# Potential short-term losses of N2O and N2 from high

# 2 concentrations of biogas digestate in arable soils

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**Abstract.** Biogas digestate (BD) is increasingly used as organic fertiliser, but has a high potential for NH<sub>3</sub> losses. Its proposed injection into soils as a counter-measure has been suggested to promote the generation of N<sub>2</sub>O, leading to a potential trade-off. Furthermore, the effect of high nutrient concentrations on N<sub>2</sub> losses as they may appear after injection of BD into soil has not yet been evaluated. Hence, we performed an incubation experiment with soil cores in a helium-oxygen atmosphere to examine the influence of soil substrate (loamy sand, clayey silt), waterfilled pore space (WFPS; 35, 55, 75%) and application rate [0, 17.6 and 35.2 mL BD per soil core (250 cm³)] on the emission of N2O, N2 and CO2 after the usage of high loads of BD. To determine the potential capacity for gaseous losses, we applied anaerobic conditions by purging with helium for the last 24 h of incubation. Immediate N<sub>2</sub>O and N<sub>2</sub> emissions as well as the N<sub>2</sub>/(N<sub>2</sub>O+N<sub>2</sub>) product ratio depended on soil type and increased with WFPS indicating a crucial role of soil gas diffusivity for the formation and emission of nitrogenous gases in agricultural soils. However, emissions did not increase with the application rate of BD. This is probably due to an inhibitory effect of the high NH<sub>4</sub><sup>+</sup> content in BD on nitrification. Our results suggest a larger potential for N<sub>2</sub>O formation immediately following BD injection in the fine-textured clayey silt compared to coarse loamy sand. By contrast, the loamy sand showed a higher potential for N2 production under anaerobic conditions. Our results suggest that short-term N losses of N<sub>2</sub>O and N<sub>2</sub> after injection may be higher than probable losses of NH<sub>3</sub> following surface application of BD.

## 1 Introduction

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Nitrous oxide (N2O) is a potent greenhouse gas (Myhre et al., 2013), with agriculture being its largest single 27 28 anthropogenic source, contributing about 4.1 Tg N<sub>2</sub>O-N yr<sup>-1</sup> or 66% of total gross anthropogenic emissions, mainly 29 as a result of mineral nitrogen (N) fertiliser and manure application (Davidson and Kanter, 2014). The generation 30 of nitrogen gas (N<sub>2</sub>) is of agronomic interest in terms of nutrient management, since such gaseous losses may imply 31 a significant loss of N from the soil/plant system (Cameron et al., 2013; Friedl et al., 2016). However, from an 32 environmental stance, N<sub>2</sub> is innocuous and, thus, the preferred type of gaseous N-loss from soil (Davidson et al., 33 2015). In general, the improvement of N use efficiency and thus the decrease of N losses in crop production are 34 paramount in the presence of challenges like food security, environmental degradation and climate change (Zhang 35 36 Digestion residues (biogas digestate, BD) from biogas plants are used as organic amendment in agriculture. But, 37 compared to undigested amendments, digestion results in an increased pH, a higher proportion of ammonium (NH<sub>4</sub><sup>+</sup>) and a narrowed carbon (C)/N ratio of BD (Möller and Müller, 2012). These altered chemical properties 38 39 may promote biochemical reactions in the soil that are responsible for the formation of gaseous N species like 40 N<sub>2</sub>O, nitric oxide (NO), N<sub>2</sub> and ammonia (NH<sub>3</sub>) (Nkoa, 2013). 41 Significant losses of N as NH<sub>3</sub> may occur within the first hours after manure application (Quakernack et al., 2012). 42 To reduce NH<sub>3</sub> losses, the application of BD by injection is recommended, but this measure can simultaneously 43 increase the potential for N<sub>2</sub>O losses compared to surface-application (Velthof and Mosquera, 2011; Wulf et al., 2002). On the one hand, high NH<sub>4</sub><sup>+</sup> concentrations in the injection band promote nitrification, consuming O<sub>2</sub> and 44 45 releasing N<sub>2</sub>O (Christensen and Rowe, 1984). On the other hand, increased amounts of C in the injection band also 46 promote respiration, additionally depleting O<sub>2</sub> supply (Dell et al., 2011). Altogether, the conditions during the

- 47 initial phase after injection of BD foster microsites favourable for microbial denitrification, which promote also
- the formation of  $N_2$  due to anaerobic conditions (Köster et al., 2015; Webb et al., 2010).
- 49 There is a wealth of biotic and abiotic processes in soils that produce N<sub>2</sub>O and N<sub>2</sub>, depending on mineral N content,
- 50 C availability as well as on temperature, most of which are enhanced by anoxic or at least suboxic conditions
- 51 (Butterbach-Bahl et al., 2013). The amounts and the relative share of N<sub>2</sub> and N<sub>2</sub>O in the overall gaseous N
- 52 emissions depend among other factors on the degree of O<sub>2</sub> restriction (Firestone and Davidson, 1989). Soil
- 53 physical and biotic factors [i.e. diffusion permitted by soil porosity in conjunction with water-filled pore space
- 54 (WFPS) as well as consumption of O<sub>2</sub> by heterotrophic respiration and nitrification] control the aerobic status of
- a soil (Ball, 2013; Maag and Vinther, 1999; Uchida et al., 2008). In general, clayey soils exhibit a lower gas
- diffusivity compared to coarse textured soils. This regularly results in higher denitrification in the former with
- 57 higher N<sub>2</sub>O emission rates, but also a higher probability for the consecutive reduction to N<sub>2</sub> (Ball, 2013; Gu et al.,
- 58 2013; Senbayram et al., 2014).
- 59 There is a general lack of knowledge about the effects of high BD concentration on gaseous N-losses as they might
- appear after injection into soils and their interactions with O<sub>2</sub> limiting factors like soil texture and WFPS, as well
- as temperature and heterotrophic respiration. Thus, we applied the helium-oxygen (He-O<sub>2</sub>) incubation technique
- 62 (Butterbach-Bahl et al., 2002) in a laboratory experiment to evaluate the effect of the factors suggested above on
- the emission of N<sub>2</sub>O and N<sub>2</sub> from different soils. Simultaneously, CO<sub>2</sub> flux was determined as an indicator for
- 64 microbial O<sub>2</sub> consumption, O<sub>2</sub> diffusion and also for the degradability of organic C applied with BD (Blagodatsky
- and Smith, 2012). We hypothesised that (1) N<sub>2</sub>O and N<sub>2</sub> emissions will increase with WFPS, (2) gaseous N losses
- 66 will also be affected by BD application rate, i.e. the hypothetical concentration of C and N resulting from injection,
- and (3) the clayey silt will induce higher gaseous N losses than the coarse loamy sand.

## 2 Material and Methods

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## 2.1 Selected soils, sampling of soil cores and biogas digestate

- 70 Two soils were selected and both were adjusted to three levels of WFPS and three quantities of BD (Table 1),
- 71 resulting in 18 factor combinations with three replicates. Temperature was increased from 2 °C during the first
- two days to 15 °C for the last three days of the incubation. Intact soil cores (diameter 7.2 cm, height 6.1 cm, volume
- 73 250 cm<sup>3</sup>) were taken with sample rings in the range from 0–0.10 m depth from two sites with different textures,
- 74 i.e. sandy loam and clayey silt. The sandy loam samples were collected from a stagnic luvisol (IUSS Working
- 75 Group WRB, 2006) located in Gülzow (North-East Germany) in the ground moraine of the Weichselian glacial
- period at 53° 48' 35" N and 12° 4' 20" E. The clayey silt samples were collected from a haplic luvisol located in
- Dornburg between the foothills and the lowlands of Central Germany at 51° 0′ 8″ N and 11° 39′ 25″ E (see Table
- 78 2 for more details on soil characteristics). After field sampling, the soil cores were dried for 48 h at 40 °C to
- 79 facilitate adjustment of WFPS.
- 80 Both sites have been cultivated with similar crop rotations used as feedstock for biogas production and have been
- 81 amended with biogas digestate for the past nine years. The crop rotation on the sandy loam consisted of maize
- 82 (Zea mays L.), rye (Secale cereale L.), sorghum (Sorghum bicolor (L.) MOENCH), winter triticale (×
- 83 Triticosecale Wittmack), ryegrass (Lolium perenne L.) and winter wheat (Triticum aestivum L.). The only

- 84 difference in the crop rotation on the clayey silt was the cultivation of sudan grass (Sorghum × drummondii)
- instead of sorghum.

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- 86 The biogas digestate used for the incubation was obtained from a biogas plant at 'Gut Dalwitz', an organic farm
- 87 in northeast Germany. The feedstock for the anaerobic fermentation in the plant consisted of 60 % maize, 20 %
- 88 solid cattle manure, 10 % dry chicken manure and 10 % rye. The digestate was analysed by LUFA Rostock,
- 89 Germany and had a pH of 8.3, 2.91% organic C, 0.16% dissolved organic C (DOC), 0.54% N and 0.27% NH<sub>4</sub>-N
- 90 in undried material with a dry matter content of 9.4%.

#### 2.2 Adjustment of WFPS and addition of N

- 92 For adjustment of WFPS, the dry and undisturbed soil cores were moistened dropwise. The respective quantities
- 93 of water were calculated based on the bulk density, an assumed particle density of 2.65 g cm<sup>-1</sup> and reduced by the
- 94 expected moisture input from subsequent addition of BD. The soil cores were then mixed with BD and finally
- 95 repacked to reach nutrient concentrations comparable to that in injection bands. The amounts of added BD were
- calculated with an assumed injection of 160 kg N ha<sup>-1</sup> into soil with row spaces of 0.15 m (narrow injection bands
- 97 with low BD concentration, LOBD) and 0.30 m (wide injection bands with high BD concentration, HIBD). These
- 98 are common ranges used by injection machinery and correspond to 17.6 and 25.3 mL BD per sample ring. After
- 99 this procedure, the soil cores were sealed with plastic lids and stored immediately at 2 °C until the beginning of
- the incubation within a week.

#### 2.3 Determination of gas fluxes

- The measurements of N<sub>2</sub>, N<sub>2</sub>O and CO<sub>2</sub> fluxes were applied following the He-O<sub>2</sub> method (Butterbach-Bahl et al.,
- 2002; Scholefield et al., 1997). Six soil cores (i.e. the repetitions of two factor combinations at a time, Table 3)
- were placed simultaneously in special gas-tight incubation vessels inside a climate chamber. Analyses were
- 105 conducted in the laboratory of the Institute for Landscape Biogeochemistry, Leibniz Centre for Agricultural
- Landscape Research (ZALF), Müncheberg, Germany. Before flux measurements, the vessels were evacuated to
- 107 0.047 bar and flushed with an artificial He/O<sub>2</sub> gas mixture (20.49 %  $O_2$ , 345.5 ppm  $CO_2$ , 359 ppb  $N_2O$ , 1863 ppb
- 108 CH<sub>4</sub>, 2.46 ppm N<sub>2</sub>, rest He) four times consecutively to remove ambient N<sub>2</sub>. Subsequently, the air temperature of
- the climate chamber was set to 2 °C and a continuous He/O<sub>2</sub> gas flow rate of 15 ml min<sup>-1</sup> was applied to the vessel
- headspace for 72 h to remove residues of  $N_2$  from soil cores by diffusion, including a restricted  $N_2$  production by
- decreased microbial activity. After this pre-incubation in the following two days the headspace concentration of
- N<sub>2</sub>O and CO<sub>2</sub> was measured once daily in the morning. To compensate for the lower precision of the detector for
- $N_2$  in relation to the detector for  $N_2O$  and  $CO_2$  (Eickenscheidt et al., 2014),  $N_2$  concentrations were measured
- consecutively three times daily in the morning. Immediately after the last measurement on the second day, the
- temperature was set to 15 °C and the measurements were continued for another two days. Finally, the He/O<sub>2</sub> gas
- mixture was substituted by pure He and, following 24 h of acclimatisation, gas measurements were carried out
- once again (Figure 1) to determine the generation of N<sub>2</sub>O and N<sub>2</sub> in a completely anaerobic soil matrix. The latter
- step is important to get a clue about the actual potential for gaseous N losses after highly concentrated BD
- application. The settings of the chromatographs for gas analyses are described in Eickenscheidt et al. (2014). Gas
- 120 fluxes were calculated according to Eq. (1):

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$$f = \frac{M \times p \times v \times dc}{R \times T \times A},$$
 (1)

where f is the flux (N<sub>2</sub> and CO<sub>2</sub>: mg m<sup>-2</sup> h<sup>-1</sup>, N<sub>2</sub>O:  $\mu$ g m<sup>-2</sup> h<sup>-1</sup>), M the molar mass in g mol<sup>-1</sup> (N<sub>2</sub>: 28, CO<sub>2</sub>: 44, N<sub>2</sub>O:

44), p the air pressure (Pa), v the air flow (L  $h^{-1}$ ), R the gas constant (8.31 J mol<sup>-1</sup> K<sup>-1</sup>), T the temperature inside the

chamber (K), A the area of the incubation vessel ( $m^2$ ), and dc the difference of gas concentrations ( $N_2$  and  $CO_2$ :

ppm, N<sub>2</sub>O: ppb) between inlet and outlet of a vessel.

To enhance the tightness against atmospheric  $N_2$  contamination, the lids of the incubation vessels were permanently purged with helium. We obtained blank values by inserting aluminium blocks into the vessels before each measurement cycle. Since this blank values were usually steady with means of 1.9 ( $1\sigma = 0.9$ ) ppm  $N_2$ , 349.6 ( $1\sigma = 11.4$ ) ppb  $N_2$ O and 353.9 ( $1\sigma = 13.5$ ) ppm  $CO_2$ , we suggest that the vessels were tight. Derived from the blank values, lowest detectable fluxes were on average 0.427 ( $1\sigma = 0.271$ ) mg  $N_2$ -N m<sup>-2</sup> h<sup>-1</sup>, 3.6 (3.1)  $\mu$ g  $N_2$ O-N m<sup>-2</sup> h<sup>-1</sup> and 0.918 (0.693) mg  $CO_2$ -C m<sup>-2</sup> h<sup>-1</sup>. For flux estimation, the blank values were subtracted from the values measured at the respective outlet. Estimated fluxes from the soil cores smaller than the respective blank fluxes of

each day were set to zero.

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#### 2.4 Soil analyses after incubation

- After incubation, the soil cores were stored at 2 °C until they were extracted with 0.1 M KCl solution (soil to extract
- 137 ratio 1:4, standardised extraction method of the commissioned laboratory at Leibniz Centre for Agricultural
- Landscape Research e. V.) and analysed for NH<sub>4</sub><sup>+</sup> and nitrate (NO<sub>3</sub><sup>-</sup>) by spectrophotometry according to DIN ISO
- 139 14256 with a continuous flow analyser 'CFA-SAN', Skalar Analytical B.V., the Netherlands and for DOC by
- combustion according to DIN ISO 10694 with an analyser 'RC 612', Leco Instruments GmbH, Germany.

# 2.5 Statistical analysis

- All statistical analyses were done using R version 3.2.3 (R Core Team, 2015) with the data of the measuring days
- under He-O<sub>2</sub> atmosphere. Data from the vessels with the factor combination of 35% WFPS and LOBD with clayey
- silt were omitted due to technical reasons during sample preparation. For the final period of pure He headspace,
- some gas concentration data could not be documented. For loamy sand, this affects all WFPS levels with LOBD
- 146 (N<sub>2</sub> and N<sub>2</sub>O), the treatment 75% WFPS with 320 kg N h<sup>-1</sup> (N<sub>2</sub>O and CO<sub>2</sub>) and for the clayey silt the treatment
- 147 35% WFPS without amendment (N<sub>2</sub>O and CO<sub>2</sub>).
- To account for repeated measurement of vessels, linear mixed effect models were applied with package 'lmerTest'
- version 2.0-33 (Kuznetsova et al., 2016) for fluxes of each gas type. The three pseudo-replicated fluxes from the
- N<sub>2</sub> measurements of each vessel were averaged for each day to obtain the same number of observations as for N<sub>2</sub>O
- and CO<sub>2</sub> fluxes. The fixed structure of models included soil type, WFPS, amount of digestate, temperature, NO<sub>3</sub>
- and DOC contents after incubation as well as the fluxes of  $N_2O$  (in the model for  $N_2$ ) and  $CO_2$  (in the models for
- 153 N<sub>2</sub>, N<sub>2</sub>O and N<sub>2</sub>/[N<sub>2</sub>+N<sub>2</sub>O] product ratio). Soil NH<sub>4</sub><sup>+</sup> was omitted since it showed high autocorrelation with the
- amount of BD applied. The individual soil cores in the vessels were set as random effect (nested within the week
- of incubation and with allowance for a variable slope of the effect each day) with regard to lack of independence
- of consecutive measurements. The model responses for  $N_2$ ,  $N_2O$  and  $CO_2$  where log transformed (ln[value + 1])

- since gas fluxes from soils usually show lognormal distributions (Kaiser et al., 1998). The function 'step' was used
- for automatic backward selection of models based on AIC (Akaike's 'An Information Criterion'). The skewness
- 159 (γ) was calculated with R package 'moments' version 0.14 (Komsta and Novomestky, 2015) to check residuals for
- normal distribution and  $|y| \le 2$  was assumed as appropriate (West et al., 1995). For mixed effects models, p-values
- of the ANOVA (type 2) were calculated based on Satterthwaite's approximation.
- 162 Cumulated gas fluxes were estimated with a bootstrap method using function 'auc.mc' of R package 'flux' version
- 163 0.3-0 (Jurasinski et al., 2014) for the R statistical software version 3.2.3 (R Core Team, 2015). In short, the fluxes
- for the period of aerobic headspace were cumulated in 100 iterations, while for each run two fluxes were omitted
- randomly. Then, the resulting data were used to calculate means and standard deviations.

# 166 3 Results

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#### 3.1 Soil NH<sub>4</sub>+, NO<sub>3</sub>- and DOC contents

- 168 The calculated application of NH<sub>4</sub><sup>+</sup>-N from BD per kg soil approximated for the sandy loam 135.8 mg (LOBD)
- and 271.5 mg (HIBD), and for the clayey silt 126.7 mg (LOBD) and 253.4 mg (HIBD). The NO<sub>3</sub> content of BD
- was negligible. In general, the NH<sub>4</sub>+ content of the soils after incubation increased with digestate application with
- lower amounts detected in the clayey silt. Nitrate was found almost exclusively in the latter soil. (Fig. 2).
- 172 The amounts of measured DOC increased with the application rate of BD, but with higher magnitudes for the
- loamy sand than for the clayey silt (Table 4).

# 174 **3.2 CO<sub>2</sub> fluxes**

- 175 CO<sub>2</sub> fluxes showed clear differences between the soils: under all combinations of temperature and oxygen, the
- fluxes were always larger from loamy sand compared to clayey silt (Table A1). In general, mean fluxes from
- 177 loamy sand increased with the amount of digestate during each of the different periods regarding temperature and
- headspace aerobicity, but showed no obvious pattern with WFPS. There was no clear trend of fluxes with the
- amount of amendment, but a slight trend of decreasing fluxes with increasing WFPS for the clayey silt. However,
- 180 the predictive power of WFPS on CO<sub>2</sub>-C fluxes was minor since it was eliminated during stepwise regression
- 181 fitting. By contrast, soil type, amount of digestate, temperature as well as the DOC content after the incubation
- had significant (p < 0.01) effects (Table 5).

#### 3.3 N<sub>2</sub>O fluxes

- 184 The mean N<sub>2</sub>O fluxes from the loamy sand in the He-O<sub>2</sub> headspace were virtually zero, independent of temperature
- and WFPS as well as the amount of BD application (Fig. 3, Table A2). In contrast, the emissions of the clayey silt
- increased with temperature and were highest at 15 °C with intermediate WFPS and amount of BD, i.e. 6.2 mg
- 187 N<sub>2</sub>O-N m<sup>-2</sup> h<sup>-1</sup> at 55% with LOBD, respectively. Surprisingly, at 15 °C, increasing the amount of BD up to HIBD
- did not increase the observed  $N_2O$  efflux; rather it decreased the efflux significantly (p < 0.05, Tuckey's HSD) at
- 189 55% and also, but not significantly, at 75% WFPS (Fig. 3, Table A2). According to the linear mixed model for

- 190 N<sub>2</sub>O fluxes in aerobic conditions, WFPS, amount of digestate, temperature, DOC content of soil after incubation
- and  $CO_2$  fluxes had significant (p < 0.001) effects on  $N_2O$  flux (Table 5).
- 192 Under anaerobic headspace conditions, the overall highest mean N<sub>2</sub>O flux was observed from the clayey silt at
- 193 35% WFPS with HIBD (11.7 mg N<sub>2</sub>O-N m<sup>-2</sup> h<sup>-1</sup>). The same soil showed a tendency of decreasing N<sub>2</sub>O fluxes with
- increasing WFPS and amendment. In the loamy sand, the pure He-atmosphere induced increasing mean N<sub>2</sub>O fluxes
- 195 (up to 1.3 mg N<sub>2</sub>O-N m<sup>-2</sup> h<sup>-1</sup>) with increasing WFPS (Fig. 3, Table A2). Thus, the anaerobic headspace induced a
- change only in the loamy sand by increasing emissions.

#### 3.4 N<sub>2</sub> fluxes

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- From the loamy sand, no or only small rates of N<sub>2</sub> were detected at both temperatures under He-O<sub>2</sub> atmosphere
- 199 (Fig. 4, Table A3). The clayer silt showed mean fluxes of up to 1.4 mg N<sub>2</sub> m<sup>-2</sup> h<sup>-1</sup> at 2 °C (all incubations with
- 200 75% WFPS) and up to 3.8 mg  $N_2$  m<sup>-2</sup> h<sup>-1</sup> at 15 °C (75% WFPS with LOBD), but no fluxes in all BD treatments
- with 35% WFPS. Put simply, temperature had a small effect on N<sub>2</sub> emissions from the sandy loam with no
- 202 consistent influence of WFPS and the amount of BD. In contrast, the clayey silt emitted increasing fluxes with
- 203 increasing temperature and WFPS. However, the application rise from LOBD up to HIBD at 15 °C resulted in
- slightly, but not significantly (p > 0.05, Tuckey's HSD) decreased fluxes (Fig. 4, Table A3). The summary of the
- linear mixed model for  $N_2$  fluxes under aerobic conditions revealed significant effects (p < 0.05) of soil type,
- WFPS, the amount of digestate, temperature, DOC content after incubation and N<sub>2</sub>O flux (Table 5).
- 207 After switching the atmosphere to pure He, the N<sub>2</sub> fluxes from the sandy loam increased more than 60-fold. In
- 208 contrast to aerobic conditions, all measured factor combinations showed mean fluxes up to 35.1 mg N<sub>2</sub> m<sup>-2</sup> h<sup>-1</sup>
- 209 (55% with 320 kg N ha<sup>-1</sup>) (Fig. 2, Day 5 in Table A3). Mean fluxes from clayey silt increased only up to 9.3 mg
- $N_2 \, \text{m}^{-2} \, \text{h}^{-1}$  in amended treatments. Thus, the loamy sand exhibited a much more intense reaction under anaerobic
- 211 headspace conditions.

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## 3.5 $N_2/(N_2 + N_2O-N)$ product ratio

- No clear trend of the product ratio of  $N_2/(N_2 + N_2O N)$  was found for incubations of the loamy sand. However,
- there was a clear distinction of the ratios for this soil under aerobic and anaerobic atmospheres: while the ratios
- were close to zero in the former, they were close to one in the latter (Fig. 5). In contrast, in the clayey silt the ratios
- 216 increased with WFPS and were affected by digestate amendment under both the aerobic and the anaerobic
- atmospheres, where the highest ratios (up to 0.8) were found in treatments without digestate and at least 55%
- WFPS. The digestate-amended treatments showed ratios around or above 0.5, with exception of the 35% WFPS
- treatments, which had ratios close to zero. According to the linear mixed model, the product ratio under aerobic
- 220 conditions was affected significantly (p < 0.01) by soil type and the amount of digestate (Table 5).

#### 4 Discussion

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223 of high NH<sub>4</sub><sup>+</sup> concentrations In the loamy sand, the higher NH<sub>4</sub>+ content measured after the incubation cycle compared to the calculated NH<sub>4</sub>+ 224 application rates may result from heterogeneity of BD itself (Andruschkewitsch et al., 2013). By contrast, the 225 226 considerable lower values after incubation in the clayey silt could be attributed to a higher fixation of NH<sub>4</sub>+ as NH<sub>3</sub> 227 by clay minerals, enhanced by the increased pH of BD (Kissel et al., 2008). 228 The overall N<sub>2</sub>O fluxes corresponded well with those from other studies with similar incubation conditions and 229 application rates of BD in terms of N ha<sup>-1</sup> (Köster et al., 2015; Senbayram et al., 2012; Severin et al., 2015). 230 However, the latter studies assumed a distribution of BD into soil by a cultivator, which implies a smaller 231 concentration of BD than we actually applied. Although we observed differences in N<sub>2</sub>O emissions between soils, soil type was not confirmed as a significant effect. Nevertheless, WFPS and temperature, which are well known 232 controls of N<sub>2</sub>O generation (Maag and Vinther, 1999), showed significant influences. Both are physical (by gas 233 234 diffusion) and biological (by increased metabolic activity and consequently increased O<sub>2</sub> consumption by respiration) drivers for O2 availability (Ball, 2013; Maag and Vinther, 1999). Accordingly, CO2 flux (resulting 235 236 from respiration of O<sub>2</sub>) generally increased with temperature and was also identified as significant by regression 237 selection. 238 The mean N<sub>2</sub> fluxes of up to 0.5 (loamy sand) and 3.8 mg N m<sup>-2</sup> h<sup>-1</sup> (clayey silt) at 15° C (Fig. 5, Table A3) were 239 considerably smaller than the mean fluxes of up to 13.0 mg m<sup>-2</sup> h<sup>-1</sup> observed by Köster et al. (2015) during the first five days of their incubation. Although the amount of BD in terms of applied N (250 kg ha<sup>-1</sup>) was comparable, 240 241 Köster et al. (2015) used a higher WFPS of 90%, which may have increased the generation of N<sub>2</sub>. In contrast to N<sub>2</sub>O emission rates, the observed N<sub>2</sub> fluxes depended not only on WFPS, but also on soil type (Table 5). This is 242 most likely due to the direct influence of soil structure on diffusivity and the resulting supply with O2 (Balaine et 243 al., 2016; Butterbach-Bahl et al., 2013). N<sub>2</sub>O flux showed also a significant effect during regression selection for 244 245 N<sub>2</sub>. N<sub>2</sub>O is the direct precursor of N<sub>2</sub> in denitrification, so the flux of the latter depends on the availability of the 246 former. However, temperature showed no significant effect. N<sub>2</sub>/(N<sub>2</sub>+N<sub>2</sub>O) product ratios were significantly determined only by soil type and WFPS: while no clear trend was 247 248 observable for the loamy sand, there was a pronounced effect in the clayey silt (Fig 4). We attribute the lack of a 249 trend in loamy sand to generally adverse conditions for the formation of N<sub>2</sub>O and N<sub>2</sub>, i.e. a sufficient supply of O<sub>2</sub> (see section 4.2). Contrary, the influence of WFPS apparently mirrored favourable conditions in the clayey silt 250 251 (Table 5). Simultaneously, with increasing WFPS, the reduction of N<sub>2</sub>O as an alternative electron acceptor under 252 reduced O2 supply accelerates (Tiedje, 1988). Accordingly, no or rather small fluxes of the investigated gaseous 253 N species were generally found in our presumably well aerated treatments with 35% WFPS. 254 In our study, one treatment (clayey silt, 55% WFPS, LOBD) showed exceptionally large mean N<sub>2</sub>O fluxes of up to 7.1 mg N m<sup>-2</sup> h<sup>-1</sup> (Fig. 3, Table A2). This could be evidence that the injection of such commonly applied amounts 255 256 of BD-N (i.e., 160 kg N ha<sup>-1</sup>) may favour much larger losses of N<sub>2</sub>O compared to an even distribution of BD in a soil surface due to larger substrate concentration in injection slits. However, with higher amendments (i.e. HIBD), 257

Increased BD application rate did not increase N2O and N2 losses probably due to inhibitory effect

we observed partially significant (p < 0.05, Tuckey's HSD) reductions of N<sub>2</sub>O and a decreasing tendency of N<sub>2</sub>

259 emissions (Table A2, Table A3). In line with this, the amount of BD showed a significant effect during the regression selection on N<sub>2</sub>O, but not on N<sub>2</sub> fluxes (Table 5). A coherent reasoning for the rather smaller emissions 260 261 of highly amended HIBD treatments might lie in an inhibitory effect of NH3 on nitrification. Accordingly, <mark>Kim et</mark> al. (2006) found a selective inhibition of NO<sub>2</sub> oxidation in the presence of 14 to 17 mg NH<sub>3</sub>-N L<sup>-1</sup>. Our calculated 262 263 application rates in the treatments with HIBD amounted to at least 253.4 mg NH<sub>4</sub><sup>+</sup>-N (kg soil)<sup>-1</sup> for the clayey silt 264 (Fig. 3) which corresponds to 13.0 mg NH<sub>3</sub>-N (kg soil)<sup>-1</sup> at 15 °C when applying the pH of the BD and assuming 265 all extractable NH<sub>4</sub>+-N to be in solution (Emerson et al., 1975). Hence, we consider this inhibitory effect as the reason for the missing increase of N<sub>2</sub>O and N<sub>2</sub>. Nevertheless, because we mixed the BD with the soil, one would 266 expect a lower NH<sub>3</sub> fixation by clay minerals in tubular injection slits in situ (Kissel et al., 2008), resulting in 267 probably lower N<sub>2</sub>O and N<sub>2</sub> fluxes from clayey soils due to an more marked inhibitory effect. 268 High NH<sub>4</sub><sup>+</sup> loads in conjunction with an increased pH favour NO<sub>2</sub><sup>-</sup> accumulation because NO<sub>2</sub><sup>-</sup> oxidising bacteria 269 270 are less resilient against high concentrations of NH<sub>3</sub> than NH<sub>3</sub> oxidising bacteria (Anthonisen et al., 1976). This 271 NO<sub>2</sub>- should have protonated then partly to toxic and unstable HNO<sub>2</sub>, which drives biological and chemical 272 production of NO and N2O for detoxification (Venterea et al., 2015). Although we did not determine NO2, we 273 suggest a dominant role of nitrifier denitrification, i.e., NO<sub>2</sub>- reduction, in the generation of N<sub>2</sub>O during our 274 experiment, especially during the anaerobic headspace conditions at the end of the incubation, resulting in the relatively small NO<sub>3</sub>- recovery in both soils. Accordingly, coupled nitrification-denitrification and bacterial 275 276 denitrification have been found to dominate the production of N<sub>2</sub>O directly after application of BD (Köster et al., 277 2011; Senbayram et al., 2009). However, N<sub>2</sub>O-N losses were clearly larger than N<sub>2</sub> losses under aerobic headspace 278 in the clayey silt. This indicates that much of the N gas loss was driven by processes other than canonical 279 denitrification. Under the above-mentioned conditions, NO-N losses may exceed N2O losses (Venterea et al., 2015), calling for taking account of NO measurements in future studies. 280 281 Supposed that 15 % of NH<sub>4</sub><sup>+</sup>-N is volatilised as NH<sub>3</sub> within the first ten hours after surface application of BD (Quakernack et al., 2012), the losses from the NH<sub>4</sub><sup>+</sup> amounts we applied would averaged to 80 mg NH<sub>3</sub>-N m<sup>-2</sup> h<sup>-1</sup> 282 283 (LOBD) and 160 mg NH<sub>3</sub>-N m<sup>-2</sup> h<sup>-1</sup> (HIBD). The actual losses of up to 11.7 mg N<sub>2</sub>O-N m<sup>-2</sup> h<sup>-1</sup> at 30 % WFPS in 284 the clayey soil (Table A2) or of up to 35.1 mg N<sub>2</sub> m<sup>-2</sup> h<sup>-1</sup> at 55 % WFPS in the sandy loam (Table A3) from our 285 HIBD treatments add up to 117 mg N<sub>2</sub>O-N and 351 mg N<sub>2</sub>, respectively, for the same period. Hence, increased 286 N<sub>2</sub>O and N<sub>2</sub> emissions following injection of BD might effectively cause higher N losses compared to a surface 287 application and deserve closer attention in future.

## 4.2 Different effects of soil diffusivity on N<sub>2</sub>O and N<sub>2</sub> fluxes

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Apparently, the tested factors affected the  $N_2O$  and  $N_2$  fluxes from both soils in a different way. A specific soil characteristic that exhibits such a fundamental control on biogeochemical processes such as denitrification is the diffusivity for  $O_2$  (Ball, 2013; Letey et al., 1980; Parkin and Tiedje, 1984), which is a main soil characteristic responsible for the appearance of anaerobic microsites. In general, diffusivity integrates the soil porosity, i.e., pore continuity and size as well as WFPS, which control both soil  $N_2O$  and  $N_2$  emissions (Balaine et al., 2016; Ball, 2013; Letey et al., 1980). Soils with a coarser texture like the loamy sand have a higher proportion of macropores and thus a higher gas diffusion compared with fine textured soils like the clayey silt we used (Groffman and Tiedje, 1991). This lets us expect conditions that are more favourable for  $N_2O$  and  $N_2$  generation in the latter due to

297 relatively poor diffusion characteristics and, thus, a smaller O2 supply. Actually, although we incubated the soils 298 at comparable levels of WFPS and BD amendments, the apparent lower diffusivity led to larger N<sub>2</sub>O and N<sub>2</sub> 299 production in the treatments with the clayey silt in relation to the loamy sand. The role of the distinct diffusivities of both soils is corroborated by our observations of the gas fluxes in anaerobic 300 301 headspace. With switching the He-O<sub>2</sub> atmosphere in the headspace to pure He, the denitrification potential can be 302 tested because anaerobicity eliminates respiration processes that use O<sub>2</sub> as electron acceptor (Parkin and Tiedje, 303 1984). We acknowledge e.g. DNRA and anammox as possible additional sources of N2O and N2 under such 304 conditions but we were not able to quantify their contribution. The anaerobic headspace induced a considerable 305 increase of N<sub>2</sub>O fluxes in the loamy sand, but not in the clayey silt. Concurrently, the N<sub>2</sub> fluxes increased in both 306 soils, but pronounced, i.e. more than 60-fold, in the sandy loam. These observed changes resulting from oxygen deprivation imply that, during the previous aerobic conditions, the diffusivity of the sandy loam was too high to 307 308 allow for a sufficient establishment of anaerobic microsites while the clayey silt ensured a moderate diffusional 309 constraint to maintain suboxic conditions. In general, only N<sub>2</sub>O fluxes from treatments with negligible fluxes 310 during the previous aerobic period increased under anaerobic conditions, including all treatments with loamy sand 311 (Fig. 3, Table A2). At the same time, there was a reduction of N<sub>2</sub>O fluxes in most clayey silt treatments. However, 312 a closer look reveals that virtually all of the latter treatments showed increased N<sub>2</sub> flux rates. Hence, there was an enhanced reduction of N<sub>2</sub>O to N<sub>2</sub>, which is reflected in the increased N<sub>2</sub>/(N<sub>2</sub> + N<sub>2</sub>O) product ratio (Fig. 5) and 313 314 points to intensified reduction of N<sub>2</sub>O due to the lack of oxygen (Parkin and Tiedje, 1984). The much larger N<sub>2</sub> 315 fluxes from the loamy sand compared to the clayey silt might have been caused as well by poor NO<sub>3</sub>- availability 316 (Fig. 2) and a high availability of C (Table 4), which promoted the reduction of N<sub>2</sub>O to N<sub>2</sub> (Senbayram et al., 317 2012). Further, we found no evidence for any shortage of substrate in the clayey silt during the subsequent anaerobic headspace conditions. However, the cumulated fluxes of both N2 and N2O amounted to a maximum 318 319 absolute loss of 9.4 ( $1\sigma = 0.3$ ) mg N per kg soil in the clayey silt with LOBD and 55% WFPS, which was 7.4 % 320 of the calculated  $NH_4^+$ -N applied with BD (Fig. 2). On the other hand, the  $N_2/(N_2+N_2O)$  product ratios increased 321 only slightly (Fig. 5) and, in contrast to the loamy sand, there were still significant N<sub>2</sub>O fluxes in the clayey silt 322 (Fig. 3). This points to still sufficient stocks of NO<sub>3</sub><sup>-</sup> in the latter (Senbayram et al., 2012). In fact, the NO<sub>3</sub><sup>-</sup> stock was greater in the clayey silt than in loamy sand after incubation (Fig. 2). Thus, we suggest that the gas fluxes 323 324 were unaffected by the change to anaerobic headspace in the clayey silt due to already low O<sub>2</sub> concentrations as a result of poor diffusivity. In conclusion, distinct gas diffusivities of both soils can be proposed as the main reason 325 for the differing N<sub>2</sub>O and N<sub>2</sub> fluxes. 326 327 In interaction with soil diffusivity, respiration affects the aerobicity of a soil matrix by concurrent consumption 328 and formation of O2 and CO2 as well. Depending on microbial C availability, respiration could be indicated by 329 DOC, though not all DOC might be readily degradable (Cook and Allan, 1992). Generally, DOC content after our incubation increased with application rate of BD (Table 4), but DOC content was always smaller in clayey silt. 330 331 This might reflect a stronger sorption of C and thus a lower availability for respiration in the clayey silt compared 332 to loamy sand (Kaiser and Guggenberger, 2000). If we compare DOC concentrations with cumulated flux rates of 333  $CO_2$  over the period of aerobic headspace, we find a good regression fit ( $R^2 = 0.91$ , p < 0.001) for both soils (Fig. 334 6) indicating a sufficient availability of C from BD for respiration and, thus, implicitly also for denitrification 335 (Reddy et al., 1982). Moreover, as increased DOC enhanced respiration (Table A1), it consequently affected O2 consumption and, thus, also the emergence of anaerobic microsites (Azam et al., 2002). Accordingly, there is also a good correlation between cumulated  $CO_2$  and  $N_2O + N_2$  fluxes for the same period from the clayey silt ( $R^2 = 0.93$ , p = 0.001), when the treatments with 35 % WFPS (which showed virtually no N emissions) are omitted (Fig. 7). However, there was no such a correlation for the loamy sand. This confirms the interactive effect of diffusivity (induced by both the soils and WFPS) and C availability on the emissions of  $N_2O$  and  $N_2$ , which, nevertheless, interacted with the inhibitory effect of high  $NH_4^+$  loads on nitrification (see chapter 4.1).

#### 5 Relevance and implications

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Our aim was to estimate the effect of differing soil environmental conditions on gaseous N losses - and not to draw conclusions about the long-term dynamics of N<sub>2</sub> and N<sub>2</sub>O emissions after BD application in concentrations similar to injection. In another laboratory study at a WFPS of 65%, Senbayram et al. (2009) measured only one peak within two days without a repeated increase later, regardless the amount of applied BD. Thus, we assume a single peak shortly after application holds also true for our incubation. We assume also the measurements after only 24 hours of anaerobicity in the headspace to be representative for the emission potential since Wang et al. (2011; 2013) showed in similar studies that the emission of N<sub>2</sub> and N<sub>2</sub>O peaked within less than 24 hours after switching their headspace from aerobic to anaerobic conditions. In summary, as hypothesised, N<sub>2</sub>O and N<sub>2</sub> emissions as well as the N<sub>2</sub>/(N<sub>2</sub>O+N<sub>2</sub>) product ratio increased with WFPS, most probably due to restricted supply of O<sub>2</sub>. Contrary to our second hypothesis, the gaseous losses of N<sub>2</sub>O and N<sub>2</sub> did not increase with the application rate of BD. This indicates an inhibitory effect of high NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup> concentrations on nitrification, which are found typically in BD. At the same time, the N<sub>2</sub>/(N<sub>2</sub>O+N<sub>2</sub>) product ratio tended to decrease with application rate as supposed, probably due to a copious supply with NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup> from oxidised BD-NH<sub>4</sub><sup>+</sup>. Confirming our third hypothesis, the fine textured clayey silt induced larger gaseous N losses and a higher N<sub>2</sub>/(N<sub>2</sub>O+N<sub>2</sub>) ratio than the coarse loamy sand by the apparent distinct diffusivities of both soils. Overall, there was a larger potential for formation of N<sub>2</sub>O in the fine-textured clayey silt compared to the coarse loamy sand after the application of high concentrations of BD as they may appear after injection. However, the loamy sand showed a large potential for N<sub>2</sub> formation under anaerobic headspace conditions. Since coupled nitrification-denitrification N losses from injected BD seem to be massive in your study, the shortterm emissions of N<sub>2</sub>O and N<sub>2</sub> after injection appear to offset the reduced NH<sub>3</sub>-N losses that would have arose hypothetically from surface application. Further investigations are needed in regarding the dynamics and the

#### 6 Data availability

The data underlying the figures are accessible publicly via the Supplement.

duration of the observed effects and their reliability for field conditions.

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Table 1: The examined factors soil texture, water-filled pore space (WFPS), and amount (i.e., concentration) of nitrogen (N) applied with biogas digestate (BD) with their respective levels applied in the present study, resulting in 18 treatments with three replicates each. The temperature was manipulated consecutively during the incubation.

Factor [n]	Levels				
Soil texture [2]	loamy sand		clayey silt		
WFPS (%) [3]	35 55		75		
BD-N (kg ha <sup>-1</sup> ) [3]	0 16		320		
Temperature (°C) [2]	2		15		

Table 2: Characteristics of both soils. Texture and mean values with standard deviations (in brackets) for carbon (C, n = 9), nitrogen (N, n = 9), pH (n = 3), bulk density (BD, n = 3) and mineral N (NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup>, n = 3) of both soils in 0–10 cm depth after field sampling.

Texture	C (mg g <sup>-1</sup> ) <sup>a</sup>	N (mg g <sup>-1</sup> ) <sup>a</sup>	$pH^b$	Bulk density (g cm <sup>-3</sup> ) <sup>c</sup>	NO <sub>3</sub> - (mg kg-1) <sup>d</sup>	NH <sub>4</sub> + (mg kg <sup>-1</sup> ) <sup>d</sup>
Loamy sand	6.99 (0.29)	0.67 (0.05)	7.2 (0.1)	1.4 (0.0)	1.0 (0.2)	0.6 (0.3)
Clayey silt	10.77 (0.28)	1.19 (0.06)	7.2 (0.0)	1.5 (0.0)	1.8 (0.2)	0.3 (0.2)

<sup>527</sup> a measured with analyser "Truspec CNS", Leco Instruments GmbH, Germany, performed according to ISO 10694 ("elemental analysis") for C and according to ISO 13878

<sup>528 (&</sup>quot;elemental analysis") for N

b measured in H<sub>2</sub>O with TitraMaster85, Radiometer Analytical SAS, France, performed according to VDLUFA Methodenbuch, Vol. 1, chap. 5.1.1

<sup>530</sup> c measured on 250 cm<sup>3</sup> soil cores

d measured with analyser "CFA-SAN", Skalar Analytical B.V., the Netherlands, performed according to ISO 14256

Table 3: Chronological order of the incubated factor combinations. Two different factor combinations with their respective repetitions (n = 3) were placed together for each weekly incubation course (cf. Fig. 1). The factors were combined by (1) soil (loamy sand: LS, clayey silt: CS), (2) amount (kg) of applied N from digestate per ha and (3) WFPS (%).

Week	Factor combination 1	Factor combination 2
1	LS - 0 N - 35%	LS - 0 N - 55%
2	LS - 0 N - 75%	LS - 160 N - 35%
3	LS - 160 N - 55%	LS - 160 N - 75%
4	LS - 320 N - 35%	LS - 320 N - 55%
5	LS - 320 N - 75%	CS - 0 N - 35%
6	CS - 0 N - 55%	CS - 0 N - 75%
7	CS - 160 N - 35%	CS - 160 N - 55%
8	CS - 160 N - 75%	CS - 320 N - 35%
9	CS - 320 N - 55%	CS - 320 N - 75%

Table 4: Mean DOC values from soils, measured after incubation with standard deviations in brackets for the respective treatments differing in amount of applied biogas digestate (BD) and water-filled pore space (WFPS).

kg digestate-N ha <sup>-1</sup>	WFPS (%)	mg DOC (kg soil) <sup>-1</sup>		
kg digestate-iv na	W115 (%)	Loamy sand	Clayey silt	
	35	41.4 (2.7)	18.9 (1.1)	
0	55	38.6 (3.1)	19.8 (1.4)	
	75	43.7 (1.4)	19.0 (1.8)	
	35	197.4 (20.7)	n.a.	
160	55	190.5 (19.3)	68.3 (12.7)	
	75	362.2 (40.0)	63.2 (9.6)	
	35	316.8 (25.3)	358.1 (26.3)	
320	55	312.5 (14.3)	94.8 (13.6)	
	75	500.1 (33.4)	105.9 (14.8)	

n.a.: data not available

5	4	4	

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	Fixed effects								
Response	Soil type	WFPS	Digestate amount	Temperature	NO <sub>3</sub> - post	DOC post	N <sub>2</sub> O flux	CO <sub>2</sub> flux	
$\overline{N_2}$	0.026	< 0.001	0.008	0.037	<b>†</b>	0.001	< 0.001	†	
$N_2O$	†	< 0.001	< 0.001	< 0.001	†	< 0.001	*	< 0.001	
$N_2/(N_2+N_2O)$	0.005	0.004	†	†	†	†	*	†	
$CO_2$	< 0.001	†	< 0.001	< 0.001	†	0.007	†	*	

<sup>†</sup> Variable eliminated during stepwise regression selection

<sup>\*</sup> Variable was not included into original regression

Figure 1: Course of incubation and gas measurements with respect to atmosphere and temperature of the headspace after two days of pre-incubation at 2  $^{\circ}$ C in He/O<sub>2</sub> gas mixture. Gas concentrations of the headspace were determined on five consecutive days, i.e. Monday to Friday in the morning. After the first two measurement days, the headspace temperature was increased from 2 to 15  $^{\circ}$ C. Additionally, after the fourth measurement day, the aerobic Helium/oxygen gas mixture in the headspace was replaced by a pure Helium atmosphere.

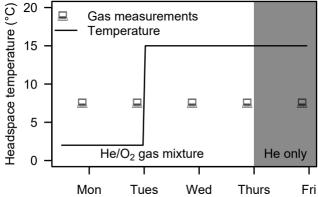


Fig. 2: Ammonium and nitrate contents from loamy sand and clayey silt after incubation with different water-filled pore spaces (WFPS, %) and amounts of digestate (0 mL per sample ring: 'No BD', 17.6 mL: 'Low BD' and 35.2 mL: 'High BD'). Error bars denote standard deviations. In general, the ammonium content increased with digestate application with lower amounts detected in the clayey silt. Nitrate was found almost exclusively in the latter soil. For comparison, inverted triangles show calculated amounts of applied ammonium, which may differ from actual rates due to heterogeneity of biogas digestate. One treatment (\*) was omitted from all analyses due to technical reasons.

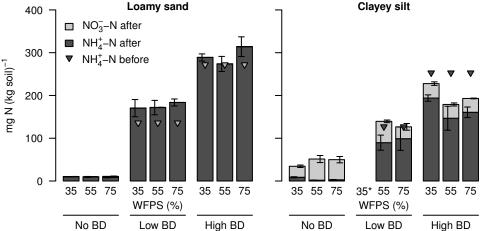


Fig. 3: Mean N<sub>2</sub>O fluxes (mg N m<sup>-2</sup> h<sup>-1</sup>) from a loamy sand and a clayey silt incubated under different water-filled pore spaces (WFPS, %) with different amounts of digestate (0 mL per sample ring: 'No BD', 17.6 mL: 'Low BD' and 35.2 mL: 'High BD'). The first till the fourth day of the incubation were measured in an aerobic He-O<sub>2</sub> headspace (with two days at 2 °C followed by another two days at 15 °C) while on the fifth day measurements where conducted in an anaerobic headspace with pure He (at 15 °C). Error bars show standard deviations; if bars are not visible, they are smaller than the symbols of the means. Under aerobic atmosphere, N<sub>2</sub>O fluxes from loamy sand were negligible, while fluxes from clayey silt showed an increase with temperature, especially with higher WFPS and intermediate amounts of digestate. Under anaerobic atmosphere, mean fluxes from loamy sand increased slightly, but significantly (Tukey's HSD, p < 0.05). The fluxes from clayey silt showed no significant differences (Tukey's HSD, p < 0.05) compared to the day before, with the exception of 35% WFPS, where mean flux increased strongly in the treatment with 320 kg digestate-N ha<sup>-1</sup>.

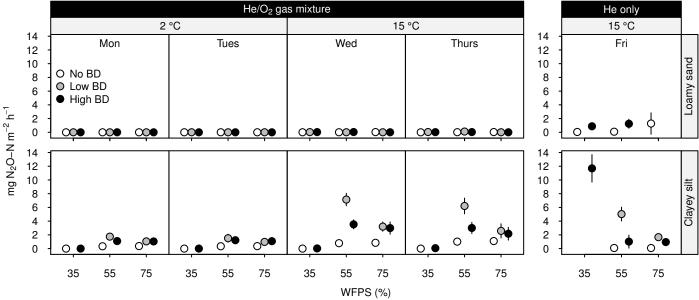


Fig. 4: Mean N<sub>2</sub> fluxes (mg m<sup>-2</sup> h<sup>-1</sup>) from a loamy sand and a clayey silt incubated under different water-filled pore spaces (WFPS, %) with different amounts of digestate (0 mL per sample ring: 'No BD', 17.6 mL: 'Low BD' and 35.2 mL: 'High BD'). The first till the fourth day of the incubation were measured in an aerobic He-O<sub>2</sub> headspace (with two days at 2 °C followed by another two days at 15 °C) while on the fifth day measurements where conducted in an anaerobic headspace with pure He (at 15 °C). Error bars show standard deviations; if bars are not visible, they are smaller than the symbols of the means. The dotted horizontal lines depict the average blank value; single flux rates lower than the respective lank value were set zero. Under aerobic atmosphere, N<sub>2</sub> fluxes from loamy sand were zero or rather negligible, while fluxes from clayey silt show a distinct increase with WFPS and higher fluxes at 15 °C. Under anaerobic atmosphere, mean fluxes from loamy sand increased by orders of magnitude, while the fluxes from clayey silt increased as well, but more gently compared to the sand.

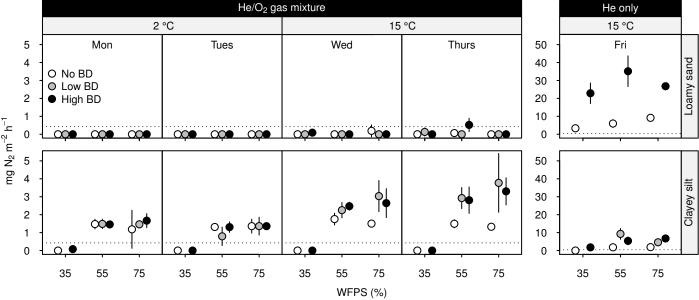


Fig. 5: Mean  $N_2/(N_2 + N_2O-N)$  product ratio from a loamy sand and a clayey silt incubated under different water-filled pore spaces (WFPS, %) with different amounts of digestate (0 mL per sample ring: 'No BD', 17.6 mL: 'Low BD' and 35.2 mL: 'High BD'). The first till the fourth day of the incubation were measured in an aerobic He-O<sub>2</sub> headspace (with two days at 2 °C followed by another two days at 15 °C) while on the fifth day measurements where conducted in an anaerobic headspace with pure He (at 15 °C). Error bars show standard deviations; if bars are not visible, they are smaller than the symbols of the means. For the loamy sand, there was a clear distinction of the ratios between aerobic and anaerobic atmospheres: while the ratios tended to 0 in the former, they tended to 1 in the latter, irrespectively of temperature or amount of digestate. For the clayey silt, ratios increased with WFPS and were highest from the unamended treatments under both the aerobic and the anaerobic atmospheres.

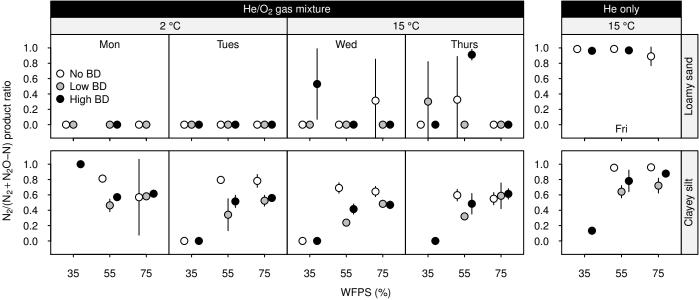


Fig. 6: Regression between DOC (mg per 100 g soil) measured after the incubation and the respective cumulated CO<sub>2</sub> emissions (g C m<sup>-2</sup>) during the period of aerobic headspace with their standard deviations and confidence interval (95%). If error bars are not visible, they are smaller than the symbols of the means. Both soils showed increasing emissions with increasing soil DOC contents as well a good regression fit ( $R^2 = 0.91$ , p < 0.001).

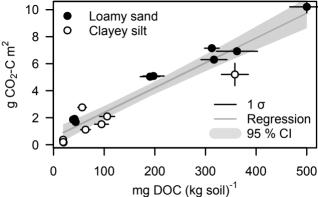


Fig. 7: Regression between cumulated  $CO_2$  emissions (g C m<sup>-2</sup>) and the respective cumulated  $N_2O + N_2$  emissions (g N m<sup>-2</sup>) from the clayey silt with WFPS > 35 % during the period of aerobic headspace with their standard deviations and confidence interval (95%). If error bars are not visible, they are smaller than the symbols of the means. The proportional increase of  $CO_2$  and the N gas species shows a good regression fit of  $R^2 = 0.93$ , (p = 0.001).

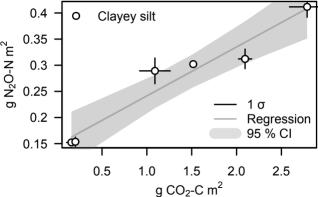


Table A1: Mean CO<sub>2</sub>-C fluxes with standard deviations in mg m<sup>-2</sup> h<sup>-1</sup> from the loamy sand and the clayey silt, treated with different water-filled pore spaces (WFPS, %), amounts of digestate (kg N ha<sup>-1</sup>) as well as different temperature regimes ( $^{\circ}$ C) under aerobic (He-O<sub>2</sub>) and anaerobic (He) atmosphere. Column 'Day' denotes the consecutive measuring days of the respective incubation cycle. Different letters after fluxes indicate significant differences (Tukey's HSD, p < 0.05) within each soil and measuring day. Zeros as last digits were omitted.

Day Atmosphere		Tommonotore (°C)	WFPS	1-a N 1-a-1	mg CO <sub>2</sub>	-C m <sup>-2</sup> h <sup>-1</sup>
		Temperature (°C)	(%)	kg N ha <sup>-1</sup>	Loamy sand	Clayey silt
1	He-O <sub>2</sub>	2	35	0	$6.8 \pm 2.4 \text{ cd}$	0 ± 0 c
1	He-O <sub>2</sub>	2	35	160	$22 \pm 3.5 \text{ bcd}$	NA
1	He-O <sub>2</sub>	2	35	320	$23.3 \pm 9.3$ bc	$22.8 \pm 2.8 \text{ ab}$
1	He-O <sub>2</sub>	2	55	0	$6 \pm 0.7  d$	$4.6 \pm 7.9 \text{ bc}$
1	He-O <sub>2</sub>	2	55	160	$34.4 \pm 3.1 \text{ b}$	$34.5 \pm 11.6 a$
1	He-O <sub>2</sub>	2	55	320	$28 \pm 3.2  b$	$15.9 \pm 3.4 \text{ abc}$
1	He-O <sub>2</sub>	2	75	0	$9.4 \pm 1.4 \text{ cd}$	$0 \pm 0$ c
1	He-O <sub>2</sub>	2	75	160	$37.5 \pm 6 \text{ b}$	$15.5 \pm 12.1$ abc
1	He-O <sub>2</sub>	2	75	320	$68.3 \pm 12.1$ a	$24.5 \pm 2.7$ a
2	He-O <sub>2</sub>	2	35	0	$9.8 \pm 3.5 \text{ c}$	1.3 ± 1.4 b
2	He-O <sub>2</sub>	2	35	160	$23 \pm 3.9 \text{ bc}$	NA
2	He-O <sub>2</sub>	2	35	320	$30.9 \pm 2.2 \text{ b}$	$22.2 \pm 2.4$ a
2	He-O <sub>2</sub>	2	55	0	$8.7 \pm 1.5 \text{ c}$	$0.6 \pm 1 \text{ b}$
2	He-O <sub>2</sub>	2	55	160	$33.4 \pm 0.9 \text{ b}$	$27.6 \pm 12.3$ a
2	He-O <sub>2</sub>	2	55	320	$35.9 \pm 2.7 \text{ b}$	$14.4 \pm 1.9 \text{ ab}$
2	He-O <sub>2</sub>	2	75	0	$8.3 \pm 1.5 \text{ c}$	$0 \pm 0$ b
2	He-O <sub>2</sub>	2	75	160	$31.9 \pm 3 \text{ b}$	$13 \pm 9.3 \text{ ab}$
2	He-O <sub>2</sub>	2	75	320	$57.6 \pm 14.8 \text{ a}$	$18.3 \pm 4 a$
3	He-O <sub>2</sub>	15	35	0	42.5 ± 4.5 c	$6.7 \pm 0.7 \text{ b}$
3	He-O <sub>2</sub>	15	35	160	$114.3 \pm 12.2 \text{ b}$	NA
3	He-O <sub>2</sub>	15	35	320	$149.5 \pm 9.4 \text{ b}$	$130.9 \pm 105 \text{ a}$
3	He-O <sub>2</sub>	15	55	0	$41.3 \pm 3.5$ c	$3.2 \pm 0.4 \text{ b}$
3	He-O <sub>2</sub>	15	55	160	$108.7 \pm 10.1 \text{ b}$	$57.8 \pm 12.2 \text{ bc}$
3	He-O <sub>2</sub>	15	55	320	162.1 ± 9.6 b	$26.8 \pm 0.7 \text{ bc}$
3	He-O <sub>2</sub>	15	75	0	$44.1 \pm 9.8 \text{ c}$	$3.2 \pm 0.7 \text{ b}$
3	He-O <sub>2</sub>	15	75	160	$150.4 \pm 19 \text{ b}$	$26.4 \pm 11.8$ bc
3	He-O <sub>2</sub>	15	75	320	$249.7 \pm 53.5 \text{ a}$	$35.3 \pm 6 \text{ bc}$
4	He-O <sub>2</sub>	15	35	0	48.7 ± 6 c	$15.1 \pm 4.9 \text{ cd}$
4	He-O <sub>2</sub>	15	35	160	$114.3 \pm 6.4 \text{ b}$	NA
4	He-O <sub>2</sub>	15	35	320	$156.9 \pm 15.4$ a	$65.7 \pm 2.2 \text{ a}$
4	He-O <sub>2</sub>	15	55	0	$48 \pm 3.4 \text{ c}$	$4.2 \pm 0.2 d$
4	He-O <sub>2</sub>	15	55	160	$109 \pm 14.4 \text{ b}$	$51.2 \pm 15.1$ ab

4	He-O <sub>2</sub>	15	55	320	$177.7 \pm 7.5$ a	$26.6 \pm 2.3 \text{ cd}$
4	He-O <sub>2</sub>	15	75	0	$34 \pm 7.8 \text{ c}$	$6.7 \pm 4 d$
4	$\text{He-O}_2$	15	75	160	$168.7 \pm 0.4 \text{ a}$	$22.1 \pm 14.8 \text{ cd}$
4	He-O <sub>2</sub>	15	75	320	$166.3 \pm 23.1 \text{ a}$	$34.1 \pm 5.7$ bc
5	Не	15	35	0	$11.2 \pm 0.6 d$	NA
5	He	15	35	160	$54.8 \pm 9.3 \text{ c}$	NA
5	He	15	35	320	$149.3 \pm 3.9 \text{ a}$	$45.8 \pm 2.1 \text{ a}$
5	He	15	55	0	$13.6 \pm 1.9 d$	$3.4 \pm 0.6 \text{ c}$
5	He	15	55	160	$55.2 \pm 4.4 \text{ bc}$	$32 \pm 11.4 \text{ ab}$
5	He	15	55	320	$164.5 \pm 3.5$ a	$15.2 \pm 10.7$ bc
5	He	15	75	0	$20.9 \pm 2.3 d$	$3.6 \pm 0.1 \text{ c}$
5	He	15	75	160	$75 \pm 7.3 \text{ b}$	$20.6 \pm 8.5 \text{ bc}$
5	He	15	75	320	NA	$26.1 \pm 2.6 \text{ ab}$

Table A2: Mean N<sub>2</sub>O-N fluxes with standard deviations in mg m<sup>-2</sup> h<sup>-1</sup> from the loamy sand and the clayey silt, treated with different water-filled pore spaces (WFPS, %), amounts of digestate (kg N ha<sup>-1</sup>) as well as different temperature regimes ( $^{\circ}$ C) under aerobic (He-O<sub>2</sub>) and anaerobic (He) atmosphere. Column 'Day' denotes the consecutive measuring days of the respective incubation cycle. Different letters after fluxes indicate significant differences (Tukey's HSD, p < 0.05) within each soil and measuring day. Zeros as last digits were omitted.

Dor	A top c == 1- == :	Temperature WEDG (01)		1-a N 11	mg N <sub>2</sub> O-	mg N <sub>2</sub> O-N m <sup>-2</sup> h <sup>-1</sup>		
Day	Atmosphere	(°C)	WFPS (%)	kg N ha <sup>-1</sup>	Loamy sand	Clayey sil		
1	He-O <sub>2</sub>	2	35	0	$0 \pm 0$	$0 \pm 0$ c		
1	$He-O_2$	2	35	160	$0 \pm 0$	NA		
1	$He-O_2$	2	35	320	$0 \pm 0$	$0 \pm 0$ c		
1	$He-O_2$	2	55	0	$0 \pm 0$	$0.3 \pm 0.1$		
1	$He-O_2$	2	55	160	$0 \pm 0$	$1.7 \pm 0.4$		
1	$He-O_2$	2	55	320	$0 \pm 0$	$1.1 \pm 0.11$		
1	$He-O_2$	2	75	0	$0 \pm 0$	$0.4 \pm 0.1$		
1	$\text{He-O}_2$	2	75	160	$0 \pm 0$	$1 \pm 0.1 \text{ b}$		
1	He-O <sub>2</sub>	2	75	320	$0 \pm 0$	$1 \pm 0.2 \text{ b}$		
2	He-O <sub>2</sub>	2	35	0	$0 \pm 0$	0 ± 0 d		
2	$He-O_2$	2	35	160	$0 \pm 0$	NA		
2	$He-O_2$	2	35	320	$0 \pm 0$	$0 \pm 0$ cd		
2	$He-O_2$	2	55	0	$0 \pm 0$	$0.3 \pm 0.1$ b		
2	$\text{He-O}_2$	2	55	160	$0 \pm 0$	$1.5 \pm 0.6$		
2	$\text{He-O}_2$	2	55	320	$0 \pm 0$	$1.2 \pm 0.2$		
2	$\text{He-O}_2$	2	75	0	$0 \pm 0$	$0.4 \pm 0.1$ b		
2	He-O <sub>2</sub>	2	75	160	$0 \pm 0$	$1 \pm 0.1 \text{ at}$		
2	He-O <sub>2</sub>	2	75	320	$0 \pm 0$	$1.1 \pm 0.2$		
3	He-O <sub>2</sub>	15	35	0	$0 \pm 0$ cd	0 ± 0 c		
3	$He-O_2$	15	35	160	$0 \pm 0$ abc	NA		
3	$He-O_2$	15	35	320	$0 \pm 0$ ab	$0 \pm 0$ c		
3	$He-O_2$	15	55	0	$0 \pm 0$ bcd	$0.8 \pm 0.2$		
3	$He-O_2$	15	55	160	$0 \pm 0$ bcd	$7.1 \pm 0.9$		
3	$He-O_2$	15	55	320	$0 \pm 0$ a	$3.5 \pm 0.7$		
3	$He-O_2$	15	75	0	$0 \pm 0$ ab	$0.8 \pm 0.2$		
3	$He-O_2$	15	75	160	$0 \pm 0 d$	$3.2 \pm 0.7$		
3	He-O <sub>2</sub>	15	75	320	$0 \pm 0$ cd	$3 \pm 0.9 \text{ b}$		
4	He-O <sub>2</sub>	15	35	0	$0 \pm 0  b$	0 ± 0 c		
4	He-O <sub>2</sub>	15	35	160	$0 \pm 0$ ab	NA		
4	He-O <sub>2</sub>	15	35	320	$0 \pm 0$ ab	$0.1 \pm 0.1$		
4	He-O <sub>2</sub>	15	55	0	$0 \pm 0$ b	$1 \pm 0.2 \text{ bc}$		
4	He-O <sub>2</sub>	15	55	160	$0.1 \pm 0.1 \text{ a}$	$6.2 \pm 1.1$		

4	$He-O_2$	15	55	320	$0 \pm 0$ ab	$3 \pm 0.8 \text{ b}$	
4	He-O <sub>2</sub>	15	75	0	$0 \pm 0$ ab	$1.1 \pm 0.3 \text{ bc}$	
4	He-O <sub>2</sub>	15	75	160	$0 \pm 0$ b	$2.6 \pm 1 \text{ b}$	
4	He-O <sub>2</sub>	15	75	320	$0 \pm 0$ b	$2.2 \pm 0.9 \text{ b}$	
5	Не	15	35	0	$0.1 \pm 0$	NA	
5	He	15	35	160	NA	NA	
5	He	15	35	320	$0.9 \pm 0.1$	$11.7 \pm 2 a$	
5	He	15	55	0	$0.1 \pm 0$	$0.1 \pm 0$ c	
5	He	15	55	160	NA	$5 \pm 1 \text{ b}$	
5	He	15	55	320	$1.2 \pm 0.7$	$1.4 \pm 0.8 \; c$	
5	He	15	75	0	$1.3 \pm 1.6$	$0.1 \pm 0$ c	
5	He	15	75	160	NA	$1.7 \pm 0.3 \text{ c}$	
5	He	15	75	320	NA	$1 \pm 0.3 \text{ c}$	

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	A + l	Temperature	WEDC (OL)	1 NI 11	mg N <sub>2</sub>	m <sup>-2</sup> h <sup>-1</sup>
	Atmosphere	(°C)	WFPS (%)	kg N ha <sup>-1</sup>	Loamy sand	Clayey silt
1	He-O <sub>2</sub>	2	35	0	$0 \pm 0$	$0 \pm 0$ bc
1	He-O <sub>2</sub>	2	35	160	$0 \pm 0$	NA
1	He-O <sub>2</sub>	2	35	320	$0 \pm 0$	$0.1 \pm 0.1$ bo
1	He-O <sub>2</sub>	2	55	0	$0 \pm 0$	$1.5 \pm 0.3$ a
1	He-O <sub>2</sub>	2	55	160	$0 \pm 0$	$1.5 \pm 0.3$ a
1	He-O <sub>2</sub>	2	55	320	$0 \pm 0$	$1.5 \pm 0$ a
1	He-O <sub>2</sub>	2	75	0	$0 \pm 0$	$1.2 \pm 1.1 \text{ a}$
1	He-O <sub>2</sub>	2	75	160	$0 \pm 0$	$1.5 \pm 0.2$ a
1	He-O <sub>2</sub>	2	75	320	$0 \pm 0$	$1.7 \pm 0.4 \text{ a}$
2	He-O <sub>2</sub>	2	35	0	$0 \pm 0$	$0 \pm 0$ c
2	He-O <sub>2</sub>	2	35	160	$0 \pm 0$	NA
2	He-O <sub>2</sub>	2	35	320	$0 \pm 0$	$0 \pm 0$ c
2	He-O <sub>2</sub>	2	55	0	$0 \pm 0$	$1.3 \pm 0.1 a$
2	He-O <sub>2</sub>	2	55	160	$0 \pm 0$	$0.8 \pm 0.5 \text{ b}$
2	He-O <sub>2</sub>	2	55	320	$0 \pm 0$	$1.3 \pm 0.3 \text{ a}$
2	He-O <sub>2</sub>	2	75	0	$0 \pm 0$	$1.4 \pm 0.4$ a
2	He-O <sub>2</sub>	2	75	160	$0 \pm 0$	$1.4 \pm 0.5 \text{ a}$
2	He-O <sub>2</sub>	2	75	320	$0 \pm 0$	$1.4 \pm 0.1$ a
3	He-O <sub>2</sub>	15	35	0	0 ± 0 b	0 ± 0 e
3	He-O <sub>2</sub>	15	35	160	$0 \pm 0$ b	NA
3	He-O <sub>2</sub>	15	35	320	$0.1 \pm 0.1 \text{ ab}$	$0 \pm 0$ e
3	He-O <sub>2</sub>	15	55	0	$0 \pm 0$ b	$1.8 \pm 0.3$ cc
3	He-O <sub>2</sub>	15	55	160	$0 \pm 0$ b	$2.3 \pm 0.4$ bo
3	He-O <sub>2</sub>	15	55	320	$0 \pm 0$ b	$2.5 \pm 0.2$ at
3	He-O <sub>2</sub>	15	75	0	$0.2 \pm 0.3 \text{ a}$	$1.5 \pm 0.2 d$
3	He-O <sub>2</sub>	15	75	160	$0 \pm 0$ b	$3 \pm 0.9 \text{ a}$
3	He-O <sub>2</sub>	15	75	320	$0 \pm 0$ b	$2.6 \pm 0.8$ at
4	He-O <sub>2</sub>	15	35	0	0 ± 0 b	0 ± 0 c
4	He-O <sub>2</sub>	15	35	160	$0.1 \pm 0.2 \text{ b}$	NA
4	He-O <sub>2</sub>	15	35	320	$0 \pm 0$ b	$0 \pm 0$ c
4	He-O <sub>2</sub>	15	55	0	$0.1 \pm 0.1 \text{ b}$	$1.5 \pm 0.2 \text{ b}$
4	$\text{He-O}_2$	15	55	160	$0 \pm 0  b$	$2.9 \pm 0.6$ a

4	He-O <sub>2</sub>	15	55	320	$0.5 \pm 0.4 \text{ a}$	$2.8 \pm 0.7 \text{ a}$	
4	$He-O_2$	15	75	0	$0 \pm 0$ b	$1.3 \pm 0.2 \text{ bc}$	
4	He-O <sub>2</sub>	15	75	160	$0 \pm 0$ b	$3.8 \pm 1.6 a$	
4	$He-O_2$	15	75	320	$0 \pm 0$ b	$3.3 \pm 0.8 a$	
5	Не	15	35	0	$3.3 \pm 0.4 d$	$0 \pm 0$ c	•
5	Не	15	35	160	NA	NA	
5	Не	15	35	320	$22.9 \pm 5.7 \text{ b}$	$1.8 \pm 0.1 \text{ c}$	
5	Не	15	55	0	$6 \pm 2.2 \text{ cd}$	$1.8 \pm 0.2$	
5	Не	15	55	160	NA	$9.5 \pm 2.7 \text{ a}$	
5	Не	15	55	320	$35.1 \pm 8.6$ a	$5.1 \pm 1.8 \text{ bc}$	
5	Не	15	75	0	$9.2 \pm 0.4 c$	$1.9 \pm 0.1 \text{ c}$	
5	Не	15	75	160	NA	$4.8 \pm 1.6 \text{ bc}$	
5	Не	15	75	320	$26.8 \pm 1.1 \text{ b}$	$6.7 \pm 0.8 \text{ b}$	