC did not cause inhibition of its growth, therefore, as in the case of Cd, Cu and Cu for TC,
234 the value of >2.6 (maximum concentration tested) was assigned.

235 Finally, the Log IC₅₀ value obtained for the control soil added with CTC was 1.52±0.10. After
236 exposure of the bacterial suspensions to CTC, the Log IC₅₀ values obtained for the samples polluted with heavy
237 metals showed an increase with respect to the control, going from 1.7 to 2.0 for all heavy metals except for Cr,
238 which did not show inhibition of bacterial growth when exposed to different concentrations of CTC. Therefore,
239 it can be indicated that bacterial communities exposed to high concentrations of heavy metals show great co-
240 tolerance to CTC. In general, the increase in tolerance obtained for TC, OTC and CTC cannot be related to the
241 toxicity of heavy metals in the soil, at 1 and 42 days of incubation (data not show).

242 Co-tolerance to TC generated by soil bacterial communities exposed to heavy metals has been
243 previously demonstrated (Berg et al., 2010; Fernández-Calviño and Bååth, 2013; Song et al. 2017). Song et al.
244 (2017) observed the existence of co-tolerance to TC in a microcosm exposed to concentrations of Cu≥333 mg
245 kg⁻¹ and Zn≥500 mg kg⁻¹. Fernández-Calviño and Bååth (2013) found co-tolerance of soil bacterial
246 communities to TC in contaminated soils with concentrations of Cu ≥500 mg kg⁻¹. The results shown in these
247 studies agree with what was observed in the present work. Soil bacterial communities exposed to high
248 concentrations of heavy metals (1000 mg kg⁻¹) generate co-tolerance to tetracycline. However, Zhong et al.
249 (2021) did not observe co-tolerance to TC in soils contaminated with Cu, Zn and Pb, while they found co-
250 tolerance of soil bacterial communities to vancomycin.

251 This co-tolerance of soil bacterial communities to TC at high concentrations of heavy metals is also
252 observed for OTC and CTC. The results obtained in this study agree with the data reported in studies previously
253 carried out by Santás-Miguel et al. (2020c), where co-tolerance of soil bacterial communities to tetracycline
254 antibiotics (TC, OTC and CTC) was observed in soils with concentrations of Cu ≥1000 mg kg⁻¹ for 42 days of
255 incubation.



256 It is relevant that co-tolerance studies, in general, focus on a reduced number of heavy metals
257 (generally Cu) and tetracycline antibiotics, while the present work shows for the first time the effect of 7
258 different heavy metals as regards the co-tolerance to the 3 most used tetracycline antibiotics (TC, OTC and
259 CTC).

260 It is also remarkable that the role of heavy metals as a co-selection factor to generate antibiotic
261 resistance genes has been studied for a wide variety of heavy metals and antibiotics (Berg et al., 2010;
262 Dickinson et al., 2019; Cao et al., 2020; Mazhar et al., 2021), and, although in the present study antibiotic
263 resistance genes (ARGs) have not been studied directly, the resistance of soil bacterial communities to heavy
264 metals and tetracycline antibiotics (TC, OTC, CTC) could be detected using the Leucine incorporation method.
265 The heavy metal levels of the tested microcosms (1000 mg kg^{-1}) are really very high, but they should not be
266 considered unrealistic, since heavy metals are not easily biodegradable, and therefore their concentration can
267 increase over time (Li et al., 2015), reaching very high levels in the soil (Smolders et al., 2004; Spiteri et al.,
268 2005; Zhuang et al., 2009; Komárek et al., 2010). Therefore, soils contaminated with high concentrations of
269 heavy metals could become hotspots of ARGs, which would pose a potential danger to human health, since
270 they could be transmitted to humans (Forsberg et al., 2012).

271

272 **4. Conclusions**

273 In the current study it was evidenced that heavy metals present in soils show wide variability as regards
274 the toxicity they exert on bacterial communities, with Cd and Cu being the most toxic. Specifically, soil samples
275 contaminated with a high concentration (1000 mg kg^{-1}) of each of 7 different heavy metals (As, Cd, Zn, Cu,
276 Ni, Cr and Pb), added separately, showed tolerance of the soil bacterial communities to the heavy metals
277 themselves, compared to the control soil. Besides, it was shown that the higher the effect of heavy metals in
278 the short and long terms, the greater the increase in tolerance to metals generated in the soil bacterial
279 communities (except for Ni). In addition, soil samples contaminated with high concentrations of heavy metals
280 also showed co-tolerance of soil bacterial communities to tetracycline, oxytetracycline and chlortetracycline
281 for all the metals tested, compared to the control soil. However, in general, there was no correlation between
282 the initial damage exerted by heavy metals in the short and long terms and the co-tolerance shown by soil
283 bacterial communities to tetracycline antibiotics. Future studies could give further details about the potential
284 interactions of different pollutants present in agricultural soils, which can be considered relevant as regards
285 environmental preservation and public health.



286

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Tables and figures

425

426

Table 1. Toxicity index values \pm error of soil samples containing 0 mg kg⁻¹ (control soil) and 1000 mg kg⁻¹ of

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As, Cd, Zn, Cu, Ni, Cr or Pb (individually), after 1 and 42 days of incubation.

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Heavy metal	Day 1	Day 42
	Toxicity Index \pm error	Toxicity Index \pm error
As	0.69 \pm 0.04	2.39 \pm 0.18
Cd	0.12 \pm 0.01	0.15 \pm 0.00
Zn	0.15 \pm 0.04	2.36 \pm 0.04
Cu	0.03 \pm 0.01	0.06 \pm 0.01
Ni	0.05 \pm 0.02	2.18 \pm 0.39
Cr	0.01 \pm 0.01	0.01 \pm 0.00
Pb	0.72 \pm 0.04	2.76 \pm 0.06

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430



431 **Table 2.** Estimated values of $\text{LogIC}_{50 \pm \text{error}}$ and R^2 from fits to a logistic model of soil samples containing 0 mg
432 kg^{-1} (control soil) and 1000 mg kg^{-1} of As, Cd, Zn, Cu, Ni, Cr or Pb (individually). The increases in tolerance ($\Delta \text{log IC}_{50}$
433 $\text{IC}_{50} = \text{LogIC}_{50 \text{ polluted}} - \text{LogIC}_{50 \text{ control}}$) were calculated for each heavy metal

Heavy metal	Control soil		Heavy metal-polluted soil (1000 mg kg^{-1})		$\Delta \text{log IC}_{50}$
	$\text{Log IC}_{50 \pm \text{error}}$	R^2	$\text{Log IC}_{50 \pm \text{error}}$	R^2	
As	-3.63 ± 0.11	0.970	-2.28 ± 0.04	0.987	1.4
Cd	-3.37 ± 0.07	0.988	-2.21 ± 0.06	0.966	1.2
Zn	-2.89 ± 0.06	0.988	-1.63 ± 0.09	0.966	1.3
Cu	-4.17 ± 0.05	0.991	-2.61 ± 0.12	0.973	1.6
Ni	-2.56 ± 0.08	0.966	-1.82 ± 0.11	0.861	0.7
Cr	-3.44 ± 0.07	0.987	-1.86 ± 0.19	0.938	1.6
Pb	-3.42 ± 0.07	0.989	-2.34 ± 0.06	0.977	1.1

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435



436 **Table 3.** Estimated values of $\text{LogIC}_{50} \pm \text{error}$ and R^2 from fits to a logistic model of soil samples polluted with 1000
 437 mg kg^{-1} of As, Cd, Zn, Cu, Ni, Cr and Pb. The increases in tolerance ($\Delta \log \text{IC}_{50} = \text{LogIC}_{50} \text{ polluted} - \text{LogIC}_{50} \text{ control}$)
 438 were calculated for tetracycline (TC), oxytetracycline (OTC), and chlortetracycline (CTC)

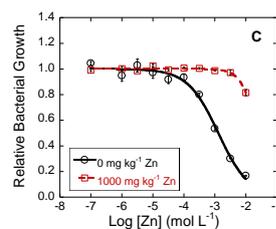
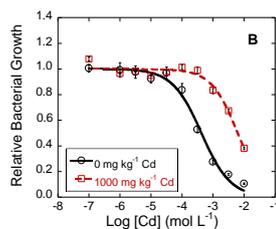
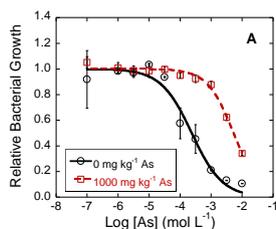
	TC		OTC			CTC		$\Delta \log \text{IC}_{50}$	
	$\text{Log IC}_{50} \pm \text{error}$	R^2	$\Delta \log \text{IC}_{50}$	$\text{Log IC}_{50} \pm \text{error}$	R^2	$\Delta \log \text{IC}_{50}$	$\text{Log IC}_{50} \pm \text{error}$		R^2
Control	1.83±0.06	0.975	--	1.11±0.08	0.982	--	1.52±0.10	0.959	--
As	2.26±0.09	0.925	0.4	1.65±0.02	0.997	0.5	1.91±0.04	0.985	0.4
Cd	≥2.6*	--	0.8	1.47±0.09	0.973	0.4	1.91±0.01	0.982	0.4
Zn	2.28±0.05	0.971	0.4	1.64±0.04	0.991	0.5	2.19±0.11	0.977	0.7
Cu	≥2.6*	--	0.8	2.67±0.37	0.771	1.6	1.82±0.08	0.959	0.3
Ni	2.33±0.04	0.982	0.5	1.68±0.07	0.968	0.6	1.75±0.04	0.992	0.2
Cr	≥2.6*	--	0.8	≥2.6*	--	1.5	≥2.6*	--	1.1
Pb	2.23±0.08	0.926	0.4	1.34±0.08	0.981	0.2	1.99±0.03	0.976	0.5

439 * The value of 2.6 is the maximum value tested

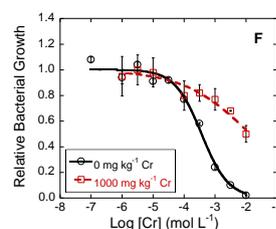
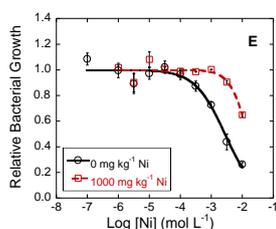
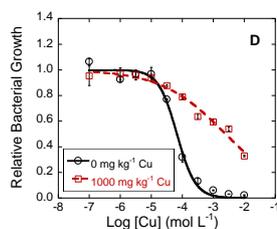
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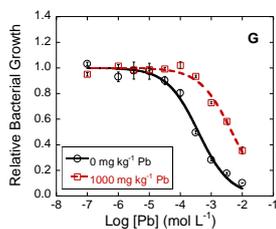


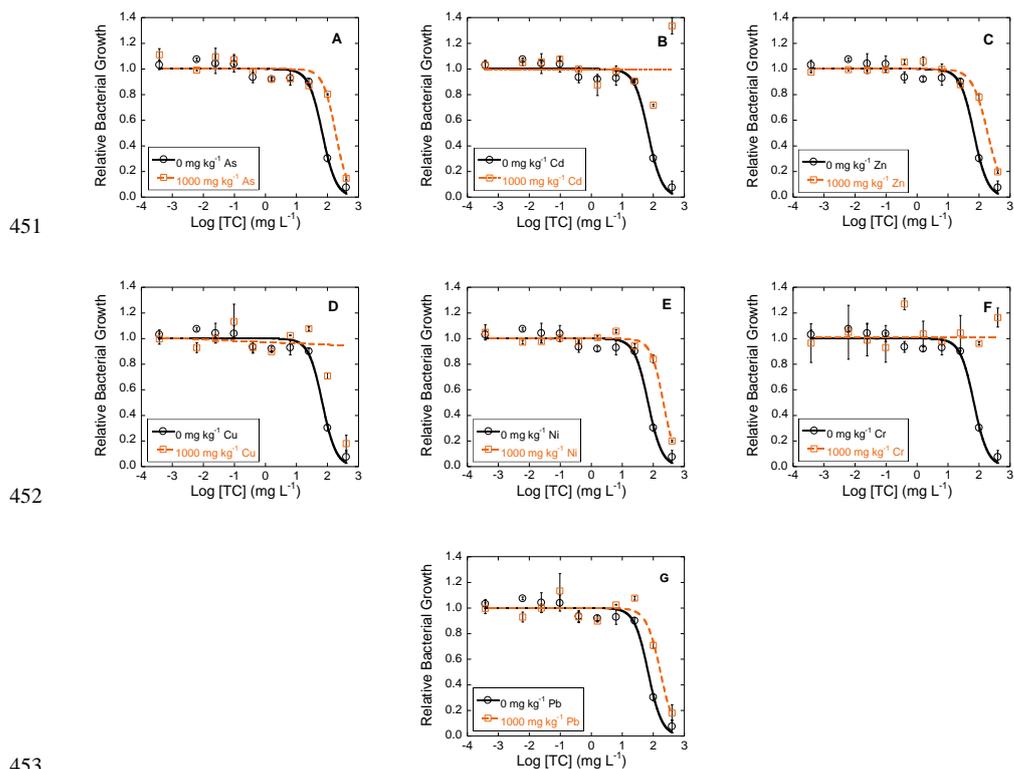
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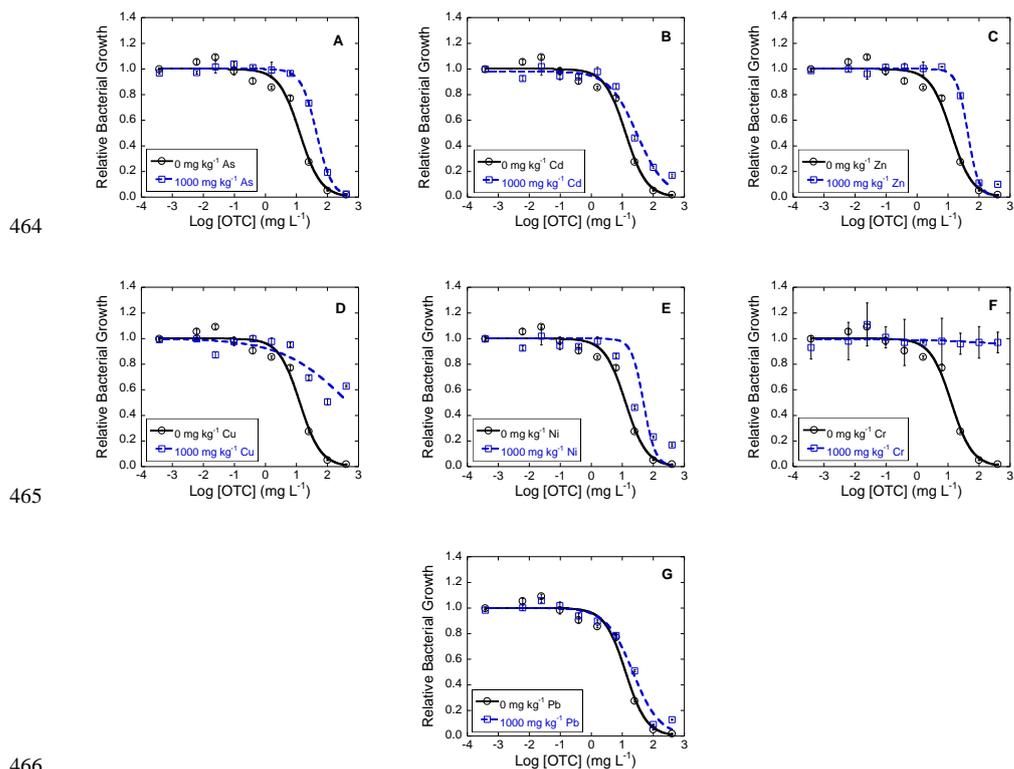
446 **Figure 1.** Dose-response curves obtained after the addition of 10 different concentrations of the heavy metals As (A), Cd
447 (B), Zn (C), Cu (D), Ni (E), Cr (F) and Pb (G) to bacterial suspensions extracted from soils polluted with 1000 mg kg⁻¹
448 of each of the same heavy metals, separately, after 42 days of incubation. Black line represented the dose-response curve
449 corresponding to the unpolluted soil (control) and discontinued red lines is dose-response curve corresponding to the soil
450 polluted with 1000 mg kg⁻¹ of heavy metal. Average values (n=3) with coefficients of variation always <5%





456 **Figure 2.** Dose-response curves obtained after the addition of 10 different concentrations of tetracycline (TC) to
457 bacterial suspension extracted from soil samples polluted with 1000 mg kg⁻¹ of As (A), Cd (B), Zn (C), Cu (D), Ni
458 (E), Cr (F) or Pb (G), each of them added separately, after 42 days of incubation. Black lines represent the dose-
459 response curve obtained for the unpolluted soil (control) and discontinued orange lines are dose-response curves
460 for soils polluted with 1000 mg kg⁻¹ of one of the heavy metals. Average values (n=3) with coefficients of variation
461 always <5%.

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468 **Figure 3.** Dose-response curves obtained after the addition of 10 different concentration of oxytetracycline (OTC)
469 to bacterial suspension extracted from soil samples polluted with 1000 mg kg⁻¹ of As (A), Cd (B), Zn (C), Cu (D),
470 Ni (E), Cr (F) or Pb (G), each of them added separately, after 42 days of incubation. Black lines represent the dose-
471 response curve obtained for the unpolluted soil (control) and discontinued blue lines are dose-response curves for
472 soils polluted with 1000 mg kg⁻¹ of one of the heavy metals. Average values (n=3) with coefficients of variation
473 always <5%.

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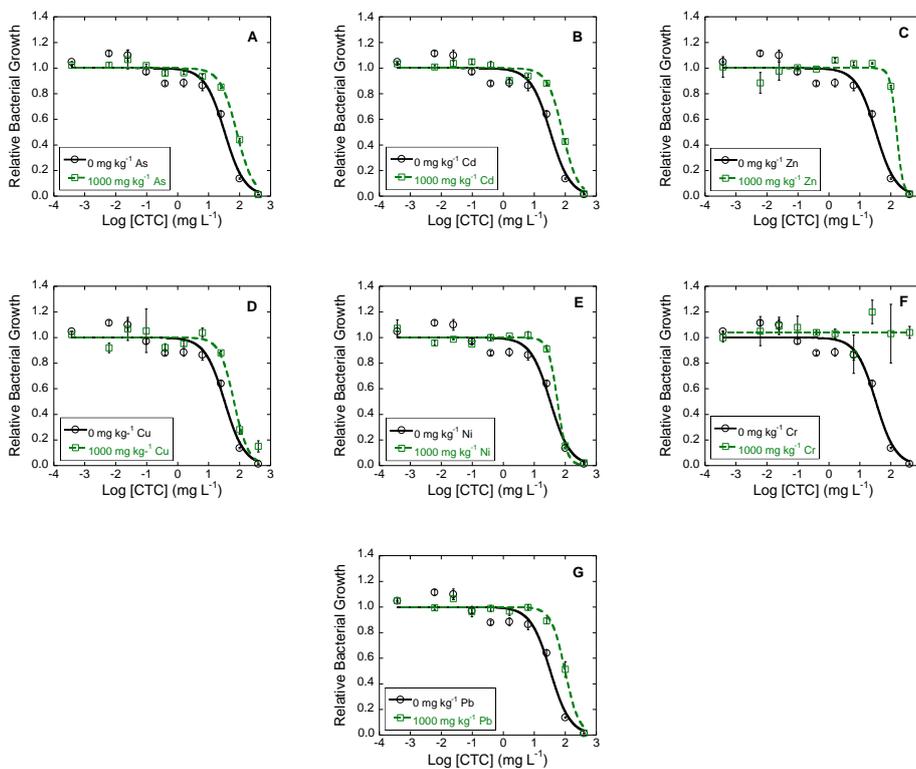
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483 **Figure 4.** Dose-response curves obtained after the addition of 10 different concentration of chlortetracycline (CTC)

484 to bacterial suspension extracted from soil samples polluted with 1000 mg kg⁻¹ of As (A), Cd (B), Zn (C), Cu (D),

485 Ni (E), Cr (F) or Pb (G), each of them added separately, after 42 days of incubation. Black lines represent the dose-

486 response curve obtained for the unpolluted soil (control) and discontinued green lines are dose-response curves for

487 soils polluted with 1000 mg kg⁻¹ of one of the heavy metals. Average values (n=3) with coefficients of variation

488 always <5%.

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