Geogenic organic carbon in terrestrial sediments and its contribution to total soil carbon

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Abstract

Geogenic organic carbon (GOC) from sedimentary rocks is an overlooked fraction in soils that has not yet been quantified, but influences the composition, age and stability of total organic carbon (OC) in soils. In this context, GOC is the OC in bedrocks deposited during sedimentation. The contribution of GOC to total soil OC may vary depending on the type of bedrock. However, no studies have been carried out to investigate the contribution of GOC derived from different terrestrial sedimentary rocks to soil OC contents.

In order to fill this knowledge gap, 10-m long sediment cores from three sites recovered from Pleistocene Loess, Miocene Sand and Triassic Red Sandstone were analysed in 1 m depth intervals and the amount of GOC calculated based on ^{14}C measurements. The ^{14}C ages of bulk sedimentary OC revealed that OC is comprised of both biogenic and geogenic components. The biogenic component relates to OC that entered the sediments from plant sources since soil development started. Assuming an average age for this biogenic component ranging from 1,000-4,000 years BP we calculated average amounts of GOC in the sediments starting at 1.5 m depth based on measured ^{14}C ages. The median amount of GOC in the sediments was then taken and its proportion of soil mass (g GOC per kg^{-1} fine soil) calculated in the soil profile. All the sediments contained considerable amounts of GOC (median amounts of 0.10 g kg^{-1} in Miocene Sand, 0.27 g kg^{-1} in Pleistocene Loess and 0.17 g kg^{-1} in Red Sandstone) compared with subsoil OC contents (between 0.53-15.21 g kg^{-1}). Long-term incubation experiments revealed that the GOC appeared comparatively stable against biodegradation. Its possible contribution to subsoil OC stocks (0.3-1.5 m depth) ranged from 1 to 26 % in soil developed in the Miocene Sand, from 16 to 21 % in the Loess soil and from 6 to 36 % at the Red Sandstone site. Thus GOC with no detectable ^{14}C content influenced the ^{14}C ages of subsoil OC, and may partly explain the strong increase in ^{14}C ages observed in many subsoils. This could be particularly important in young soils on terrestrial sediments with comparatively low amounts of OC, where GOC can make a large contribution to total OC stocks.

Keywords Geogenic organic carbon; sedimentary organic carbon; ^{14}C; terrestrial sediments; incubation experiment

1. Introduction

On average, the world's soils store more than 50 % of organic carbon (OC) in the subsoil below 30 cm depth (Batjes, 2014). This type of carbon is considered a highly stable carbon pool due to its apparently high ^{14}C ages (Mathieu et al., 2015; Schrumpf et al., 2013). However, another explanation for this could be the contribution of geogenic organic carbon (GOC), which is defined here as OC that originates from deposition during sedimentation and rock formation, and may increasingly influence subsoil OC with depth (Graz et al., 2010; Kögel-Knabner et al., 2008; Schrumpf et al., 2013; Trumbore, 2009). GOC in most cases is devoid of ^{14}C and thus may lead to an overestimation of ancient OC sources although a number of studies showed the importance of root derived, young OC inputs to subsoils (Angst et al., 2016; Crow et al., 2009). Therefore GOC may significantly influence and affect the overall ^{14}C signal, particularly in OC-poo subsoils. Vindušková et al. (2015) investigated the contribution of GOC to soils in reclaimed mine soils, and found GOC contributions to total soil OC of between 26 and 99 %. Furthermore OC-rich sediments with contents of 2-7 g kg^{-1} (Hemingway et al., 2018) and 28-105 g kg^{-1} (Frouz et al., 2011) have been investigated with regard to the stability of OC in these sediments, but no conclusion reached about GOC contributions in soils. However, the impact of GOC on soils derived from sediments or sedimentary rocks with lower OC contents has not yet been investigated. Considering that approximately 65 % of the continental earth's surface is covered with sediments and sedimentary rocks (Amiotte Suchet et al., 2003), a potentially large fraction of soils could contain GOC that contributes to soil OC stocks, even though a large portion might be derived from recent sedimentation processes. There is not yet much literature about sediments containing only low amounts of OC. There are estimations that assume sandstones to be free of GOC(van der Voort et al., 2018) or, in contrast, a storage model that assumes generally high GOC amounts of 2.4 g kg^{-1} for all sandy deposits (Copard et al., 2007). Therefore, more information about the amounts of GOC in sediments is needed.

To estimate the possible contribution of GOC to subsoil OC stocks, it is important to establish the amount of OC in sediments that comes from sedimentation (GOC) and to distinguish it from OC derived from current vegetation (biogenic OC). There are many soil- and substrate-specific factors that might influence the OC contribution from current vegetation to sedimentary OC, such as potential rooting depth or pore distribution. No method has yet been established to allow a direct quantification of GOC in different soils or sediments, apart from promising methods to quantify the graphitic part of GOC in soils (Zethof et al., 2019). The only reliable approach to distinguish between both sources is the use of ^{14}C. Since deposition of sediments mostly took place > 50,000 years BP, they do not contain ^{14}C, which has a mean half-life time of 5,730 years (Libby, 1952). In addition, the δ^{13}C values of OC in the sediments allow carbonaceous sources with δ^{13}C values around 0 ‰ to be distinguished from organic sources with δ^{13}C values < -22 ‰. Thus, the use of both carbon isotopes could reveal wether the OC is a mixture of GOC and OC from vegetation that is less than 50,000 years old.

One important question about the possible contribution of GOC in soils is whether the GOC is mineralised when it becomes part of the soil. As GOC resists degradation once it has been deposited, it can be assumed that it already exhibits a strong inherent recalcitrance. Nevertheless, this could also be due to a physical protection that prevents microbial accessibility. However, when it becomes part of the subsoil during soil development, this OC pool could be degraded by the infiltration of water, oxygen, fresh nutrients and microorganisms. Direct microbial coal degradation has already been observed in incubation experiments on mine soils (Rumpel and Kögel-Knabner, 2002; Waschkies and Huttl, 1999) or shale bedrocks directly exposed to the surface (Soulet et al., 2017). There has been no study to establish whether GOC is degradable in OC-poor sediments or sedimentary rocks has not been investigated, but it could differ since the amount of available OC can also drive microbial respiration (Colman and Schimel, 2013). Therefore these sediments might contain fewer microorganisms that can be spatially separated from the GOC, which may hamper its respiration.

To the best of the author's knowledge, there has only been one study by van der Voort et al. (2018) that has investigated the amount of GOC in soils. They estimated that GOC makes up around 80 % of soil OC in a moraine-derived soil, suggesting that GOC's contribution to soil OC is large. However, apart from this study by van der Voort et al. (2018) on a very specific sediment, there have been no direct calculations of the amount of GOC in soils.

The aim of this study was to quantify GOC in different terrestrial sediments and a sedimentary rock, and investigate its stability in incubation experiments in order to make assumptions about its possible contribution to soil OC stocks in soil profiles at the same site. The main research questions were: i) What is the relationship between sedimentary and subsoil OC contents? ii) Is OC in sediments ^{14}C-free and how much is really geogenic? iii) Will sedimentary GOC be degraded? and iv) How much does GOC contribute to soil OC?

2. Material and methods

2.1 Site description

Three sites were selected with different sedimentary bedrocks derived from a single geologic substrate that can be found close to the surface and is homogeneous down to 10 m depth. The sites represented one sedimentary rock and two soft sediments. The sedimentary rock was a sandstone (Solling Formation, Triassic) under European beech forest (*Fagus sylvatica*) 11.5 km north-east of Göttingen (51°35.012' N; 10°3.960' O), referred to below as "Red Sandstone". The soil is classified as a Folic Brunic Arenosol according to the World Reference Base for Soil Resources (WRB, 2006). The sediments were loessic deposits (Weichselian Glacial) that have been under agricultural landuse for the past decades, 30 km north of Göttingen (51°48.101 N; 9°58.002' O), referred to as "Loess",and terrestrial sandy deposits from the Miocene (Neogene formerly named Tertiary) in a European beech forest 13 km south-west of Göttingen (51°28.673 N; 9°45.323' O) referred to as "Miocene Sand". The respective soils have been classified as a Haplic Luvisol and a Dystric Chromic Arenosol accordingly. The sediments at the Loess site were deposited between the last glacial and interglacial periods between 115,000 and 400,000 years BP, according to Jordan and Schwartau (1993). To the best of the author's knowledge, these forest sites have never been under agricultural use. The associated soils are classified as a Luvisol and a Cambisol respectively. Mean annual air temperature and precipitation were 9.2 °C and 647 mm (1981-2010) according to the nearby weather station that covers all three sites.

2.2 Sampling and sample preparation

Two 10-m long sediment cores with a diameter of 15 cm were drilled at each site in April 2017. For the soft sediments (Loess and Miocene Sand), drilling was conducted as percussion drilling, and for the hard sediment (Red Sandstone) as cable core drilling with water as the flushing solution. The replicates per site were drilled approximately 10 m apart. The sampled cores were subdivided into 1-m increments. For further chemical analysis, material was taken from a depth of 85-95 cm for each 1-m increment, and the outer 5 cm removed to avoid possible contamination. Thus the increment from 1-2 m for example, was represented by a sample from a 1.85-1.95 m depth.

One sample from each depth increment (1-2 m, 2-3 m, 3-4 m , 4-5 m, 5-6 m, 6-7 m, 7-8 m, 8-9 m, 9-10 m) for chemical analysis was oven dried at 60 °C and sieved to pass 2 mm. The Red Sandstone samples were crushed with a hammer before being dried and sieved. Additionally, approximately 1-m deep soil profiles were dug and soil samples taken from the different classified layers to obtain corresponding soil parameters (Fig. 1).

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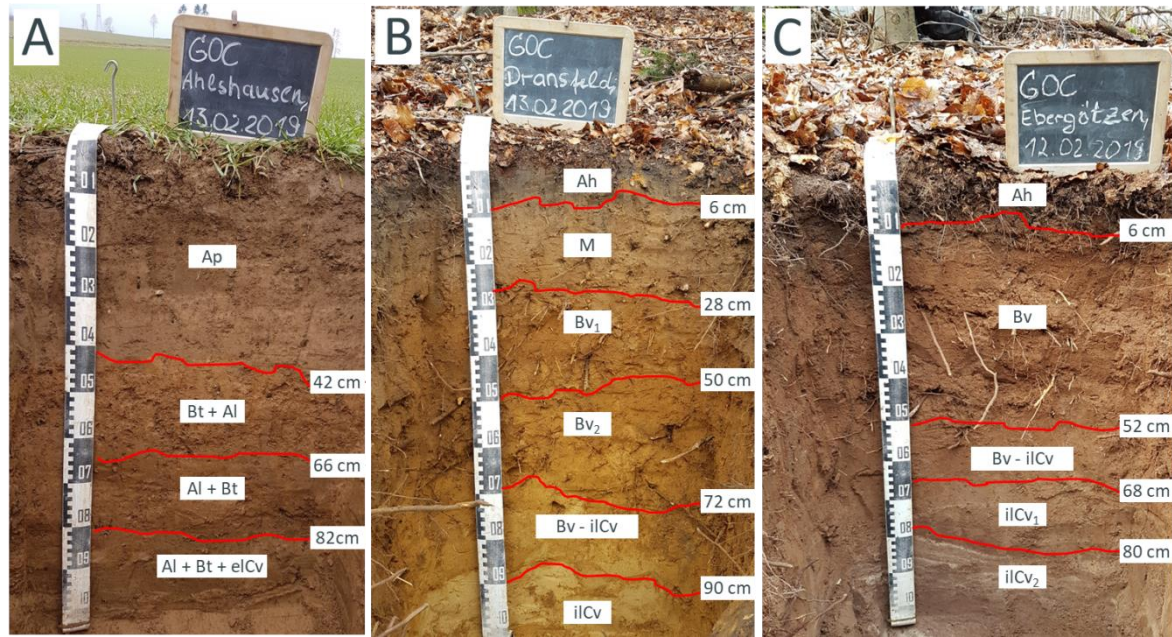


Fig. 1: Respective soil profiles at the Loess (A), Miocene Sand (B) and Red Sandstone (C) sites. Soil classification was conducted using the German classification system for soil horizons (Eckelmann et al., 2006) and later transferred to the WRB (2006). Depth transitions to the starting sediment were 82 cm for Loess (Al + Bt + elCv), 72 cm for Miocene Sand (Bv - ilCv) and 68 cm for Red Sandstone (ilCv$_1$).

Samples were oven dried at 60°C and stored and sieved to pass 2 mm. For the determination of OC and ^{13}C, samples were ground in a planetary ball mill.

2.3 Chemical analysis and calculations

Three aliquots of each sieved sample were analysed by dry combustion for total C and total N content (TruMac CN LECO, St. Joseph, MI, USA). Samples with a pH value of > 6 (measured in 0,01 mol L^{-1} CaCl) were analysed for carbonates after the sample was ignited for 16h in a muffle kiln at 450 °C according to Nelson and Summers (1983), to remove the organic part of total C. The OC concentration was calculated by subtracting carbonaceous C from total C, expressed as g OC kg^{-1} dry matter. Homogenised samples were further analysed for δ^{13}C values after carbonate removal in an isotope ratio mass spectrometer (Delta Plus, Thermo Fisher, Waltham, MA, USA) coupled to an elemental analyser (FLASH EA 1122 NA 1500; Wigan, United Kingdom). Resulting δ^{13}C values (‰) were expressed relative to the international standard of Vienna Pee Dee Belemnite. The bulk densities for the soil samples were obtained with 250 cm³ sampling rings from each layer of the soil profile. For the sedimentary Loess and Miocene Sand samples, the bulk density of the deepest soil sample (94 and 100 cm respectively) was used. The bulk densities and densities without pore space of the intact Red Sandstone cores were determined on four subsamples (from 1.6, 3.6, 7 and 9 m depth) with a Dryflow-pycnometer (GeoPyc 1360) and a gas pycnometer (AccuPyc 1330) respectively. Bulk densities for the missing depth increments were linearly interpolated. For radiocarbon (^{14}C) analysis, the sediment samples were first treated with 1% HCl to remove inorganic C, and then transferred into pre-

167   combusted quartz ampoules containing copper oxide and silver wool. The ampoules were evacuated,

168   flame-sealed, then combusted at 900 °C, and the CO_2 evolved was purified on a vacuum ring

169   (Rethemeyer et al., 2019). The ^{14}C contents were measured with the MICADAS accelerator mass

170   spectrometry (AMS) system at ETH Zürich, Switzerland. Where possible, one sedimentary sample per

171   depth increment and site and one sample per soil layer was analysed. Due to the very low OC contents

172   in some sediment samples, ^{14}C contents could only be determined for three samples from the Miocene

173   Sand (from 1.9, 4.9 and 7.9 m depth) and four from the Red Sandstone (1.9, 4.9, 7.9 and 9.9 m depth).

174   For the Loess, ^{14}C of bulk OC was measured in all depth intervals (1.9-9.9 m).

175   Total OC stocks (Mg ha^{-1}) were calculated according to Eq. 1:

$$OC\ stock = OC \cdot BD \cdot (1 - stone\ content) \cdot depth \cdot 0.1 \hspace{3cm} \text{Eq. 1}$$

176   where *OC* is the weight-based OC content, either in the fine soil <2-mm fraction of the soil profiles or

177   in the sediments (g kg^{-1}), *BD* is the bulk density of the fine soil (g cm^{-3}), stone content is the volume-

178   based proportion of stones (cm³ cm^{-3}) and depth is the thickness of the depth increment (cm). To be

179   able to compare OC stocks and contributions from GOC, it was decided to set the borders between the

180   topsoils- and subsoils at 0.3 m and the transition from subsoils to the sediments at 1.5 m. According to

181   Richter and Markewitz (1995) this represents a common border for the transition from soil to

182   sediment. The sediments where further subdivided into an upper and a lower part at a depth of 4 m.

183   In a second step, the amount of GOC and biogenic OC in the sediments was calculated, considering

184   GOC as one carbon pool free of [14]C. For the sediments, the proportion of biogenic OC ($f_{biogenic}$) of the

185   total amount of OC was calculated in a two pool model (Eq. 2) used by Cerri et al. (1985):

$$f_{biogenic}\ (\%) = \frac{F_{biogenic\ OC} - F_{GOC}}{F_{sample} - F_{GOC}} \cdot 100 \hspace{3cm} \text{Eq. 2}$$

186   where *F* represents the ^{14}C content in the fraction modern carbon (F[14]C) from a source compared with

187   the ^{14}C content of an oxalic standard (Stuiver and Polach, 1977; Torn et al., 2009). The sources were

188   the GOC fraction (F_{GOC}), the sample (F_{sample}) and the biogenic OC fraction ($F_{biogenic\ OC}$). Since the [14]C

189   content of the GOC fraction can be set to zero, this equation could be simplified to:

$$f_{biogenic}\ (\%) = \frac{F_{biogenic\ OC}}{F_{sample}} \cdot 100 \hspace{3cm} \text{Eq. 3}$$

190   For the biogenic OC in the sediments, an average ^{14}C age ranging from 1,000-4,000 years BP was

191   assumed, based on published ^{14}C results of dissolved OC reaching greater depths (Artinger et al.,

192   1996; Schiff et al., 1997). The ^{14}C contents in the sediment from 2 to 4 m depth of the loess led to ages

193   < 3,000 years BP, and were therefore even younger than at 74 cm depth (4,413 years BP). Thus, they

194   were treated like the soil part for the calculation below of a GOC fraction. Respective times were
195   converted into ^{14}C contents ($F_{biogenic\ OC}$) according to Torn et al. (2009):

$$F_{biogenic\ OC} = e^{\left(\frac{t}{-8033}\right)} \qquad \text{Eq. 4}$$

196   where t represents the ^{14}C age (1,000 or 4,000 years BP respectively) and 8033 represents the mean
197   life of radiocarbon in years. The proportion of GOC in the sediments (f_{GOC}) is therefore the portion left
198   (Eq. 5):

$$f_{GOC} = 100\ \% - f_{biogenic} \qquad \text{Eq. 5}$$

199   For the depth increments without measured ^{14}C ages (Table S1), the calculated amounts were linearly
200   interpolated with measured ^{14}C ages from samples above and below. This was done by assuming a
201   depth-dependent correlation and using the adjacent values. To calculate the amount of GOC in the soil
202   profiles, the weight-based amount of GOC in the sediments was first calculated by multiplying its
203   fraction (f_{GOC}) by the respective OC content (in g OC per kg^{-1} dry mass). The median amount of GOC
204   (g GOC per kg^{-1} dry mass) of these sedimentary values was then taken and its proportion of soil OC
205   content calculated (g OC per kg^{-1} fine soil) in the soil profile. This was done for the proportion of
206   GOC in the sediments calculated with a 1,000 and a 4,000-year-old biogenic OC fraction ($F_{biogenic\ OC}$ in
207   Eq. 3) to obtain a range of GOC contributions. It was assumed that the GOC fraction resisted
208   degradation during soil formation. Therefore, this proportion represents the largest possible amount of
209   GOC that could contribute to soil OC stocks. On this assumption, it was also possible to define the
210   influence of GOC in the soil profile on the resulting ^{14}C ages. Since the calculated ^{14}C ages represent a
211   mixture of the ^{14}C content from GOC and the biogenic fraction (Eq. 5), the GOC fraction has the same
212   influence on the soil ^{14}C age as on bulk OC (according to Eq. 3). Reducing the age by this fraction
213   would therefore represent an "unbiased" age of soil-derived-OC.

214   *2.4 Incubation experiment*

215   To assess the potential stability of OC in the sediments against microbial decay, two laboratory
216   incubation experiments were conducted at 20 °C for 50 and 63 days respectively. This was done to
217   reveal the potential degradation of OC from the sediments. A temperature of 20 °C was chosen
218   because only very low degradation rates were expected at lower temperatures. The first, 50–day
219   experiment was conducted with intact Red Sandstone core samples, while the second experiment was
220   performed after the Red Sandstone was crushed to pieces < 2 mm to simulate the process of
221   weathering when the intact sediment or sedimentary rock becomes part of the (sub)soil. For the
222   incubations, four subsamples of 1,340-6,890 g per sample from different depth intervals were used.
223   The sample material was stored at room temperature until the start of the incubation experiment. Four
224   samples were taken from each sediment and from each of four depth ranges for Miocene Sand (1.2-
225   2.8, 3.2-4.8, 6.1-7.8 and 8.2-9.9 m), Red Sandstone (2-3, 4-5.8, 7-8 and 8.4-9.8 m) and Loess (1.4-2,

226 2.7-3, 4.8-6 and 9.1-10 m). Water content was adjusted to correspond to 40 % of water-holding
227 capacity based on the poured bulk density, determined by filling the loose material into a defined
228 volume and measuring its weight. The material for the four repetitions was mixed in large plastic vats
229 and water added before the respective four subsamples were transferred to incubation vessels. There
230 were also four blank samples with no material. Based on preliminary tests and their calculated bulk
231 density and porosity, the intact Red Sandstone samples were kept in a barrel with pure water for 14
232 hours to reach a water content of nearly 40 %. Samples were placed in polycarbonate vessels with a
233 volume of 7069 cm³ and made airtight. The lids contained two tube connectors so that the samples
234 could be flushed with ambient air. After flushing, samples were set to a starting pressure of about
235 1,300 mbar and kept closed until the end of the incubation. Nine gas samples were taken in evacuated
236 glass vials (20 mL), 0, 3, 7, 13, 21, 30, 59 and 63 days after the start of incubation. Samples were
237 analysed for CO_2 concentrations by gas chromatography (Agilent 7890A, GC, Agilent Technologies,
238 Santa Clara, USA) to account for the amount of accumulated CO_2. Three additional gas samples were
239 taken 0, 30 and 63 days after the incubation started and analysed with an isotope ratio mass
240 spectrometer (Delta Plus XP, Thermo Fisher Scientific, Bremen, Germany) to account for the
241 development of $\delta^{13}C$ of CO_2 during the respiration. The corresponding pressure was measured on each
242 sampling date. When the over-pressure of a vessel was lost due to leakages, it was removed from the
243 sampling because contamination with ambient air could no longer be ruled out. This happened for a
244 third of all the samples.

245 The amount of respired CO_2-C (mg CO_2-C d^{-1}) was calculated using Eq 6.

$$CO_2 - C = \frac{0.1 \cdot p \cdot x_i \cdot M \cdot V}{R \cdot T \cdot t}$$

Eq. 6

246 where p is the pressure (mbar), x_i is the difference in CO_2 concentration between the samplings (ppm),
247 M is the molar mass of C (g mol^{-1}), V is the air volume of the sample (m³), R is the molar gas constant
248 (J kmol^{-1} K^{-1}), T is the incubation temperature (K) and t is the elapsed time (d) between the samplings.
249 Based on the $\delta^{13}C$ values of CO_2 the proportion of CO_2 derived from carbonates was subtracted where
250 necessary according to Bertrand et al. (2007). This respiration rate was related to the OC content of the
251 samples (called "OC-normalised respiration") by being divided by the total amount of OC in g in the
252 sample.

253 Since we observed an almost linear respiration behaviour was observed in both incubation
254 experiments, a simple linear regression model was fitted to describe the mineralisation rate per time
255 (mineralised OC (%) = $k \cdot t$).

256

257

 2.5 Statistics

259 Statistical analyses were conducted using the statistical environment R (R Core Team (2018))
260 including the function "*lm*" to fit linear models and the package *ggplot2* (Wickham, 2016) for
261 graphical presentation. The models were tested for deviations from homoscedasticity, normality of
262 residuals and absence of collinearity. The tests revealed heteroscedasticity of the residuals, which can
263 be explained by an increasing standard deviation with time. It subsequently became apparent that the
264 residuals were not normally distributed since the dependent variable, representing the proportion of
265 OC being mineralised, only allows values between 0 and 1. Bearing this in mind, the results should be
266 treated as an indicator of differences between the samples and a scale for the mineralisation of the OC
267 pool. Therefore calculations of standard errors and significance of the parameters were omitted since
268 this would not produce reasonable results with the model used.

269 **3. Results**

270 *3.1 Relationship between sedimentary and subsoil organic carbon*

271 Using three laboratory replicates per sample and comparing them with the muffled samples,
272 measurable OC contents were detected in all the sediments analysed. The mean relative standard
273 deviation of the laboratory replicates was 9.5 %. In terms of the detection limit, the sample with the
274 lowest total C content (mean of 0.04 g C kg^{-1} soil) showed values between 0.00 and 0.01 g C kg^{-1} soil
275 after removal of OC at 450 °C. Thus the range of 0.00 to 0.01 g C kg^{-1} soil was assumed to be the
276 mean standard error from the measurement. Despite having the same material down to 10 m depth at
277 each site, there were still some inhomogeneities that were visible and measurable. This was especially
278 true for Loess and Red Sandstone. The amount of OC in the sediments from 1 to 10 m depth was
279 comparatively low in Miocene Sand and Red Sandstone (0.04-0.71 g kg^{-1} and 0.01-0.53 g kg^{-1},
280 respectively), while much higher OC contents of 0.21-9.71 g C kg^{-1} were found in Loess (Fig. 2 a).
281 The median OC content of the sediments was within a range comparable to that of the respective
282 deepest subsoil horizon. This deepest horizon was a Cv horizon at 94 cm depth for Loess, 100 cm
283 depth for Miocene Sand and 74 cm depth for Red Sandstone. In detail, the median OC content in the
284 sediments, compared with the respective Cv horizons, corresponded to 27 % for Loess, 29 % for
285 Miocene Sand and 39 % for Red Sandstone. The Loess OC contents were highly variable, highlighting
286 the changing sedimentary conditions during the past glacial and interglacial periods (Jordan and
287 Schwartau, 1993). At 4-5 m depth, the OC contents of Loess were even higher (9.7 g kg^{-1}) than in the
288 subsoil (3 g kg^{-1}). This was not an outlier because the high OC content could be visually confirmed by
289 the very dark colour of the sample. In Miocene Sand and Red Sandstone, no clear depth gradient of
290 OC was found at 2-10 m depth (Fig. 2 a). Even though the OC content in the sediments were low, the
291 OC stocks could be very large. A comparison of OC stocks in topsoils (0-0.3 m), subsoils (0.3-1.5 m)
292 and the sediments down to 10 m depth revealed quite high OC stocks in the sediments. For Loess, OC

in the sediment contributed up to 71 % of the total OC amount while it was 51 % for Red Sandstone and 21 % for Miocene Sand (Table 1).

The distribution of the $\delta^{13}C$ values of OC in the soil and sediment profiles showed an increase in $\delta^{13}C$ with depth in the soil down to 1 m (Fig. 2 b). In contrast, the $\delta^{13}C$ values of OC in the sediments showed no clear trend with increasing depth, but they all were within the range of C_3 plant material. A value above -25 ‰ for Red Sandstone at 4 m depth could be explained by the high values of inorganic carbon (IC) at this depth. It can be assumed that decarbonisation of this sample was not completely successful. Unexpectedly high amounts of IC were found in parts of Red Sandstone, indicating the presence of calcareous deposits in this terrestrial material (Fig. 2 c). The Loess also included some distinct calcareous layers at 5 and 10 m depth, while there were only small amounts of around 0.1 mg IC g^{-1} soil at the other depths. This could be due to the fact that the investigated loess deposits belong to the "Leine Ilme Basin", a region with aoelian loamy loess that has been decalcified during weathering and soil genesis (Wagner, 2011), while the contents in the soil profile can be explained by liming. No IC was present in Miocene Sand.

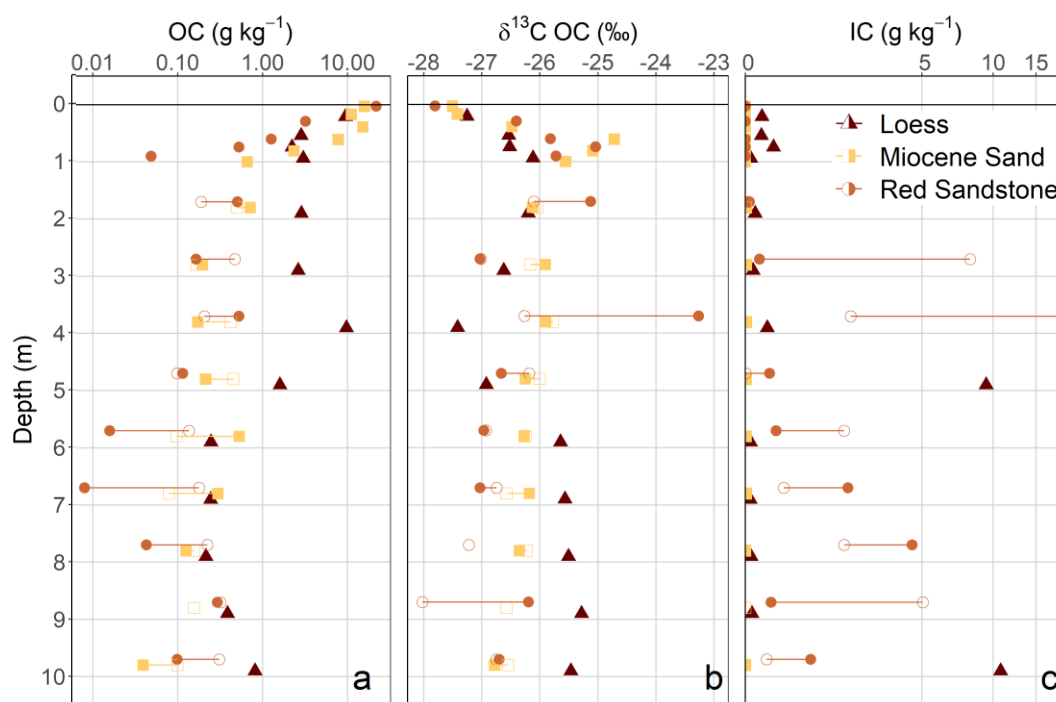



Fig. 2: Depth distribution of different bulk properties of the soil profiles and deep drilling cores. Presented parameters include the log scale organic carbon (OC) and inorganic carbon (IC) content (a and c) and the $\delta^{13}C$ values of the organic carbon (b) related to the amount of fine soil or dry mass respectively. Filled and unfilled symbols represent the two different cores. For the Loess, only one core could be analysed.

Table 1: OC stocks and proportions for the three sites down to 10 m depth. Proportions of biogenic and geogenic OC were calculated based on ^{14}C results and assumptions described in the material and methods section. Represented ranges are calculated based on the assumption of a 1,000 or 4,000-year-old biogenic OC fraction reaching the sediments.

Substrate	Layer	Depth (m)	TOC (Mg·ha^{-1})	TOC (%)[a]	OC stocks (Mg ha^{-1}) geogenic 4,000 yrs.	geogenic 1,000 yrs.	biogenic 1,000 yrs.	biogenic 4,000 yrs.	Proportion of OC (%)[b] geogenic 4,000 yrs.	geogenic 1,000 yrs.	biogenic 1,000 yrs.	biogenic 4,000 yrs.
Loess	Topsoil	0.0 - 0.3	40.6	11	1.1 - 1.2		40 - 39		3 - 3		97 - 97	
	Subsoil	0.3 - 1.5	66.0	17	5.1 - 5.3		61 - 61		8 - 8		92 - 92	
	Upper Sediment	1.5 - 4.0	218.6	57	10.6 - 11.0		208 - 208		5 - 5		95 - 95	
	Lower Sediment	4.0 - 10.0	55.1	15	42.9 - 46.7		8 - 12		78 - 85		15 - 22	
Red Sandstone	Topsoil	0.0 - 0.3	22.2	30	0.6 - 0.7		22 - 22		3 - 3		97 - 97	
	Subsoil	0.3 - 1.5	13.7	19	1.6 - 1.7		12 - 12		12 - 12		88 - 88	
	Upper Sediment	1.5 - 4.0	18.7	25	13.1 - 14.9		4 - 6		70 - 80		20 - 30	
	Lower Sediment	4.0 - 10.0	19.3	26	14.1 - 15.7		4 - 5		73 - 82		18 - 27	
Miocene Sand	Topsoil	0.0 - 0.3	39.1	31	0.3 - 0.3		39 - 39		1 - 1		99 - 99	
	Subsoil	0.3 - 1.5	60.8	48	1.3 - 1.6		59 - 60		2 - 3		97 - 98	
	Upper Sediment	1.5 - 4.0	10.6	8	3.7 - 5.8		5 - 7		34 - 55		45 - 66	
	Lower Sediment	4.0 - 10.0	16.3	13	8.3 - 10.8		6 - 8		51 - 66		34 - 49	

[a] % of total OC stock from 0-10 m

[b] % of OC stock in the respective depth increment

3.2 Ages of organic carbon in soils and sediments and contributions from geogenic organic carbon

The ages of OC in the Loess soil profiles revealed a modern carbon signature (0 years BP) at 0.3 m depth, with a sharp increase up to 4,413 ± 51 years BP at 0.7 m depth (Fig. 3 a). For the Red Sandstone soil profile there was only an increase in the ages from a modern signature at 0.04 m to 532 ± 41 years BP at 0.3 m depth. The Miocene Sand soil profile at 0.4 m depth showed an increase from 1,277 ± 41 to 1,771 ± 44 years BP at 0.6 m depth. Thus, OC of the subsoil (around 0.6 m depth) in Loess was more than twice as old as in Miocene Sand. In contrast, Loess had a modern signature at 0.3 m, while the soil developed in Red Sandstone showed an average age of 532 ± 41 years BP at 0.3 m depth.

The ages of OC in the sediments ranged from 2,200-30,730 years BP, with respective mean ages of 9,077 ± 3,234 years BP for Miocene Sand, 13,674 ± 9,632 years BP for Loess, and 14,463 ± 1,992 years BP for Red Sandstone. For all sediments, 11 out of 16 samples had a ^{14}C content that led to an ^{14}C age older than 11,600 years BP, which we assumed to be the time after the latest glacial period when soil development started (Litt et al., 2007). Therefore the sediments contained a mixture of geogenic (^{14}C-free) and biogenic (with ^{14}C) OC. Despite being the youngest sediment, Loess partly revealed the highest apparent ^{14}C ages up to 30,730 ± 631 years BP (Fig. 3 a). The ages of OC in the sediment of Red Sandstone and Miocene Sand ranged from 12,940 ± 132 to 17,390 ± 206 years BP

337   and from 6,750 ± 86 to 12,770 ± 151 years BP respectively, revealing no depth trend with age in the

338   deeper sediment. The calculated GOC fraction in the sediments was highest for Red Sandstone,

339   ranging from 67 to 87 %, with a mean of 77 % (Fig. 3 b). For the three Miocene Sand samples, the

340   GOC fraction ranged from 29 to 77 % with a mean of 53 %. Loess showed a sharp increase at a depth

341   of five metres where GOC contribution rising to between 71 to 98 % while it was only 5 to 19 % at the

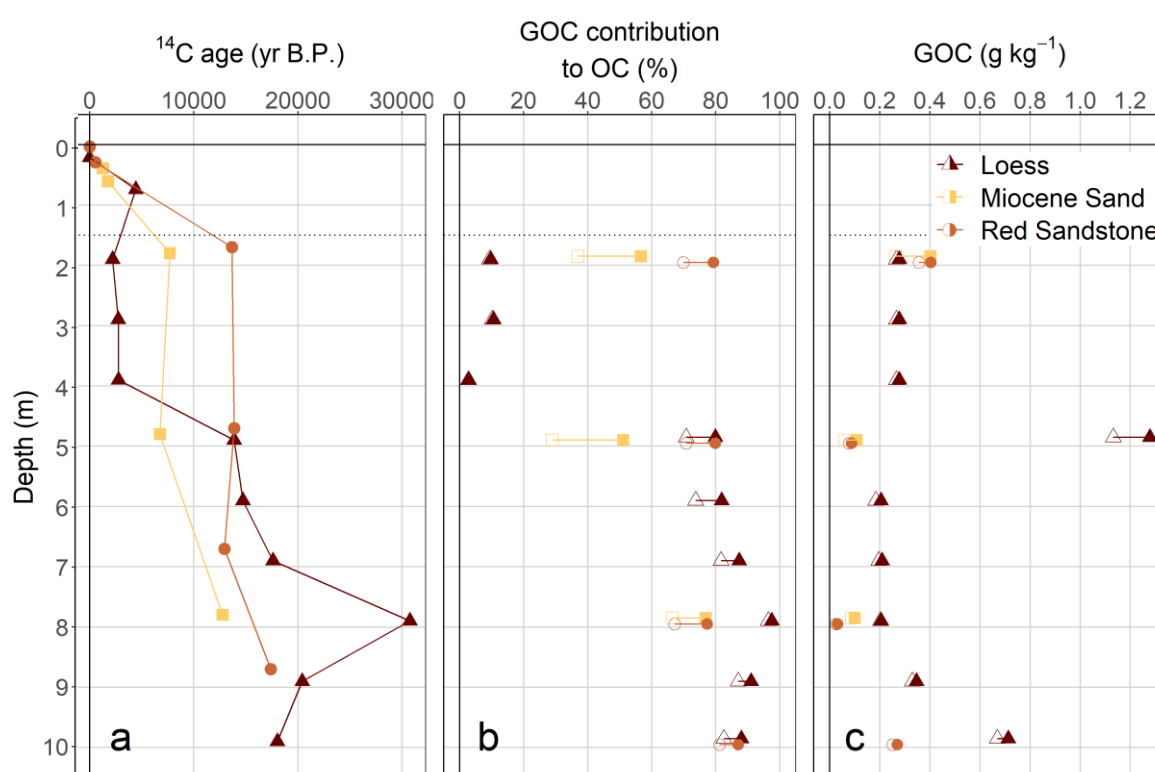
342   2 to 4 m depth.

343   The calculated weight-based content of GOC in the sediment revealed a comparatively uniform

344   distribution for all the sediments with depth, except the extremely high contents of Loess at 5 m depth

345   (Fig. 3 c). The investigated sediment depths revealed quite a narrow range of GOC contents with 0.10

346   ± 0.03 g kg^{-1} for Miocene Sand, 0.17 ± 0.12 g kg^{-1} for Red Sandstone and 0.27 ± 0.08 g kg^{-1} for Loess.

347

348   **Fig. 3:** Depth distribution of apparent ^{14}C ages (in years BP) (a), GOC contribution to OC contents in the sediments (b) and

349   resulting weight based amounts of GOC (c) in the sediments. The range in the contribution from GOC is due to the

350   assumption of an average biogenic OC age of 4,000 (empty shapes) and 1,000 years (filled shapes).

351   *3.3 Biodegradability of sedimentary derived organic carbon*

352   The incubation experiment revealed a potential, but low biodegradability of OC for all samples, but

353   without a clear depth gradient from 1 to 10 m (Fig. 4). To compare the effect of crushing on the

354   respiration of Red Sandstone samples, the mineralisation rates from the first incubation experiment

355   were compared with those of the second incubation experiment (Table 2). While Red Sandstone

356   showed very low mineralisation when the samples were incubated as intact cores (0.3-1.0 mg CO_2-C

357 g^{-1} OC y^{-1}), the mineralisation rate constants were up to five times higher when the samples were

358 crushed (1.0-2.1 mg CO_2-C g^{-1} OC y^{-1}). In a comparison of Loess and Miocene Sand, both revealed

359 large differences between samples with the lowest (1.2 and 1.8 mg CO_2-C g^{-1} OC y^{-1}) and highest (34.2

360 and 12 mg CO_2-C g^{-1} OC y^{-1}) respiration rate constants. Interestingly, there was no depth gradient for

361 the different substrates, but samples from 7 and 9 m depth of Miocene Sand tended to have up to 6.8

362 times lower respiration rate constants than samples from the 2 and 4 m depth. There was also a very

363 wide variation between samples from the same depth (see Fig. S1) as revealed by the high standard

364 deviations. Assuming a constant mineralisation rate over time, the results of the incubation experiment

365 would result in mean residence times of between 29 and 135 years for the Loess, 83 and 556 years for

366 the Miocene Sand and between 476 and 1,000 years for the crushed Red Sandstone. Since the CH_4

367 levels of the samples remained at a low level (Fig. S3) there were no indications of oxygen-limited

368 conditions during the incubation.

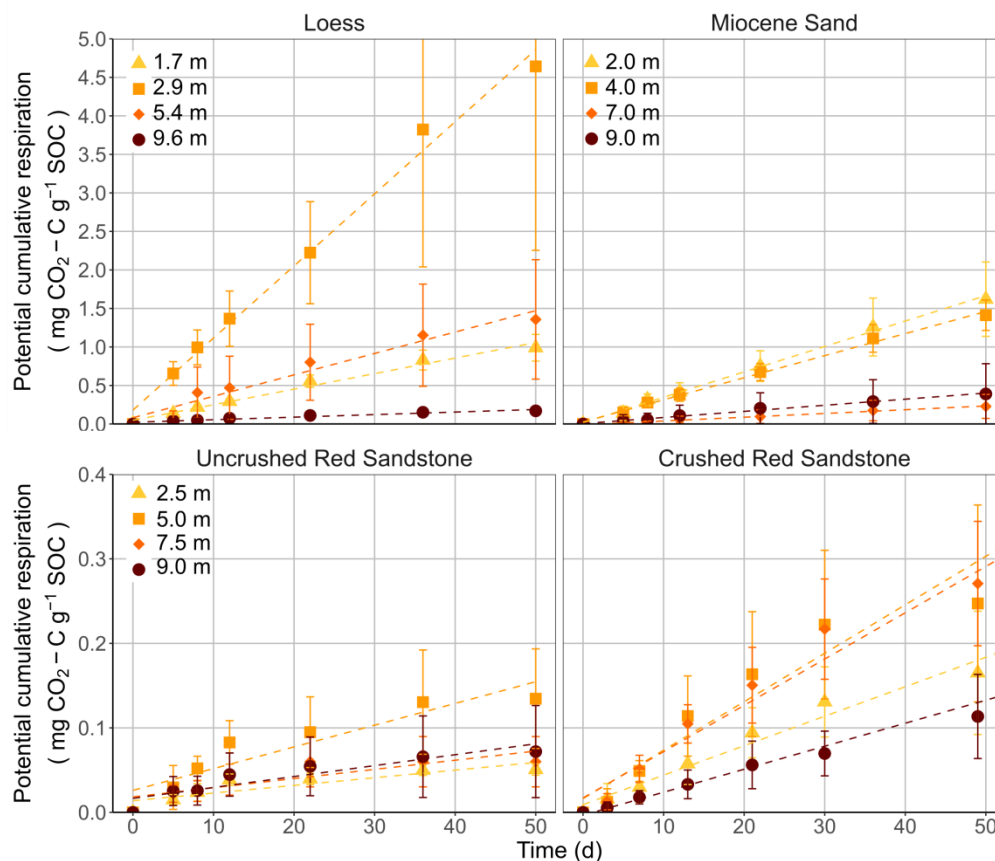
369

370 **Table 2:** Comparison of mineralisation rate constants between the first (intact Red Sandstone samples) and second incubation

371 experiments (crushed Red Sandstone samples) calculated using linear models.

Substrate	Depth (m)	Mineralisation rate constant (mg CO_2-C g^{-1} OC y^{-1})[a]
Miocene Sand	2	12.0
	4	10.5
	7	1.8
	9	2.9
Loess	1.7	7.4
	2.9	34.2
	5.4	10.2
	9.6	1.3
Uncrushed Red Sandstone	2.5	0.3
	5	1.0
	7.5	0.4
	9	0.4
Crushed Red Sandstone	2.5	1.3
	5	2.1
	7.5	2.0
	9	1.0

372 [a] represents the slope of the fitted linear model for respiration during the incubation experiment

373

374
375 **Fig. 4:** Potential degradability of sedimentary OC from three sites. Results represent cumulative respiration from
376 Loess and Miocene Sand samples and both Red Sandstone incubation experiments with uncrushed and crushed
377 samples, with respective standard deviations (n = 4). Dashed lines represent a fitted linear model to the
378 respiration data.

379 *3.4 Possible contribution from geogenic organic carbon to soil organic carbon*

380 As bedrock weathers, it becomes part of the soil and GOC also becomes soil OC. The potential
381 amount of GOC in the subsoil was dependent on the sedimentary bedrock (Fig. 5). In the subsoil of
382 Miocene Sand the proportion of GOC amounted to 2-3 %, at the Loess site it was 8 %, and at the Red
383 Sandstone site it was 12 % (Table 1). For the defined topsoils (0-30 cm depth), contributions of GOC
384 to soil OC were smaller with 0.7-0.9 % for Miocene Sand, 2.8-2.9 % for Loess and 2.8-3.0 % for Red
385 Sandstone The possible contribution of GOC to soil OC was also calculated under the assumption that
386 the biogenic fraction had an unrealistically high average age of 10,000 years (Fig. S4). This resulted in
387 the GOC fraction falling to 0 for Miocene Sand. For Loess the geogenic fraction fell by ~ 25 %, and
388 for Red Sandstone by ~ 16 %.

389 The presence of ^{14}C-free GOC in soil OC reduced the mean bulk soil OC ^{14}C ages, depending on its
390 proportion in soil OC content. Topsoils that developed in Loess and Red Sandstone had a modern [14]C
391 content of 1.029 and 1.035 $F^{14}C$, similar to the atmospheric ^{14}C content in 1950. Due to the large
392 proportion of biogenic OC, no influence of a geogenic fraction was detected in the topsoil of these
393 sites. No ^{14}C data were available for the topsoil of Miocene Sand. For all subsoils, the influence of

394   GOC on bulk soil OC ^{14}C contents depended greatly on the depth and corresponding OC contents. For

395   Loess, the possible influence on ^{14}C ages in the subsoils was quite high, with an average of 10 %

396   reduction in mean apparent ^{14}C ages in the subsoil. Thus it would reduce the measured age of 4,413

397   years BP in 74 cm depth by 532-555 years BP. Geogenic OC potentially reduced the mean apparent

398   radiocarbon age of 1,277 years BP at 0.39 m depth in Miocene Sand by about 7-9 years, and the

399   radiocarbon age of 1,771 years BP at 0.61 m depth by 20-24 years. These reductions are below the

400   respective standard deviations of the measurement. Nevertheless, at 1 m depth a given possible

401   proportion of 13.1-16.3 % would reduce an interpolated ^{14}C age of 3,053 years BP by 399-497 years

402   For Red Sandstone the influence of GOC on ^{14}C ages would be highest in the subsoil. At 74 cm depth

403   it would increase an age of 1,453 years BP by 451-490 years BP. Due to the low amounts of soil OC at

404   90 cm depth at the Red Sandstone site, the weight-based median amount of GOC in the sediments was

405   four times higher than the biogenic amount of soil OC.

406

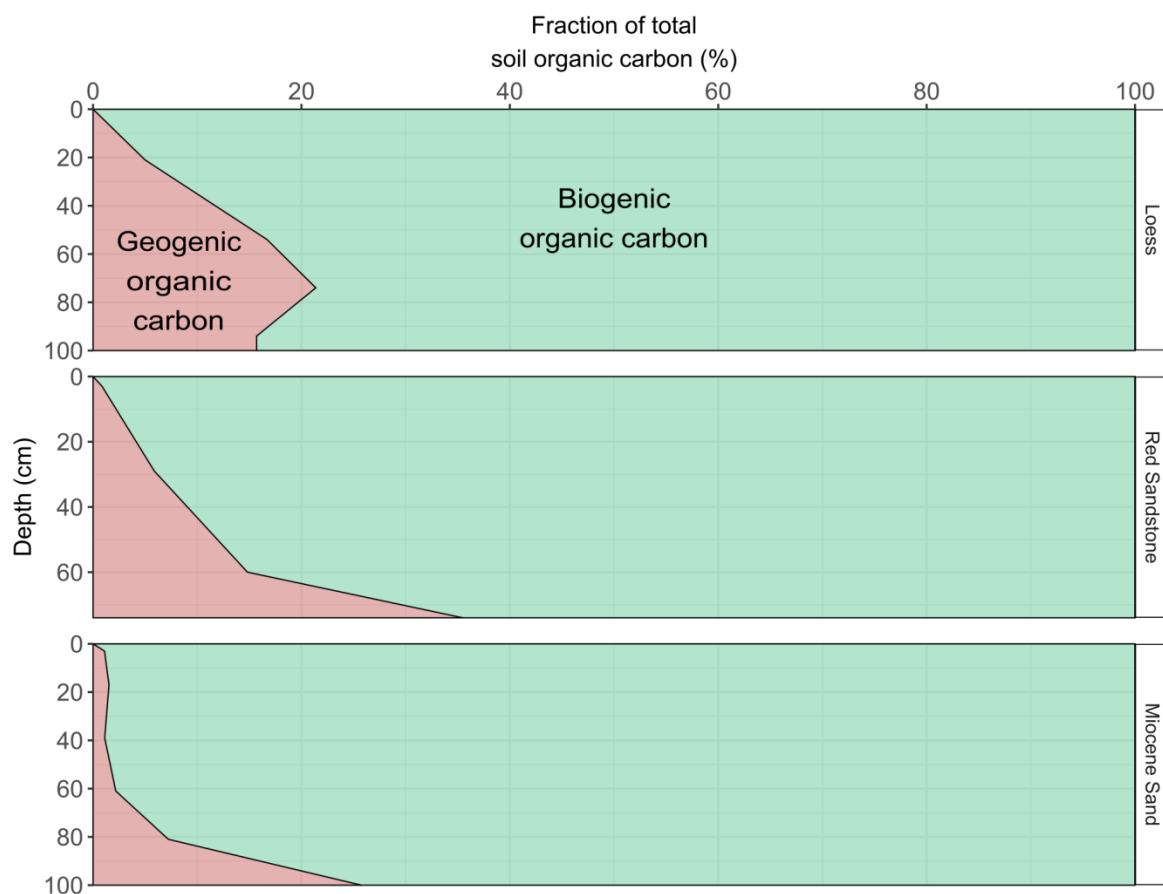



Fig. 5: Largest possible contribution of GOC to OC (red area) in relation to bulk OC content, taking into account the median
GOC contents of the sediments for the respective horizontal weight based OC contents. The contribution of GOC is the
mean amount based on the assumption of an average biogenic OC age of between 4,000 and 1,000 years

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4. Discussion

4.1 Geogenic OC in the sediments

It should be noted that these GOC calculations in sediments were based on the assumption of biogenic OC in the sediments not being older than 4,000 years BP on average. The influence of a biogenic OC fraction derived from soils that developed before the latest glacial period was also excluded. Thus it is possible that the biogenic OC fraction in the sediments is even older. Nevertheless, even with an assumed age of 10,000 years for the biogenic OC fraction, the greatest possible contribution was 15.4 % for Loss (94 cm depth) and 21.5 % for Red Sandstone (74 cm depth) (Fig. S4). Considering that modern OC entering the sediment has a quantitatively greater influence on measured ^{14}C ages than very old OC, such high ages for the biogenic OC fraction seem unrealistic and would mean that almost no young OC enters the sediment. Given the high ^{14}C ages, it can therefore be assumed that some of the OC in the sediments comes from sedimentation as geogenic OC.

4.1.1 Site dependent contents of GOC

Regarding the calculated contribution of GOC to OC in the sediments, the assumed range of biogenic ^{14}C ages from 1,000-4,000 years BP was within the typical range for ages of dissolved OC leaching from soils (Artinger et al., 1996; Jia et al., 2019). Nevertheless, the range of 1,000-4,000 years BP did not greatly influence the range of calculated sedimentary contribution from GOC, especially for Loess (~ 6 % difference) and Red Sandstone (~ 9 % difference). The calculated GOC contribution for Miocene Sand was comparatively low (29-77 %) compared with the contribution for Red Sandstone and Loess, especially at 5 m depth. Compared with the Red Sandstone, this could be due to deep biogenic carbon inputs, such as roots and root exudates from the trees (Angst et al., 2016; John et al., 2016; Kirfel et al., 2017; Tückmantel et al., 2017), since the loosely bedded Miocene Sand allows for deep infiltration compared with the Red Sandstone site with its shallow bedrock as a root-restricting layer (Schneider and Don, 2019). Although it was not possible to define the exact rooting depth for Miocene Sand, this depth probably does not exceed 4 m, according to Schenk and Jackson (2005). Therefore the increase in sedimentary contribution from 5 to 8 m depth of Miocene Sand could be due to the decreasing influence of roots and root exudates at 8 m depth.

In Loess, the low ^{14}C ages at 2-4 m depth (2,200-2,770 years BP) in contrast to a ^{14}C age of 4,413 years BP at 0.74 m depth were surprising. This might indicate past anthropogenic activities or erosion-driven material movement that might have led to a mixing of the upper part of the profile. Furthermore, the modern ^{14}C signature at 21 cm depth could be due to the plough layer at the Loess site mixing the upper 30 cm. Nevertheless, at depths below 4 m the high ^{14}C ages of OC in the sediments indicated a large proportion of GOC. This could be due to different sedimentation periods, soil forming and also soil burial processes (Chaopricha and Marin-Spiotta, 2014) that took place

448   during the Pleistocene. These processes can lead to the presence of buried layers in Loess with varying
449   amounts of rather recalcitrant OC, as shown by Marin-Spiotta et al. (2014). For the investigated Loess
450   site, different sedimentation and soil forming processes can be expected due to the presence of
451   completely different material in the cores in terms of colour and measured OC contents. For example,
452   the very dark Loess at 4 m depth with its high OC content support the assumption of sedimentation
453   circumstances that favoured the accumulation and preservation of OC. This is in accordance with
454   Jordan and Schwartau (1993) who investigated the same site and assigned the different layers to
455   specific Pleistocene sedimentation periods.

456   In summary, the contribution of GOC to sedimentary OC was substrate dependent. A loosely bedded
457   sediment like Miocene Sand with extremely low concentrations of OC could be more prone to
458   infiltration of biogenic OC and dilution of GOC. This resulted in contributions of biogenic OC to the
459   sediments of about 50 %. In contrast, the Loess site with comparatively low infiltration rates or the
460   Red Sandstone site with reduced possibilities for deep rooting seemed to contain relative constant
461   contributions from GOC of around 80 %.

462   The GOC contribution within the sediments was not found to increase with soil depth. This is in
463   contrast to the results of Frouz et al. (2011) which showed that different sediment types from a
464   Miocene clay sediment had higher weight-based carbon contents at 150 m compared with 30 m depth.
465   However, a comparison with the present study is difficult since Frouz et al. (2011) did not distinguish
466   between the geogenic and biogenic OC fractions, and OC contents were much higher (28-112 g kg^{-1}
467   dry mass compared to 0.008-10 g kg^{-1} dry mass in our results). However, it also underlines the
468   importance of different sedimentation processes for the amount and depth distribution of OC in
469   sediments and sedimentary rocks. *4.1.2 Geogenic OC in Miocene Sand and Red Sandstone compared*
470   *with other studies*

471   It is hard to compare the weight-based amounts of GOC in terrestrial sediments since most studies in
472   this field rarely determine the amounts of OC in terrestrial sediments, or presume that sandy
473   sediments, for example, do not contain large amounts of OC (Artinger et al., 1996). Quite high
474   amounts of OC have been found in the skeleton part of different soils on sandstones with 0.61-1.97 t
475   ha^{-1} (Corti et al.,2002). Nevertheless, they mentioned the possible strong influence of organic
476   substances from the soil solution without quantifying it, and did not directly investigate OC in the
477   sediments. Additionally Copard et al. (2007) assumed an OC amount of 2.4 g kg^{-1} from an unknown
478   source for all sandy sediments in a global storage modelling approach for the first metre of sediments.
479   This would fit with the amounts of around 1 and 5 mg OC g^{-1} found by Krummholz et al. (1997) for
480   Dakota sandstone layers at > 180 m depth, but is much greater than the median GOC amount in Red
481   Sandstone (0.2 g kg^{-1}) and Miocene Sand (0.1 g kg^{-1}) in the present study.

482   *4.1.3 Geogenic OC in Loess compared with other studies*

Loess deposits are relatively well investigated because they provide a record of paleoenvironmental conditions (Hatté et al., 1998; Head et al., 1989; Murton et al., 2015; Wang et al., 1996). The median amount of 0.27 g kg^{-1} from the present study was low compared with studies by Hatté et al. (1998), Wang et al. (1996) and Strauss et al. (2012). Hatté et al. (1998) investigated 20-m deep loess deposits in the Rhine valley and found OC contents between 1.0-8.6 g kg^{-1}, Wang et al. (1996) investigated 12-m deep loess deposits in China and found OC contents of 31.2 ± 30.5 g kg^{-1} and Strauss et al. (2012) found OC contents of 15 ± 14 g kg^{-1} in Yedoma loess deposits in Siberia. This shows that the deposits from the site in the present study stored comparatively low GOC contents, although most of the above-mentioned studies do not distinguish between a biogenic and geogenic OC pool. Nevertheless, as also shown in the present study, Loess had a high OC content compared with other sediments.. This was in line with highest OC contents in subsoils at the Loess site and may indicate the importance and contribution of bedrock OC to subsoil OC.

4.2 Is sedimentary derived organic carbon biodegradable?

The incubation experiment revealed a mineralisation of OC within the sediments with values between 0.1 and 3.4 % of total OC being mineralised after one year, assuming a constant mineralisation rate. The incubation temperature of 20 °C should therefore be considered to be above the typical mean temperatures in the subsoils. For subsoils with comparable climatic conditions, Wordell-Dietrich et al. (2019) found temperatures at 150 cm depth ranging from 4 to 14.4°C over a two-year period. Assuming a Q_{10} value of around two for the assumable difference (Hamdi et al., 2013), the respiration rate at typical subsoil temperatures would be roughly half that. A direct mineralisation of OC from sediments is in agreement with several studies investigating the direct mineralisation from outcrops (Copard et al., 2007; Horan et al., 2017; Petsch et al., 2000; Soulet et al., 2017; Seifert et al., 2001). The difference to the present study is, that those studies observed this mineralisation when the sediments were directly exposed to the surface or/and were part of a very rapidly eroding area. Thus GOC from the sediments is already in touch with the atmosphere and inputs of recent vegetation. However, Frouz et al. (2011) conducted an incubation experiment with sedimentary samples from OC-rich Miocene clay sediments and found quite high respiration rate constants, with values between 3.5-12.3 mg CO_2-C g^{-1} OC y^{-1} during a 91 day incubation experiment. They attribute this to the prevailing presence of aliphatic compounds in their samples being decomposed. Kieft and Rosacker (1991) also found high respiration rates of sedimentary samples, with values between 0.9-9.5 mg CO_2-C g^{-1} OC y^{-1}, which they primarily attribute to the physiological status of the soil microbial community expressed as adenylate energy charge. Those results are in fairly good agreement with the respiration rate constants observed in the present study (1 - 34.2 mg CO_2-C g^{-1} OC y^{-1}). Meanwhile, compared with the subsoil incubation experiments, the mineralisation found in the present incubation experiment was quite low. For example, in subsoil incubation experiments at 20 °C, Wordell-Dietrich et al. (2017) found that between 5-9.5 mg CO_2-C g^{-1} OC of OC are mineralised after incubation for 63 days, Wang

519   et al. (2013) report values between 5-15 mg CO_2-C g^{-1} OC after 28 days, and Soucemarianadin et al.

520   (2018) report values between 10- and 12.5 mg CO_2-C g^{-1} OC after 70 days. The difference between

521   the respiration rates observed in the present incubation experiment and the results from sedimentary

522   and subsoil incubation experiments could be due to different microbial communities, OC quality and

523   the physical connection between OC and potential decomposers. As shown in a meta-analysis by

524   Colman and Schimel (2013), different microbial compositions, their abundance and the quality of OC,

525   strongly affects respiration rates. This might also be indicative of extreme differences in respiration

526   rates even within the same substrate and sample.

527   The results of the present study indicated that there was a considerable portion of biogenic OC in the

528   sediments. The observed mineralisation could therefore be due primarily to the consumption of this

529   biogenic OC part. Furthermore, the low mineralisation rate of Red Sandstone during the incubation as

530   intact cores (Fig. S1) promoted the stability of GOC when it is part of the sediments. This might be

531   due tothe low accessibility of OC in the sediments for microorganisms and the low availability of

532   water due to a preferential flow through the sandstone (Swanson et al., 2006).

533   Altogether, the low mineralisation rates of the OC in the sediments might be caused by a lack of fresh

534   substrates and/or microorganisms that could enhance the degradation of OC (Fontaine et al., 2007).

535   Seifert et al. (2011) have shown that microorganisms are able to degrade sedimentary OC after the

536   addition of glucose in black slate outcrops. Nevertheless, mineralisation of OC could only be observed

537   with the addition of water, which indicates the widely recognised assumption of the presence of an

538   active microbial community in the sediments (Bomberg et al., 2017; Joergensen and Wichern, 2018;

539   Magnabosco et al., 2018). Since fresh substrates were not added during the incubation, respiration

540   rates could be even higher if a fresh substrate-induced priming effect occurs. Furthermore the inherent

541   and active microbial communities in the sediments might have assimilated ^{14}C-free GOC into their

542   biomass, as shown by Schwab et al. (2019). Thus part of the labile OC pool in the sediments might

543   also be derived from metabolised ^{14}C-free microbial biomass.

544   The incubation experiment was unable to answer the question of whether GOC is mineralised when it

545   becomes part of the (sub-)soil. Assuming that a large part of the biogenic OC in the sediments was

546   mineralised during the incubation experiment, geogenic OC could still be preserved during soil

547   formation especially in the subsoils. This is in accordance with the indirect approach taken by Graz et

548   al. (2010) to determine the mineralisation of sedimentary OC when it becomes part of the subsoil .

549   They stated that 30 % of GOC resists degradation when it becomes part of the soil due to the results of

550   a quantitative palynofacies analysis of bedrock and soil samples. Hemingway et al. (2018) found that

551   sedimentary OC directly exposed to the surface in a rapidly eroding tropical mountain area exhibits

552   considerable mineralisation down to 1 m below the surface, also leading to around 30 % of GOC

553   remaining in the soil. Based on ^{14}C measurements they found out that on average 67 ± 11 % of the OC

554   fraction in the sediments could be lost during soil formation, but no distinction was made between

555 biogenic and geogenic OC fractions. This indicates that a microbial mineralisation of bedrock OC

556 takes place but may be partly restricted to biogenic OC.

557 Regarding the depth distribution of GOC in the sediments, the amount of GOC (in g kg^{-1}) did not

558 increase with depth, but there were clear differences. This reflects the sedimentation history with

559 different initial amounts of OC and degradation during sedimentation, which is particularly evident in

560 the high amounts of GOC at 5 m depth in Loess. However, the contents of GOC, especially in Red

561 Sandstone and Miocene Sand, were within the same range for the whole depth. This might indicate

562 that degradation of GOC is not depth-dependent within the sediments. If there were a stronger

563 degradation of GOC with decreasing depth, a decreasing amount of GOC could be expected due to the

564 input of water, microorganisms and fresh nutrients from above. Furthermore, there was a relatively

565 constant contribution of biogenic OC within the sediments, meaning that if biogenic OC enters the

566 sediments with possibly degrading microorganisms, this biogenic OC might also be largely

567 mineralised. A study by Heitkötter et al. (2018) demonstrates effectively that degradation of OC in

568 subsoils is primarily limited to small hotspot areas. Also in sediments, Krummholz et al. (1997) show

569 that microbial communities can particularly be found in spatially discrete areas. Nevertheless,

570 Heitkötter et al. (2018) also show that microorganisms outside the hotspots can be activated when

571 substrate is supplied. Thus the bioavailability of GOC might be very site-dependent since root

572 channels as microbial hotspots, for example, are less abundant and stable in sandy soils (Schneider and

573 Don, 2019). With regard to the sites investigated here, the solid Red Sandstone might only obtain

574 water in preferential flow paths (Swanson, 2006) leading to a comparatively stable OC pool in the

575 sediment. Meanwhile for weathered, poorly structured sandy soils, prevalent matrix flow conditions

576 can be assumed (Flury et al., 1994). Thus the bulk GOC might be supplied with fresh substrates and

577 water from above more often than for the well-structured Loess soil with more frequent and stable

578 preferential flow paths (Schneider and Don, 2019). This might lead to a lower accessibility and

579 therefore slower turnover rates of OC (Dungait et al., 2012).

580 The conclusion can be drawn that GOC can and will be degraded when it becomes part of the subsoil

581 but probably to a comparatively limited extent.

582 *4.3 How much does GOC contributes to soil organic carbon?*

583 The contribution of GOC to soil OC stocks in this study was driven by the amount of OC in the soil,

584 the amount of GOC in the respective sediment and also the turnover of GOC when it becomes part of

585 the soil. Our results revealed that despite differing between sediments, GOC content varied in a quite

586 narrow range between 0.1 and 0.3 g kg^{-1}. The contribution of GOC to topsoil OC was negligible.

587 Assuming no degradation of GOC, the greatest possible contributions to total subsoil OC were found

588 for Red Sandstone (~30 %) and the lowest for Miocene Sand (0.6 %). This was due to the range of

589 OC contents in the subsoils (0.53 g kg^{-1}-15.21 g kg^{-1}). When soil OC contents were low, the possible

590   contributions from GOC were high and vice versa For the investigated soils, the OC contents of 3 g

591   kg^{-1} soil allowed for possible GOC contributions of between 5- and 10 %. For OC contents around 1 g

592   kg^{-1} soil, a GOC contribution of between 10- and 20 % seemed possible. Thus, greater contributions

593   were made by GOC rich sediments such as Loess, while smaller contributions from sandy sediments.

594   In comparison, van der Voort et al. (2018) estimated the contribution of GOC in a soil derived from

595   glacial deposits (flysch) at between 80-100 cm depth to be around 40 %. For a soil developed from a

596   poorly consolidated sedimentary rock (calcareous and shaly moraine), they calculate the contribution

597   of GOC to range from 20 % at 145 cm depth to 80 % at 310 cm depth. An attempt has also been made

598   to fractionate subsoils to extract the most stable OC that may be derived from GOC. Paul et al. (2001)

599   investigated a soil developed on loess over till with 30 % of subsoil OC as a non-hydrolysable fraction

600   showing a ^{14}C age of 13,000 years BP.They also concluded that this high ^{14}C age can partly be

601   explained by a GOC fraction. These results indicate that deposits from the past glacial periods, such as

602   flysch or till in particular, have much greater potential for OC contributions from GOC, possibly due

603   to the higher amounts of GOC in their sediments. Since only terrestrial sediments were investigated in

604   the present study, it should be noted that marine sediments or shales also contain much higher amounts

605   of OC, up to 250 g kg^{-1} (Hemingway et al., 2018; Petsch et al., 2000). Thus, the amount of GOC they

606   conatin and their possible contribution to subsoil OC stocks might therefore be much higher.

607   Nevertheless, the ^{14}C ages of OC in the subsoil can be high in soils derived from igneous parent

608   materials without GOC (Rumpel et al., 2002), although even crystalline bedrocks contain microbial

609   communities (Purkamo et al., 2020). Furthermore, on a global scale, the ^{14}C ages of soil OC are

610   primarily driven by climatic conditions, clay content and age of the soil (Mathieu et al., 2015).

611   However, for terrestrial sediments with comparatively low amounts of GOC that started their soil

612   development after the latest glacial period, a scale of possible contributions could be obtained when

613   the amount of OC is known. Thus, on a global scale, the high ^{14}C age of subsoils is not driven just by

614   the GOC fraction but the presence of GOC may greatly influence subsoil ^{14}C.

615   5. Conclusions

616   In this study the amount of GOC in sediments and in the soil was analysed by radiocarbon dating. The

617   aim was to find out if GOC from different terrestrial sediments can have an influence on soil OC

618   stocks. This approach of estimating the GOC contribution to soil OC showed that common and

619   abundant terrestrial sediments with low amounts of sedimentary OC can make a considerable

620   contribution to soil OC stocks. One fraction of OC in the sediments is of geogenic origin and could

621   therefore influence measured ^{14}C ages in soil, particularly in subsoils. Subsoils are known for their

622   high ^{14}C ages and slow turnover rates and slow reaction to changing environmental condition. These

623   subsoil OC properties may partly be derived from the GOC in the subsoil. The sediments at the

624   investigated sites contained OC in a range of 0.1-0.3 g kg^{-1}. These amounts allowed for contributions

625   from GOC of between 10-30 % in subsoils, defined here as soil horizons ranging from 0.3 to 1.5 m

626 depth. Incubation of sediments indicated that this geogenic contribution presents a quite stable OC
627 pool, especially for subsoils.

628 **Data availability**

629 The data will be made available on request

630 **Author contribution**

631 AD conceived of and designed the study, FK performed the sampling and analysis, and wrote the first
632 draft. All the authors contributed to generating and reviewing the subsequent versions of the
633 manuscript.

634 **Competing interest**

635 The authors declare that they have no conflict of interest

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646   **References**

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