



# 1 Geogenic organic carbon in terrestrial sediments and its 2 contribution to total soil carbon

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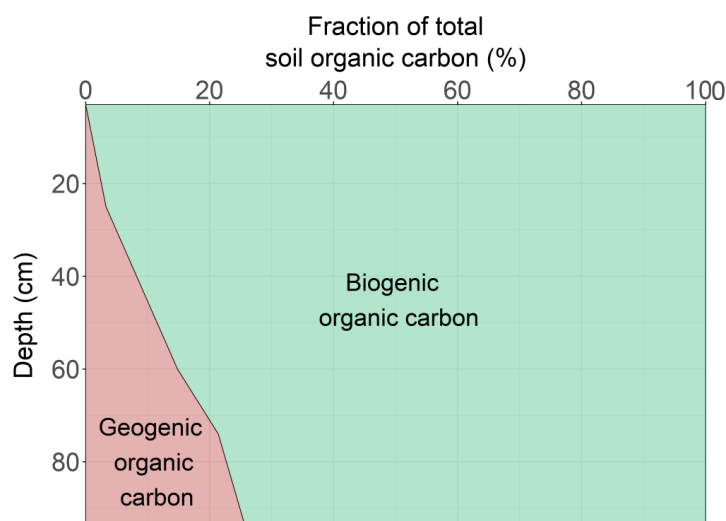
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16 **Graphical abstract:** The median GOC contribution (based on all three sediments) to soil OC with depth based on <sup>14</sup>C  
 17 measurements.



18 **Abstract**

19 Geogenic organic carbon (GOC) from sedimentary rocks is an overlooked fraction in soils that has not  
20 been quantified yet, influencing the composition, age and stability of total organic carbon (OC) in  
21 soils. In this context GOC is referred to as the OC in bedrocks deposited during sedimentation.  
22 However, the contribution of GOC to total soil OC varies with the type of bedrock. So far studies  
23 investigating the contribution of GOC derived from different terrestrial sedimentary rocks to soil OC  
24 contents are missing.

25 In order to fill this gap, we analysed 10 m long sediment cores at three sites recovered from  
26 Pleistocene Loess, Miocene Sand and Triassic Red Sandstone and calculated the amount of GOC  
27 based on  $^{14}\text{C}$  measurements.  $^{14}\text{C}$  ages of bulk sedimentary OC revealed that OC represents a mixture of  
28 biogenic and geogenic components. Biogenic refers to OC that entered the sediments recently from  
29 plant sources. All sediments contain considerable amounts of GOC (median amounts of  $0.10 \text{ g kg}^{-1}$  at  
30 the Miocene Sand,  $0.27 \text{ g kg}^{-1}$  at the Pleistocene Loess and  $0.17$  at Red Sandstone) in comparison to  
31 subsoil OC contents (between  $0.53$ - $15.21 \text{ g kg}^{-1}$ ). Long-term incubation experiments revealed that this  
32 GOC seemed to be comparatively stable against biodegradation. Its possible contribution to subsoil  
33 OC stocks ( $0.3$ - $1.5 \text{ m}$  depth) is  $\sim 2.5 \%$  in soil developed in the Miocene Sand,  $\sim 8 \%$  in the Loess soil  
34 and  $\sim 12 \%$  at the Red Sandstone site. Thus GOC having no detectable  $^{14}\text{C}$  contents influences  $^{14}\text{C}$   
35 ages of subsoil OC and thus may partly explain the strong  $^{14}\text{C}$  ages increase observed in many  
36 subsoils. This is particularly important in soils on terrestrial sediments with comparatively low  
37 amounts of OC, where GOC can considerably contribute to total OC stocks

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48 **Keywords** Geogenic organic carbon; sedimentary organic carbon;  $^{14}\text{C}$ ; terrestrial sediments; incubation  
49 experiment

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## 51 1. Introduction

52 On a global average soils store more than 50 % of OC in the subsoil below 30 cm depth (Batjes,  
 53 2014). This carbon is considered as a highly stabilised carbon pool due to its high apparent  $^{14}\text{C}$  ages  
 54 (Schrumpf et al., 2013, Mathieu et al., 2015). This, however, may also be explained by an contribution  
 55 from geogenic organic carbon (GOC), which is defined here as OC that is deposited during  
 56 sedimentation and rock formation, and may increasingly influence subsoil OC with increasing depth  
 57 (Schrumpf et al., 2013, Graz et al., 2010, Kögel-Knabner et al., 2008, Trumbore, 2009). GOC in most  
 58 cases is devoid of  $^{14}\text{C}$  and thus may lead to an overestimation of ancient OC sources although a  
 59 number of studies showed the importance of root derived, young OC inputs to subsoils (Angst et al.,  
 60 2016, Crow et al., 2009). The contribution of GOC to soils has been investigated in reclaimed mine  
 61 soils where Vindušková et al. (2015) found contributions from GOC between 26 and 99 % to total soil  
 62 OC. Furthermore especially OC rich sediments with contents of 2-7 g kg $^{-1}$  (Hemingway et al., 2018) or  
 63 28-105 g kg $^{-1}$  (Frouz et al., 2011) have been investigated. The impact of GOC on soils derived from  
 64 sediments or sedimentary rocks with lower OC contents, however, has not been investigated so far.  
 65 Considering the fact that approximately 65 % of the continental earth's surface is covered with  
 66 sediments and sedimentary rocks (Amiotte Suchet et al., 2003) a potentially large fraction of soils  
 67 could contain GOC that contributes to soil OC stocks, even though a lot of them might be derived  
 68 from recent sedimentation processes. So far there is not much literature about sediments with only low  
 69 amounts of OC. There are estimations that assume sandstones to be GOC free (van der Voort et al.,  
 70 2018) or, in contrast, a storage model that assumes generally high GOC amounts of 2.4 g kg $^{-1}$  for all  
 71 sandy deposits (Copard et al., 2007). Therefore, more information about the amounts of OC in  
 72 sediments is needed.

73 To estimate the possible contribution of GOC to subsoil OC stocks, it is further necessary to know  
 74 about the amount of OC in sediments that comes from sedimentation (GOC) and to distinguish it from  
 75 OC that is derived from current vegetation (biogenic OC). There are many soil- and substrate-specific  
 76 factors that might influence the OC contribution from current vegetation to sedimentary OC like  
 77 potential rooting depth or hydraulic conductivity. So far no method could be established that would  
 78 allow a direct quantification of GOC in different soils or sediments, apart from promising methods to  
 79 quantify the graphitic part of GOC in soils (Zethof et al., 2019). The only reliable approach to  
 80 distinguish both sources is the use of  $^{14}\text{C}$ . Because deposition of sediments mostly took place > 50,000  
 81 yrs. BP, they do not contain any  $^{14}\text{C}$ , which has a mean half life time of 5,730 yrs. (Libby, 1952). In  
 82 addition,  $\delta^{13}\text{C}$  values of OC in the sediments allow to distinguish carbonaceous from organic sources.  
 83 Thus, using both carbon isotopes can reveal if the OC is a mixture of GOC and OC from the  
 84 vegetation that is younger than 50,000 yrs. A quantification of the geogenic part of OC in the  
 85 sediments is only possible if the average  $^{14}\text{C}$  age of biogenic OC is known or can be estimated.



86 One important question regarding a possible contribution from GOC in soils is, if this GOC will be  
87 mineralised when it becomes part of the soil. Due to the fact that GOC resists degradation since it has  
88 been deposited, it can be assumed that it already exhibits a strong inherent recalcitrance. Nevertheless,  
89 this could also be due to a physical protection that prevented microbial accessibility. However, when it  
90 becomes part of the subsoil during progressing soil development, the infiltration of water, oxygen,  
91 fresh nutrients and microorganisms might cause the degradation of this OC pool. The direct microbial  
92 coal degradation has already been observed via incubation experiments in mine soils (Waschkies and  
93 Huttel, 1999, Rumpel and Kögel-Knabner, 2002) or in shale bedrocks directly exposed to the surface  
94 (Soulet et al., 2017). If GOC is degradable in OC-poor sediments or sedimentary rocks has not been  
95 investigated so far but might be different since the amount of OC can also drive microbial respiration  
96 (Colman and Schimel, 2013).

97 To the best of our knowledge there is only one study by van der Voort et al. (2018) investigating the  
98 amount of GOC in soils and estimating this to make up about 80 % of soil OC in a moraine derived  
99 soil. This reveals that GOC might considerably contribute to soil OC. But beside the study from van  
100 der Voort et al. (2018) on a very specific sediment, further direct calculations of the amount of GOC in  
101 soils are missing.

102 Our aim was to quantify GOC in different terrestrial sediments and a sedimentary rock and investigate  
103 its stability in incubation experiments to make assumptions about its possible contribution to soil OC  
104 stocks in soil profiles at the same site. Our main research questions were i) what is the relation  
105 between sedimentary and subsoil OC contents? ii) is OC in sediments <sup>14</sup>C free and how much is really  
106 geogenic? iii) will sedimentary GOC be degraded? and iv) how much does GOC contribute to soil  
107 OC?

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## 2. Material and methods

### 2.1 Site description

Three sites were selected with different sedimentary bedrocks derived from a single geologic substrate, that can be found close to the surface and that is homogeneous down to 10 m depth. The sites represented one sedimentary rock and two soft sediments. The sedimentary rock was a sandstone (Solling Formation, Triassic) under European beech forest (*Fagus sylvatica*) 11.5 km north-east of Göttingen (51°35.012' N; 10°3.960' O) in the following referred to as “Red Sandstone”. The soil was classified as a Cambisol according to World Reference Base for Soil Resources (WRB, 2006). The sediments were loessic deposits (Weichselian Glacial) under an agricultural field, 30 km north of Göttingen (51°48.101' N; 9°58.002' O) referred as “Loess” and terrestrial sandy deposits from the Miocene (Neogene formerly named Tertiary) in a European beech forest 13 km south-west of Göttingen (51°28.673' N; 9°45.323' O) referred to as “Miocene Sand”. The associated soils are classified as a Luvisol and a Cambisol respectively. Mean annual air temperature and precipitation were 9.2 °C and 647 mm (1981-2010) at the nearby weather station including all three sites.

### 2.2 Sampling and sample preparation

Two 10 m long sediment cores with a diameter of 15 cm were drilled at each site in April 2017. For the soft sediments, drilling was conducted as percussion drilling and for the hard sediments as cable core drilling with water as flushing solution. We subdivided the sampled cores into 1 m increments. The replicates per site were drilled in a distance of approximately 10 m.

One sample from each depth increment (1-2 m, 2-3 m, 3-4 m, 4-5 m, 5-6 m, 6-7 m, 7-8 m, 8-9 m, 9-10 m) for chemical analysis was oven dried at 60°C and sieved to pass 2 mm. The Red Sandstone samples were crushed with a hammer before drying and sieving was conducted. To avoid a possible contamination the outer 5 cm of the drilling cores were removed. Additionally, 1 m deep soil profiles were dug and soil samples were taken from the different classified layers to obtain corresponding soil parameters. Samples were oven dried at 60°C and stored and sieved to pass 2 mm. For the determination of OC and <sup>13</sup>C samples were ground in a planetary ball mill. For <sup>14</sup>C analysis, subsamples were decarbonized with 1M HCl and heating for 1 h at 80 °C followed by 10 h at room temperature.

### 2.3 Chemical analysis and calculations

Three aliquots of each sieved sample was analysed by dry combustion for total C and total N content (TruMac CN LECO, St. Joseph, MI, USA). Samples with a pH value of > 6 were analysed for carbonates after ignition of the sample at 450 °C for 16 h in a muffle kiln. The OC concentration was calculated by subtracting carbonaceous C from total C and expressed as g OC kg<sup>-1</sup> dry matter. Homogenised samples were further analysed for δ<sup>13</sup>C values after removing carbonates in an isotope



ratio mass spectrometer (Delta Plus, Thermo Fisher, Waltham, MA, USA) coupled to an elemental analyser (FLASH EA 1122 NA 1500; Wigan, United Kingdom). Resulting  $\delta^{13}\text{C}$  values (‰) were expressed relative to the international standard of Vienna Pee Dee Belemnite. The bulk densities for the soil samples were obtained with 250 cm<sup>3</sup> sampling rings from each layer of the soil profile. For the sedimentary Loess and Miocene Sand samples, the bulk density of the deepest respective soil sample was used. The bulk densities and the densities without pore space of the intact Red Sandstone cores were determined on four subsamples (from 1.6, 3.6, 7 and 9 m depth) with a Dryflow-pycnometer (GeoPyc 1360) and a gas pycnometer (AccuPyc 1330) respectively. Bulk densities for the missing depth increments were linearly interpolated. For radiocarbon (<sup>14</sup>C) analysis, the sediment samples were first treated with acid to remove inorganic C and were then transferred into pre-combusted quartz ampoules together with copper oxid and silver wool. The ampoules were evacuated, flame sealed, then combusted at 900°C, and the CO<sub>2</sub> evolved was purified on a vacuum ring (Rethemeyer et al., 2019). <sup>14</sup>C contents were measured with the MICADAS accelerator mass spectrometry (AMS) system at the ETH Zürich, Switzerland. If possible one sedimentary sample per depth increment and site and one sample per soil layer was analysed. Due to the very low OC contents in some sediment samples, <sup>14</sup>C contents could only be determined for three samples from the Miocene Sand (from 1.9, 4.9 and 7.9 m depth) and four from the Red Sandstone (1.9, 4.9, 7.9 and 9.9 m depth). For the Loess, <sup>14</sup>C of bulk OC was measured in all depth intervals (1.9-9.9 m).

Total OC stocks (Mg ha<sup>-1</sup>) were calculated according to Eq. 1:

$$OC\ stock = OC \cdot BD \cdot (1 - stone\ content) \cdot depth \cdot 0.1 \quad \text{Eq. 1}$$

where *OC* is the weight based OC content, either in the fine soil <2-mm fraction of the soil profiles, or in the sediments (g kg<sup>-1</sup>), *BD* is the bulk density of the fine soil (g cm<sup>-3</sup>), *stone content* is the volume based proportion of stones (cm<sup>3</sup> cm<sup>-3</sup>) and *depth* is the thickness of the depth increment (cm). Due to the fact that we did not determined the transition from soil to sediment exactly we set it to 1.5 m depth for all sites. This represents a common boarder for the transition from soil to sediment according to Richter and Markewitz (1995). This was done to be able to compare OC stocks and contributions from GOC later on. We further subdivided the sediments into an upper and a lower part at 4 m depth.

In a second step we calculated the amount of GOC and biogenic OC in the sediments considering GOC as one carbon pool free of <sup>14</sup>C. For the sediments we calculated the proportion of biogenic OC (*f<sub>biogenic</sub>*) on the total amount of OC with a two pool model (Eq. 2) used by Cerri et al. (1985):

$$f_{biogenic} (\%) = \frac{F_{biogenic\ OC} - F_{GOC}}{F_{sample} - F_{GOC}} \cdot 100 \quad \text{Eq. 2}$$

where *F* represents the <sup>14</sup>C content in fraction modern carbon (F<sup>14</sup>C) from a source compared to the <sup>14</sup>C content of an oxalic standard (Torn et al., 2009, Stuiver and Polach, 1977). Sources were the GOC



174 fraction ( $F_{GOC}$ ), the sample ( $F_{sample}$ ) and the biogenic OC fraction ( $F_{biogenic\ OC}$ ). Since the  $^{14}C$  content of  
 175 the GOC fraction can be set to zero, this equation can be simplified to:

$$f_{biogenic} (\%) = \frac{F_{biogenic\ OC}}{F_{sample}} \cdot 100 \quad \text{Eq. 3}$$

176 For the biogenic OC in the sediments we assumed an average  $^{14}C$  age ranging from 1,000-4,000 yrs.  
 177 BP. We assumed this range based on published  $^{14}C$  results of dissolved OC reaching greater depths  
 178 (Schiff et al., 1997, Artinger et al., 1996). The  $^{14}C$  contents in the sediment from 2 to 4 m depth of the  
 179 loess led to ages < 3,000 yrs. BP and were therefore even younger than in 74 cm depth (4,413 yrs.  
 180 BP). Thus, they were treated like the soil part for the calculation of a GOC fraction in the following.  
 181 Respective times were converted into  $^{14}C$  contents ( $F_{biogenic\ OC}$ ) according to Torn et al. (2009):

$$F_{biogenic\ OC} = e^{\left(\frac{t}{-8033}\right)} \quad \text{Eq. 4}$$

182 where  $t$  represents the  $^{14}C$  age (1,000 or 4,000 yrs. BP respectively) and 8033 yrs. the mean life of  
 183 radiocarbon. The proportion of GOC in the sediments ( $f_{GOC}$ ) consequently is the remaining portion  
 184 (Eq. 5).

$$f_{GOC} = 100 \% - f_{biogenic} \quad \text{Eq. 5}$$

185 For the depth increments without measured  $^{14}C$  ages (Tab. S1) we linearly interpolated the calculated  
 186 amounts with measured  $^{14}C$  ages from over- and underlying samples. This was done by assuming a  
 187 depth dependent correlation and using the adjacent values. To calculate the amount of GOC in the soil  
 188 profiles we first calculated the weight-based amount of GOC in the sediments by multiplying its  
 189 fraction ( $f_{GOC}$ ) with the respective OC content (in g OC per  $kg^{-1}$  dry mass). We then took the median  
 190 amount of GOC (g GOC per  $kg^{-1}$  dry mass) of these sedimentary values and calculated its proportion  
 191 on the soil OC content (g OC per  $kg^{-1}$  fine soil) in the soil profile. This was done for the proportion of  
 192 GOC in the sediments calculated with a 1,000 and a 4,000 year old biogenic OC fraction ( $F_{biogenic\ OC}$  in  
 193 Eq. 3) to obtain a range of GOC contributions. We assumed that the GOC fraction resisted degradation  
 194 during soil formation. Therefore, this proportion represents the highest possible amount of GOC that  
 195 may contribute to soil OC stocks. Under this assumption we were also able to define the influence of  
 196 GOC in the soil profile on the resulting  $^{14}C$  ages. Since the calculated  $^{14}C$  ages represent a mixture of  
 197 the  $^{14}C$  content from the GOC and the biogenic fraction (Eq. 5), the GOC fraction has the same  
 198 influence on the soil  $^{14}C$  age as on bulk OC (according to Eq. 3). Reducing the age by this fraction  
 199 would therefore represent an “unbiased” age of soil derived OC.

## 200 2.4 Incubation experiment

201 To assess the potential stability of OC in the sediments against microbial decay, two laboratory  
 202 incubation experiments were conducted at 20°C for 50 and 533 days respectively. This was done to



203 reveal the potential degradation of OC from the sediments under optimal conditions. The first, 50 days  
 204 lasting experiment was conducted with intact Red Sandstone core samples, while the second  
 205 experiment was performed after crushing the Red Sandstone to sizes < 2 mm. This was done to  
 206 simulate the process of weathering when the intact sediment or sedimentary rock becomes part of the  
 207 (sub)soil. The Loess and Miocene Sand samples were flushed with fresh air between both incubations.  
 208 For the incubations, four subsamples with 1,340-6,890 g per sample from different depth intervals  
 209 were used. The sample material was stored at room temperature until the incubation experiment  
 210 started. We took four samples from each sediment and from each of four depth ranges for the Miocene  
 211 Sand (1.2-2.8, 3.2-4.8, 6.1-7.8 and 8.2-9.9 m), the Red Sandstone (2-3, 4-5.8, 7-8 and 8.4-9.8 m) and  
 212 the Loess (1.4-2, 2.7-3, 4.8-6 and 9.1-10 m). Additionally four blank samples with no material were  
 213 installed. A water content corresponding to 40 % of the water holding capacity based on the pouring  
 214 density was adjusted. Based on preliminary tests and its calculated bulk density and porosity, the intact  
 215 Red Sandstone samples were kept in a barrel with pure water for 14 hours to reach a water content of  
 216 nearly 40 %. Samples were placed in polycarbonate vessels with a volume of 7069 cm<sup>3</sup> and closed air-  
 217 tight. The lids contained two tube connectors so that the samples could be flushed with ambient air.  
 218 After flushing, samples were set to a starting pressure of about 1,300 mbar and kept closed until the  
 219 end of the incubation. Nine gas samples were taken in evacuated glass vials (20 mL), 0, 3, 7, 13, 21,  
 220 30, 59, 63 and 533 days after the incubation started. Samples were analysed for CO<sub>2</sub> concentrations by  
 221 gas chromatography (Agilent 7890A, GC, Agilent Technologies, Santa Clara, USA) to account for the  
 222 amount of accumulated CO<sub>2</sub>. Three additional gas samples were taken 0, 30 and 63 days after the  
 223 incubation started and analysed with an isotope ratio mass spectrometer (Delta Plus XP, Thermo  
 224 Fisher Scientific, Bremen, Germany) to account for the development of δ<sup>13</sup>C of CO<sub>2</sub> during the  
 225 respiration. Corresponding pressure was measured at each sampling date. When the over-pressure of a  
 226 vessel was lost due to leakages, it was removed from the sampling because a contamination with  
 227 ambient air could not be excluded. This happened for one third of all samples.

228 The amount of respired CO<sub>2</sub>-C (mg CO<sub>2</sub>-C d<sup>-1</sup>) was calculated with Eq 6.

$$CO_2 - C = \frac{0.1 \cdot p \cdot x_i \cdot M \cdot V}{R \cdot T \cdot t} \quad \text{Eq. 6}$$

229 where  $p$  is the pressure (mbar),  $x_i$  is the difference of the CO<sub>2</sub> concentration between the samplings  
 230 (ppm),  $M$  is the molar mass of C (g mol<sup>-1</sup>),  $V$  the air volume of the sample (m<sup>3</sup>),  $R$  is the molar gas  
 231 constant (J kmol<sup>-1</sup> K<sup>-1</sup>),  $T$  is the incubation temperature (K) and  $t$  is the elapsed time (d) between the  
 232 samplings. Based on the δ<sup>13</sup>C values of CO<sub>2</sub> the proportion of CO<sub>2</sub> derived from carbonates was  
 233 subtracted according to Bertrand et al. (2007) if necessary. This respiration rate was related to the OC  
 234 content of the samples (called “OC-normalised respiration”) by dividing it by the total amount of OC  
 235 in g in the sample.





236 Assuming that OC in the sediments represents a mixture of a stable and a labile pool with different  
 237 extents of degradability, we fitted a double exponential model to this OC-normalised respiration with  
 238 different rates according to Qualls and Haines (1992) :

$$\text{mineralised OC (\%)} = (100 - a) \cdot (1 - e^{-k_1 \cdot t}) + a \cdot (1 - e^{-k_2 \cdot t}) \quad \text{Eq. 7}$$

239 where  $a$  represents the proportion of the stable OC pool (%),  $k_1$  and  $k_2$  the associated mineralisation  
 240 rate constants of the labile and stable OC ( $\text{year}^{-1}$ ) and  $t$  the elapsed time (years). We also calculated the  
 241 mean residence time (years) of the labile and the stable OC pool ( $1/k_1$  and  $1/k_2$ ). If it was not possible  
 242 to fit a double exponential model we used a simple linear regression ( $\text{mineralised OC (\%)} = k \cdot t$ ) to  
 243 obtain a mineralisation rate. This was the case for the Red Sandstone samples, showing a linear  
 244 mineralisation rate during the incubation experiment.

## 245 2.5 Statistics

246 Statistical analyses were conducted using the statistical environment R (R Core Team (2018) including  
 247 the base function “*nls*” (nonlinear least-squares estimates) to fit non-linear models, “*lm*” to fit linear  
 248 models and the package *ggplot2* (Wickham, 2016) for graphical presentation. The non-linear models  
 249 were used to model the amount of mineralised OC during the incubation experiment using Eq. 7.  
 250 Models were tested for deviations from homoscedasticity, normality of residuals and absence of  
 251 collinearity. The tests revealed heteroscedasticity of the residuals. This can be explained by an  
 252 increasing standard deviation with time. We are further aware of not having normally distributed  
 253 residuals since the dependent variable, representing the proportion of OC being mineralised, only  
 254 allows values between 0 and 1. Keeping this in mind the results from the double exponential model  
 255 have to be treated as an indicator for the differences between the samples and a scale for the  
 256 mineralisation of a stable and a labile OC pool. We therefore also omitted calculations of standard  
 257 errors and significance of the parameters since this would not lead to reasonable results with the used  
 258 models.

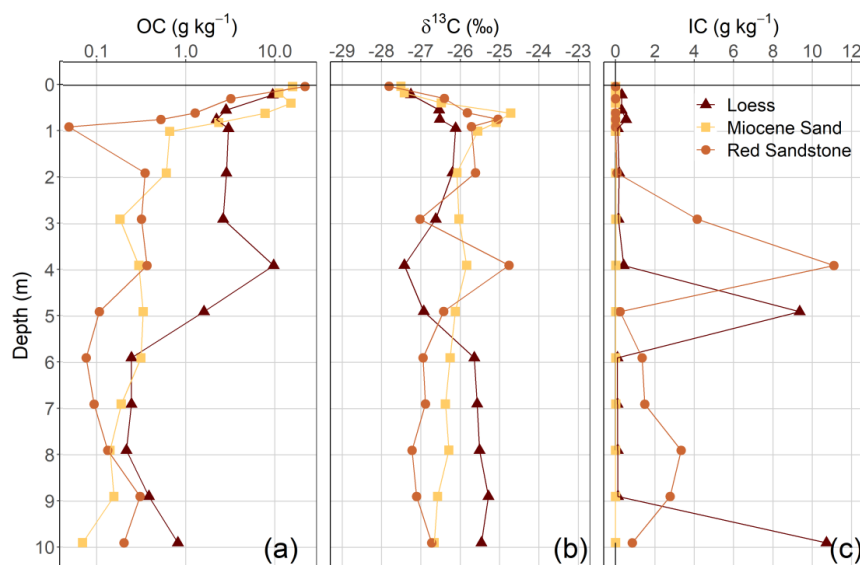


### 259 3. Results

#### 260 3.1 Relation between sedimentary and subsoil organic carbon

261 In all analysed sediments OC contents were measured above the detection limit. The amount of OC in  
 262 the sediments from 1 to 10 m depth was comparatively low in the Miocene Sand and Red Sandstone  
 263 (0.04-0.71 g kg<sup>-1</sup> and 0.01-0.53 g kg<sup>-1</sup>, respectively). Considerably higher OC contents of 0.21-9.71 g  
 264 C kg<sup>-1</sup> were found in the Loess (Fig. 1 a). The median OC content of the sediments was in a  
 265 comparable range like those in the respective deepest subsoil horizon. This deepest horizon was a Cv  
 266 horizon in 94 cm depth for the Loess, 100 cm depth for the Miocene Sand and 74 cm depth for the Red  
 267 Sandstone. In detail, the median OC content in the sediments, compared to the respective Cv horizons,  
 268 corresponded to 27 % for the Loess, 29 % for the Miocene Sand and 39 % for the Red Sandstone. The  
 269 Loess OC contents were highly variable, highlighting the changing sedimentary conditions during the  
 270 past glacial and interglacial periods (Jordan and Schwartau, 1993). In 4-5 m depth, OC contents of the  
 271 Loess were even higher (9.7 g kg<sup>-1</sup>) than in the subsoil (3 g kg<sup>-1</sup>). In the Miocene Sand and the Red  
 272 Sandstone no clear depth gradient of OC was found in 2-10 m depth (Fig. 1 a). Even though OC  
 273 content in the sediments are low, OC stocks can be considerable large. A comparison of OC stocks in  
 274 topsoils (0-0.3 m), subsoils (0.3-1.5 m) and the sediments down to 10 m depth revealed quite high OC  
 275 stocks in the sediments. For the Loess, OC in the sediment contributed up to 71 % of the total OC  
 276 amount while it was 51 % for the Red Sandstone and 21 % for the Miocene Sand (Table 1).

277 The distribution of the  $\delta^{13}\text{C}$  values of OC in the soil and sediment profiles showed an increase of  $\delta^{13}\text{C}$   
 278 with increasing depth in the soil down to 1 m depth (Fig. 1 b). Contrastingly, the  $\delta^{13}\text{C}$  values of OC in  
 279 the sediments showed no clear trend with increasing depth but they all were in the range of C<sub>3</sub> plant  
 280 material. A value above -25 ‰ for the Red Sandstone in 4 m depth can be explained by corresponding  
 281 high values of inorganic carbon in this depth. It can be assumed that decarbonisation of this sample  
 282 was not completely successful. Unexpectedly high amounts of inorganic carbon were found in parts of  
 283 the Red Sandstone, indicating the presence of calcareous deposits in this terrestrial material (Fig. 1 c).  
 284 The Loess also includes some distinct calcareous layers in 5 and 10 m depth, while there was no  
 285 inorganic carbon present in the Miocene Sand.



**Fig. 1:** Depth distribution of different bulk properties of the soil profiles and the deep drilling cores. Presented parameters include the log scale organic carbon (OC) content (a),  $\delta^{13}\text{C}$  values of the organic carbon (b) and the inorganic carbon (IC) content (c) related to the amount of fine soil or dry mass respectively.

**Table 1:** OC stocks and proportions for the three sites down to 10 m depth. Proportions of biogenic and geogenic OC were calculated based on  $^{14}\text{C}$  results and assumptions described in the material and methods section. Represented ranges are calculated based on the assumption of a 1,000 or 4,000 year old biogenic OC fraction reaching the sediments.

Substrate	Layer	Depth (m)	TOC		OC stocks ( $\text{Mg ha}^{-1}$ )				Proportion of OC (%) <sup>b</sup>			
			( $\text{Mg} \cdot \text{ha}^{-1}$ )	(% ) <sup>a</sup>	geogenic		biogenic		geogenic		biogenic	
					4,000 yrs.	1,000 yrs.	1,000 yrs.	4,000 yrs.	4,000 yrs.	1,000 yrs.	1,000 yrs.	4,000 yrs.
Loess	Topsoil	0.0 - 0.3	40.6	11	1.1 - 1.2		40 - 39		3 - 3		97 - 97	
	Subsoil	0.3 - 1.5	66.0	17	5.1 - 5.3		61 - 61		8 - 8		92 - 92	
	Upper Sediment	1.5 - 4.0	218.6	57	10.6 - 11.0		208 - 208		5 - 5		95 - 95	
	Lower Sediment	4.0 - 10.0	55.1	15	42.9 - 46.7		8 - 12		78 - 85		15 - 22	
Red Sandstone	Topsoil	0.0 - 0.3	22.2	30	0.6 - 0.7		22 - 22		3 - 3		97 - 97	
	Subsoil	0.3 - 1.5	13.7	19	1.6 - 1.7		12 - 12		12 - 12		88 - 88	
	Upper Sediment	1.5 - 4.0	18.7	25	13.1 - 14.9		4 - 6		70 - 80		20 - 30	
	Lower Sediment	4.0 - 10.0	19.3	26	14.1 - 15.7		4 - 5		73 - 82		18 - 27	
Miocene Sand	Topsoil	0.0 - 0.3	39.1	31	0.3 - 0.3		39 - 39		1 - 1		99 - 99	
	Subsoil	0.3 - 1.5	60.8	48	1.3 - 1.6		59 - 60		2 - 3		97 - 98	
	Upper Sediment	1.5 - 4.0	10.6	8	3.7 - 5.8		5 - 7		34 - 55		45 - 66	
	Lower Sediment	4.0 - 10.0	16.3	13	8.3 - 10.8		6 - 8		51 - 66		34 - 49	

<sup>a</sup> % of total OC stock from 0-10 m

<sup>b</sup> % of OC stock in the respective depth increment

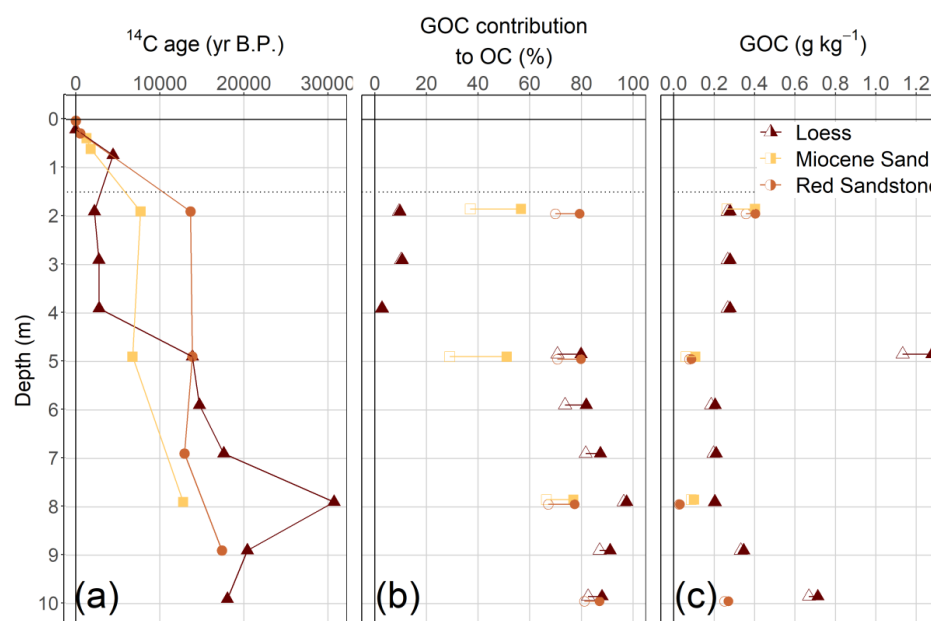


### 297 3.2 Ages of organic carbon in soils and sediments and contributions from geogenic organic carbon

298 The ages of OC in the Loess soil profiles revealed a modern carbon like signature (0 yrs. BP) in 0.3 m  
 299 depth with a sharp increase up to  $4,413 \pm 51$  yrs. BP in 0.7 m depth (Fig. 2 a). For the Red Sandstone  
 300 soil profile there was only an increase in the ages from a modern like signature in 0.04 m to  $532 \pm 41$   
 301 yrs. BP in 0.3 m depth. The Miocene Sand soil profile in 0.4 m depth showed an increase from  $1,277 \pm$   
 302  $41$  to  $1,771 \pm 44$  yrs. BP in 0.6 m depth. Thus, OC of the subsoil (around 0.6 m depth) in the Loess  
 303 was more than twice as old as in the Miocene Sand. Contrastingly, the Loess has a modern like  
 304 signature in 0.3 m while the soil developed in Red Sandstone showed an average age of  $532 \pm 41$  yrs.  
 305 BP in 0.3 m depth. This could be due to the observed plough layer at the Loess site mixing up the  
 306 upper 30 cm with a predominantly modern  $^{14}\text{C}$  signature.

307 The ages of OC in the sediments ranged from 2,200-30,730 yrs. BP, with respective mean ages of  
 308  $9,077 \pm 3,234$  yrs. BP for the Miocene Sand,  $13,674 \pm 9,632$  yrs. BP for the Loess, and  $14,463 \pm 1,992$   
 309 yrs. BP for the Red Sandstone. For all sediments, 11 out of 16 samples contained a  $^{14}\text{C}$  content that led  
 310 to an apparent  $^{14}\text{C}$  age older than the soil age of 11,600 yrs. BP, assuming that soil development  
 311 started after the latest glacial period at this time (Litt et al., 2007). Therefore sediments contain a  
 312 mixture of geogenic ( $^{14}\text{C}$  free) and biogenic (with  $^{14}\text{C}$ ) OC. Despite being the youngest sediment, the  
 313 Loess partly revealed the highest apparent  $^{14}\text{C}$  ages with up to  $30,730 \pm 631$  yrs. BP (Fig. 2 a). The  
 314 ages of OC in the sediment of the Red Sandstone and the Miocene Sand ranged from  $12,940 \pm 132$  to  
 315  $17,390 \pm 206$  yrs. BP and from  $6,750 \pm 86$  to  $12,770 \pm 151$  yrs. BP, respectively revealing no depth  
 316 trend with higher ages in the deeper sediment. The calculated GOC fraction in the sediments was  
 317 highest for the Red Sandstone, ranging from 67 to 87 % with a mean of 77 % (Fig. 2 b). For the three  
 318 samples of the Miocene Sand the GOC fraction ranges from 29 to 77 % with a mean of 53 %. The  
 319 Loess showed a sharp increase at a depth of five metres where GOC contribution went up to 71 to 98  
 320 % while it was only 5 to 19 % in 2 to 4 m depth.

321 The calculated weight based content of GOC in the sediment revealed a comparatively uniform  
 322 distribution for all sediments with depth, except the extremely high contents of the Loess in 5 m depth  
 323 (Fig. 2 c). The investigated sediment depths revealed a quite narrow range of GOC contents with  $0.10$   
 324  $\pm 0.03 \text{ g kg}^{-1}$  for the Miocene Sand,  $0.17 \pm 0.12 \text{ g kg}^{-1}$  for the Red Sandstone and  $0.27 \pm 0.08 \text{ g kg}^{-1}$  for  
 325 the Loess.



**Fig. 2:** Depth distribution of apparent  $^{14}\text{C}$  ages (in yrs. BP) (a), GOC contribution to OC contents in the sediments (b) and resulting weight based amounts of GOC (c) in the sediments. The range in the contribution from GOC is due to the assumption of an average biogenic OC age of 4,000 (empty shapes) and 1,000 years (filled shapes).

### 3.3 Biodegradability of sedimentary derived organic carbon

The incubation experiment revealed a potential but low biodegradability of OC for all samples, but without a clear depth gradient from 1 to 10 m (Fig. 3). For all substrates, the highest mineralisation for a stable OC pool occurred for the shallowest and the deepest depth increment (Table 2). While the Red Sandstone showed very low mineralisation when the samples were incubated as intact cores ( $0.03\text{--}0.9 \text{ mg CO}_2\text{-C g}^{-1} \text{ OC y}^{-1}$ ), the mineralisation rate constants were approximately four times higher when the samples were crushed ( $0.7\text{--}2.2 \text{ mg CO}_2\text{-C g}^{-1} \text{ OC y}^{-1}$ ) (Tab. S1). Comparing the two incubation experiments, the Miocene Sand showed an even higher increase in mineralisation rate constants (from  $1.2\text{--}12$  to  $9.8\text{--}17.6 \text{ mg CO}_2\text{-C g}^{-1} \text{ OC y}^{-1}$ ) while the Loess revealed reduced mineralisation rate constants for three out of four samples (from  $1\text{--}34.2$  to  $3.5\text{--}11 \text{ mg CO}_2\text{-C g}^{-1} \text{ OC y}^{-1}$ ). After 533 days, at the end of the second incubation experiment, the amount of mineralised OC decreased in the order Miocene Sand ( $0.79\text{--}1.52 \%$ ) > Loess ( $0.42\text{--}0.85 \%$ ) > Red Sandstone ( $0.17\text{--}0.42 \%$ , Table 2). In contrast, the fitted models for the second incubation experiment with crushed Red Sandstone samples revealed that mineralisation rate constants were by far highest for the Red Sandstone. They were up to 100 times higher than respective mineralisation rate constants for the stable pool of the Loess and Miocene Sand. For the Loess and the Miocene Sand, mineralisation was to a large extent affected by the high mineralisation rates at the start of the incubation which is shown by the high mineralisation



347 rates of the labile OC fraction (Fig. 3, Table 2). Nevertheless, the two-pool model revealed that this  
 348 labile OC fraction ( $100-a$  in Eq. 5) only represents an extremely low portion of mineralised OC ( $< 1$   
 349 %). This resulted in calculated mean residence times for the stable OC pool of the Miocene Sand and  
 350 the Loess of  $> 10,000$  yrs. compared to residence times  $< 1,000$  yrs. for the Red Sandstone.

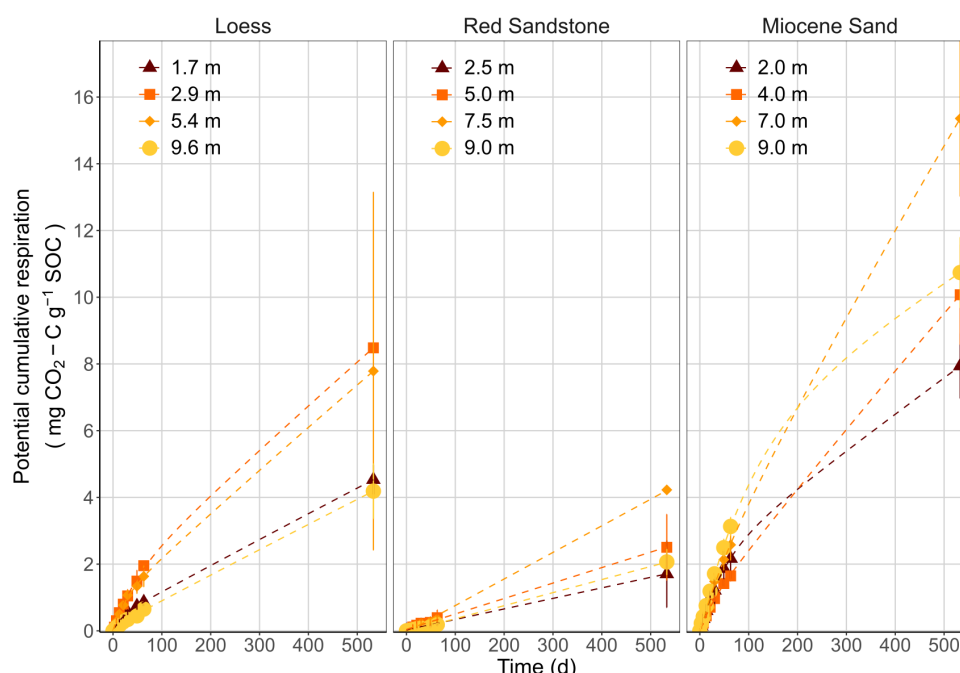
351

352 **Table 2:** Results from the incubation experiment: amount of mineralised OC within 533 days and the proportion,  
 353 mineralisation rate constant and mean residence time of the fast ( $k1$ ) and slow ( $k2$ ) degrading OC pool. Results  
 354 for the Red Sandstone are derived from the incubation experiment with crushed samples.

Substrate	Depth (m)	Mineralised OC	Labile OC	Stable OC	Labile OC	Stable OC	Labile OC	Stable OC
		(% of total OC)			Mineralisation rate ( $\text{mg CO}_2\text{-C g}^{-1}\text{ OC y}^{-1}$ ) <sup>a</sup>		Mean residence time (years) <sup>b</sup>	
Miocene Sand	2	0.79	0.20	99.80	74	0.04	13	24,672
	4	1.01	0.06	99.94	127	0.07	8	15,381
	7	1.52	0.11	99.89	77	0.10	13	10,250
	9	1.06	0.54	99.46	35	0.04	29	27,742
Loess	1.7	0.45	0.04	99.96	200	0.03	5	35,148
	2.9	0.85	0.14	99.86	81	0.05	12	20,379
	5.4	0.78	0.09	99.91	146	0.05	7	21,038
	9.6	0.42	0.01	99.99	211	0.03	5	35,950
Red Sandstone	2.5	0.17	-		1.15		867	
	5	0.25	-		1.69		590	
	7.5	0.42	-		2.88		348	
	9	0.21	-		1.43		701	

355 <sup>a</sup> Derived from the mineralisation rate constant  $k1$  and  $k2$

356 <sup>b</sup> Assuming a constant mineralisation with the rate constants  $k1$  and  $k2$



**Fig. 3:** Potential degradability of sedimentary OC from three sites. Results represent cumulative respiration from a 533 days incubation experiment at 20°C with respective standard deviations (n = 4). Dashed lines represent a fitted model to the respiration data. For the Loess and the Miocene Sand a non-linear regression model was fitted while for the Red Sandstone a linear model was used.

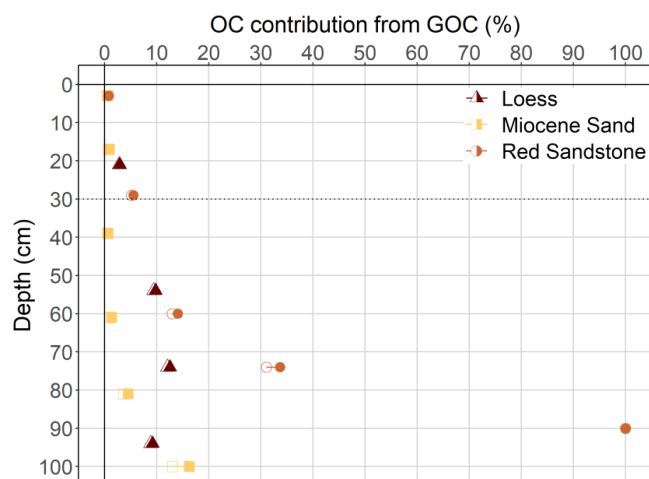
### 3.4 Possible contribution from geogenic organic carbon to soil organic carbon

When the bedrock is weathering it becomes part of the soil and also GOC becomes soil OC. The potential amount of GOC in the subsoil was dependent on the sedimentary bedrock (Fig. 4). In the subsoil of the Miocene Sand it added up to 2-3 %, at the Loess site it was 8 % and at the Red Sandstone site it was 12 % (Table 1). For the defined topsoils (0-30 cm depth), contributions of GOC to soil OC were smaller with 0.7-0.9 % for the Miocene Sand, 2.8-2.9 % for the Loess and 2.8-3.0 % for the Red Sandstone.

The presence of <sup>14</sup>C-free GOC to soil OC reduces the mean bulk soil OC <sup>14</sup>C ages depending on its proportion on soil OC content. Topsoils that developed in the Loess and the Red Sandstone had a modern <sup>14</sup>C content of 1.029 and 1.035 F<sup>14</sup>C similar to the atmospheric <sup>14</sup>C content in 1950. Because of the large proportion of biogenic OC, an influence of a geogenic fraction is not detectable in the topsoil of these sites. No <sup>14</sup>C data are available for the topsoil of the Miocene Sand. For all subsoils, influence of GOC on bulk soil OC <sup>14</sup>C contents depends strongly on the depth and the corresponding OC contents. For the Loess the possible influence on <sup>14</sup>C ages in the subsoils would be quite high, with



an average of 10 % reduction in mean apparent  $^{14}\text{C}$  ages in the subsoil. It would therefore reduce the measured age from 4,413 yrs. BP in 74 cm depth by 532-555 yrs. BP. Geogenic OC potentially reduces the mean apparent radiocarbon age of 1,277 yrs. BP in 0.39 m depth in the Miocene Sand by about 7-9 yrs and the radiocarbon age of 1,771 yrs. BP in 0.61 m depth by 20-24 yrs.. This reduction is below the respective standard deviations of the measurement. Nevertheless, in 1 m depth a given possible proportion of 13.1-16.3 % would reduce an interpolated  $^{14}\text{C}$  age of 3,053 yrs. BP by 399-497 yrs. For the Red Sandstone the influence of GOC on  $^{14}\text{C}$  ages would be highest in the subsoil. In 74 cm depth it would influence an age of 1,453 yrs. BP by 451-490 yrs. BP. Due to the low amounts of soil OC in 90 cm depth at the Red Sandstone site, the weight based median amount of GOC in the sediments is even four times higher than the biogenic amount of soil OC.



**Fig. 4:** Highest possible contribution from GOC to OC in relation to the bulk OC content. Here the median GOC contents of the sediments were taken into account for the respective horizontal weight based OC contents. The range in the contribution from GOC is due to the assumption of an average biogenic OC age of 4,000 (empty shapes) and 1,000 years (filled shapes). The dotted line represents the defined boarder between top- and subsoils.





## 393 4. Discussion

### 394 4.1 Geogenic organic carbon in the sediments

395 Regarding our calculated contribution from GOC to OC in the sediments, the assumed range of  
 396 biogenic  $^{14}\text{C}$  ages from 1,000-4,000 yrs. BP is within the typical range for ages of dissolved OC  
 397 leaching from soils (Artinger et al., 1996, Jia et al., 2019). Despite this, the range from 1,000-4,000  
 398 yrs. BP did not greatly influence the range of calculated sedimentary contribution from GOC,  
 399 especially for the Loess (~ 6 % difference) and the Red Sandstone (~ 9 % difference). The calculated  
 400 GOC contribution for the Miocene Sand was comparatively low (29-77 %) compared to the  
 401 contribution for the Red Sandstone and the Loess. This could be due to deep biogenic carbon inputs  
 402 e.g. as roots and root exudates from the trees (Angst et al., 2016, John et al., 2016, Tückmantel et al.,  
 403 2017, Kirfel et al., 2017), since the loosely bedded Miocene Sand allows for a deep infiltration  
 404 compared to the Red Sandstone site with its shallow bedrock as a root restricting layer (Schneider and  
 405 Don, 2019). Compared to the Loess site, the texture driven lower water conductivity (Saxton and  
 406 Rawls, 2006) could be responsible for lower contributions from biogenic OC. In the Loess, the low  $^{14}\text{C}$   
 407 ages in 2-4 m depth (2,200-2,770 yrs. BP) in contrast to the high  $^{14}\text{C}$  age of 4,413 yrs. BP in 0.74 m  
 408 depth were surprising. This might indicate past anthropogenic activities or erosion driven material  
 409 movement that might have mixed up the upper part of the profile. Nevertheless, at depths below 4 m  
 410 the high  $^{14}\text{C}$  ages of OC in the sediments indicated a large proportion of GOC. This could be due to  
 411 different sedimentation periods, soil forming and also soil burial processes (Chaopricha and Marin-  
 412 Spiotta, 2014) that took place during the Pleistocene. These processes can lead to the presence of  
 413 buried layers in the Loess with varying amounts of rather recalcitrant OC as it was shown by Marin-  
 414 Spiotta et al. (2014) and is indicated by the high variance in OC amounts within the Loess sediment.

415 In summary, the contribution from GOC to sedimentary OC was substrate dependent. A loosely  
 416 bedded sediment like the Miocene Sand with extremely low concentrations of OC is more prone for  
 417 infiltration of biogenic OC. This resulted in contributions of biogenic OC to the sediments of about 50  
 418 %. In contrast, the Loess site with comparatively low infiltration rates or the Red Sandstone site with  
 419 reduced possibilities for deep rooting seemed to contain relative constant contributions from GOC of  
 420 around 80 %.

421 We found that the GOC contribution within the sediments did not increase with increasing soil depth.  
 422 This is in contrast to the results of Frouz et al. (2011) showing that different sediment types from a  
 423 Miocene clay sediment had higher weight based carbon contents in 150 m compared to 30 m depth.  
 424 However, a comparison with the present study is rather difficult since Frouz et al. (2011) did not  
 425 distinguish between the geogenic and the biogenic OC fraction and OC contents were by far higher  
 426 (28-112 g  $\text{kg}^{-1}$  compared to 0.008-10 g  $\text{kg}^{-1}$  in our results). But this also underlines the importance of



different sedimentation processes for the amount and depth distribution of OC in sediments and sedimentary rocks.

A comparison of the weight based amounts of GOC in terrestrial sediments is difficult, since most studies in that field rarely determine the amounts of OC in terrestrial sediments, or pre-assume that e.g. sandy sediments contain no considerable amounts of OC (Artinger et al., 1996). Quite high amounts of OC were found in the skeleton part of different soils on sandstones with 0.61-1.97 t ha<sup>-1</sup> Corti et al. (2002). Nevertheless, they mentioned the possible high influence of organic substances from the soil solution without quantifying it and they did not directly investigated OC in the sediments. Additionally Copard et al. (2007) assumed an OC amount of 2.4 g kg<sup>-1</sup> from an unknown source for all sandy sediments in a global storage modelling approach for the first metre of the sediments. This number is between 14-24 times higher than the median GOC amount in the Red Sandstone (0.2 g kg<sup>-1</sup>) and the Miocene Sand (0.1 g kg<sup>-1</sup>). More data are thus needed to constrain the OC pool in sedimentary bedrock since it will also influence the soil carbon pool in our study.

Loess deposits are comparably well investigated as archive for paleoenvironmental conditions (Wang et al., 1996, Head et al., 1989, Hatté et al., 1998, Murton et al., 2015). The median amount of 0.27 g kg<sup>-1</sup> from our study is low compared with studies from Hatté et al. (1998), Wang et al. (1996) and Strauss et al. (2012). Hatté et al. (1998) investigated 20 m depth loess deposits in the Rhine valley and found OC contents between 1.0-8.6 g kg<sup>-1</sup>, Wang et al. (1996) investigated 12 m depth loess deposits in China and found OC contents of 31.2 ± 30.5 g kg<sup>-1</sup> and Strauss et al. (2012) found OC contents of 15 ± 14 g kg<sup>-1</sup> in Yedoma loess deposits in Siberia. This shows that the deposits from our investigated site stored comparatively low GOC contents, although most of the studies mentioned above did not distinguish between a biogenic and a geogenic OC pool. Nevertheless, also in our study Loess is, compared to other sediments, a sedimentary bedrock with a high OC content. This was in line with highest OC contents in subsoils at the Loess site and may indicate the importance and contribution of bedrock OC to subsoil OC.

#### 4.2 Is sedimentary derived organic carbon biodegradable?

The incubation experiment revealed a mineralisation of OC within the sediments with values between 0.02-0.3 % of total OC being mineralised after 63 days, and values between 0.2-1.5 % after 533 days. A direct mineralisation of OC from sediments is in accordance with several studies investigating the direct mineralisation from outcrops (Copard et al., 2007, Soulet et al., 2017, Horan et al., 2017, Petsch et al., 2000). The difference to our study is, that they observed this mineralisation when the sediments were directly exposed to the surface or/and part of a very fast eroding area. Thus GOC from the sediments already is in touch with the atmosphere and inputs of the recent vegetation. However, Frouz et al. (2011) conducted an incubation experiment with sedimentary samples from OC rich Miocene clay sediments. They found quite high respiration rate constants with values between 3.5-12.3 mg



CO<sub>2</sub>-C g<sup>-1</sup> OC y<sup>-1</sup>. They attributed this to the prevailing presence of aliphatic compounds in their samples being decomposed. Also Kieft and Rosacker (1991) found high respiration rates of sedimentary samples with values between 0.9-9.5 mg CO<sub>2</sub>-C g<sup>-1</sup> OC y<sup>-1</sup> which they primarily attributed to the physiological status of the soil microbial community expressed as adenylate energy charge. Since both studies were conducted on marine sediments with comparatively high OC concentrations of 0.52-11 g kg<sup>-1</sup>, together with potentially different microbial community compositions, this might also be a factor that drives the higher respiration (Colman and Schimel, 2013) compared to our samples. Despite this Kieft and Rosacker (1991) did not distinguish between a fast and a slow degrading OC pool. Compared to subsoil incubation experiments, the mineralisation found in our incubation experiment was also quite low. For example, in subsoil incubation experiments at 20°C, Wordell-Dietrich et al. (2017) found that between 0.5-0.95 % of OC are mineralised after incubation for 63 days, Wang et al. (2013) reported values between 0.5-1.5 % of after 28 days and Soucemarianadin et al. (2018) reported values between 1-1.25 % after 70 days. The difference is that our incubation experiment lasted at least 5 times longer and the highest respiration took place at the beginning. Comparing the modelled mineralisation rate constants for the stable OC pool from the two pool model (0.028-0.049 mg CO<sub>2</sub>-C g<sup>-1</sup> OC y<sup>-1</sup> for the Loess and 0.036-0.097 mg CO<sub>2</sub>-C g<sup>-1</sup> OC y<sup>-1</sup> for the Miocene Sand) shows that they are at least around 10 times lower compared to typical soil OC mineralisation rate constants for a slow degrading OC pool with values between 0.4-11.7 mg CO<sub>2</sub>-C g<sup>-1</sup> OC y<sup>-1</sup> (Saidy et al., 2012, Rasmussen et al., 2006, Santos et al., 2012). Also the mineralisation rate constants for the Red Sandstone (1.2-2.9 mg CO<sub>2</sub>-C g<sup>-1</sup> OC y<sup>-1</sup>) are within the lower range of these rates. Thereby the comparatively high mineralisation rates for the Red Sandstone samples might only be due to the duration of the incubation. Due to the mineralisation with time it seems not reasonable to assume a higher mineralisation rate compared to the Miocene Sand and the Loess. Furthermore, the low mineralisation rate of the Red Sandstone during the incubation as intact cores (Fig. S1) promotes the stability of GOC when it is part of the sediments. This might be due to the low accessibility of OC in the sediments for microorganisms and the low the availability of water due to a preferential flow through the sandstone (Swanson et al., 2006). Altogether the low mineralisation rates of the OC in the sediments might be due to the lack of fresh substrates and/or microorganisms that could enhance the degradation of OC (Fontaine et al., 2007). Nevertheless, a mineralisation of OC could be observed only by adding water, which indicates the presence of an active microbial community in the sediments (Joergensen and Wichern, 2018, Magnabosco et al., 2018, Bomberg et al., 2017). Beside the observed mineralisation, the extremely low calculated amounts of a labile pool (<0.22 % of total OC), according to the fitted two pool model for the Loess and the Miocene Sand, clearly differentiates from the biogenic OC pool (> 10 % of total OC) based on <sup>14</sup>C measurements. Thus, there is no correspondence between a fast mineralisable OC pool and the biogenic OC pool. This may be the result of a biogenic OC pool in the sediments being mineralised and containing a readily bioavailable and degradable as



well as a more stabilised fraction. If the total bedrock OC would show mineralisation rates as observed in our study, sediments still would become OC free within  $\leq 36,000$  years.

Taken together the long-term incubation cannot answer the question if GOC will be mineralised when it becomes part of the (sub-)soil. The high residence times of the stable OC pool, however, indicate that it might be relatively resistant against degradation when it becomes part of the subsoil. This is in accordance to an indirect approach to determine the mineralisation of sedimentary OC when it becomes part of the subsoil from Graz et al. (2010). They stated that 30 % of GOC resisted degradation when it becomes part of the soil due to the results from a quantitative palynofacies analysis of bedrock and soil samples. On the other hand Hemingway et al. (2018) found that sedimentary OC directly exposed to the surface in a rapidly eroding tropical mountain area exhibits a considerable mineralisation down to 1 m below the surface. Based on  $^{14}\text{C}$  measurements they found out that on average  $67 \pm 11$  % of the OC fraction in the sediments could be lost during soil formation but did not distinguished between a biogenic and a geogenic OC fraction. This indicates that a microbial mineralisation of bedrock OC takes place but may be partly restricted to biogenic OC.

Regarding the depth distribution of GOC in the sediments, the amount of GOC (in  $\text{g kg}^{-1}$ ) does not increase with depth but shows clear differences. On the one hand, this represents the sedimentation history with different initial amounts of OC and degradation during sedimentation. This is particularly evident by the high amounts of GOC in 5 m depth of the Loess. Meanwhile, contents of GOC especially in the Red Sandstone and Miocene Sand are in the same range over the whole depth. This might indicate that GOC is not degraded within the sediments. If there would be a degradation of GOC within the sediments one would expect a decreasing amount with decreasing depth due to the input of water, microorganisms and fresh nutrients from above. Furthermore, there is a relatively constant contribution from biogenic OC within the sediments. This means that, if biogenic OC enters the sediments, together with possibly degrading microorganisms, this biogenic OC might also be preferably mineralised. This gives a hint that GOC is not degradable when it becomes part of the (sub-)soil, since especially for the Loess and the Miocene Sand, the conditions within the sediments in  $> 1.5$  m depth do not differ so much from the conditions (e.g. oxygen and water content) within the subsoil.

#### 4.3 How much GOC contributes to soil organic carbon?

The contribution of GOC to soil OC stocks in our study is driven by the amount of OC in the soil and the amount of GOC in the respective sediment. Our results revealed that despite differences between sediments, GOC content varied in a quite narrow range between 0.1 and  $0.3 \text{ g kg}^{-1}$ . The contribution of GOC to topsoil OC was negligible. Highest possible contributions of GOC to total subsoil OC were found for the Red Sandstone ( $\sim 30$  %) and lowest for the Miocene Sand (0.6 %). This was due to the range of OC contents in the subsoils ( $0.53 \text{ g kg}^{-1}$ - $15.21 \text{ g kg}^{-1}$ ). When soil OC contents were low, the possible contributions from GOC were high and vice versa (Fig. S2). For our investigated soils OC



533 contents of  $3 \text{ g kg}^{-1}$  soil allowed for possible GOC contributions between 5-10 %. For OC contents  
 534 around  $1 \text{ g kg}^{-1}$  soil a GOC contribution between 10-20 % seems to be possible. Thereby higher  
 535 contributions came from GOC rich sediments like the Loess and lower contributions from sandy  
 536 sediments. In comparison, van der Voort et al. (2018) estimated the contribution from GOC of a soil  
 537 derived from glacial deposits (flysch) between 80-100 cm depth to be around 40 %. For a soil  
 538 developed from a poorly consolidated sedimentary rock (calcareous and shaly moraine) they  
 539 calculated the contribution from GOC to range from 20 % in 145 cm depth to 80 % in 310 cm depth.  
 540 There has further been an attempt to fractionate subsoils to extract the most stable OC that may be  
 541 derived from GOC. Paul et al. (2001) found that 30 % of subsoil OC was non-hydrolysable. The  
 542 investigated a soil developed on loess over till with this non-hydrolysable fraction showing a  $^{14}\text{C}$  age  
 543 of 13,000 yrs. BP. They also concluded that this high age can partly be explained by a GOC fraction.  
 544 These results indicate that especially deposits from the past glacial periods like flysch or till have a  
 545 much higher potential for OC contributions from GOC possibly due to their higher amounts of GOC in  
 546 the sediments. Since we only investigated terrestrial sediments it has to be taken into account that also  
 547 marine sediments or shales contain much higher amounts of OC up to  $250 \text{ g kg}^{-1}$  (Hemingway et al.,  
 548 2018, Petsch et al., 2000). Their amount of GOC and possible contribution to subsoil OC stocks might  
 549 therefore be much higher.

550 Nevertheless,  $^{14}\text{C}$  ages of OC in the subsoil can also be high in soils derived from igneous parent  
 551 materials without GOC (Rumpel et al., 2002). Furthermore, on a global scale  $^{14}\text{C}$  ages of soil OC are  
 552 primarily driven by climatic conditions, clay content and age, since soil development started (Mathieu  
 553 et al., 2015). But for terrestrial sediments with comparatively low amounts of GOC that started their  
 554 soil development after the latest glacial period, we could obtain a scale for possible contributions when  
 555 the amount of OC is known. Thus, at a global scale the high  $^{14}\text{C}$  age of subsoils is not only driven by  
 556 the GOC fraction but the presence of GOC may considerable influence subsoil  $^{14}\text{C}$ .

## 557 5. Conclusion

558 With our approach of estimating the GOC contribution to soil OC, we could show that common and  
 559 abundant terrestrial sediments, with low amounts of sedimentary OC, can contribute considerably to  
 560 subsoil OC stocks. One fraction of OC in the sediments is of geogenic origin and could therefore  
 561 influence measured  $^{14}\text{C}$  ages in soil, in particular in subsoils. Subsoils are known for their high  $^{14}\text{C}$   
 562 ages and slow turnover rates and slow reaction to changing environmental condition. These properties  
 563 of subsoil OC may partly be derived from the GOC in the subsoil. The sediments at the investigated  
 564 sites contained OC in a range from  $0.1\text{-}0.3 \text{ g kg}^{-1}$ , allowing for contributions from GOC between 10-30  
 565 % in subsoils. We have also shown that this geogenic contribution presents a quite stable OC pool,  
 566 especially for subsoils. Thus, also sediments with comparatively low amounts of OC, could show  
 567 considerable contributions from GOC.



568 **Data availability**

569 The data will be made available on request

570 **Author contribution**

571 AD conceived of and designed the study, FK performed the sampling and analysis, and wrote the  
572 first draft. All the authors contributed to generating and reviewing the subsequent versions of the  
573 manuscript.

574 **Competing interest**

575 The authors declare that they have no conflict of interest

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