

1 **Modelling of long term Zn, Cu, Cd, Pb dynamics from soils fertilized with organic**
2 **amendments.**

3

4 Running head: Modelling of long-term trace element dynamics

5

6 Claudia Cagnarini^{1,2}, Stephen Lofts³, Luigi Paolo D'Acqui⁴, Jochen Mayer⁵, Roman Grüter⁶,
7 Susan Tandy^{6,8}, Rainer Schulin⁶, Benjamin Costerousse⁷, Simone Orlandini¹, Giancarlo Renella⁹

8

9 1, Dipartimento di Scienze e Tecnologie Agrarie, Alimentari, Ambientali e Forestali (DAGRI),

10 Piazzale delle Cascine, 18, 50144 Firenze, Italy

11 2, UK Centre for Ecology & Hydrology, Environment Centre Wales, Deiniol Road, Bangor,

12 Gwynedd, LL57 2UW, United Kingdom

13 3, UK Centre for Ecology & Hydrology, Lancaster Environment Centre, Library Avenue, Bailrigg,

14 Lancaster, LA1 4AP, United Kingdom

15 4, Istituto di Ricerca sugli Ecosistemi Terrestri, CNR, Area della Ricerca di Firenze, Via Madonna

16 del Piano 10, 50019 Sesto Fiorentino (Firenze), Italy

17 5, Agroscope, Dept Agroecology and Environment, Reckenholzstrasse 191, 8046 Zurich,

18 Switzerland

19 6, Institute of Terrestrial Ecosystems, ETH Zurich, Universitätstrasse 16, 8092 Zurich, Switzerland

20 7, Institute of Agricultural Sciences, ETH, Universitätstrasse 2, 8092 Zurich, Switzerland

21 8, Rothamsted Research, North Wyke, Okehampton, Devon, EX20 2SB, United Kingdom

22 9, Department of Agronomy, Food, Natural Resources, Animals and Environment (DAFNAE),

23 viale dell'Università 16, 35020 Legnaro (Pd), Italy

24

25 Corresponding author: Claudia Cagnarini, Email: ccagnar@ceh.ac.uk

26

27 Paper type: Research Article

28

29 Key words: trace elements, farmyard manure, sewage sludge, compost, sustainability, Switzerland

30

31 Declaration of interest: none.

32

33 **Abstract**

34 Soil contamination by trace elements (TEs) is a major concern for sustainable land management.
35 One potential source of excessive inputs of TEs into agricultural soils are organic amendments.
36 Here, we used dynamic simulations carried out with the Intermediate Dynamic Model for Metals
37 (IDMM) to describe the observed trends of topsoil Zn, Cu, Pb and Cd concentrations in a long-
38 term (>60 year) crop trial in Switzerland, where soil plots have been treated with different organic
39 amendments (farmyard manure, sewage sludge and compost).

40 The observed EDTA-extractable concentrations ranged between 2.6-27.1 mg kg⁻¹ for Zn, 4.9-29.0
41 mg kg⁻¹ for Cu, 6.1-26.2 mg kg⁻¹ for Pb and 0.08-0.66 mg kg⁻¹ for Cd. Metal input rates were
42 initially estimated based on literature data. An additional, calibrated metal flux, tentatively
43 attributed to mineral weathering, was necessary to fit the observed data. Dissolved organic carbon
44 fluxes were estimated using a soil organic carbon model. The model adequately reproduced the
45 EDTA-extractable (labile) concentrations when input rates were optimised and soil lateral mixing
46 was invoked to account for the edge effect of mechanically ploughing the trial plots: the global
47 average RMSE was 2.7 and the average bias (overestimation) was -1.66, -2.18, -4.34 and -0.05 mg
48 kg⁻¹ for Zn, Cu, Pb and Cd, respectively. The calibrated model was used to project the long term
49 metal trends in field conditions (without soil lateral mixing), under stable climate and management
50 practices, with soil organic carbon estimated by modelling and assumed trends in soil pH. Labile
51 metal concentrations to 2100 were largely projected to remain near constant or to decline, except
52 for some metals in plots receiving compost. Ecotoxicological thresholds (critical limits) were
53 predicted to be exceeded presently under sewage sludge inputs, and to remain so to 2100.
54 Ecological risks were largely not indicated in the other plots, although some minor exceedances
55 of critical limits were projected to occur for Zn before 2100.

56 This study advances our understanding TEs long-term dynamics in agricultural fields, paving the
57 way to quantitative applications of modelling at field scales.

58

59 **1. Introduction**

60 Trace elements (TEs) are naturally present in soils due to mineral weathering and biogeochemical
61 cycles. Several TEs such as zinc (Zn), copper (Cu) and nickel (Ni) play important roles in
62 biochemical processes and are essential for living organisms at low concentrations, though they
63 can become toxic to biota at high concentrations; therefore, their presence in soil can be tolerable
64 in a relatively narrow range of values (Adriano, 2005). Wan et al. (2020) and Wang et al. (2015)
65 reported that the concentrations providing ecological safety in China ranged from 38.3 to 263.3
66 mg kg⁻¹ for Zn and from 13.1 to 51.9 mg kg⁻¹ for Cu when shifting from acidic soils to alkaline
67 non/calcareous soils. In contrast, other TEs such as lead (Pb) and cadmium (Cd), which are not
68 physiologically active, may be toxic to living organisms at low concentrations and their
69 accumulation in soil is of particular concern. For example, background cadmium levels in world
70 topsoils range from 0.01 to 2.7 mg kg⁻¹, and in Europe the mean cadmium concentration in
71 cultivated soils is 0.5 mg kg⁻¹ (EFSA, 2009). Excessive uptake of trace elements by crop plants
72 and enrichment in edible parts can pose significant risks to human health by entering into the food
73 chain (McGrath et al., 2015).

74 Accumulation of TEs in cultivated soils is widespread and is mainly caused by application of low
75 grade agrochemicals, organic fertilizers and sewage sludge (Toth et al., 2016). In an European
76 Union-wide survey, Ballabio et al. (2018) reported that agricultural soils have among the highest

77 potential to become enriched in Cu compared with other land uses, and that land cover and
78 management are better predictors of soil Cu concentrations than natural soil formation factors.

79 Organic amendments are considered more sustainable than inorganic mineral fertilizers (Diacono
80 & Montemurro, 2010): in fact, current industrial processes for N-fertilizer production are based on
81 fossil fuels, and P-fertilizers are manufactured from phosphate rocks which are naturally limited
82 (Roberts, 2014). However, the application of organic amendments, such as farmyard manure,
83 compost and digestates of bio-wastes, can also introduce TEs into agricultural soils. In the
84 European Union, typical levels of trace elements in cattle manure are 63-175 mg kg⁻¹ for Zn, 15-
85 75 mg kg⁻¹ for Cu, 1.4-4.3 mg kg⁻¹ for Pb and 0.1-0.4 mg kg⁻¹ for Cd (NEBRA, 2015), but pig and
86 poultry manure can be much more enriched in trace metals. Mean concentrations of Zn, Cu, Pb
87 and Cd in greenwaste compost in Germany were reported to be 168, 33, 61 and 0.7 mg kg⁻¹,
88 respectively (NEBRA, 2015). Application of sewage sludge into agricultural soils can be even
89 more problematic as sewage sludge can contain trace metals up to 30 times their concentrations in
90 soil (Hudcova et al., 2019).

91 Once in the soil, multiple factors may control TE speciation, solubility and mobility (Gu & Evans,
92 2008), such as soil pH, soil and dissolved organic matter (SOM, DOM) contents, the quantity and
93 chemical composition of clay minerals and metal (oxy)hydroxides. Furthermore, speciation can
94 influence the TE toxicological hazard, particularly to organisms that are directly exposed to soils
95 such as plants and earthworms: models, such as the Biotic Ligand Model (Paquin et al. 2002),
96 postulate that metal toxicity is related to uptake of specific metal species in competition with other
97 solution ions, rather than to total dissolved metals. Repeated applications of organic amendments
98 can lead to the accumulation of TEs in agricultural soils, particularly through direct reactions with
99 the soil solids (adsorption), formation of precipitates, or physical occlusion within the organo-

100 mineral aggregates (fixation). This can lead to TE concentrations exceeding environmental
101 legislation thresholds.

102 In the context of long term TE accumulation due to regular application of organic amendments or
103 other additions, predicting the long term speciation and dynamics of TEs is useful to support
104 decisions on ecosystem management and human health protection. Dynamic models, if reliable,
105 are essential for this purpose. Models for TE dynamics exist at a number of levels of complexity,
106 from those with a mechanistic approach requiring highly detailed input information and calibration
107 (Bonten et al., 2011), to simple mass balance approaches generally applicable at large scales but
108 relatively unsuitable to understand and unravelling complex metal dynamics (Six & Smolders,
109 2014). Empirical models have been used to simulate dynamics and uptake of TEs at specific
110 agricultural sites based on site-specific calibration, (Bergkvist & Jarvis, 2004; Ingwersen &
111 Streck, 2006), but such models can lack a reliable generalisation of the parameters to different
112 climatic conditions and hydrological and soil physio-chemical properties.

113 Among models for determining TE dynamics in soil, the Intermediate Dynamic Model for Metals
114 (IDMM) (Lofts et al., 2013; Xu et al., 2016) is an example of a semi-empirical dynamic model
115 which allows general application, given a reasonably parsimonious set of input data. It is intended
116 for long term application from decades to centuries. The IDMM describes metal dynamics from a
117 past year at which metal inputs can be assumed to be uninfluenced by anthropogenic activities.
118 Processes influencing metal dynamics, including solid/solution partitioning, fixation into soil solid
119 phases, and leaching, are described in a semi-empirical manner. Hydrological variables (e.g.
120 annual volume of soil drainage) are specified as time series. Similarly, key properties influencing
121 metal dynamics, such as the pH of the soil solution, dissolved organic carbon (DOC) flux, SOM
122 content and soil erosion rate, may be fixed to single values or varied annually.

123 The objective of this study was to assess the capability of the IDMM to reproduce metal dynamics
124 at a well-characterized location receiving a range of organic amendments. This application of the
125 model is at a smaller and more detailed scale than previous evaluations (Lofts et al., 2013; Xu et
126 al., 2016). We simulated the dynamics of Zn, Cu, Pb and Cd in the topsoil of a long term (>60 yr)
127 agricultural trial in Switzerland, comprising a series of plots receiving either farmyard manure,
128 sewage sludge, green waste compost, or no amendment. After optimising the model parameters
129 and driving data for the site, we then made projections of metal dynamics into the future under the
130 same agronomic practices, in order to assess their sustainability in terms of environmental risk.

131 2. Materials and Methods

132 2.1 The study site

133 The Zurich Organic Fertilization Experiment (ZOFE) is a long-term agricultural plot trial started
134 in 1949 by the Swiss Federal Agricultural Research Institute (Agroscope) at Zurich-Reckenholz,
135 Switzerland, to compare different fertilization schemes in an 8-year crop rotation: 1) winter
136 wheat/intercrop, 2) maize, 3) potato, 4) winter wheat/intercrop, 5) maize, 6) summer barley, 7)
137 clover grass ley, 8) clover grass ley (Oberholzer et al., 2014). Ploughing has been carried out to a
138 depth of at least 20 cm, from north to south and *vice versa*, alternating the direction of adjacent
139 passes (Figure 1). The site is located at 420 m a.s.l., the mean annual precipitation is 1054 mm,
140 and the mean annual temperature is 9.4°C. The soil is a carbonate-free, loamy (14% clay) Luvisol
141 (IUSS, 2006), with a SOC content of 1.43% w/w and a pH (H₂O) value of 6.5 prior to the
142 experiment. The field trial consists of twelve treatments replicated in five blocks of dimensions
143 7m × 5m in a systematic design. The same cultivation and plant protection regimes have been
144 applied to all the treatments. In the present study we investigated the following four treatments:
145 control (NIL#1) with no fertilization and no amendment, farmyard manure (FYM#2) with
146 application of 5 t OM ha⁻¹ every second year, sewage sludge (SS#3) with application of 2.5 t OM
147 ha⁻¹ every year, and green waste compost (COM#4) with application of 2.5 t OM ha⁻¹ every year
148 (Figure 1).

149

150 2.2 Trace element time series

151 The NIL, FYM, SS and COM soils (top 20cm, sieved to 2mm) were sampled from the Agroscope
152 ZOFE soil archive and analysed for total and EDTA-extractable concentrations of Zn, Cu, Pb and
153 Cd. Analysed soils were sampled from the years 1972, 1979, 1982, 1991, 1995, 2000, 2003, 2007

154 and 2011. Before 2011, samples from the five replicate plots per treatment had been bulked, so
155 variability among replicate plots could not be assessed. To determine the total and EDTA-
156 extractable soil TE concentrations respectively, soils were digested in aqua regia and extracted
157 using EDTA, and analysed by ICP-OES (Dv sequential Perkin Elmer Optima 2000). The EDTA-
158 extractable pools were obtained with the extraction protocol described by (Quevauviller, 1998).
159 The total metal concentrations were compared with the one-point-in-time measurements from the
160 same plots carried out with ICP-MS from an independent laboratory, so that interferences of As
161 with Cd in the readings (McBride, 2011) were ruled out as shown in Figure S1 of the Supporting
162 Information for Cd. Quality control of the ICP-OES was done every ten readings on the calibration
163 curve by measuring the TES concentrations in the blank samples and in the standard sample at
164 concentration of 1 ppm. The limits of quantification of total and EDTA-extractable concentrations
165 for each TE are given in Table S1 (Supporting Information).

166 Samples of farmyard manure from 2011 and 2014, sewage sludge from 2008 and 2012 and
167 compost from 2011, 2013 and 2014 were also analysed for total and EDTA-extractable
168 concentrations of Zn, Cu, Pb and Cd as described above.

169

170 **2.3 The Intermediate Dynamic Model for Metals with lateral mixing**

171 A detailed description of the Intermediate Dynamic Model for Metals (IDMM) is given in the
172 Supporting Information; here we present an overview of the most relevant concepts. The IDMM
173 simulates annual concentrations of total and geochemically active metals within a topsoil, and
174 metal fluxes from the soil due to porewater leaching and crop uptake. It distinguishes between a
175 pool of geochemically active (labile) TEs, comprising dissolved and adsorbed forms, and a non–

176 labile (aged) pool that comprises chemically less reactive and occluded forms (Figure 2A). The
177 labile metal pool is partitioned into dissolved and adsorbed forms assuming chemical equilibrium.
178 A Freundlich-type isotherm (Groenenberg et al., 2010) describes the relationship between free and
179 adsorbed TE ions, and the relationship between free TE ions and TEs complexes in the porewater
180 is computed using WHAM/Model VI (Tipping, 1998). Transformations between labile and aged
181 pools follow first-order kinetics. Two pools of aged metal are defined, termed ‘weakly aged’ and
182 ‘strongly aged’. The weakly aged pool has relatively rapid, reversible transformation kinetics with
183 the labile pool. Weakly aged metal may transfer into the strongly aged pool, and strongly aged
184 metal may transfer back into the labile pool. Both these transformations have relatively slow
185 kinetics compared to the transformations occurring between the labile and weakly aged pools.

186 The IDMM is driven by annual TE input rates, which we assume to be entirely in labile form.
187 Simulations start from a past year (1750 in this case), in which all metal inputs are assumed to be
188 natural, and where the soil is in steady state, i.e. metal input and output fluxes balance (Tipping,
189 1998). Metals can be removed from the soil due to leaching of the dissolved form in drainage
190 water, erosion of the soil, and uptake into the harvestable parts of crops. In this study erosion was
191 neglected in consideration of the site geomorphological characteristics (i.e. the site has negligible
192 slope). Since the soil samples were relevant to the homogenised ploughing depth, the soil was
193 modelled as a single well-mixed layer of 20 cm.

194 The work of McGrath (1987) and McGrath & Cegarra (1992) on plot experiments at Rothamsted
195 Experimental Station (U.K.) showed that lateral movement of soil among adjacent plots, due to
196 regular ploughing, exerted a significant influence on the temporal trends of metal concentrations
197 within the plough layer. Therefore, we extended the IDMM to enable the lateral transfer of metal
198 across the plots to be simulated. Figure 2B illustrates the lateral mixing model setup. The plots are

199 subdivided into a number of strips, within which the soil composition is physicochemically
200 homogeneous. Each year, a proportion of the topsoil, and its associated TEs, in each strip is
201 exchanged with the adjacent strip. Mean soil TE concentrations in each plot are computed as the
202 sum of the TE mass in each strip, divided by the sum of the soil mass in each strip. The number of
203 strips per plot, and the amount of soil exchanging, can be varied. Figure 2B shows an example
204 setup with five strips of 1m width and an exchange width of 0.2m, equal to the plough depth. For
205 simulations, we ordered the plots NIL-FYM-SS-COM. We fixed the exchange width to this value
206 for all simulations. A sensitivity analysis was carried out on the number of strips per plot, in order
207 to understand the impact of the choice of strip number on the predicted TE trends.

208

209 **2.4 Metal input rate estimation**

210 Besides TE inputs from the organic amendments, TE inputs to the plots were assumed to comprise
211 geogenic and anthropogenic deposition from the atmosphere, and mineral weathering from the
212 coarse (>2mm) fraction of soil. Estimation and optimisation of the input rates, as based on
213 available measurements and data from literature, are described in detail in the Supporting
214 Information. For the sewage sludge amendment only, two approaches were used to estimate the
215 metal inputs. The first approach, termed ‘Swiss Trend’, was derived from literature trends specific
216 to Switzerland; considering the unsatisfactory simulation results, the metal input rates were
217 adjusted to the ZOFÉ plots and this approach was termed ‘Idealised Trend’ (more details in the
218 Supporting Information).

219

220 **2.5 Model parameterisation**

221 The IDMM requires time series inputs of annual soil porewater pH, SOM, DOC fluxes in
222 porewater, and annual drainage volume, in order to simulate soil TE concentrations. For computing
223 TE losses by crop uptake and removal, annual crop yields and TE concentrations in the harvestable
224 parts of crops are required.

225 Soil porewater pH trends for 1949–2014 were obtained by firstly converting measurements in
226 aqueous extracts to porewater pH values, according to the formula provided by de Vries et al.
227 (2008) (Figure S7 in the Supporting Information). The trends in pH were then smoothed to reduce
228 the effect of noise on the simulations; a five-year period, four pass averaging approach was used
229 for smoothing. For future projections, we ran simulations assuming either (i) no change in pH from
230 2014 to 2100, or (ii) an exponential decline in pH from 2014 onwards to reach a constant pH 0.5
231 units below the 2014 value.

232 Prior to 1949, SOM was assumed to be constant and set to the mean of the first measurement in
233 each plot. Fitting of SOM observations before 2014 and projections to 2100 were done by applying
234 the two-pool SOM model ICBM (Andrén et al., 2004). Briefly, after fitting the NIL treatment with
235 the coefficients obtained by Menichetti et al. (2016) for ZOFÉ, the humification coefficient was
236 calibrated to fit the other amended treatments. Each plot was divided into five strips and the soil
237 lateral mixing approach described above was incorporated into the modelling. The resulting trends
238 are shown in Figure S7 (Supporting Information). The SOC was converted to SOM by assuming
239 the latter to be 50% C by mass.

240 No porewater dissolved organic carbon (DOC) concentration or flux data were available for ZOFÉ.
241 The total carbon respiration flux from 1949 onwards, obtained from the ICBM modelling, was
242 assumed to be proportional to the DOC loss flux. The annual DOC flux prior to the start of the
243 experiment was estimated, and DOC fluxes during the experiment were then estimated by scaling.

244 As the site was grassland prior to the experiment, we collated published data on DOC fluxes from
245 improved grasslands (Fu et al., 2019; Kindler et al., 2011; Buckingham et al., 2008), and derived
246 a median flux of $8.1 \text{ g m}^{-2} \text{ a}^{-1}$ for scaling. The computed flux trends were divided by the annual
247 drainage to obtain DOC concentration trends (Figure 3). The annual drainage was assumed to be
248 constant, and calculated by averaging the difference between rainfall measurements at local
249 stations and evapotranspiration estimated with a locally calibrated Primault equation.

250 Crop metal removal was assumed to be a function of crop biomass (Figure S7 in the Supporting
251 Information), as crop metal concentrations were assumed not to vary. Crop yields have been
252 measured in ZOFE yearly, based on the harvest from a sub-plot in each plot. Shoot biomass was
253 estimated by scaling the crop yields linearly according to the functions provided by Bolinder et al.
254 (2007). The Zn, Cu and Cd concentrations in winter wheat grains and shoots were measured at
255 harvest in 2014 and 2015 and the average values were taken to represent the respective metal
256 contents in the grains and shoots of wheat and barley, over the entire simulation period. The TE
257 concentrations for Pb, and for the other crops in the 8-year rotation, were estimated from previous
258 reports (de Vries et al., 2008; EFSA, 2009, 2010; SAEFL, 2003; SCAN, 2003a,b). Table S8
259 (Supporting Information) reports the assumed TE concentrations for all crops.

260

261 **2.6 Analysis of the soil and organic amendment FTIR and XRD**

262 NIL, FYM, SS and COM amended soil samples from 1972 and 2011 and SS sample from 2013
263 were analysed by Fourier Transform Infrared Spectroscopy (FTIR) to detect any change in time of
264 the soil organic fraction between the treatments. The DRIFT spectra were obtained using a rapid-
265 scan Spectrum-GX (Perkin Elmer, Monza, Italy) Fourier transform infrared spectrometer (FTIR)

266 in the mid-infrared spectral range (4000 to 450 cm^{-1}). The spectrometer was equipped with a
267 Peltier-cooled deuterated triglycine sulphate (DTGS) detector and an extended range KBr beam
268 splitter. Soil samples of 50 mg were placed in a stainless steel sample cup, located in a Perkin
269 Elmer diffuse reflectance accessory and scanned for 60 s. A silicon carbide (SiC) reference disk
270 was used as the background sample (Perkin-Elmer). The most noticeable peaks were attributed
271 according to D'Acqui et al. (2015) and Niemeyer et al. (1992), as reported in the Supporting
272 Information. The same soil samples were also analysed by X Ray Diffraction (XRD) to investigate
273 the soil mineral fraction, as also described in the Supporting Information.

274

275 **2.7 Projections of the long-term effects of organic amendment applications**

276 After evaluating and calibrating the IDMM for the period 1949-2014, the model was run in
277 predictive mode from 1750 to 2100 to derive projections of the long term influence of organic
278 amendment application on topsoil TE concentrations. For the period 2015–2100, the following
279 modelling assumptions were made: *i*) stable annual temperature and topsoil drainage; *ii*) constant
280 rates of TE input rates *via* all sources, at the 2014 rates, including the ‘Idealised Trend’ inputs *via*
281 sewage sludge addition; *iii*) constant crop yields at the 2014 values. Future SOM, DOC flux and
282 pH trends were generated as previously described.

283 The projected labile TE concentrations were compared against ecotoxicological critical limits
284 calculated for each metal according to the methodology of Lofts et al. (2004). This method assumes
285 that the free metal ion concentration is the most appropriate indicator of toxicity, combined with a
286 protective effect of the soil porewater pH. The resulting critical limit functions, expressed as labile

287 metal concentrations and functions of soil pH and SOM, aim to be protective of 95% of soil
288 species. The critical limit functions are reported in the Supporting Information.

289

290 **2.8 Statistical analysis**

291 All statistical analyses were carried out in R (version 3.5.0). The Mann-Kendall test (package
292 “Kendall”) was used to assess monotonic trends in the TE time series. Increasing trends (Kendall’s
293 tau statistic > 0) and decreasing trends (Kendall’s tau statistic < 0) were considered significant
294 when the two-sided *P*-value was less than 0.05. The *bias* function and the “dplyr” package were
295 used to calculate the bias and the root mean squared error, respectively, of the simulated labile
296 concentrations versus the observed data.

297 3. Results and Discussion

298 3.1 TE measurements in soil and organic amendments

299 Despite the continuous application of organic amendments, total TE concentrations (Figure 4)
300 showed no significant ($P < 0.05$) accumulation patterns over time in the topsoil according to the
301 Mann-Kendall trend test, except for the Pb concentration in the NIL treatment topsoil, which
302 increased significantly due to atmospheric deposition. While the total concentrations were
303 basically stable, the trends were decreasing for Zn in the SS treatment and for Cu in all the
304 treatments, with Cu displaying large decreases, from 60–102 mg kg⁻¹ in 1972 to 30–57 mg kg⁻¹ in
305 1995. The concentrations measured in 1972 were clearly elevated when compared to Ballabio et
306 al. (2018), in which an average total Cu concentration of ca.17 mg kg⁻¹ was reported from more
307 than 21,000 topsoils of EU countries. The observed higher Cu concentration in 1972 could be
308 ascribed to past anthropogenic contamination, such as application of Cu-based fungicides. The
309 reasons for the extensive loss of total Cu are unknown, as the rate of loss (5-8 mg kg⁻¹ in 35 years)
310 was larger than expected by leaching, which is of the order of 0.2-0.3 mg kg⁻¹ per year (Vulkan
311 et al., 2000 For comparison, the temporal trend of the total P concentration in the topsoil (Figure
312 4) exhibited lower temporal variability than the TE concentrations. Phosphorus showed significant
313 accumulation over time in the FYM and SS treatments, but not in the NIL and COM treatments.

314 Increases in the EDTA-extractable concentrations (Table 1) were significant for Zn and Cd in the
315 NIL and FYM treatments, and for Cu in the NIL treatment ($P < 0.05$). Conversely, in the SS
316 treatment all the metals showed steady declines in their EDTA-extractable concentrations over the
317 measurement period. Observed metal lability, calculated as the ratio of the EDTA-extractable
318 concentration and total concentration in the same year, is reported in Figure S8.

319 The total TE concentrations of the organic amendments are reported in Table 2. The farmyard
320 manure and compost samples had comparable levels of total TE concentrations, with the green
321 waste compost presenting higher Pb enrichments. The sewage sludge had higher total TE
322 concentrations than the compost and farmyard manure, explaining why the magnitude of all total
323 TE concentrations ranked in the order NIL < COM = FYM < SS in 2011. The analysed sewage
324 sludge showed also the highest variability of the TE lability (Figure S8), with Zn lability varying
325 from 0.39 to 0.15 and Cu from 0.48 to 0.22 in the samples from 2008 and 2012. The lower lability
326 of Cu and Pb in the organic amendments than in the topsoil could be ascribed to their stronger
327 affinity for organic matter (McBride et al., 1999).

328

329 **3.2 Simulations of the metal trends**

330 The IDMM model was run from 1750 (pristine conditions) to simulate the observed metal
331 concentrations data between up to 2014. The observed and simulated total concentrations are
332 shown for reference in the Supporting Information (Figure S9). Here we focus firstly on the
333 predicted labile metal concentrations, and briefly discuss the modelling of the total metal
334 concentrations.

335 From a mechanistic point of view, the IDMM does not directly simulate the total soil metal
336 concentration, but explicitly considers separate labile, weakly aged and strongly aged forms, each
337 with their own dynamics. The IDMM sets the initial labile concentration assuming a steady state
338 balance of natural input and output metal fluxes, including kinetic transfers to and from the
339 nonlabile pool. In a dynamic simulation, changes in the labile concentration are then driven by
340 changes in inputs and the soil parameters influencing solid-solution partitioning and rate of net

341 aging (pH, soil organic matter, DOC flux). The initial aged concentrations are set by a combination
342 of equilibrium with the initial labile concentration, coupled with the adjust of one kinetic constant
343 (see Supporting Information for details). Changes in the aging concentration are then driven by
344 aging transformations of the labile form. Transformations between the labile and weakly aged
345 forms are sufficiently rapid that the weakly aged form responds rapidly to changes in the labile
346 concentration, but the slower transformations involving the strongly aged pool result in a highly
347 delayed response to changes in the labile pool. An example of predicted trends in the three pools
348 is shown in Figure S10 (Supporting Information).

349 Using five strips of dimension 1m × 7m to simulate each plot when considering lateral mixing, the
350 IDMM gave the predictions shown in Figure 5 when using the ‘Swiss Trend’ and ‘Idealised Trend’
351 inputs, both with and without lateral mixing.

352 Fitting of apparent model weathering rates was done for each scenario. Only small differences
353 were seen among the fitted rates, so here we focus only on the rates obtained when assuming lateral
354 mixing and using the ‘Idealised Trend’. Fitted apparent weathering rates were 9.28, 1.62, 0.0798
355 and 0.914 mg m⁻² a⁻¹ for Cu, Zn, Cd and Pb respectively. These are generally higher than the
356 natural atmospheric deposition rates of 0.0699, 0.0752, 0.0109 and 0.522 mg m⁻² a⁻¹. Clearly the
357 contribution of the apparent weathering rate is not negligible relative to other natural inputs.
358 Estimates of topsoil metal weathering inputs are uncommon in the literature, but Imseng and co-
359 workers (Imseng et al., 2018, 2019) have published estimates of Zn and Cd weathering at Swiss
360 grassland locations. Generally, the weathering rates they estimate (0.001–0.5 mg m⁻² a⁻¹ for Zn,
361 <0.01–0.39 mg m⁻² a⁻¹ for Cd) are lower than our fitted Zn rate but comparable for Cd. It is
362 reasonable to assume our fitted rates to be apparent weathering rates only, subject to a number of
363 uncertainties, particularly the lack of knowledge regarding anthropogenic metal additions to the

364 field prior to setting up the ZOFE experiment. This is particularly clear for copper, for which the
365 apparent rate looks unrealistically high and is likely to be compensating for unquantified past
366 inputs, as per what is known of the site history. Nonetheless, the fitting exercise is useful in
367 allowing us both to calibrate model predictions to observations, and to emphasise where the key
368 gaps in knowledge exist, that should be tackled to allow more comprehensive and plausible future
369 applications of the IDMM.

370 The ‘Swiss Trend’ of metal inputs predicted neither the magnitude nor the trends in labile metal
371 concentrations in the SS plot with or without lateral mixing enabled, with the partial exception of
372 Cd (Figure 5). The downward trend in labile metal concentrations in SS cannot be explained due
373 to metal removal in leaching and/or crop uptake using the key variables (porewater pH and DOC
374 concentration, metal contents of harvestable crops). Simulations were therefore repeated with
375 lateral mixing and optimisation of the inputs to SS (the ‘Idealised Trend’). The optimised inputs
376 (total inputs from 1949 to 2014) were factors of 3.2, 2.3, 1.5 and 6.6 times higher than the
377 literature-derived inputs. With lateral mixing invoked, the model was reasonably successful in
378 describing the downwards observed trends in labile metal in the SS plot, although there was a
379 tendency to overestimate post-1990 observed concentrations, particularly for Pb. To illustrate the
380 influence of lateral mixing on predictions, we also ran a simulation using the ‘Idealised Trend’
381 inputs without lateral mixing. The results suggest that lateral mixing due to ploughing has been a
382 key determinant of the observed metal concentrations in SS. In the absence of lateral mixing,
383 higher labile metal concentrations are consistently predicted from 1949 onwards (Figure 5), and
384 the observations are consistently overestimated. For example, in 2011, the observed labile metal
385 concentrations were 17.6, 18.5, 0.36 and 12.8 mg/kg for Cu, Zn, Cd and Pb respectively. The
386 model predictions when invoking lateral mixing were 21, 20, 0.46 and 25.2 mg/kg respectively,

387 which are reasonable agreement with the exception of Pb. By contrast, the predicted concentrations
388 when lateral mixing was not invoked were 40.6, 42.2, 0.87 and 49.7 mg/kg respectively, which are
389 all approximately double the concentrations predicted in the presence of lateral mixing. These
390 results are consistent with the lateral distribution of metals in ploughed experimental plots at
391 Rothamsted Experimental Station previously (McGrath, 1987). This influence of ploughing on
392 metal distribution is, however, likely to be confined to the specific management conditions of
393 experimental plots such as ZOFÉ, and not to be relevant for modelling commercial agricultural
394 systems at field and higher scales.

395 Observed labile metal concentrations in NIL, FYM and COM are consistently lower than those in
396 SS due to the lower rates of input. The IDMM generally reproduces the magnitude and trends in
397 concentrations reasonably well in these plots, but tends to slightly underestimate concentrations,
398 particularly in the FYM and COM plots. In the absence of lateral mixing, there is negligible
399 distinction between the ‘Swiss Trend’ and ‘Idealised Trend’ for these plots. Invoking lateral
400 mixing increases the predicted labile concentrations somewhat due to the predicted net transfer of
401 metal from the SS plot to the other plots. This results in an overestimation of the observed labile
402 metal in FYM and COM. The effect of lateral mixing on the predictions for NIL is smaller, since
403 it is not immediately adjacent to SS. There are marginal improvements to the predictions,
404 particularly for Cu and to an extent Zn, when lateral mixing is invoked and the ‘Idealised Trend’
405 SS input trend is applied.

406 Model performance with soil lateral mixing under both the ‘Swiss Trend’ and the ‘Idealized
407 Trend’ is reported in Table 3: the model adequately reproduced the EDTA-extractable
408 concentrations when the input rates were optimised (‘Idealized Trend’), resulting in a global

409 average RMSE of 2.7 and an average bias (overestimation) of -1.66, -2.18, -4.34 and -0.05 mg kg-
410 1 for Zn, Cu, Pb and Cd, respectively.

411 Total metal simulations are shown in Figure S9. Generally, the model predictions are similar in
412 trend to the prediction of labile metal, with increases after 1949 due to ZOFE inputs, and in SS, a
413 decline from the late 1970s onwards as a result of lateral mixing. Trends in total zinc are well
414 reproduced, although there is a bias to low predictions in SS and marginally high predictions in
415 COM. For Pb, particularly in FYM, SS and COM, temporal trends in observed concentrations are
416 unclear. For example, there is no clear trend to decreasing concentrations in SS, as predicted by
417 modelling. Generally, concentrations in FYM, SS and COM are overestimated, while the trend in
418 NIL is well reproduced. Observed cadmium concentrations also show appreciable noise; a clear
419 (declining) trend is seen only in SS. Modelled cadmium concentrations are almost consistently
420 biased high, although the trend in SS is reproduced. Inspection of the fit for the NIL plot suggests
421 that the initial total Cd may be overestimated as a result of the fitting approach, as the mean of the
422 first three observations is greater than the subsequent measurements. Total copper exhibits a
423 distinctive declining trend in all the plots, as noted previously, but the reason for these observed
424 trends is unclear. The rates of decline cannot be explained by any of the metal loss processes in
425 the IDMM. The concentrations in SS, while showing the same declining trend as the other plots,
426 stabilise at higher concentrations, reflecting the higher copper input.

427

428 **3.3 Lateral mixing sensitivity analysis**

429 We investigated the sensitivity of the predictions with lateral mixing to the number of
430 homogeneous soil strips used per plot (Section 2.3), by running simulations using the ‘Idealised

431 Trend' and either two or ten soil strips per plot, maintaining a margin width of 0.2m. Figure 6
432 shows the simulated labile concentrations of Zn, Cu, Pb and Cd in 2014 across the transect. With
433 two strips per plot, the labile concentrations in SS were on average 28% lower than with five strips
434 when considering all the metals; the concentrations were comparable in FYM and COM but were
435 higher in NIL by on average 81%. With 10 strips per plot, the labile concentrations in SS were on
436 average 40% higher than with five strips, but lower in NIL, FYM and COM. Therefore, increasing
437 the number of strips per plot reduced the predicted redistribution of the TEs from the SS to the
438 adjacent plots. The choice of the number of strips per plot has clearly an effect on both the
439 determination of the 'Idealised Trend' inputs and the simulation performance. The pragmatic
440 choice of five strips for modelling provides an example of the magnitude of the effect of ploughing
441 on the redistribution of metals across the treatments, but in general the presence of the lateral
442 mixing effect does somewhat limit the usefulness of the plot data for model evaluation, by
443 requiring additional modelling and parameterisation not necessary for true field application.

444

445 **3.4 Soil spectroscopy analysis and long-term effects of organic amendment applications**

446 The IDMM model assumes that the TEs present in the organic amendments are fully labile, so they
447 are added to the labile pool when they are introduced in the soil. While metal lability observations
448 in the organic amendment samples (Figure S8) show that the metals are not entirely labile, they
449 also indicate high variability of the metal lability in the SS amendment samples from 2008 and
450 2012. Additionally, Figure S9 shows that Pb lability in the SS-amended soils decreased over time,
451 while for the other metals in the SS treatment the lability remained constant despite a significant
452 pH decrease. At least for Pb, a consistent decrease of the incoming- metal lability would improve
453 the simulations of the labile metal concentrations in the SS treatment, also under the 'Idealised

454 Trend' inputs and soil lateral mixing. To support this hypothesis, the soil mineral and organic
455 fractions of soil samples from 1972 and 2011 were investigated to detect eventual changes between
456 treatments and over time caused by the application of the organic amendments.

457 The XRD diffractograms from 1972 and 2011 (and 2013 only for SS) did not reveal noticeable
458 differences among the soil samples (Figure S11 in the Supporting Information). This would
459 indicate that the long-term application of the organic amendments, including the sewage sludge,
460 did not introduce any exogenous minerals, such as clay minerals and Fe-(oxy) hydroxides, capable
461 to modify the TE lability in the soil, with particular reference to the decrease in the SS plots.

462 The FTIR spectra from 1972 and 2011 were also not suggestive of differing organic fraction
463 composition between treatments and over time, except for a varying peak in the sewage sludge
464 samples. To focus on the time change of the soil organic fraction composition, the FTIR spectra
465 from the 2011 and 2013 SS samples were subtracted from the 1972 spectra, thus resulting in
466 differential spectra. As shown in Figure 7, the differential spectra from 1972-2011 evidenced a
467 peak at 1040 cm^{-1} , associated to the functional group of polysaccharide-like compounds, that was
468 not present in the differential spectra from 1972-2013. This means that polysaccharide-like
469 compounds, which are reported to have high affinity for TEs (Geddie & Sutherland, 1993; Veglio,
470 Beolchini, & Gasbarro, 1997), were present in the soil in 1972 and 2013, but not in 2011.

471 In conclusion, the scarce observations available are more suggestive of organic fraction variability
472 in the sewage sludge amendments applied to ZOFE rather than to any consistent (in time) metal
473 lability trend. This organic matter composition change could have an impact on the lability of the
474 incoming TEs, and hence on simulation of labile TE dynamics, but more research is needed on the
475 lability of metal in organic amendments and the response of that lability following addition to soil.

476 **3.5 Long-term effects of organic amendment applications**

477 Figure 8 presents the projected labile metal concentrations from 1940 to 2100 under the model
478 conditions described in Section 2.7. In NIL, FYM and SS, labile metal concentrations are projected
479 to be steady or declining, regardless of the pH trend. In COM, steady increases of Cu, Zn and Pb
480 are projected, but a small decline in Cd. The projected increases in COM are likely to be due at
481 least in part to the smaller decline in soil OM in this plot, relative to the other plots. Soil–porewater
482 partitioning in the IDMM is a function of soil OM, and so a temporal decline in organic matter
483 will lower soil-solution partition coefficients and increase metal leaching. The proportional
484 declines in labile metal concentration in NIL, FYM and SS are consistently in the order Cd > Zn
485 > Pb ~ Cu. This reflects the higher tendency of Zn and Cd to be lost from the soils, due to their
486 lower binding affinities for the soil OM.

487 Projecting a decline in porewater pH to 2100 produces varying results across different metals and
488 plots. In NIL, FYM and SS, Zn, Cu and Cd are consistently projected to be lower in 2100 as a
489 result of lower pH. Conversely, in COM the same metals are projected to increase slightly under
490 the declining pH trend. This is also observed for Pb in SS. In a number of further cases, notably
491 Zn in FYM and SS, Cu in SS and Pb in FYM, the labile metal concentration under declining pH
492 is projected to be higher than that under constant pH initially, but then to drop to or below the
493 constant pH projection in 2100. These observations can be rationalised by considering the interplay
494 among the metal input rate, the rate of labile metal loss in leaching, and the pH dependence of the
495 aging rate of added labile metal. In COM, the metal input rates are sufficiently high to project
496 continued net metal accumulation, given the soil chemistry, while in the other plots input rates are
497 insufficient to allow continued accumulation under the changing soil chemical conditions.
498 Increased leaching losses with declines in pH are expected to be most important for Zn and Cd.

499 For Cu, and to an extent Pb, the distribution of organic matter between solid and solution is
500 expected to drive partitioning and hence leaching losses. In all the plots except NIL, the ratio of
501 DOC flux to SOC pool is projected to increase between 2014 and 2100. This trend will have a
502 positive influence on metal leaching by driving an increase in the proportion of metal complexed
503 with DOC in soil porewaters, relative to the proportion adsorbed to the soil solids. In contrast to
504 the other factors, declining pH will reduce the rate at which input metal is projected to age and so
505 drive greater retention within the labile pool. This latter process is projected to be particularly
506 important for Pb, but can also be seen influence projections of the other metals, particularly in
507 COM.

508 Figure 9 clearly shows that the largest projected risks occur in SS, driven by the relatively high
509 metal inputs. In particular, it is notable that for all metals exceedance of the critical limit is
510 predicted to have occurred prior to 2014 and to be maintained until 2100. The risk characterisation
511 ratios (RCRs; the ratio of modelled to critical labile concentration) for Zn are projected to exceed
512 11 around the present day and then to decline under projected constant pH, but to remain relatively
513 high (~9) in 2100. A similar pattern is seen for Cd, but the RCRs are smaller, approaching ~1.2
514 around the present day and declining to ~0.7 in 2100. In contrast, RCRs for Cu and Pb are predicted
515 to continuously increase from 1949 to 2100, although the rate of increase is projected to be
516 relatively low under the assumptions of future conditions.

517 The declining future pH trend results in higher projected risks for Pb in SS after 2014. The labile
518 Pb concentration is projected to be lower after 2014 in the declining pH scenario (Figure 8).
519 Therefore, the increased risk must be due to the lowering of the critical limit concentration due to
520 the decline in pH. Similar considerations can explain the slightly higher risk predicted for Zn and
521 Cd, but the interplay among the driving factors appears more complex, since the projected labile

522 metals are not initially lower in the declining pH scenario (Figure 8). For Cu, the decline in pH has
523 a relatively small influence on the critical limit concentration and so only marginal differences in
524 risk are projected.

525 Predicted risks to 2100 in NIL, FYM and COM are small and only limited to Zn; only marginal
526 differences between the constant and declining pH scenarios are seen. Notably, exceedance for Zn
527 is projected to occur in the future (around 2030), but RCR remains relatively small ($RCR < 1.5$ in
528 2100).

529

530 4 Conclusions

- 531 • We applied the dynamic soil metal model IDMM to simulate long term soil metal trends
532 in the Zurich Organic Fertiliser Experiment trial, Switzerland, using metal chemistry
533 parameters derived from independent datasets;
- 534 • Following calibration of inputs and account for lateral soil mixing, the IDMM reproduced
535 the observed EDTA-extractable concentrations of Zn, Cu, Pb and Cd with average
536 overestimation by -1.66, -2.18, -4.34 and -0.05 mg kg⁻¹ for Zn, Cu, Pb and Cd, respectively.
- 537 • Considerable amounts of data are required to run the model. In particular, we used
538 modelling to fit observations and estimate future trends in the soil OM content, and to
539 estimate DOC fluxes from the topsoil;
- 540 • Estimating historic TE inputs as robustly as possible is important for the modelling. Where
541 estimated inputs give biased predictions of soil TEs, optimisation may be useful within
542 plausible limits if it aids in evaluation of the model;
- 543 • Mineral weathering inputs of TEs to soils are key in supporting the modelling. Knowledge
544 of metal weathering rates is generally poor, however. Optimisation of weathering rates is
545 possible, but must be interpreted cautiously. More research is required on robust
546 determination and modelling of weathering rates in the field;
- 547 • Under the specific conditions of the ZOFÉ experiment, lateral mixing of soil due to
548 ploughing appears to be a significant influence on the observed metal concentrations, due
549 to net redistribution of TEs from plots with relatively high concentrations to those with
550 relatively low concentrations;

- 551 • Changes in the soil chemistry of the plots since the inception of the ZOFÉ experiment,
552 notably acidification and loss of soil OM, make the soils more vulnerable to the ecological
553 impacts of metals;
- 554 • Projections of metal concentrations, including under future conditions of constant climate
555 and metal inputs, suggest that historic inputs of sewage sludge would result in present day
556 exceedances of threshold concentrations (critical limits) for all the TEs and that
557 exceedances would remain until at least 2100. Some minor exceedance of Cu and Zn
558 critical limits would be expected by 2100 under manure and compost application.
- 559 • The application of the IDMM model to the ZOFÉ trial is a promising step towards
560 understanding the key processes controlling past, current and future TE dynamics at field
561 scale;
- 562 • While long term trials such as ZOFÉ have the advantage of being well characterized, their
563 soil metal concentrations can be influenced by their distinctive management practice, as
564 this example shows;
- 565 • This study has identified some key knowledge gaps that need to be addressed for large
566 scale model application: characterization metal weathering rates and DOC fluxes, and the
567 metal contents of organic amendments;
- 568 • Because of the complexity of the model data requirements and likely resulting uncertainties
569 in predictions, large scale application is likely to be most useful in assessing broad spatial
570 and temporal trends in metal concentrations and risks.

571

572

573 **Author Contribution**

574 CC and SL run the model and wrote the ms. RS, ST, RG and BC run the soil experiments and LD
575 the FTIR and XRD analysis. GR and SO set up the research. All the co-authors revised the ms.

576

577 **Acknowledgements**

578 We thank Lucie Gunst from Agroscope for making possible the resampling of soils and
579 amendments from the archive. Also, we gratefully acknowledge Dr. Susanna Pucci and Dr. Luisa
580 Andrenelli who carried out the TE concentration analysis, Mr. Alessandro Dodero for sample
581 preparation for spectroscopic analysis and Dr. Alessandra Bonetti for FTIR analysis. We are
582 indebted to two anonymous reviewers for providing insightful comments. This research has been
583 partially funded by the Doctoral Program of the DISPAA of the University of Florence.

584

585 **Competing Interests**

586 The authors declare that they have no conflict of interest

587

Year	EDTA-extractable concentrations [mg kg ⁻¹]			
	Zn	Cu	Pb	Cd
NIL Treatment				
1972	3.1	6.2	7.5	0.14
1979	2.6	6.3	7.8	0.13
1982	2.9	6.1	8.8	0.08
1991	4.5	6.2	8.0	0.13
1995	4.3	6.2	8.9	0.14
2000	6.5	6.9	9.6	0.16
2003	6.4	6.7	8.4	0.17
2007	6.1	7.1	7.8	0.15
2011*	6.3	7.6	9.2	0.18
FYM Treatment				
1972	3.5	8.3	8.9	0.17
1979	5.1	10.1	4.9	0.22
1982	5.1	9.6	11.5	0.23
1991	6.6	9.0	9.9	0.23
1995	8.5	11.0	11.5	0.24
2000	8.4	10.9	11.8	0.25
2003	7.7	10.5	10.8	0.25
2007	11.2	10.5	9.7	0.24
2011*	8.4	9.7	10.2	0.25
SS Treatment				
1972	20.7	24.6	28.2	0.61
1979	26.7	26.2	29.0	0.66
1982	27.1	24.2	26.0	0.58
1991	24.3	22.7	19.8	0.50
1995	20.4	19.5	16.2	0.40
2000	19.6	20.7	16.7	0.42
2003	13.8	14.8	12.9	0.31
2007	17.5	17.3	12.2	0.33
2011*	17.6	18.5	12.8	0.36
COM Treatment				
1972	3.7	8.6	9.7	0.18
1979	4.6	9.5	10.3	0.20
1982	8.6	13.3	11.7	0.24
1991	10.3	13.4	14.3	0.29
1995	6.5	8.5	9.5	0.18
2000	6.9	10.4	11.7	0.21
2003	5.0	8.7	9.4	0.18
2007	n.a.	n.a.	n.a.	n.a.
2011*	7.3	10.0	10.8	0.28

588 *mean of five replicates

589

590 **Table 1:** Soil EDTA-extractable concentrations of Zn, Cu, Pb, Cd in ZOFÉ topsoil (0-20 cm) for

591 the treatments NIL, FYM, SS and COM.

592

593

	Total Concentrations [mg kg ⁻¹]					EDTA-extracted Concentrations [mg kg ⁻¹]			
	Farmyard Manure								
Years	Zn	Cu	Pb	Cd	P	Zn	Cu	Pb	Cd
2011	109.5	23.6	traces*	traces*	6915	32.5	5.6	traces*	traces*
2014	158.9	27.8	traces*	traces*	8146	53.1	6.7	traces*	traces*
Sewage Sludge									
2008	447.5	165.8	24.8	3.18	16110	173.1	78.8	9.1	0.26
2012	715.0	301.2	35.4	7.00	28870	105.7	66.2	8.6	0.16
Compost									
2011	130.2	40.0	37.5	traces*	2238	30.8	6.1	12.8	traces*
2013	122.9	39.7	31.9	traces*	2118	30.2	6.5	13.0	traces*
2014	124.8	43.7	65.4	traces*	2242	28.7	5.6	12.5	traces*

594

595 *lower or close to the limit of detection.

596

597

598 **Table 2:** Total and EDTA-extractable concentrations of Zn, Cu, Pb, Cd and P relative to the total
599 dry matter of the organic amendment samples from available years.

Metals	Swiss Trend		Idealised Trend		Treatments
	RMSE	Bias	RMSE	Bias	
Zn	2.14	-1.71	1.73	1.45	NIL
Zn	1.87	-0.84	3.20	-2.95	FYM
Zn	11.03	10.00	3.62	-1.42	SS
Zn	2.45	-1.59	4.05	-3.72	COM
Cu	3.83	3.82	0.48	-0.36	NIL
Cu	4.00	3.93	2.98	-2.86	FYM
Cu	10.66	9.90	3.25	-2.25	SS
Cu	4.64	4.21	3.96	-3.23	COM
Pb	8.34	8.31	1.12	-0.61	NIL
Pb	9.90	9.70	5.47	-5.13	FYM
Pb	19.96	18.90	9.09	-7.28	SS
Pb	10.81	10.69	4.59	-4.32	COM
Cd	8.07	-8.07	0.03	0.01	NIL
Cd	8.53	-8.52	0.05	-0.05	FYM
Cd	9.40	-9.36	0.10	-0.08	SS
Cd	8.59	-8.59	0.10	-0.09	COM
<i>Global Avg</i>	7.76	2.55	2.74	-2.06	
<i>Zn Avg</i>	4.37	1.47	3.15	-1.66	
<i>Cu Avg</i>	5.78	5.47	2.67	-2.18	
<i>Pb Avg</i>	12.25	11.90	5.07	-4.34	
<i>Cd Avg</i>	8.65	-8.63	0.07	-0.05	

600
601 **Table 3:** Model performance in terms of Root Mean Square Error (RMSE) and Bias of simulated
602 labile concentrations vs observed EDTA-extractable concentrations with lateral mixing under the
603 ‘Swiss Trend’ and the ‘Idealised Trend’.

604
605

606 **Figure Captions**

607 **Figure 1:** The experimental design of ZOFE with 12 treatments replicated in five blocks. Only
608 the 4 treatments highlighted in grey were investigated.

609 **Figure 2:** The IDMM model with lateral mixing. (A) The description of the model structure; (B)
610 the description of the soil lateral mixing method.

611 **Figure 3:** Estimated DOC concentration time trends in ZOFE topsoil (20 cm) for the treatments
612 NIL, FYM, SS and COM.

613 **Figure 4:** Total concentration time trends of Zn, Cu, Pb, Cd and P in ZOFE topsoil (20 cm) for
614 the treatments NIL, FYM, SS and COM.

615 **Figure 5:** Observed EDTA-extractable concentrations (bullets), simulated labile concentrations
616 with mixing (solid line) and without mixing (dashed line) with ‘Swiss Trend’ inputs (top) and
617 ‘Idealised Trend’ inputs (bottom) for Zn, Cu, Pb, Cd in ZOFE topsoil (0-20 cm) for the treatments
618 NIL, FYM, SS and COM.

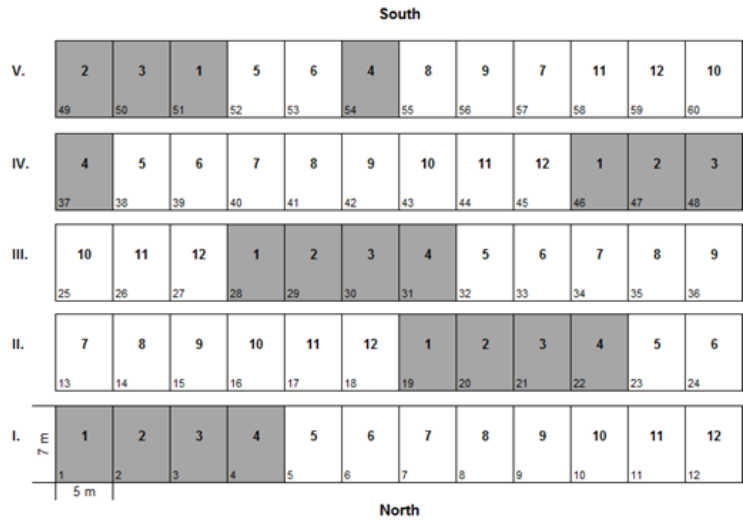
619 **Figure 6:** Simulated labile concentrations of Zn, Cu, Pb and Cd in 2014 across a transect
620 comprising the plots NIL, FYM, SS and COM in series with five and 10 strips per plot with
621 ‘Idealised Trend’ inputs.

622 **Figure 7:** Differential FTIR spectra of the SS samples: SS 2011 spectra subtracted from the 1972
623 spectra (red line), SS 2013 spectra subtracted from the 1972 spectra (blue line).

624 **Figure 8:** Projected labile concentrations for Zn, Cu, Pb and Cd to 2100 without lateral mixing
625 and with ‘Idealised Trend’ inputs. Two scenarios are considered: constant pH (solid line), pH
626 decreasing exponentially to 0.5 units below 2014 value (dotted line).

627 **Figure 9:** Projected Risk Characterisation Ratios (RCR) for Zn, Cu, Pb and Cd to 2100 without
628 lateral mixing and with 'Idealised Trend' inputs. Two scenarios are considered: constant pH (solid
629 line), pH decreasing exponentially to 0.5 units below 2014 value (dotted line).

630



- | | | |
|---|---|-----|
| 1 | = | NIL |
| 2 | = | FYM |
| 3 | = | SS |
| 4 | = | COM |

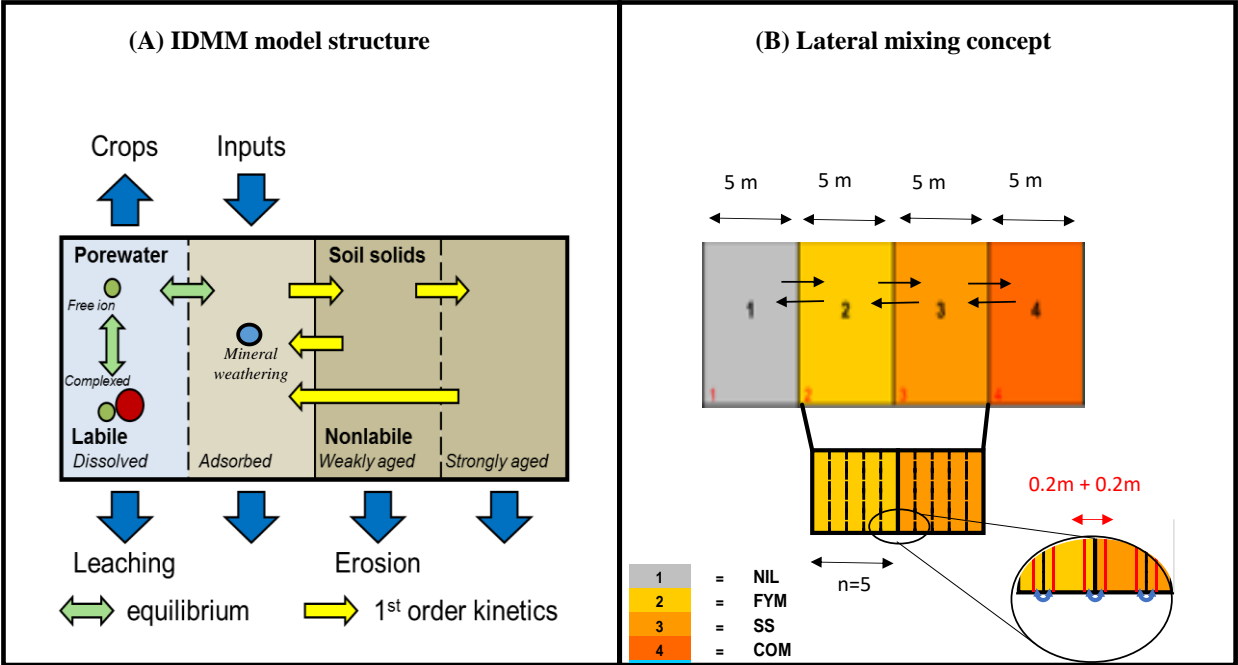
631

632 **Figure 1**

633

634

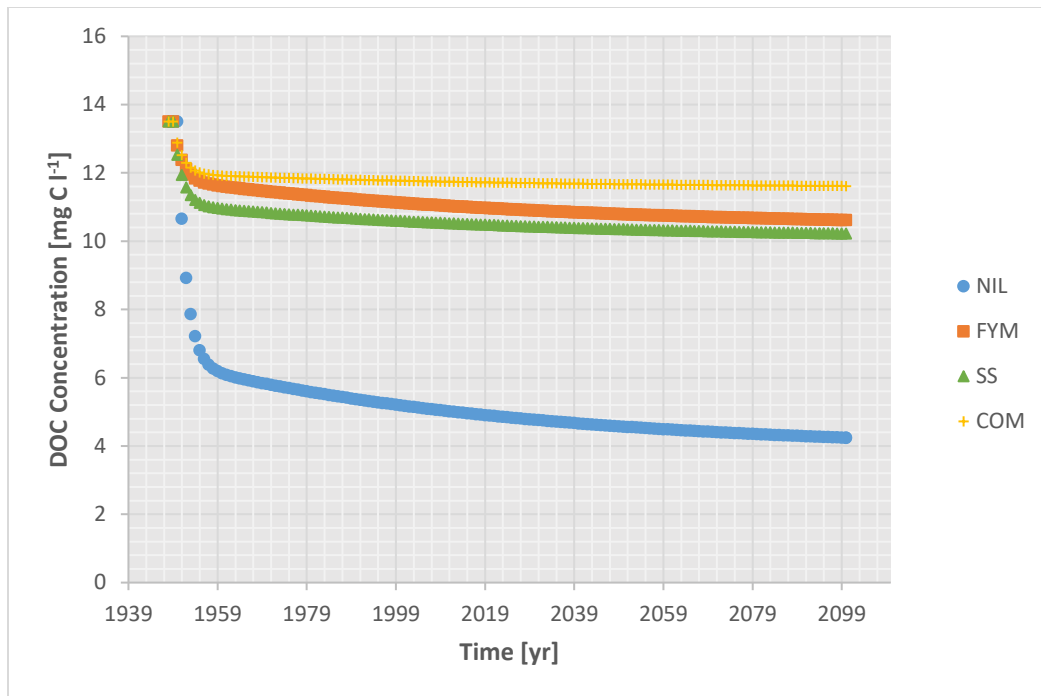
635



636

637

638 **Figure 2**



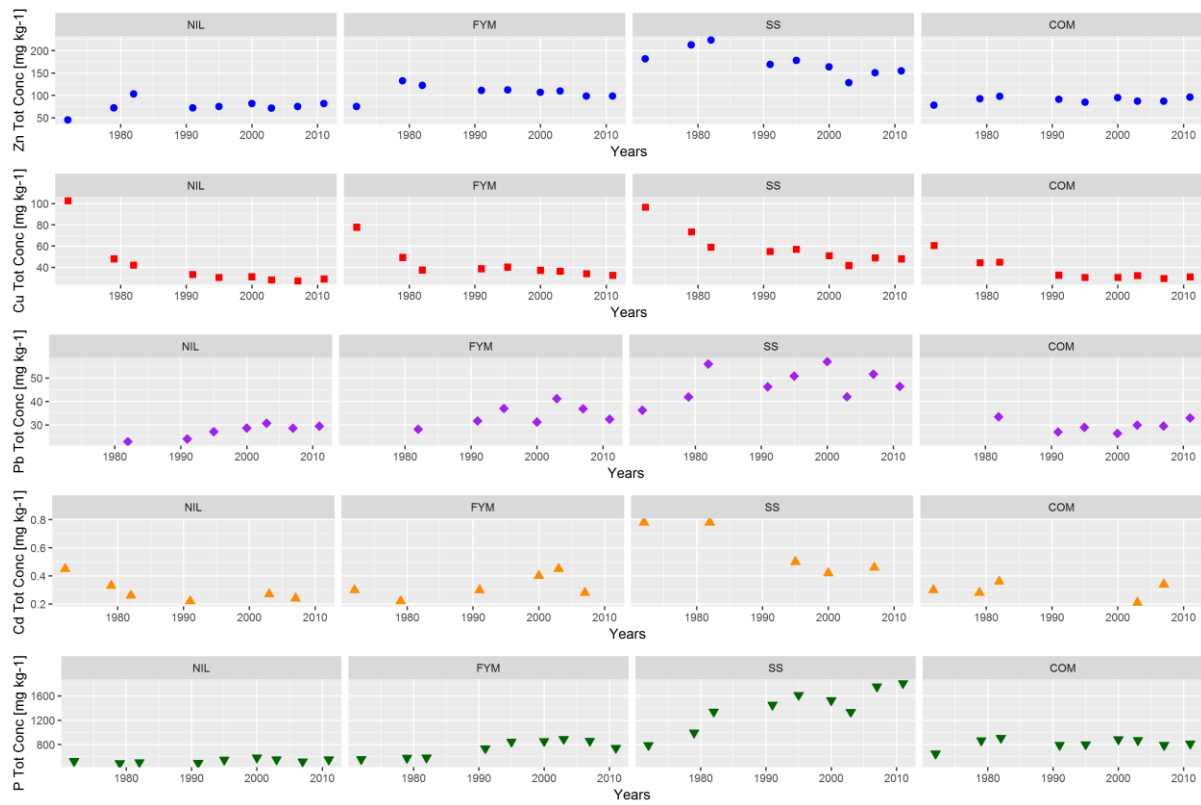
639

640 **Figure 3**

641

642

643



644

645 **Figure 4**

646

647

648

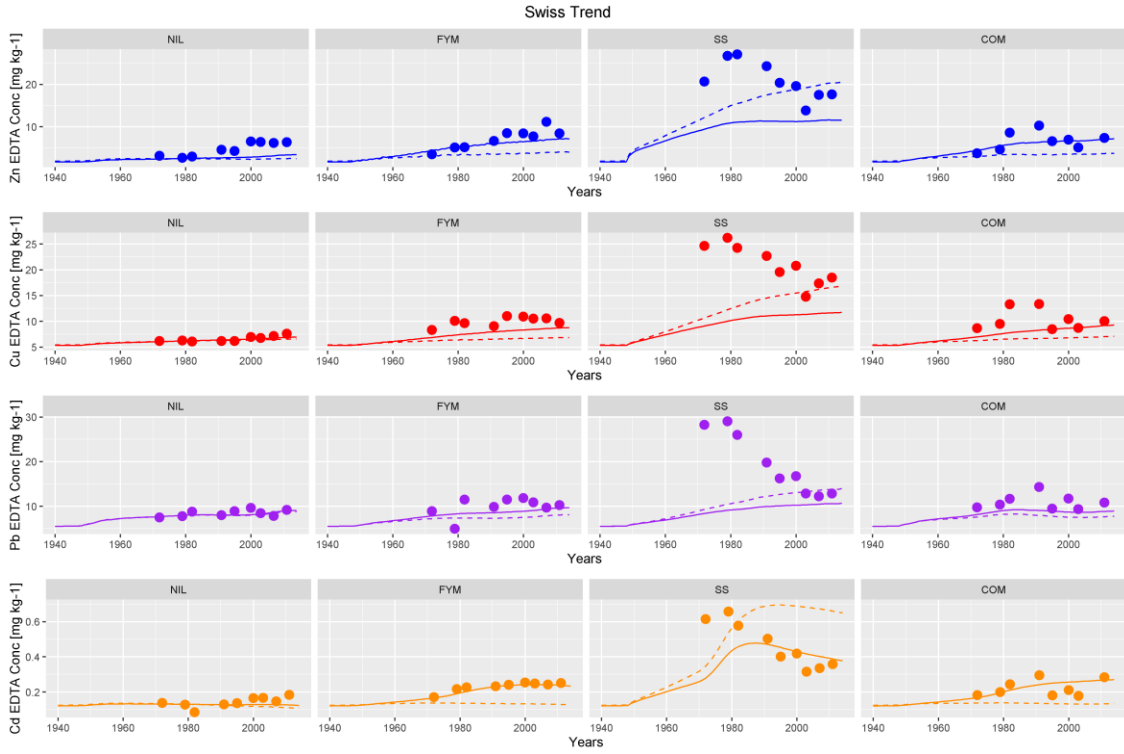
649

650

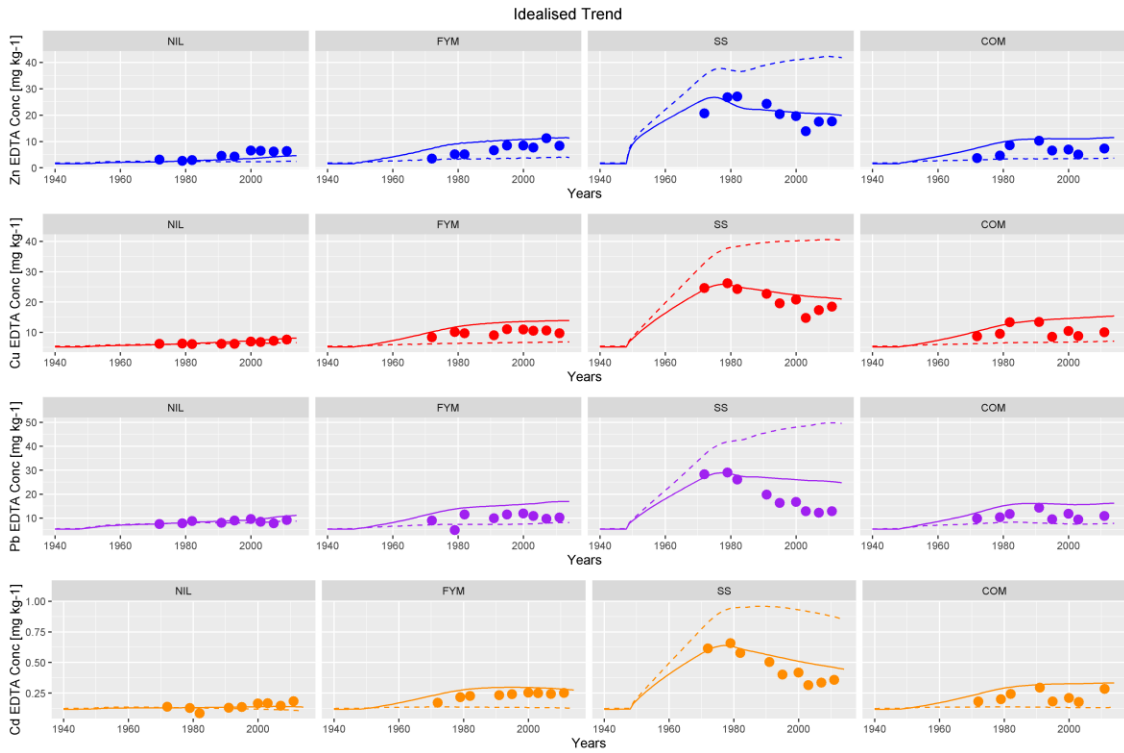
651

652

653



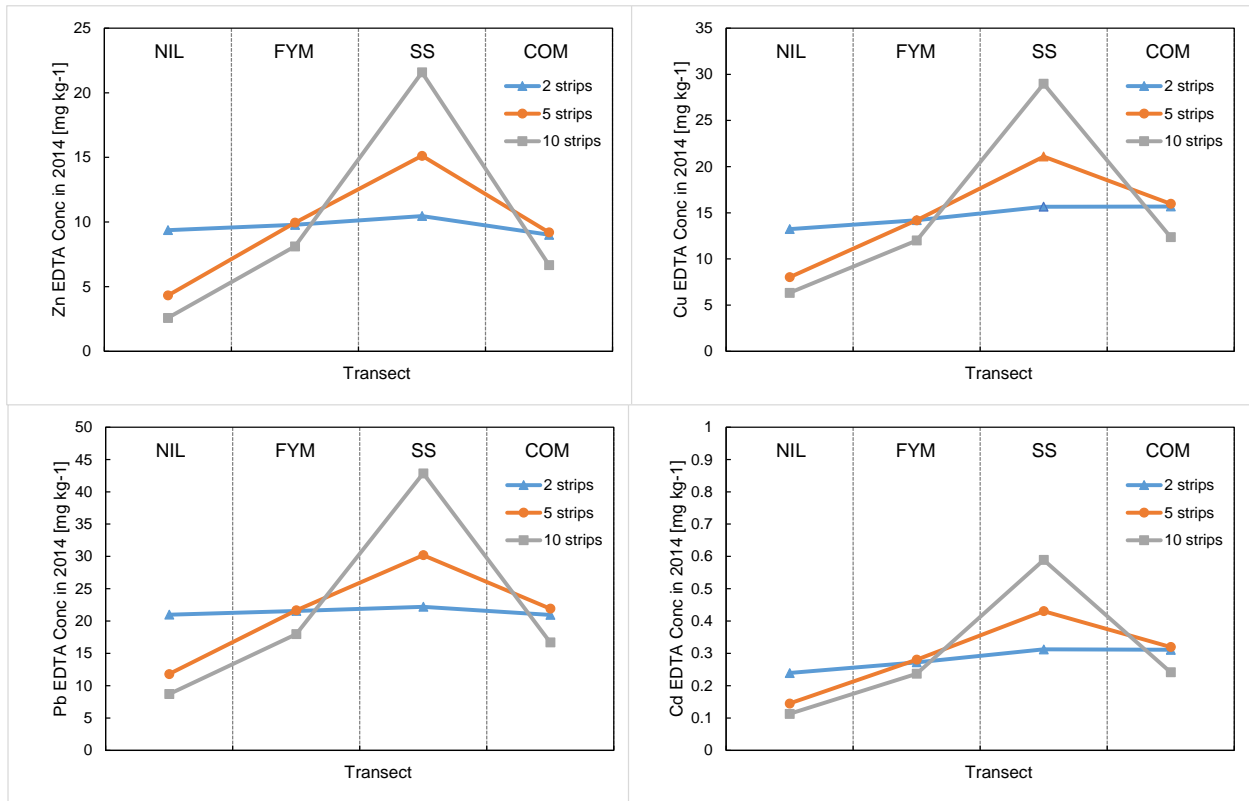
654



655

656 **Figure 5**

657



658

659 **Figure 6**

660

661

662

663

664

665

666

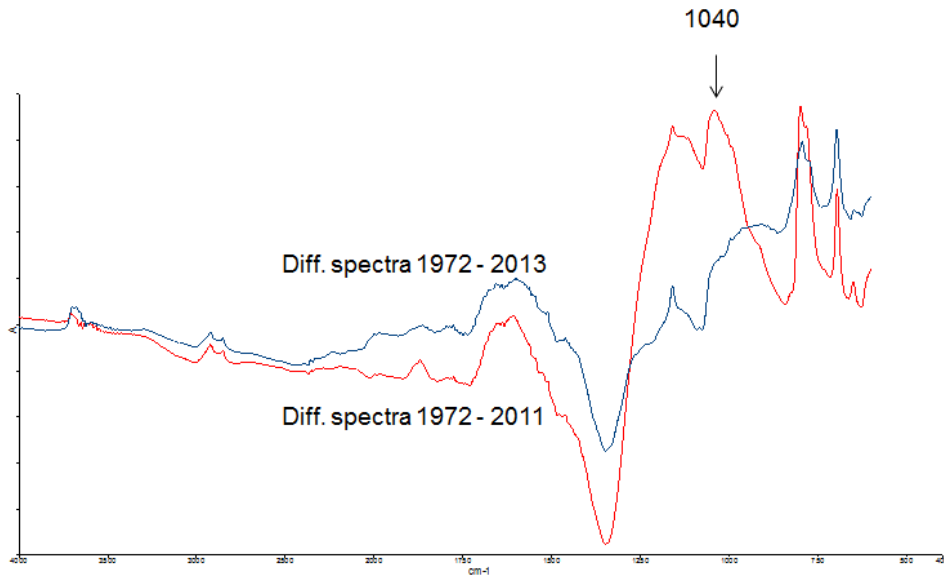
667

668

669

670

671



672

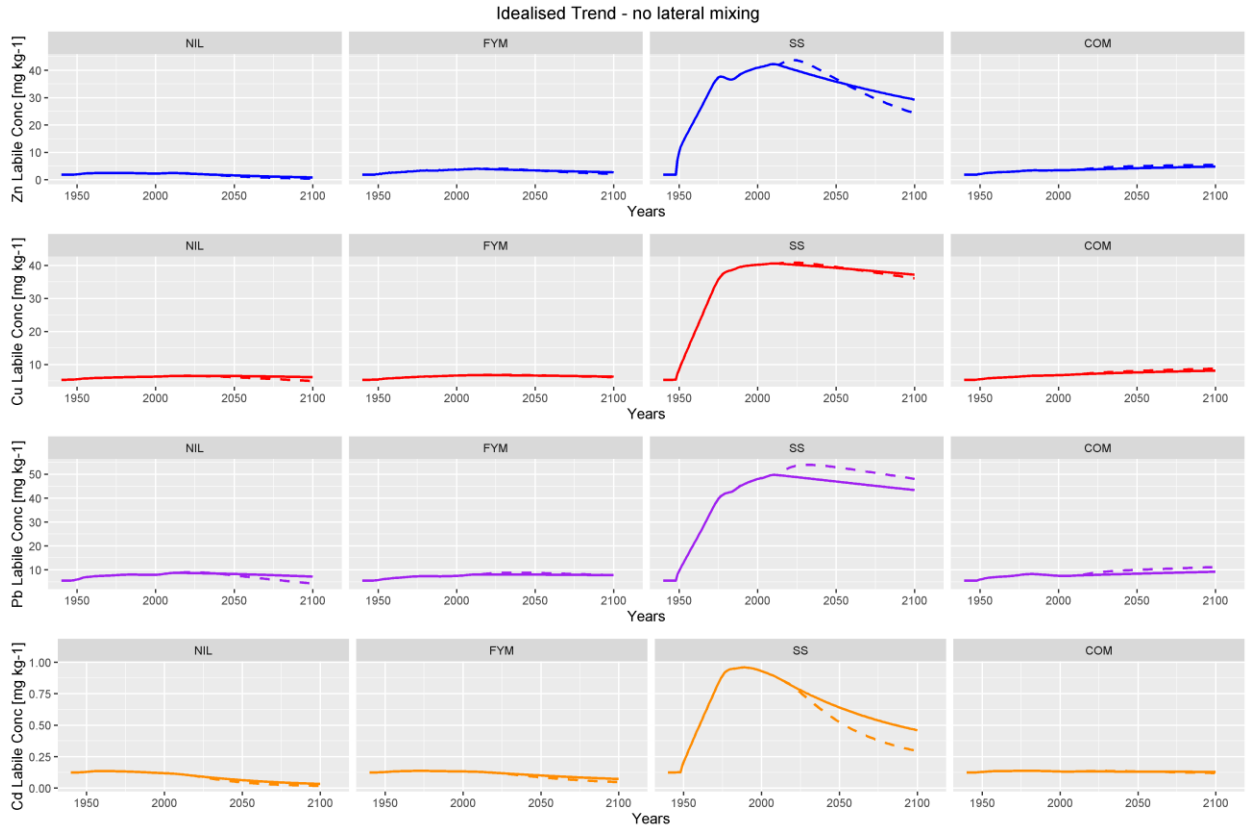
673 **Figure 7**

674

675

676

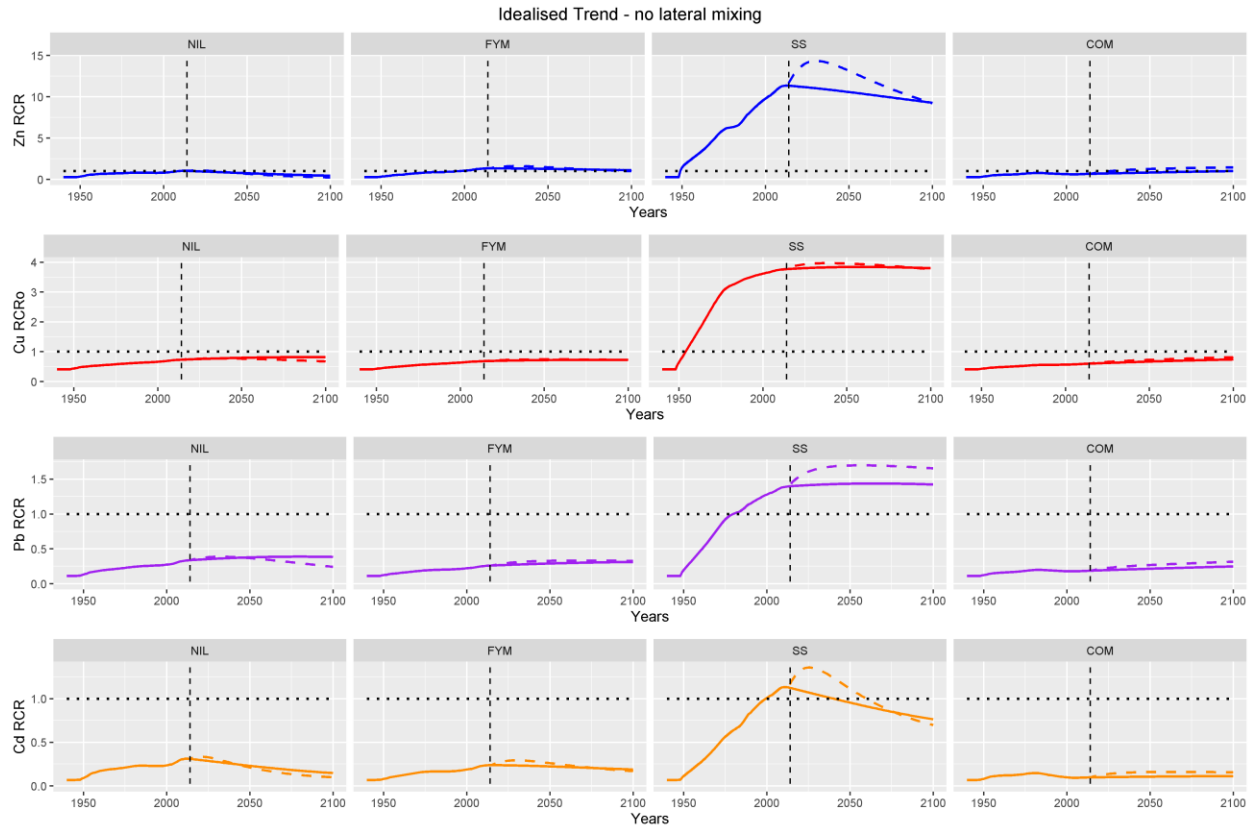
677



678

679 **Figure 8**

680



681

682 **Figure 9**

683

684 **Bibliography**

- 685 Adriano, D.C., 2005. Trace elements in terrestrial environments. Biogeochemistry, Bioavailability, and
 686 Risks of Metals. Springer, New York.
- 687 Andrén, O., Kätterer, T. & Karlsson, T. (2004). ICBM regional model for estimations of dynamics of
 688 agricultural soil carbon pools. *Nutrient Cycling in Agroecosystems* 70, 231–239 (2004).
 689 <https://doi.org/10.1023/B:FRES.0000048471.59164.ff>
- 690 Ballabio, C., Panagos, P., Lugato, E., Huang, J. H., Orgiazzi, A., Jones, A., . . . Montanarella, L. (2018). Copper
 691 distribution in European topsoils: An assessment based on LUCAS soil survey. *Science of the Total*
 692 *Environment*, 636, 282-298. doi:10.1016/j.scitotenv.2018.04.268
- 693 Bergkvist, P., & Jarvis, N. (2004). Modeling organic carbon dynamics and cadmium fate in long-term
 694 sludge-amended soil. *Journal of Environmental Quality*, 33(1), 181-191. doi:DOI
 695 10.2134/jeq2004.0181
- 696 Bolinder, M. A., Janzen, H. H., Gregorich, E. G., Angers, D. A., & VandenBygaart, A. J. (2007). An approach
 697 for estimating net primary productivity and annual carbon inputs to soil for common agricultural
 698 crops in Canada. *Agriculture Ecosystems & Environment*, 118(1-4), 29-42.
 699 doi:10.1016/j.agee.2006.05.013
- 700 Bonten, L. T. C., Groenenberg, J. E., Meesenburg, H., & de Vries, W. (2011). Using advanced surface
 701 complexation models for modelling soil chemistry under forests: Solling forest, Germany.
 702 *Environmental Pollution*, 159(10), 2831-2839. doi:10.1016/j.envpol.2011.05.002
- 703 Buckingham S., Tipping E., Hamilton-Taylor J. (2008). Concentrations and fluxes of dissolved organic
 704 carbon in UK topsoils. *Sci total Envir.*, 407(1):460-70
- 705 D'Acqui, L. P., Santi, C. A., Vizza, F., & Certini, G. (2015). Living and dead soil organic matter under different
 706 land uses on a Mediterranean island. *European Journal of Soil Science*, 66(2), 298-310.
 707 doi:10.1111/ejss.12219
- 708 de Vries, W., Romkens, P. F. A. M., & Bonten, L. T. C. (2008). Spatially explicit integrated risk assessment
 709 of present soil concentrations of cadmium, lead, copper and zinc in the Netherlands. *Water Air*
 710 *and Soil Pollution*, 191(1-4), 199-215. doi:10.1007/s11270-008-9617-z
- 711 Diacono, M., & Montemurro, F. (2010). Long-term effects of organic amendments on soil fertility. A
 712 review. *Agronomy for Sustainable Development*, 30(2), 401-422. doi:10.1051/agro/2009040
- 713 EFSA. 2009. Scientific Opinion of the Panel on Contaminants in the Food Chain on a request from the
 714 European Commission on cadmium in food. *The EFSA Journal* (2009) 980, 1-139.
- 715 EFSA Panel on Contaminants in the Food Chain (CONTAM); Scientific Opinion on Lead in Food. *EFSA*
 716 *Journal* 2010; 8(4):1570. [151 pp.]. doi:10.2903/j.efsa.2010.1570. Available online:
 717 www.efsa.europa.eu
- 718 Fu, J., Gasche, R., Wang, N., Lu, H. Y., Butterbach-Bahl, K., & Kiese, R. (2019). Dissolved organic carbon
 719 leaching from montane grasslands under contrasting climate, soil and management conditions.
 720 *Biogeochemistry*, 145(1-2), 47-61. doi:10.1007/s10533-019-00589-y
- 721 Geddie, J. L., & Sutherland, I. W. (1993). Uptake of Metals by Bacterial Polysaccharides. *Journal of Applied*
 722 *Bacteriology*, 74(4), 467-472. doi:DOI 10.1111/j.1365-2672.1993.tb05155.x
- 723 Groenenberg, J. E., Romkens, P. F. A. M., Comans, R. N. J., Luster, J., Pampura, T., Shotbolt, L., . . . de Vries,
 724 W. (2010). Transfer functions for solid-solution partitioning of cadmium, copper, nickel, lead and
 725 zinc in soils: derivation of relationships for free metal ion activities and validation with
 726 independent data. *European Journal of Soil Science*, 61(1), 58-73. doi:10.1111/j.1365-
 727 2389.2009.01201.x

728 Gu, X. Y., & Evans, L. J. (2008). Surface complexation modelling of Cd(II), Cu(II), Ni(II), Pb(II) and Zn(II)
729 adsorption onto kaolinite. *Geochimica Et Cosmochimica Acta*, 72(2), 267-276.
730 doi:10.1016/j.gca.2007.09.032

731 Hudcova, H., Vymazal, J., & Rozkosny, M. (2019). Present restrictions of sewage sludge application in
732 agriculture within the European Union. *Soil and Water Research*, 14(2), 104-120.
733 doi:10.17221/36/2018-Swr

734 Imseng, M., Wiggerhauser, M., Keller, A., Muller, M., Rehkamper, M., Murphy, K., . . . Bigalke, M. (2018).
735 Fate of Cd in Agricultural Soils: A Stable Isotope Approach to Anthropogenic Impact, Soil
736 Formation, and Soil-Plant Cycling. *Environmental Science & Technology*, 52(4), 1919-1928.
737 doi:10.1021/acs.est.7b05439

738 Imseng, M., Wiggerhauser, M., Muller, M., Keller, A., Frossard, E., Wilcke, W., & Bigalke, M. (2019). The
739 Fate of Zn in Agricultural Soils: A Stable Isotope Approach to Anthropogenic Impact, Soil
740 Formation, and Soil-Plant Cycling. *Environmental Science & Technology*, 53(8), 4140-4149.
741 doi:10.1021/acs.est.8b03675

742 Ingwersen, J., & Streck, T. (2006). Modeling the environmental fate of cadmium in a large wastewater
743 irrigation area. *Journal of Environmental Quality*, 35(5), 1702-1714. doi:DOI
744 10.2134/jeq2005.0412

745 IUSS Working Group WRB. 2006. World reference base for soil resources 2006. World Soil Resources
746 Reports No. 103. FAO, Rome

747 Jarvis, N. J., Taylor, A., Larsbo, M., Etana, A., & Rosen, K. (2010). Modelling the effects of bioturbation on
748 the re-distribution of ¹³⁷Cs in an undisturbed grassland soil. *European Journal of Soil Science*,
749 61(1), 24-34. doi:10.1111/j.1365-2389.2009.01209.x

750 Kindler, R., Siemens, J., Kaiser, K., Walmsley, D. C., Bernhofer, C., Buchmann, N., . . . Kaupenjohann, M.
751 (2011). Dissolved carbon leaching from soil is a crucial component of the net ecosystem carbon
752 balance. *Global Change Biology*, 17(2), 1167-1185. doi:10.1111/j.1365-2486.2010.02282.x

753 Lofts, S., Tipping, E., Lawlor, A. J., & Shotbolt, L. (2013). An intermediate complexity dynamic model for
754 predicting accumulation of atmospherically-deposited metals (Ni, Cu, Zn, Cd, Pb) in catchment
755 soils: 1400 to present. *Environmental Pollution*, 180, 236-245. doi:10.1016/j.envpol.2013.05.030

756 Lofts, S., Spurgeon, D. J., Svendsen, C., & Tipping, E. (2004). Deriving soil critical limits for Cu, Zn, Cd, and
757 Ph: A method based on free ion concentrations. *Environmental Science & Technology*, 38(13),
758 3623-3631. doi:DOI 10.1021/es030155h

759 McBride, M. B., Richards, B. K., Steenhuis, T., & Spiers, G. (1999). Long-term leaching of trace elements in
760 a heavily sludge-amended silty clay loam soil. *Soil Science*, 164(9), 613-623. doi:Doi
761 10.1097/00010694-199909000-00001

762 McBride, M. B. (2011). A Comparison of Reliability of Soil Cadmium Determination by Standard
763 Spectrometric Methods. *Journal of Environmental Quality*, 40(6), 1863-1869.
764 doi:10.2134/jeq2011.0096

765 McGrath, S.P., 1987. Long-term studies of metal transfers following application of sewage sludge. In:
766 Pollutant transport and fate in ecosystems, Eds. Coughtrey, P.J., Martin, M.H., Unsworth, M.H..
767 Blackwell Scientific Publications, Oxford, U.K

768 McGrath, S. P., & Cegarra, J. (1992). Chemical Extractability of Heavy-Metals during and after Long-Term
769 Applications of Sewage-Sludge to Soil. *Journal of Soil Science*, 43(2), 313-321. doi:DOI
770 10.1111/j.1365-2389.1992.tb00139.x

771 McGrath, S. P., & Zhao, F. J. (2015). Concentrations of metals and metalloids in soils that have the potential
772 to lead to exceedance of maximum limit concentrations of contaminants in food and feed. *Soil
773 Use and Management*, 31, 34-45. doi:10.1111/sum.12080

774 Menichetti, L., Katterer, T., & Leifeld, J. (2016). Parametrization consequences of constraining soil organic
775 matter models by total carbon and radiocarbon using long-term field data. *Biogeosciences*,
776 13(10), 3003-3019. doi:10.5194/bg-13-3003-2016

777 NEBRA. 2015. Metals in Biosolids, Other Soil Amendments, & Fertilizers

778 Niemeyer, J., Chen, Y., & Bollag, J. M. (1992). Characterization of Humic Acids, Composts, and Peat by
779 Diffuse Reflectance Fourier-Transform Infrared-Spectroscopy. *Soil Science Society of America
780 Journal*, 56(1), 135-140. doi:DOI 10.2136/sssaj1992.03615995005600010021x

781 Oberholzer, H. R., Leifeld, J., & Mayer, J. (2014). Changes in soil carbon and crop yield over 60 years in the
782 Zurich Organic Fertilization Experiment, following land-use change from grassland to cropland.
783 *Journal of Plant Nutrition and Soil Science*, 177(5), 696-704. doi:10.1002/jpln.201300385

784 Paquin, P.R, Gorsuch, J.W., Apte, S., Batley, G.E., Bowles, K.C., Campbell, P.G.C., Delos, C.G., Di Toro, D.M.,
785 Dwyer, R.L., Galvez, F., Gensemer, R.W., Goss, G.G., Hogstrand, C., Janssen, C.R., McGeer, J.C.,
786 Naddy, R.B., Playle, R.C., Santore, R.C., Schneider, U., Stubblefield, W.A., Wood, C.M., Wu, K.B.,
787 (2002). The biotic ligand model: a historical overview. *Comparative Biochemistry and Physiology
788 Part C*, 133, 3-35.

789 Quevauviller, P. (1998). Operationally defined extraction procedures for soil and sediment analysis - I.
790 Standardization. *Trac-Trends in Analytical Chemistry*, 17(5), 289-298. doi:Doi 10.1016/S0165-
791 9936(97)00119-2

792 Roberts, T. L. (2014). Cadmium and Phosphorous Fertilizers: The Issues and the Science. *Symphos 2013 -
793 2nd International Symposium on Innovation and Technology in the Phosphate Industry*, 83, 52-59.
794 doi:10.1016/j.proeng.2014.09.012

795 SAEFL. 2003. Assessment of Risks to Health and the Environment from Cadmium in Mineral Fertilisers.
796 Technical report.

797 SCAN. 2003a. Opinion of the Scientific Committee for Animal Nutrition on the use of copper in
798 feedingstuffs. EUROPEAN COMMISSION, HEALTH & CONSUMER PROTECTION DIRECTORATE-
799 GENERAL.

800 SCAN. 2003b. Opinion of the Scientific Committee for Animal Nutrition on the use of zinc in feedingstuffs.
801 EUROPEAN COMMISSION, HEALTH & CONSUMER PROTECTION DIRECTORATE-GENERAL.

802 Six, L., & Smolders, E. (2014). Future trends in soil cadmium concentration under current cadmium fluxes
803 to European agricultural soils. *Science of the Total Environment*, 485, 319-328.
804 doi:10.1016/j.scitotenv.2014.03.109

805 Smith, P. (2016). Soil carbon sequestration and biochar as negative emission technologies. *Global Change
806 Biology*, 22(3), 1315-1324. doi:10.1111/gcb.13178

807 Tipping, E. (1998). Humic ion-binding model VI: An improved description of the interactions of protons
808 and metal ions with humic substances. *Aquatic Geochemistry*, 4(1), 3-48. doi:Doi
809 10.1023/A:1009627214459

810 Tipping, E. CHUM: a hydrochemical model for upland catchments. *J Hydrol* 1996;174: 305–30.

811 Toth, G., Hermann, T., Da Silva, M. R., & Montanarella, L. (2016). Heavy metals in agricultural soils of the
812 European Union with implications for food safety. *Environment International*, 88, 299-309.
813 doi:10.1016/j.envint.2015.12.017

814 Veglio, F., Beolchini, F., & Gasbarro, A. (1997). Biosorption of toxic metals: An equilibrium study using free
815 cells of *Arthrobacter* sp. *Process Biochemistry*, 32(2), 99-105. doi:Doi 10.1016/S0032-
816 9592(96)00047-7

817 Vulkan, R., Zhao, F. J., Barbosa-Jefferson, V., Preston, S., Paton, G. I., Tipping, E., & McGrath, S. P. (2000).
818 Copper speciation and impacts on bacterial biosensors in the pore water of copper-contaminated
819 soils. *Environmental Science & Technology*, 34(24), 5115-5121. doi:10.1021/es/0000910

820 Wan, Y. N., Jiang, B., Wei, D. P., & Ma, Y. B. (2020). Ecological criteria for zinc in Chinese soil as affected
821 by soil properties. *Ecotoxicology and Environmental Safety*, 194. doi:ARTN
822 11041810.1016/j.ecoenv.2020.110418
823 Wang, X. Q., Wei, D. P., Ma, Y. B., & McLaughlin, M. J. (2015). Derivation of Soil Ecological Criteria for
824 Copper in Chinese Soils. *Plos One*, 10(7). doi:ARTN e013394110.1371/journal.pone.0133941
825 Xu, L., Lofts, S., & Lu, Y. (2016). Terrestrial ecosystem health under long-term metal inputs: modeling and
826 risk assessment. *Ecosystem Health and Sustainability*, 2(5), e01214. doi:10.1002/ehs2.1214

827

828