

1     **Aluminous clay and pedogenic Fe oxides modulate aggregation and**  
2                     **related carbon contents in soils of the humid tropics**

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22     **Keywords:** tropical soils, aggregate size distribution, aggregate stability, soil mineralogy, kaolinite,  
23     gibbsite, goethite, hematite, land-use change

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26 **Abstract**

27 Aggregation affects a wide range of physical and biogeochemical soil properties with positive effects on  
28 soil carbon storage. For weathered tropical soils, aluminous clays (kaolinite and gibbsite) and pedogenic  
29 Fe (oxyhydr)oxides (goethite and hematite; termed 'Fe oxides') have been suggested as important building  
30 units for aggregates. However, as aluminosilicates, aluminum hydroxides, and Fe oxides are part of the  
31 clay-sized fraction it is hard to separate, how certain mineral phases modulate aggregation. In addition, it  
32 is not known what consequences this will have for organic carbon (OC) persistence after land-use change.  
33 We selected topsoils with unique mineralogical compositions in the East Usambara Mountains of  
34 Tanzania under forest and cropland land uses, varying in contents of aluminous clay and Fe oxides.  
35 Across the mineralogical combinations, we determined the aggregate size distribution, aggregate stability,  
36 OC contents of aggregate size fractions as well as changes in aggregation and OC contents under forest  
37 and cropland land use. Patterns in soil aggregation were rather similar across the different mineralogical  
38 combinations (high level of macroaggregation and high aggregate stability). Nevertheless, we found some  
39 statistically significant effects of aluminous clay and pedogenic Fe oxides on aggregation and OC storage.  
40 An aluminous clay content > 250 g kg<sup>-1</sup> in combination with pedogenic Fe contents < 60 g kg<sup>-1</sup>  
41 significantly promoted the formation of large macroaggregates > 4 mm. In contrast, a pedogenic Fe  
42 content > 60 g kg<sup>-1</sup> in combination with aluminous clay content of < 250 g kg<sup>-1</sup> promoted OC storage and  
43 persistence even under agricultural use. The combination with low aluminous clay, and high pedogenic Fe  
44 contents, displayed the highest OC persistence, despite conversion of forest to cropland caused substantial  
45 disaggregation. Our data indicate that aggregation in these tropical soils, is modulated by the mineralogical  
46 regime, causing moderate but significant differences in aggregate size distribution. Nevertheless,  
47 aggregation was little decisive for overall OC persistence in these highly weathered soils, where OC  
48 storage is more regulated by direct mineral-organic interactions.

**Gelöscht:** feedbacks

**Gelöscht:** as both secondary

**Gelöscht:** aluminosilicates

**Gelöscht:** ,

**Gelöscht:** and what consequences this has

**Gelöscht:** . Soils are

**Gelöscht:** We found the soil aggregation patterns (high level of macroaggregation and aggregate stability) more similar than different among mineralogical combinations.

**Gelöscht:** Yet, an

**Gelöscht:** after the change in land use

**Gelöscht:** -

**Gelöscht:** contents

**Gelöscht:** is typical

**Gelöscht:** of the humid tropics

## 69 1. Introduction

70 Many functions of soils such as food production, water purification as well as climate regulation are  
71 tightly linked to soil structure (Bronick and Lal, 2005; FAO, 2015; Six et al., 2004). Aggregates are the  
72 structural backbone of soil and changes in aggregation impacts various processes such as root  
73 development, soil erosion, and soil organic carbon (OC) accumulation (Chaplot et al., 2010; Le Bissonnais  
74 et al., 2018). Based on their size, soil aggregates are typically classified into small microaggregates  
75 (< 20 µm), large microaggregates (20–250 µm), and macroaggregates (> 0.25 mm) (Tisdall and Oades,  
76 1982). Cementing agents such as clay minerals, metal (oxyhydr)oxides, as well as organic matter (OM)  
77 are considered as primary building units of microaggregates (Totsche et al., 2018), which provide the basis  
78 for the formation of larger soil structural units (Asano and Wagai, 2014). The study by Six et al. (2002)  
79 points to the special role of inorganic compounds such as clay minerals and pedogenic metal oxides in the  
80 formation of aggregates in the tropics. Pedogenic iron (Fe<sub>d</sub>) (oxyhydr)oxides (summarized as ‘Fe oxides’)  
81 have been reported to facilitate macroaggregation (Peng et al., 2015) and aggregate stability (Duiker et al.,  
82 2003). Under the acidic conditions of weathered tropical soils, Fe oxides provide positively charged  
83 surfaces capable of reacting with negatively charged inorganic constituents, like clay minerals or OM  
84 (Kaiser and Guggenberger, 2003; Kleber et al., 2015; Six et al., 2004; Totsche et al., 2018). Aggregation  
85 might be ascribed to inorganic or organic cementing agents with no consensus about the relevance of each  
86 individual agent. Understanding the effects of individual cementing agents for aggregation is needed to  
87 disentangle their potential contribution to soil aggregation. For example, the extent of aggregation has  
88 been either positively related to the contents of clay and OC (Chaplot and Cooper, 2015; Paul et al., 2008;  
89 Spaccini et al., 2001), or to differences in the clay mineral composition (Fernández-Ugalde et al., 2013).  
90 Furthermore, Barthès et al. (2008) showed that texture had no effect on macroaggregation over a range of  
91 tropical soils characterized by low-activity clay minerals. Uncertainty also derives from the fact that the  
92 clay size particle fraction (< 2-µm) not only contains OM and different types of clay minerals, but also  
93 variable contents of pedogenic Fe and aluminum (Al) oxides (Barré et al. 2014; Fernández-Ugalde et al.  
94 2013; Wagai and Mayer 2007). Denef et al. (2004) showed that significant differences in the amount of

**Gelöscht:** Especially in weathered tropical soils, aggregation depends strongly on inorganic cementing agents

**Gelöscht:** At present, however, there is little consensus to which extent aggregation can be ascribed to individual inorganic or organic cementing agents, or whether aggregation is best explained by their mutual interactions.

107 microaggregates encased in macroaggregates can be related to the clay mineral composition (2:1, mixed  
108 layer, 1:1 clays). They assume that interactions of 1:1 clay minerals with Fe oxides cause a higher  
109 aggregate stability compared to those involving 2:1 clay minerals (Denef et al., 2002, 2004). Such mutual  
110 interactions between typical aluminous clay-sized minerals (e.g. kaolinite, gibbsite) and pedogenic Fe  
111 oxides are thus possible drivers of aggregation in weathered tropical soils (Durn et al., 2019).

112 As indicated above, soil aggregation is considered to be an important process that increases OC  
113 persistence, because of the physical separation of OM from microorganisms and their exoenzymes (Six et  
114 al., 2004). Thus, improved aggregation could contribute to enhanced OC storage in soils (Kravchenko et  
115 al., 2015; Marín-Spiotta et al., 2008; Schmidt et al., 2011). Managing aggregation, e.g., for climate change  
116 mitigation, requires profound knowledge on the controls of aggregation and their effects on OC  
117 persistence (Paul et al., 2008). To the best of our knowledge there are no studies available, which  
118 investigated the influence of changes in the content of clay minerals with low activity and the content of  
119 pedogenic metal oxides on aggregation under comparable mineralogical conditions for weathered tropical  
120 soils. Macroaggregates are particularly susceptible to soil management (Six et al., 2000a; Totsche et al.,  
121 2018). Consequently, destruction of macroaggregates upon changes from forests to cropland might  
122 account for OC losses that were observed in tropical soils (Don et al., 2011; Kirsten et al., 2019; Mujuru et  
123 al., 2013). The stability of aggregates should thus determine OC losses induced by land-use change and  
124 higher losses should be related to lower aggregate stability (Denef et al., 2002; Le Bissonnais et al., 2018;

125 Six et al., 2000b). We are currently not aware of any studies that solve the puzzle to which extent the  
126 amount of aluminous clay and pedogenic Fe oxides controls soil aggregation and OC storage in highly  
127 weathered soils of the humid tropics.

128 This study takes advantage of soils under natural forest and cropland in the East Usambara  
129 Mountains of Tanzania. The mineralogical composition of the study soils is very homogeneous with  
130 kaolinite and gibbsite as the main aluminous minerals of the clay fraction and goethite and hematite as  
131 dominant pedogenic Fe oxides (Kirsten et al., 2021). Yet, the ratio of aluminous clays to Fe oxides  
132 differed strongly, giving rise to unique mineralogical combinations under both land use types. Thus, the

**Gelöscht:** At present, we are not aware of any studies resolving the puzzle to which extent aluminous clay and pedogenic Fe oxides, control soil aggregation as well as OC storage in weathered tropical soils.

**Gelöscht:** unique mineralogical combinations of

**Feldfunktion geändert**

**Gelöscht:** The soils vary in the amount of aluminous clay (kaolinite, gibbsite) and pedogenic Fe oxides (goethite, hematite) but without variation in their mineralogical composition (Kirsten et al., 2021).

147 conversion of natural forest to cropland in the study region, enables us to evaluate the effect of land-use  
148 change under each mineralogical combination on soil physical properties and related OC persistence. In  
149 the precursor study, we found a positive relationship between the storage of mineral-associated OC and  
150 the ratio of pedogenic Fe to aluminous clay under forest and cropland land use, suggesting that a larger  
151 share of Fe oxides is linked to larger OC storage and persistency against land-use change (Kirsten et al.,  
152 2021). In the present study, we test whether aggregation and its contribution to OC storage follow similar  
153 patterns, or are decoupled from the individual contribution main mineral constituents. In detail, our main  
154 research goal was to investigate the individual role of aluminous clay and pedogenic Fe oxides for  
155 determining (i) the soil aggregate size distribution, (ii) aggregate stability, (iii) the consequences for OC  
156 allocation into different aggregate size fractions, and (iv) the consequences for OC persistence related to  
157 land-use change. We hypothesize that the mineralogical combination resulting in the largest aggregate  
158 stability also results in the largest OC persistence. For this purpose, we determined the aggregate size  
159 distribution of soils under both land uses, determined the OC contents of obtained aggregate fractions, and  
160 tested the stability of the two largest aggregate size fractions (2–4 mm and > 4 mm). As a measure of OC  
161 persistence, the OC content of aggregate size fractions was compared between the two land uses in the  
162 same mineralogical combination. We generally focused on soil samples from 0–10 cm to test our current  
163 hypothesis since land-use induced OC losses from soils of the study region largely occur in this depth  
164 increment (cf. Kirsten et al., 2019).

**Gelöscht:** small-scale

**Gelöscht:** that took place in that region

**Gelöscht:** into

**Gelöscht:** In detail, our main research objective was to investigate into the individual role of aluminous clay and pedogenic Fe oxides for determining

**Gelöscht:** presume

**Gelöscht:** also results in largest OC persistence after conversion of forests into croplands.

**Gelöscht:** Since land use induced OC losses in this region largely occur in topsoils (Kirsten et al., 2019), we concentrated on samples from that part of the soil.

182 **2. Material and methods**

183 **2.1 Study area and soil sampling**

184 The study was conducted in the Eastern Usambara Mountains of Tanzania close to the village Amani  
185 (5°06'00" S; 38°38'00" E). The climate is humid monsoonal with a mean annual precipitation of  
186 1,918 mm, and a mean annual temperature of 20.6°C with low variability within the study area (*Hamilton*  
187 and *Bensted-Smith*, 1989). The dominating Acrisols and Alisols, developed from Precambrian crystalline  
188 bedrock, are deeply weathered and highly leached, with visible clay illuviation in the subsoil (*Kirsten et*  
189 *al.*, 2019). Briefly, all soil samples were collected on mid-slope position. We sampled six plots under  
190 forest and three under annual cropping. The site selection was done based on total clay amount determined  
191 in the field and the associated total Fe amount measured with a portable XRF device (*Kirsten et al.*, 2021).  
192 We did not observe systematic differences in vegetation composition of the forest sites and NMR spectra  
193 showed a similar composition of litter for each of the two land uses investigated (*Kirsten et al.*, 2021).  
194 Furthermore, several visits in the study region over the last decade (2012, 2013, 2015, and 2018)  
195 combined with personal talks to farmers and local partners working in the region, enabled us to select  
196 cropland sites with similar agricultural management (cultivation of cassava (*Manihot esculenta*), hand hoe  
197 tillage, biomass burning before seed bed preparation). At each plot, mineral soil from three adjacent and  
198 randomly distributed soil pits at mid-slope position was sampled at 0–5 and 5–10 cm depths. Living roots  
199 were removed and aliquots of the soils were sieved to < 2 mm after drying at 40°C. For each depth  
200 increment, three undisturbed soil cores (100 cm<sup>3</sup>) were collected for bulk density determination.

**Gelöscht:** Criteria for site selection and soil sampling has been described in detail by *Kirsten et al.* (2021).

**Gelöscht:** Soil from three adjacent and randomly distributed soil pits was sampled at 0–5 and 5–10 cm depth.

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202 **2.2 Soil analyses**

203 *Basic soil properties and selected mineralogical combinations*

204 Bulk density was determined after drying the soil at 105°C and corrected for coarse fragments (*Carter and*  
205 *Gregorich*, 2008). Soil pH was measured in 0.01 M CaCl<sub>2</sub> at a soil to solution ratio of 1 : 2.5. Extraction  
206 of poorly crystalline Fe and Al phases as well as of Fe and Al complexed by OM was done with  
207 ammonium oxalate according to *Schwertmann* (1964). Effective cation exchange capacity (CEC<sub>eff</sub>) and

214 base saturation (BS) were determined following the procedure provided by Trüby and Aldinger (1989).  
215 Contents of OC and total N were analyzed by high temperature combustion at 950°C and thermo-  
216 conductivity detection (Vario EL III/Elementar, Heraeus, Langensfeld, Germany). A combined  
217 dithionite-citrate-bicarbonate extraction and subsequent texture analysis was applied to determine the  
218 contents of aluminous clay and total pedogenic Fe ( $Fe_d$ ). Briefly, 5–6 g soil pre-treated with 30%  $H_2O_2$   
219 were extracted with 30 g sodium dithionite ( $Na_2S_2O_4$ ) and 1.35 L buffer solution (0.27 M trisodium citrate  
220 dihydrate ( $C_6H_5Na_3O_7 \cdot 2H_2O$ ) + 0.11 M sodium bicarbonate ( $NaHCO_3$ )) at 75°C in a water bath for  
221 15 min (Mehra and Jackson, 1958). The Fe concentration of the extracts were measured by inductively  
222 coupled plasma optical emission spectroscopy (ICP-OES) using a CIROS-CCD instrument (Spectro,  
223 Kleve, Germany). The residues of the extraction were then subjected to a texture analysis using the pipette  
224 method (Gee and Bauder, 1986). Details of the procedure are described in Kirsten et al. (2021). Based on  
225 the respective content of aluminous clay and pedogenic Fe oxide in the 5–10 cm depth increment, each  
226 sample was assigned to a certain mineralogical combination. The threshold values for aluminous clay and  
227 pedogenic Fe oxides to distinguish between “high” and “low” were set to 250 g  $kg^{-1}$  and 60 g  $kg^{-1}$ ,  
228 respectively. We differentiated four groups varying in contents of aluminous clay and pedogenic Fe oxides  
229 under forest (i.e. ‘low clay–low Fe’, ‘low clay–high Fe’, ‘high clay–low Fe’, ‘high clay–high Fe’), and  
230 three analogous groups under cropland (i.e. ‘low clay–low Fe’, ‘low clay–high Fe’, ‘high clay–high Fe’).

231

### 232 *Aggregate size distribution, aggregate stability and carbon contents*

233 Aggregate size distribution was determined by dry sieving as it most closely resembles soil conditions at  
234 the end of the long dry season. Undisturbed soil was dried at 40°C for 48 hours. Separation of aggregate  
235 sizes was conducted with a sieving machine (AS 200 control “g”, Retsch, Hanau, Germany) combined  
236 with a set of four sieves with meshes of 4, 2, 1, and 0.25 mm, respectively (Larney, 2008). The amplitude  
237 was set to 1.51 mm (7.6 g-force), which was applied over a sieving duration of three minutes. Aggregate  
238 stability was tested for the two largest aggregate size fractions (2–4 mm and > 4 mm). The fast wetting  
239 pretreatment was applied to both fractions (Le Bissonnais, 1996) using a wet-sieving apparatus

240 (Eijkelkamp, Giesbeek, Netherlands) with sieve openings of 63  $\mu\text{m}$ . This procedure simulates the  
241 transition of aggregates from dry to rainy season. Sieving was conducted in ethanol for three minutes  
242 (stroke 1.3 cm,  $f = 34 \text{ min}^{-1}$ ). All aggregates remaining on the sieve were dried at 105°C. Water-stable  
243 aggregates were subsequently introduced to a sieving apparatus with a set of five sieves with mesh sizes of  
244 4, 2, 1, 0.63, and 0.25 mm, respectively (Larney, 2008). For each obtained aggregate fraction by dry  
245 sieving, OC contents analyzed by high temperature combustion at 950°C and thermo-conductivity  
246 detection (Vario EL III/Elementar, Heraeus, Langensfeld, Germany). The mass corrected OC content of  
247 a certain aggregate fraction was calculated using equation 1 to resemble the contribution to total soil OC,

$$248 \text{ Mass - corrected } OC_{Aggregate} = \frac{m_i}{\sum_{i=0}^n m_i} \times OC_{Aggregate} \quad (\text{Eq. 1})$$

249 where  $m_i$  represents the mass of an aggregate size fraction (g),  $\sum m_i$ , the sum of masses of all size  
250 fractions (g), and  $OC_{Aggregate}$  the OC content of aggregate fraction "i".

251 The mean weight diameter (MWD) of aggregates was calculated using equation 2 for undisturbed soil to  
252 describe the initial aggregate size distribution, and for the large aggregate size fractions after exposure to  
253 the stability test to evaluate the effect of fast wetting on aggregate stability,

$$254 MWD = \sum_{i=0}^n \frac{m_i}{\sum m_i} \times d_i \quad (\text{Eq. 2})$$

255 where  $m_i$  represents the mass of an aggregate size fraction (g),  $\sum m_i$ , the sum of masses of all size  
256 fractions (g), and  $d_i$  the mean mesh diameter of fraction "i" (mm). The MWD of the aggregate fraction  
257 > 4 mm was estimated by doubling the largest sieve size diameter (Youker and McGuinness, 1957).

258

### 259 2.3 Statistics and calculations

260 The mean and standard deviation of data were calculated with the software package R (version 3.6.0). To  
261 test for significant differences between treatments, linear model function [lm()] was used in combination  
262 with analysis of variance [aov(lm())]. The Tukey-HSD test was used as a post-hoc comparison of means;  
263 the LSD-test was applied in the case of non-equality of variances. Regression analysis was used to test for  
264 relationships between mineralogical properties and MWD, masses of aggregate size fractions, aggregate  
265 stability, and OC losses due to land-use change. Statistical differences are reported at a significance level

**Gelöscht:** Linear regression and correlation analysis was used to test for relations between independent variables.



270 of  $p < 0.05$ . Based on our selected threshold values for aluminous clay and pedogenic Fe oxides, we were  
271 able to achieve the following number of replicates for the mineralogical combinations: 'low clay–low Fe'  
272 under forest ( $n = 4$ ), 'low clay–high Fe' under forest ( $n = 4$ ), 'high clay–low Fe' under forest ( $n = 3$ ), 'high  
273 clay–high Fe' under forest ( $n = 7$ ); all cropland combinations ( $n = 3$ ).

274

275 **3. Results**

276 **3.1 Mineralogical composition and general soil properties**

277 The selected mineralogical combinations represent a broad spectrum of possible combinations, mineral  
278 aluminous clay and Fe oxide constituents. Amounts of aluminous clay varied between 149 and 438 g kg<sup>-1</sup>,  
279 and Fe<sub>d</sub> between 21 and 101 g kg<sup>-1</sup> across all sites and land uses. Amorphous Fe and Al phases  
280 contributed little to pedogenic oxides as indicated by low proportions of oxalate-extractable Fe and Al  
281 (Table 1). The advanced weathering state of study soils was also reflected in low pH and CEC<sub>eff</sub> values  
282 (Table 1).

**Gelöscht:** The mineralogical composition of the study soils was very homogeneous with kaolinite and gibbsite as the main aluminous minerals of the clay fraction, and well-crystalline goethite and hematite as dominant pedogenic Fe oxides (cf. *Kirsten et al., 2021*).

**Gelöscht:** in both

**Gelöscht:** aluminous clay and Fe oxide

**Gelöscht:** ; *Kirsten et al., 2021*

**Table 1:** Basic properties of the two soil depth increments sampled along the mineralogical combinations with aluminous clay (clay), dithionite-citrate-bicarbonate-extractable Fe ( $Fe_d$ ), total soil organic carbon content (OC),  $Fe_d$  to aluminous clay ratios ( $Fe_d/clay$ ), effective cation exchange capacity ( $CEC_{eff}$ ), hydrogen peroxide- and dithionite-citrate-bicarbonate-treated sand and silt contents, and oxalate-extractable Fe and Al content ( $Fe_o$  and  $Al_o$ ). Aluminous clay represents the weight sum of kaolinite and gibbsite present in the  $< 2\text{-}\mu\text{m}$  fraction after removal of OM and pedogenic Fe oxides. Lower case letters indicate significant differences within a certain land use as separated by depth. Sample numbers for the combinations are as follows: ‘low clay–low Fe’ under forest ( $n = 4$ ), ‘low clay–high Fe’ under forest ( $n = 4$ ), ‘high clay–low Fe’ under forest ( $n = 3$ ), ‘high clay–high Fe’ under forest ( $n = 7$ ); all cropland combinations ( $n = 3$ ).

Land use	Mineralogical Combination	Depth (cm)	(g kg <sup>-1</sup> )					(0.01 M CaCl <sub>2</sub> )		pH	$CEC_{eff}$ (cmol <sub>c</sub> kg <sup>-1</sup> )	
			Sand	Silt	Clay	$Fe_d$	$Fe_d/clay$	$Fe_o$	$Al_o$			OC
<b>Forest</b>	Low aluminous clay– Low pedogenic Fe oxides	0–5	788 <sup>a</sup> (21)	63 <sup>c</sup> (24)	149 <sup>b</sup> (19)	21 <sup>d</sup> (4)	0.15 <sup>b, A</sup> (0.04)	1.4 <sup>a</sup> (0.3)	1.2 <sup>a</sup> (0.2)	76 <sup>ab, A</sup> (27)	3.5 <sup>b</sup> (0.1)	5.7 <sup>a</sup> (2.6)
		5–10	712 <sup>a</sup> (46)	107 <sup>b</sup> (57)	181 <sup>b</sup> (19)	38 <sup>b</sup> (13)	0.21 <sup>bc, A</sup> (0.09)	1.8 <sup>a</sup> (0.3)	1.4 <sup>a</sup> (0.2)	34 <sup>a, A</sup> (6)	3.7 <sup>b</sup> (0.1)	2.9 <sup>a</sup> (0.1)
<b>Forest</b>	Low aluminous clay– High pedogenic Fe oxides	0–5	617 <sup>b</sup> (36)	201 <sup>a</sup> (52)	182 <sup>b</sup> (38)	78 <sup>a</sup> (14)	0.45 <sup>a, A</sup> (0.12)	1.3 <sup>a</sup> (0.2)	1.5 <sup>a</sup> (0.2)	57 <sup>b, A</sup> (14)	3.8 <sup>a</sup> (0.2)	5.6 <sup>a</sup> (1.7)
		5–10	647 <sup>b</sup> (49)	179 <sup>a</sup> (26)	174 <sup>b</sup> (42)	77 <sup>a</sup> (4)	0.47 <sup>a, A</sup> (0.13)	1.3 <sup>b</sup> (0.1)	1.6 <sup>a</sup> (0.3)	37 <sup>a, A</sup> (7)	3.8 <sup>ab</sup> (0.1)	3.2 <sup>a</sup> (0.9)
<b>Forest</b>	High aluminous clay– Low pedogenic Fe oxides	0–5	571 <sup>c</sup> (19)	131 <sup>b</sup> (32)	298 <sup>a</sup> (41)	36 <sup>c</sup> (5)	0.12 <sup>b</sup> (0.01)	0.9 <sup>b</sup> (0.0)	1.3 <sup>a</sup> (0.2)	43 <sup>b</sup> (6)	4.0 <sup>a</sup> (0.2)	5.2 <sup>a</sup> (1.1)
		5–10	489 <sup>c</sup> (24)	137 <sup>ab</sup> (1)	374 <sup>a</sup> (24)	44 <sup>b</sup> (7)	0.12 <sup>c</sup> (0.02)	1.0 <sup>b</sup> (0.1)	1.5 <sup>a</sup> (0.3)	23 <sup>b</sup> (5)	3.9 <sup>ab</sup> (0.1)	3.0 <sup>a</sup> (0.4)
<b>Forest</b>	High aluminous clay– High pedogenic Fe oxides	0–5	530 <sup>c</sup> (28)	152 <sup>b</sup> (24)	318 <sup>a</sup> (41)	67 <sup>b</sup> (5)	0.22 <sup>b, A</sup> (0.03)	1.2 <sup>ab</sup> (0.3)	1.9 <sup>a</sup> (0.8)	95 <sup>a, A</sup> (31)	4.1 <sup>a</sup> (0.2)	7.8 <sup>a</sup> (1.8)
		5–10	473 <sup>c</sup> (35)	178 <sup>a</sup> (45)	349 <sup>a</sup> (40)	81 <sup>a</sup> (6)	0.23 <sup>b, A</sup> (0.02)	1.3 <sup>b</sup> (0.1)	1.7 <sup>a</sup> (0.2)	35 <sup>a, A</sup> (5)	4.0 <sup>a</sup> (0.1)	4.9 <sup>a</sup> (4.0)
<b>Cropland</b>	Low aluminous clay– Low pedogenic Fe oxides	0–5	670 <sup>a</sup> (8)	103 <sup>c</sup> (4)	227 <sup>b</sup> (6)	30 <sup>c</sup> (2)	0.13 <sup>b, A</sup> (0.01)	0.6 <sup>c</sup> (0.0)	1.1 <sup>c</sup> (0.1)	19 <sup>c, B</sup> (0)	5.0 <sup>b</sup> (0.1)	5.1 <sup>b</sup> (0.2)
		5–10	669 <sup>a</sup> (8)	118 <sup>b</sup> (28)	213 <sup>b</sup> (24)	29 <sup>c</sup> (4)	0.14 <sup>b, A</sup> (0.03)	0.6 <sup>c</sup> (0.0)	1.1 <sup>b</sup> (0.1)	19 <sup>c, B</sup> (1)	5.0 <sup>ab</sup> (0.1)	5.1 <sup>b</sup> (0.2)
<b>Cropland</b>	Low aluminous clay–	0–5	602 <sup>b</sup> (17)	200 <sup>a</sup> (13)	198 <sup>b</sup> (29)	101 <sup>a</sup> (4)	0.51 <sup>a, A</sup> (0.06)	1.5 <sup>a</sup> (0.0)	47 <sup>a, A</sup> (1)	4.9 <sup>c</sup> (0.1)	5.1 <sup>b</sup> (0.2)	

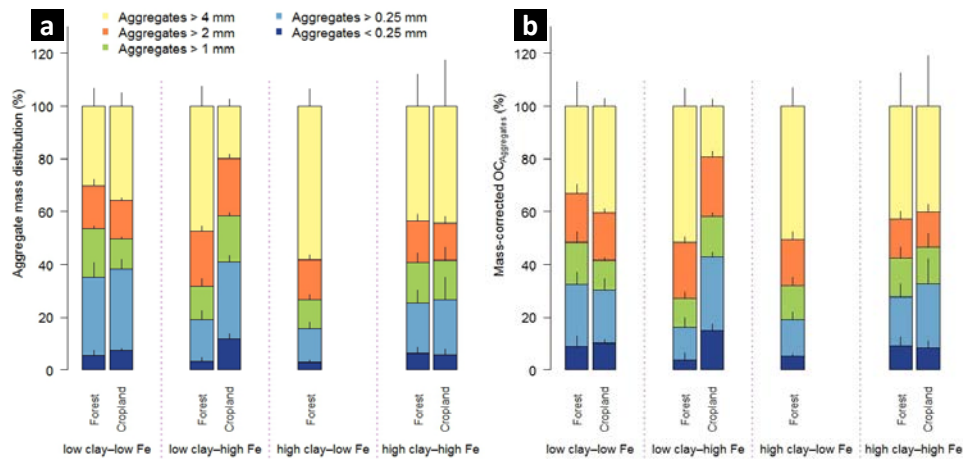
<b>High pedogenic Fe oxides</b>	5–10	<b>579<sup>b</sup></b> (19)	<b>206<sup>a</sup></b> (4)	<b>215<sup>b</sup></b> (23)	<b>100<sup>a</sup></b> (5)	<b>0.47<sup>a,A</sup></b> (0.07)	<b>1.7<sup>a</sup></b> (0.1)	<b>4.3<sup>a</sup></b> (0.6)	<b>48<sup>a,A</sup></b> (5)	<b>4.8<sup>b</sup></b> (0.1)	<b>5.0<sup>b</sup></b> (1.2)
<b>High aluminous clay–</b>	0–5	<b>437<sup>c</sup></b> (14)	<b>129<sup>b</sup></b> (12)	<b>434<sup>a</sup></b> (18)	<b>63<sup>b</sup></b> (3)	<b>0.15<sup>b,B</sup></b> (0.01)	<b>1.2<sup>b</sup></b> (0.0)	<b>1.4<sup>b</sup></b> (0.0)	<b>34<sup>b,B</sup></b> (1)	<b>5.4<sup>a</sup></b> (0.0)	<b>9.4<sup>a</sup></b> (0.5)
<b>High pedogenic Fe oxides</b>	5–10	<b>399<sup>c</sup></b> (18)	<b>163<sup>ab</sup></b> (35)	<b>438<sup>a</sup></b> (17)	<b>66<sup>b</sup></b> (4)	<b>0.15<sup>b,B</sup></b> (0.01)	<b>1.2<sup>b</sup></b> (0.1)	<b>1.3<sup>b</sup></b> (0.2)	<b>30<sup>b,A</sup></b> (3)	<b>5.2<sup>a</sup></b> (0.1)	<b>7.3<sup>a</sup></b> (0.7)

303 **3.2 Aggregate size distribution**

304 The studied soils were highly aggregated and showed significant variation in their aggregate size  
305 distribution across the mineralogical combinations (Table 2). For most combinations, about 74% of soil  
306 mass was present in aggregates > 2 mm (Figure 1a), whereas in forest soils with low contents in both  
307 aluminous clay and Fe oxides only 40% could be assigned to aggregates > 2 mm. Only 3–12% of total soil  
308 mass remained in < 0.25 mm aggregates (Table 2). The low clay–low Fe combination under forest  
309 displayed the significant smallest MWD, with 2.9 mm in 0–5 cm depth and 3.7 mm in 5–10 cm depth  
310 (Table 2). In contrast, the low clay–high Fe combination always had the largest MWD (4.8 mm in 0–5 cm  
311 depth, and 4.6 mm in 5–10 cm depth) among the other forest combinations. Our data suggest that the  
312 MWD under forest is significantly positively influenced by the Fe<sub>d</sub> content (MWD<sub>Forest 0–5 cm</sub>: r<sup>2</sup> = 0.4,  
313 p < 0.001; MWD<sub>Forest 5–10 cm</sub>: r<sup>2</sup> = 0.15, p = 0.06), whereas nearly no effect was observed for aluminous clay  
314 (MWD<sub>Forest 0–5 cm</sub>: r<sup>2</sup> < 0.01, p = 0.79; MWD<sub>Forest 5–10 cm</sub>: r<sup>2</sup> < 0.01, p = 0.30, Table S1). Contrary to the  
315 mineralogical combinations under forest, the significant smallest MWD under cropland was within the  
316 low clay–high Fe combination (2.7 mm in 0–5 cm depth and 2.7 mm in 5–10 cm depth; Table 2). The low  
317 clay–low Fe and high clay–high Fe cropland combinations showed no strong differences in their MWDs.  
318 Nonetheless, a significant negative linear relationship existed between MWD and the pedogenic-Fe to  
319 aluminous clay ratio (MWD<sub>Cropland 0–5 cm</sub>: r<sup>2</sup> = 0.47, p = 0.03; MWD<sub>Forest 5–10 cm</sub>: r<sup>2</sup> = 0.47, p = 0.02) for the  
320 mineralogical combinations under cropland (Table S1).

321 Corresponding to the smallest MWD, the low clay–low Fe forest combination contained the  
322 smallest fraction of > 4 mm aggregates. The contribution of these large aggregates under forest increased  
323 in the order: low clay–low Fe < low clay–high Fe = high clay–high Fe < high clay–low Fe (Figure 1a). For  
324 croplands, the low clay–high Fe combination comprised the smallest amount of > 4 mm aggregates  
325 whereas the high clay–high Fe combination exhibited the respective highest share (Figure 1a). The  
326 explained variance of > 4 mm aggregate mass due to aluminous clay and Fe<sub>d</sub> was generally low, except for  
327 the cropland combinations (positive effect of aluminous clay and negative effect of pedogenic Fe; Table  
328 S1).

**Gelöscht:** For the low clay–low Fe combination under forest, about 40% of the total soil mass prevailed in > 2 mm aggregates, while in the high clay–low Fe combination 74% were assigned to this fraction (Figure 1a). Furthermore, o



336  
 337 **Figure 1:** Aggregate size distribution of the combined 0–5 and 5–10 cm depth increments (a), and relative  
 338 mass-corrected OC contents (b) along the mineralogical combinations. Clay represents the weight sum of  
 339 kaolinite and gibbsite present in the < 2- $\mu$ m fraction after removal of OM and pedogenic Fe oxides, and  
 340 Fe denotes the content of pedogenic Fe oxides extracted with dithionite-citrate-bicarbonate. Sample  
 341 numbers for the combinations are as follows: 'low clay–low Fe' under forest ( $n = 4$ ), 'low clay–high Fe'  
 342 under forest ( $n = 4$ ), 'high clay–low Fe' under forest ( $n = 3$ ), 'high clay–high Fe' under forest ( $n = 7$ ); all  
 343 cropland combinations ( $n = 3$ ).  
 344

345 The mineralogical combinations affected the amounts of 2–4 mm aggregates differently than those  
 346 of > 4 mm aggregates. The low clay–high Fe combination under forest and cropland contained slightly but  
 347 significantly more 2–4 mm aggregates (Figure 1a), being associated with a significantly higher  $Fe_d$  to  
 348 aluminous clay ratio (Table 1). In fact, in a multiple regression model for the entire data set (combined  
 349 land uses and depths), we observed a positive relationship between the mass of 2–4 mm aggregates and  
 350  $Fe_d$  content, whereas the content of aluminous clay had a negative effect ( $r^2 = 0.57$ ,  $p < 0.001$ ; Table S1).  
 351 The same model separated by soil depth showed similar relationships (Table S1). Across all mineralogical  
 352 combinations, amounts of < 0.25 mm aggregates were principally comparable, despite of significantly  
 353 higher shares in the low clay–low Fe and high clay–high Fe combinations under forest. In contrast, a  
 354 significant larger amount of < 0.25 mm aggregates was observed in the low clay–high Fe combination  
 355 under cropland. In this mineralogical combination, land-use change caused a quadrupling of < 0.25 mm  
 356 aggregate mass from about 30 to nearly 120  $g\ kg^{-1}$  (Table 2). In contrast to the macroaggregate fractions

357 shown above, there was no correlation between mineralogical parameters and the mass of < 0.25 mm  
358 aggregates, neither for the entire data set (combined land uses and depths) nor when separated by soil  
359 depth (Table S1). Only under cropland we observed a negative effect of aluminous clay and a positive  
360 influence of  $Fe_d$  on microaggregate contents (aggregate mass < 0.25 mm<sub>0-5 cm</sub>:  $r^2 = 0.8$ ,  $p = 0.004$ ;  
361 aggregate mass < 0.25 mm<sub>5-10 cm</sub>:  $r^2 = 0.61$ ,  $p = 0.03$ ).

**Table 2:** Aggregate masses (mass) and OC content of aggregate size fractions (dry sieving) within different combinations of aluminous clay and pedogenic Fe oxides, OC change ( $\Delta$ OC) between land uses within a certain mineralogical combination and depth, and related mean weight diameter (MWD). Aluminous clay represents the weight sum of kaolinite and gibbsite present in the < 2- $\mu$ m fraction after removal of OM and pedogenic Fe oxides. Lower case letters indicate significant differences within a certain land use separated by depth, and capital letters denote significant differences between land uses. Sample numbers for the combinations are as follows: 'low clay-low Fe' under forest ( $n = 4$ ), 'low clay-high Fe' under forest ( $n = 4$ ), 'high clay-low Fe' under forest ( $n = 3$ ), 'high clay-high Fe' under forest ( $n = 7$ ); all cropland combinations ( $n = 3$ ).

Land use	Mineralogical Combination	Depth (cm)	> 4 mm			1-2 mm			0.25-1 mm			< 0.25 mm						
			mass (g kg <sup>-1</sup> )	OC (%)	$\Delta$ OC (%)	mass (g kg <sup>-1</sup> )	OC (%)	$\Delta$ OC (%)	mass (g kg <sup>-1</sup> )	OC (%)	$\Delta$ OC (%)	mass (g kg <sup>-1</sup> )	OC (%)	$\Delta$ OC (%)	MWD (mm)			
Forest	Low aluminous clay-pedogenic Fe oxides	0-5	249 <sup>c,A</sup> (33)	76 <sup>b,A</sup> (32)	na	144 <sup>b,A</sup> (21)	83 <sup>a,A</sup> (22)	na	191 <sup>b,A</sup> (4)	65 <sup>b,A</sup> (9)	na	345 <sup>b,A</sup> (40)	56 <sup>b,b,A</sup> (18)	na	70 <sup>b,A</sup> (15)	125 <sup>ab,A</sup> (51)	na	2.9 <sup>c,A</sup> (0.3)
		5-10	343 <sup>b,A</sup> (61)	40 <sup>b,A</sup> (8)	na	176 <sup>ab,A</sup> (21)	39 <sup>a,A</sup> (10)	na	181 <sup>b,A</sup> (15)	27 <sup>a,A</sup> (9)	na	257 <sup>b,A</sup> (36)	28 <sup>a,A</sup> (5)	na	44 <sup>a,B</sup> (11)	51 <sup>b,A</sup> (17)	na	3.7 <sup>b,A</sup> (0.4)
	Low aluminous clay-High pedogenic Fe oxides	0-5	493 <sup>ab,A</sup> (99)	68 <sup>ab,A</sup> (19)	na	210 <sup>a</sup> (20)	65 <sup>a,A</sup> (22)	na	115 <sup>b,B</sup> (38)	62 <sup>a,A</sup> (25)	na	150 <sup>c,B</sup> (42)	49 <sup>b,A</sup> (25)	na	33 <sup>b,B</sup> (14)	62 <sup>b,A</sup> (36)	na	4.8 <sup>a,A</sup> (0.7)
Forest	High pedogenic Fe oxides	5-10	451 <sup>ab,A</sup> (36)	40 <sup>b,A</sup> (11)	na	210 <sup>ab,A</sup> (27)	36 <sup>b,B</sup> (5)	na	139 <sup>ab,B</sup> (10)	29 <sup>a,A</sup> (7)	na	166 <sup>b,B</sup> (24)	31 <sup>a,A</sup> (11)	na	34 <sup>a,B</sup> (20)	44 <sup>a,A</sup> (18)	na	4.6 <sup>b,A</sup> (0.3)
		0-5	604 <sup>a</sup> (84)	38 <sup>b</sup> (5)	na	140 <sup>b</sup> (21)	63 <sup>a</sup> (34)	na	100 <sup>b</sup> (21)	80 <sup>a</sup> (51)	na	125 <sup>c</sup> (31)	62 <sup>ab</sup> (28)	na	31 <sup>b</sup> (13)	101 <sup>ab</sup> (59)	na	4.3 <sup>ab</sup> (0.4)
	Low pedogenic Fe oxides	5-10	561 <sup>a</sup> (47)	26 <sup>a</sup> (14)	na	163 <sup>b</sup> (12)	28 <sup>b</sup> (7)	na	118 <sup>b</sup> (17)	22 <sup>a</sup> (3)	na	127 <sup>b</sup> (21)	25 <sup>a</sup> (6)	na	30 <sup>a</sup> (1)	43 <sup>a</sup> (18)	na	4.1 <sup>a</sup> (0.2)
Forest	High aluminous clay-High pedogenic Fe oxides	0-5	397 <sup>b,A</sup> (91)	86 <sup>a,A</sup> (21)	na	157 <sup>b,A</sup> (27)	89 <sup>a,A</sup> (32)	na	163 <sup>a,A</sup> (32)	99 <sup>a,A</sup> (50)	na	208 <sup>b,B</sup> (36)	91 <sup>a,A</sup> (38)	na	74 <sup>a,A</sup> (14)	133 <sup>a,A</sup> (47)	na	4.0 <sup>b,A</sup> (0.6)
	High pedogenic Fe oxides	5-10	474 <sup>ab,A</sup> (139)	35 <sup>a,A</sup> (7)	na	156 <sup>b,A</sup> (27)	33 <sup>ab,A</sup> (4)	na	146 <sup>ab,A</sup> (41)	30 <sup>a,A</sup> (4)	na	172 <sup>b,A</sup> (61)	34 <sup>a,A</sup> (4)	na	52 <sup>a,A</sup> (26)	51 <sup>b,A</sup> (6)	na	4.6 <sup>a,A</sup> (1.0)
Cropland	Low aluminous clay-Low pedogenic Fe oxides	0-5	347 <sup>a,A</sup> (69)	20 <sup>b,B</sup> (3)	-73	147 <sup>b,A</sup> (13)	21 <sup>c,B</sup> (1)	-75	115 <sup>b,B</sup> (4)	17 <sup>c,B</sup> (1)	-74	318 <sup>a,A</sup> (52)	11 <sup>c,B</sup> (3)	-80	74 <sup>b,A</sup> (12)	24 <sup>c,B</sup> (1)	-81	3.6 <sup>b,A</sup> (0.5)
		5-10	368 <sup>ab,A</sup> (28)	20 <sup>b,B</sup> (1)	-50	143 <sup>b,A</sup> (8)	22 <sup>b,B</sup> (5)	-44	113 <sup>b,B</sup> (10)	17 <sup>b,A</sup> (2)	-37	299 <sup>a,A</sup> (15)	11 <sup>c,B</sup> (2)	-61	77 <sup>b,A</sup> (1)	24 <sup>c,A</sup> (3)	-53	3.7 <sup>b,A</sup> (0.2)
Cropland	Low aluminous clay-High pedogenic Fe oxides	0-5	201 <sup>b,B</sup> (39)	47 <sup>a,A</sup> (7)	-30	212 <sup>a,A</sup> (12)	49 <sup>a,A</sup> (2)	-25	173 <sup>b,A</sup> (18)	42 <sup>a,A</sup> (3)	-32	296 <sup>b,A</sup> (33)	46 <sup>a,A</sup> (1)	-6	119 <sup>a,A</sup> (4)	62 <sup>b,A</sup> (2)	$\pm 0$	2.7 <sup>b,B</sup> (0.3)



<b>High pedogenic Fe oxides</b>	5–10	<b>194<sup>c,B</sup></b> (11)	<b>47<sup>b,A</sup></b> (13)	+18	<b>224<sup>a,A</sup></b> (15)	<b>49<sup>a,A</sup></b> (4)	+36	<b>177<sup>a,A</sup></b> (1)	<b>42<sup>a,A</sup></b> (6)	+45	<b>287<sup>a,A</sup></b> (13)	<b>45<sup>a,A</sup></b> (3)	+45	<b>118<sup>a,A</sup></b> (29)	<b>58<sup>a,A</sup></b> (9)	+32	<b>2,7<sup>c,B</sup></b> (0.1)
<b>Cropland High aluminous clay–</b>	0–5	<b>296<sup>ab,A</sup></b> (40)	<b>26<sup>b,B</sup></b> (6)	-71	<b>159<sup>b,A</sup></b> (8)	<b>29<sup>b,B</sup></b> (7)	-67	<b>191<sup>a,A</sup></b> (2)	<b>28<sup>b,B</sup></b> (4)	-71	<b>278<sup>a,A</sup></b> (25)	<b>35<sup>b,A</sup></b> (2)	-62	<b>77<sup>b,A</sup></b> (10)	<b>41<sup>b,B</sup></b> (1)	-69	<b>3,3<sup>ab,A</sup></b> (0.3)
<b>High pedogenic Fe oxides</b>	5–10	<b>593<sup>a,A</sup></b> (95)	<b>25<sup>b,A</sup></b> (3)	-29	<b>118<sup>b,A</sup></b> (21)	<b>26<sup>b,B</sup></b> (2)	-21	<b>107<sup>b,A</sup></b> (29)	<b>25<sup>b,A</sup></b> (4)	-17	<b>138<sup>b,A</sup></b> (37)	<b>32<sup>b,A</sup></b> (3)	-6	<b>43<sup>b,A</sup></b> (10)	<b>41<sup>b,B</sup></b> (5)	-20	<b>5,3<sup>ab,A</sup></b> (0.6)

na = not applicable.

368

369

370 In summary, mineralogical combinations and land use significantly affected the aggregate size  
371 distribution of soils, despite quantitative relations to mineralogical proxies could not be observed for each  
372 aggregate class. In undisturbed forest soils, higher pedogenic Fe contents resulted in increasing MWD  
373 especially in 0–5 cm depth and significantly larger amounts of > 2 mm aggregates. The conversion from  
374 forest to croplands either decreased MWD, as particularly observed for the low clay–high Fe combination,  
375 or had no effect (low clay–low Fe). Overall, the observed differences in aggregate masses and MWD were  
376 surprisingly moderate, given the widely differing contents in aluminous clay and Fe oxides across the  
377 mineralogical combinations.

378

### 379 **3.3 Aggregate stability**

380 In general, there was little variation of MWD values for >4 mm aggregates over all mineralogical  
381 combinations. In fact, the MWD of this fraction was always close to its calculated mean diameter (6 mm;  
382 calculation was done after (*Youker and McGuinness, 1957*)), overall indicating a high stability.  
383 Nevertheless, there were some minor differences in aggregate stability across mineralogical combinations.  
384 The low clay–low Fe and high clay–low Fe combinations had a significantly lower aggregate stability in  
385 comparison with the two other combinations under the two land uses (Table 3). The slightly higher  
386 abundance of 2–4 mm aggregates in the low clay–high Fe combination under forest and cropland was  
387 accompanied by a significantly higher aggregate stability under both land uses (Table 2 and 3). In  
388 summary, all aggregates can be classified as stable with only minor differences imposed by the  
389 mineralogical combinations. Slightly higher aggregate stability was associated with a larger amount of  
390 pedogenic Fe, and increasing Fe<sub>d</sub> to aluminous clay ratios, whereas differences in the amount of aluminous  
391 clay had almost no effect on the aggregate stability (Table S2).

392

393

394

395

396 **Table 3:** Aggregate stability of selected aggregate size fractions after applying the fast wetting procedure  
 397 along the different combinations of aluminous clay and pedogenic Fe oxides, indicated by the resulting  
 398 mean weight diameter (MWD). Aluminous clay represents the weight sum of kaolinite and gibbsite  
 399 present in the < 2- $\mu$ m fraction after removal of OM and pedogenic Fe oxides. Lower case letters indicate  
 400 significant differences within a certain land use separated by depth, and capital letters denote significant  
 401 differences between land uses. Sample numbers for the combinations are as follows: 'low clay–low Fe'  
 402 under forest ( $n = 4$ ), 'low clay–high Fe' under forest ( $n = 4$ ), 'high clay–low Fe' under forest ( $n = 3$ ), 'high  
 403 clay–high Fe' under forest ( $n = 7$ ); all cropland combinations ( $n = 3$ ).

Land use	Mineralogical combination	Depth (cm)	MWD	
			Fast wetting > 4 mm	Fast wetting 2–4 mm
Forest	Low aluminous clay–	0–5	4.9 <sup>b,A</sup> (0.4)	2.6 <sup>b,A</sup> (0.1)
	Low pedogenic Fe oxides	5–10	5.1 <sup>a,A</sup> (0.3)	2.4 <sup>b,A</sup> (0.3)
Forest	Low aluminous clay–	0–5	5.6 <sup>a,A</sup> (0.2)	2.8 <sup>a,A</sup> (0.1)
	High pedogenic Fe oxides	5–10	4.9 <sup>a,A</sup> (0.9)	2.7 <sup>a,A</sup> (0.1)
Forest	High aluminous clay–	0–5	5.4 <sup>ab</sup> (0.4)	2.7 <sup>b</sup> (0.0)
	Low pedogenic Fe oxides	5–10	4.5 <sup>a</sup> (1.2)	2.4 <sup>b</sup> (0.3)
Forest	High aluminous clay–	0–5	5.5 <sup>a,A</sup> (0.2)	2.6 <sup>b,A</sup> (0.1)
	High pedogenic Fe oxides	5–10	5.2 <sup>a,A</sup> (0.4)	2.6 <sup>ab,B</sup> (0.1)
Cropland	Low aluminous clay–	0–5	4.4 <sup>b,A</sup> (0.1)	2.6 <sup>c,A</sup> (0.0)
	Low pedogenic Fe oxides	5–10	4.9 <sup>b,A</sup> (0.3)	2.4 <sup>b,A</sup> (0.1)
Cropland	Low aluminous clay–	0–5	5.2 <sup>a,A</sup> (0.2)	2.9 <sup>a,A</sup> (0.0)
	High pedogenic Fe oxides	5–10	5.3 <sup>ab,A</sup> (0.1)	2.8 <sup>a,A</sup> (0.0)
Cropland	High aluminous clay–	0–5	4.9 <sup>a,B</sup> (0.2)	2.7 <sup>b,A</sup> (0.1)
	High pedogenic Fe oxides	5–10	5.6 <sup>a,A</sup> (0.2)	2.8 <sup>a,A</sup> (0.0)

404

### 405 3.4 Organic carbon in soils and aggregate size fractions

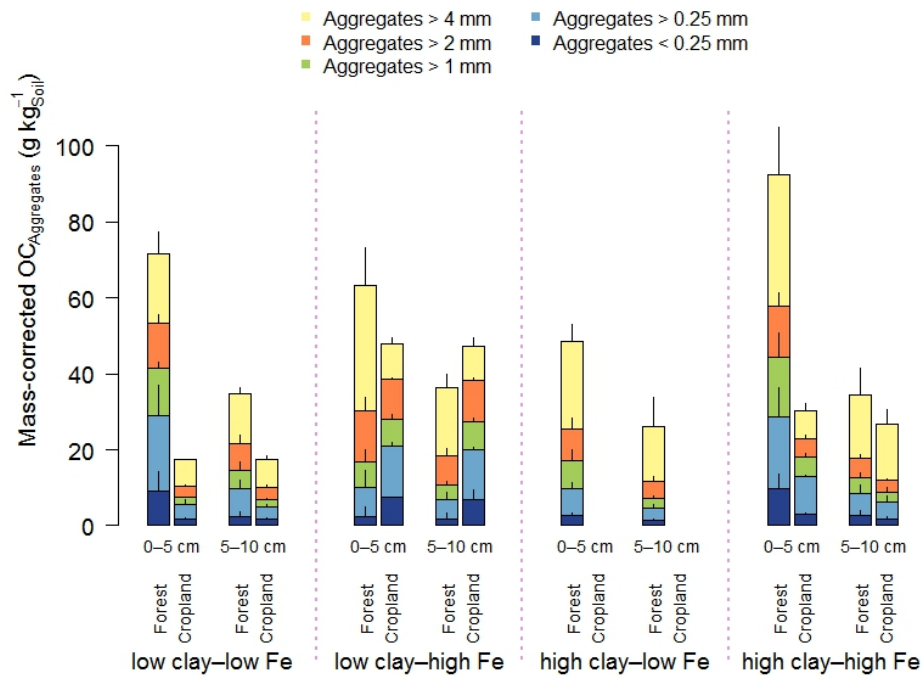
406 Variation in mineral constituents caused different soil OC contents, ranging between 19 to 95 g OC kg<sup>-1</sup>  
 407 across all sites including both land use and depth (Table 1). A significant proportion of the total OC  
 408 content of all forest soils was present in > 4 mm aggregates in both depth increments (low clay–low Fe:

**Gelöscht:** As outlined in *Kirsten et al.* (2021), significantly higher OC stocks were observed for low clay–high Fe combination under cropland and for high clay–high Fe combinations under forest. Forest conversion to cropland caused marked OC losses for the low clay–low Fe combination but no or minor losses for the low clay–high Fe combination (Table 1; *Kirsten et al.*, 2021).¶

420 33% < high clay–high Fe: 43% < high clay–low Fe: 51% < low clay–high Fe: 52%; Figure 1b). Forest to  
 421 cropland conversion caused OC losses from most aggregate size fractions (Figure 2). For the > 4 mm  
 422 aggregates this was significantly modified by the mineralogical combinations at least at 0–5 cm depth,  
 423 generally following the order: low clay–high Fe < high clay–high Fe < low clay–low Fe (Table S3).  
 424 Losses of OC from aggregate size fractions were generally higher at 0–5 than at 5–10 cm depth (Figure 2).  
 425 As mentioned above, no significant loss of total OC occurred for the low clay–high Fe combination,  
 426 irrespective of the significant mass redistribution of the > 4 mm aggregate fraction into smaller aggregate  
 427 fractions (Table 2). Although there were differences in OC losses among mineralogical combinations,  
 428 there was little indication that coarser aggregate size fractions lost more OC than smaller ones (Table 2).  
 429

**Gelöscht:** decline

**Gelöscht:** Hence, OC formerly associated with large macroaggregates persisted the land-use conversion to croplands residing in newly formed smaller aggregates. While



430 **Figure 2:** Mass-corrected OC contents of aggregate size fractions along the mineralogical combinations.  
 431 Clay represents the weight sum of kaolinite and gibbsite present in the < 2- $\mu$ m fraction after removal of  
 432 OM and pedogenic Fe oxides, and Fe denotes the content of pedogenic Fe oxides extracted with  
 433 dithionite-citrate-bicarbonate. Sample numbers for the combinations are as follows: 'low clay–low Fe'  
 434

441 under forest ( $n = 4$ ), 'low clay–high Fe' under forest ( $n = 4$ ), 'high clay–low Fe' under forest ( $n = 3$ ), 'high  
442 clay–high Fe' under forest ( $n = 7$ ); all cropland combinations ( $n = 3$ ).  
443

444 **4. Discussion**

445 The aggregate size distribution of soils along the mineralogical combinations under both land uses were in  
446 the range of values reported for African soils. For example, soils with strongly contrasting clay content  
447 (220 and 650 g kg<sup>-1</sup>) but similar clay mineralogy (kaolinite) in the central highlands of Kenya displayed  
448 macroaggregate contents of 245 and 636 g kg<sup>-1</sup> soil, respectively (*Gentile et al., 2010*). In addition, for  
449 soils from the catchment of the Riru river also located in the central highlands of Kenya it was shown that  
450 macroaggregates (2–4.2 mm) displayed a large stability (*Kamamia et al., 2021*). The reported MWD's  
451 after application of the fast-wetting stability test were 2.5 mm for cropland and 3.2 mm for indigenous  
452 forest sites (*Kamamia et al., 2021*). These values are close to those observed in our study soils for 2–4 mm  
453 aggregates. In contrast, soils in Brazil under native forest vegetation and similar mineral composition  
454 (kaolinite, gibbsite, hematite) even subsumed over 90% of total aggregate mass in > 2 mm aggregates  
455 (*Maltoni et al., 2017*). Nonetheless, reported data all point at a better soil structure and aggregate stability  
456 of tropical soils dominated by low-activity clay minerals and well-crystalline Fe oxides, which is  
457 consistent with all mineralogical combinations of this study.

**Gelöscht:** and also high aggregate stability with MWD values of the 2–4.6 mm aggregates ranging from 2.5 to 3.2 mm (*Kamamia et al., 2021*).

458  
459 **4.1 Aggregation and aggregate stability as controlled by aluminous clay and pedogenic Fe oxides**

460 Our data demonstrates relatively small differences in aggregation among the generally well-aggregated  
461 study soils, being characterized by high aggregate stability despite of large variations in aluminous clay  
462 (factor three) and pedogenic Fe (factor five) contents. Yet, we noticed some distinct modifications of the  
463 aggregation size distribution and aggregate stability in both forest and cropland soils. The low clay–low  
464 Fe soil under forest had a significantly smaller amount of > 4 mm and 2–4 mm aggregates and a  
465 significantly lower MWD than all other mineralogical combinations. Notably, a combined increase in  
466 aluminous clay and Fe oxides did not necessarily cause a shift towards larger aggregates and thus higher  
467 MWD (see low clay–high Fe forest). Furthermore, the low clay–low Fe and high clay–high Fe  
468 combinations under forest contained more < 0.25 mm aggregates. Thus, under undisturbed soil conditions  
469 it appears that the formation of larger aggregates is promoted if one of the two aggregate-forming mineral

**Gelöscht:** Our data demonstrates that mineralogical combinations, with contents of aluminous clay varying by factor three and pedogenic Fe oxides by factor five, did not result in entirely different aggregation and stability patterns in the studied weathered tropical soils.

482 fractions is more abundant than the other (high clay–low Fe and low clay–high Fe combinations). We  
483 assume that the positive effect of increasing aluminous clay content on the aggregate mass > 4 mm is  
484 related to the hybrid electrostatic properties of kaolinite on edges (variable) and surfaces (permanent  
485 negative), which enable the formation of characteristic cards-house structures (*Qafoku and Sumner, 2002*).  
486 In addition to this increase in aggregation caused by the dominance in kaolinitic properties (i.e. high  
487 clay–low Fe), we also expect that, similar to the study by *Dultz et al. (2019)*, there are mixing ratios  
488 between aluminous clay and pedogenic Fe minerals, which lead to improved aggregation (greater MWD;  
489 i.e. low clay–high Fe). This effect is probably explained by changes in the electrostatic properties of the  
490 mineralogical combinations, as was shown in the study by *Hou et al. (2007)* for kaolinite in different  
491 relative combinations with goethite and hematite. Nevertheless, aluminous clay is the decisive control for  
492 macroaggregation in these weathered tropical soils, confirming the often described promoting effect of  
493 increasing clay content on aggregation (*Feller and Beare, 1997*). Furthermore, the high clay–low Fe and  
494 high clay–high Fe combinations under forest also nicely demonstrate how nearly equal amounts of  
495 aluminous clay plus pedogenic Fe oxides (i.e. similar clay contents) cause different amounts of > 4 mm  
496 aggregates. Consequently, the connection between textural properties and aggregation can remain hidden  
497 (*Barthès et al., 2008*) without considering the mineralogical composition of the whole clay fraction  
498 (*Fernández-Ugalde et al., 2013; King et al., 2019; West et al., 2004*).

499 Land-use change had a distinct impact on aggregate distribution like indicated in other studies  
500 (*Feller and Beare, 1997; Six et al., 2002*) and depended also on the mineralogical combinations, though  
501 croplands not followed the trajectory observed under forest. A significantly lower MWD under low  
502 clay–high Fe rather than low clay–low Fe can be mainly attributed to a reduced amount of > 4 mm  
503 aggregates. We assume that differences in the ratio of pedogenic Fe to aluminous clay in the low clay–low  
504 Fe and high clay–high Fe (0.13 to 0.15) in comparison with the low clay–high Fe combination (0.47 to  
505 0.51) under cropland explains the stability of ‘card-house’ structures like described for mineralogically  
506 similar Oxisols from Brazil and India (*Bartoli et al., 1992*). Accordingly, a higher Fe<sub>d</sub> to aluminous clay  
507 ratios seems to be disadvantageous for the formation of such structures, especially in > 4 mm aggregates.

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510 The different pH-dependent charge characteristics of kaolinite and pedogenic Fe oxides (*Kaiser* and  
511 *Guggenberger*, 2003), and their relative share can lead to altered charge properties of soils (*Anda* et al.,  
512 2008). We hypothesize, that an increasing amount of Fe oxides in the investigated mineralogical  
513 combinations adds more positive charge, thus possibly reducing structural integrity and aggregate stability  
514 if not sufficiently compensated by OM or clay minerals. Furthermore, in the low clay–high Fe cropland  
515 combination, land-use change caused a significant four-fold increase of < 0.25 mm aggregates due to the  
516 breakdown of > 4 mm aggregates. Nonetheless, our results show that agricultural management does not  
517 necessarily decreases macroaggregation and related MWD's, like reported in *Rabbi* et al. (2015).

Gelöscht: elsewhere

Gelöscht: (*Rabbi* et al., 2015)

518 The dominant role of pedogenic Fe oxides for macroaggregation under undisturbed tropical soil  
519 conditions proposed by *Six* et al. (2002) cannot be confirmed in our study. This is because the low  
520 clay–high Fe forest soil contained a smaller amount of > 4 mm aggregates compared to the high clay–low  
521 Fe forest soil in both depth increments. Consequently, this rather points at the importance of kaolinite for  
522 macroaggregation, which is in line with results from two Oxisols in Brazil (*Vrdoljak* and *Sposito*, 2002),  
523 showing kaolinite being the backbone of the investigated aggregate size fractions. The less intense  
524 formation of > 4 mm aggregates in the low clay–high Fe forest combination was also observed under  
525 cropland, whereas the low clay–low Fe and high clay–high Fe croplands showed either no significant  
526 decrease or even an increase in > 4 mm aggregate mass. Thus, simultaneous abundance of large amounts  
527 of aluminous clay and pedogenic Fe oxides preserved a higher aggregate stability than under  
528 mineralogically imbalanced conditions, although no conclusions can be drawn for the high clay–low Fe  
529 combination. Nonetheless, > 4 mm aggregates had a higher resistance to field operations in mineralogical  
530 combinations with lower Fe<sub>a</sub> to aluminous clay ratios (0.13 to 0.15).

531 In contrast to the > 4 mm aggregates, 2–4 mm aggregates corresponded more clearly to the positive  
532 effect of pedogenic Fe oxides on aggregation and aggregate stability as proposed for weathered tropical  
533 soils (*Igwe* et al., 2013; *Peng* et al., 2015; *Six* et al., 2002). Both, the low clay–high Fe forest and low  
534 clay–high Fe cropland soils contained somewhat but significantly more 2–4 mm aggregates than other  
535 mineral combinations in concert with a higher aggregate stability of this particular fraction. This finding



538 also demonstrates that mineral interactions forming water-stable aggregates in tropical soils are differently  
539 affected by a given mineralogical combination. Higher  $Fe_d$  to aluminous clay ratios ( $> 0.45$ ) modulate  
540 aggregate distribution towards aggregates 2–4 mm, whereas distinctly lower values (high clay–low Fe  
541 forest: 0.12) shifted the maximum to  $> 4$  mm aggregates. Overall, the two macroaggregate fractions  
542 discussed above are differentially affected by the mineralogical combinations, although the magnitude was  
543 less than expected, given the pronounced variation in aluminous clay and Fe contents.

544

#### 545 **4.2 Importance of aggregation for OC persistence – effects of aluminous clay and pedogenic Fe ox-** 546 **ides**

547 Clay minerals and Fe oxides are considered as important mineral constituents fostering aggregation and  
548 subsequent OC storage via physical protection (*Denef et al., 2004*). The overwhelming portion of OC in  
549 the studied topsoils resided in mineral-organic associations (35–81%), whereas OC occluded in  
550 aggregates amounted to 7–24%, with a lower share under cropland than forest as determined by density  
551 fractionation (*Kirsten et al., 2021*). The low clay–high Fe cropland had an OC content more than twice  
552 larger than that of the low clay–low Fe cropland, but comprised a significantly smaller MWD. Thus, a  
553 shift towards more macroaggregation, indicated by a larger MWD in certain mineralogical combinations,  
554 did not result in higher total OC storage, like shown for other tropical soils (*Barthès et al., 2008; Bartoli et*  
555 *al., 1991; Spaccini et al., 2001*). The OC content of the  $> 4$  mm aggregate and 2–4 mm aggregate fractions  
556 accounted for 42 to 73% of the total soil OC content (Figure 1b). This, however, does not *per se* indicate  
557 the relevance of macroaggregation for OC storage in weathered tropical soils like proposed by others  
558 (*Feller and Beare, 1997; King et al., 2019; Six et al., 2002*). The high clay–low Fe forest with the highest  
559 share in  $> 4$  mm and 2–4 mm aggregates had significant lower OC contents in these fractions than most  
560 other mineralogical combinations. Comparing forest with cropland soils (Table 2), we observed  
561 significantly reduced OC contents in the majority of macroaggregate fractions of the low clay–low Fe and  
562 high clay–high Fe croplands, as reported in other studies (*Blanco-Canqui and Lal, 2004; Lobe et al.,*  
563 *2011*). In contrast, fewer changes of aggregate-associated and total soil OC contents was observed in the

**Gelöscht:** Furthermore, if land-use change is taken into account

**Gelöscht:** least

567 low clay–high Fe combination, despite it experienced the strongest disaggregation of the largest  
568 macroaggregates (Figure 1a and Figure 2). We conclude that larger amounts of > 2 mm aggregates or  
569 higher stability during wet sieving not automatically translates into higher aggregate-associated OC  
570 contents, as reported for Ferralsols (*Maltoni et al., 2017*). Given all these observations and the fact that  
571 occluded OM determined by density fractionation was mostly of subordinate relevance, particularly in  
572 croplands, OC storage in study soils seems rather disconnected from their aggregation status.  
573 Consequently, the loss of large aggregates and the mass redistribution into smaller aggregate size fractions  
574 does not automatically imply a loss of soil OC, because a substantial part of the OC in aggregate fractions  
575 is bound to minerals with a higher persistence against land-use change (*Kirsten et al., 2021*). Here, density  
576 fractionation could shed more light on the nature and quantity of OM located in certain aggregate size  
577 fractions.

578         Microaggregates contained the highest OC content per unit of mass for almost all mineralogical  
579 combinations, depth increments, and land uses (Table 2). This is in line with the findings of *Chenu* and  
580 *Plante* (2006) and *Lobe et al.* (2011) that microaggregates can significantly contribute to OC storage. As  
581 aggregates were isolated by dry sieving, these microaggregates were not located inside larger aggregates,  
582 rendering them principally better accessible for OC allocation. Particularly OC contained in the  
583 < 0.25 mm aggregates of the low clay–high Fe combination revealed a strong persistence against land-use  
584 change, which explains well the unaltered soil OC contents upon land-use change.

585

586 **5. Conclusions**

587 Classification of soils into mineralogical combinations of aluminous clay and pedogenic Fe oxides  
588 revealed significant effects of mineral constituents on soil structure and related OC storage in weathered  
589 tropical soils. Despite that, overall patterns across combinations were more similar than different, *i.e.*,  
590 always comprising a high level of macroaggregation and aggregate stability. Aggregates > 4 mm of the  
591 low clay–low Fe and high clay–high Fe combinations were less affected by land-use change, thus  
592 pedogenic Fe in a certain relation with aluminous clay (0.13 to 0.23) seems beneficial to maintain the  
593 structural integrity of macroaggregates. Despite the high physical stability, OC contents of  
594 macroaggregates declined substantially in most mineralogical combinations during forest–cropland  
595 conversion. This highlights the fact that structural integrity of macroaggregates during land-use change  
596 cannot be equated with OC persistence. For the low clay–high Fe combination, substantial destruction of  
597 > 4 mm aggregates during land-use change due to agricultural management was also not accompanied by  
598 higher OC losses. Hence, we must reject our initial hypothesis that the mineralogical combination that  
599 results in the greatest aggregate stability best preserves OC during the conversion from forest to cropland.  
600 Thus, the formation of macroaggregates cannot be considered as a main stabilization process for OC in  
601 strongly weathered soils of the humid tropics. We suggest that the formation of mineral-organic  
602 associations as part of the aggregate size fractions is the most important process that preserves OC during  
603 land-use change in these soils.

**Gelöscht:** Thus, we have to reject our initial assumption that the mineralogical combination resulting in the largest aggregate stability better preserved OC during conversion of forests into croplands. We suggest that in weathered tropical soils this is largely attributable to the importance of mineral-organic associations, where changes in aggregation do not immediately offset the stabilizing effect of soil minerals.

616 **7. Author contribution**

617 KK, RM, MK, and KHF designed the project. MK, KK, RM, DNK, and KHF collected soil or data to  
618 supported the sampling campaign. MK, KK, RM, and KHF evaluated data and all authors conducted a  
619 thorough critical review of the manuscript. MK, KK, and RM wrote the manuscript with contribution of  
620 all authors.

621

622 **8. Competing interests**

623 The authors declare that they have no conflict of interest.

624

625 **9. Acknowledgements**

626 We are grateful to the officials of Amani Nature Reserve who supported the field campaign in February  
627 2018. Aloyce Mkongewa enthusiastically assisted fieldwork. We are also indebted to Gisela Ciesielski,  
628 Manuela Unger, Mandy Meise, Tobias Krause, Thomas Klinger, Gudrun Nemson-von Koch, and  
629 Christine Krenkewitz for laboratory support and analytical work. Finally, the authors would like to thank  
630 the two reviewers for their comprehensive reading of the manuscript and constructive suggestions for its  
631 improvement. This study was supported by grants of the Deutsche Forschungsgemeinschaft (DFG): FE  
632 504/15-1, KA 1737/16-1, and MI 1377/11-1.

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