

1 **Potential short-term losses of N₂O and N₂ from high**
2 **concentrations of biogas digestate depend mainly on soil**
3 **texture and moisture, not on NH₄⁺**

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11 **Abstract.** Biogas digestate (BD) is increasingly used as organic fertiliser, but has a high potential for NH₃
12 losses. Its proposed injection into soils as a counter-measure has been suggested to promote the generation of
13 N₂O, leading to a potential trade-off. Furthermore, the effect of high nutrient concentrations on N₂ losses as they
14 may appear after injection of BD into soil has not yet been evaluated. Hence, we performed an incubation
15 experiment with soil cores in a helium-oxygen atmosphere to examine the influence of soil substrate (loamy
16 sand, clayey silt), water-filled pore space (WFPS; 35, 55, 75%) and application rate (0, 17.6 and 35.2 mL BD per
17 soil core [250 cm³]) on the emissions of N₂O, N₂ and CO₂ after the usage of high loads of BD. To determine the
18 potential capacity for gaseous losses, we applied anaerobic conditions by purging with helium for the last 24 h of
19 incubation. Immediate N₂O and N₂ emissions as well as the N₂/(N₂O+N₂) ratio depended on soil type and
20 increased with WFPS indicating a crucial role of soil gas diffusivity for the formation of nitrogenous gases in
21 agricultural soils. However, the emissions did not increase with the application rate of BD probably due to an
22 inhibitory effect of the high NH₄⁺ content in BD. Our results suggest a larger potential for N₂O formation in the
23 fine-textured clayey silt compared to the coarse loamy sand after applying high concentrations of BD as
24 appearing after injection. However, the loamy sand showed basically a large potential for N₂ formation under
25 anaerobic headspace conditions. Nevertheless, our results show the need for further investigations on the
26 dynamics and the duration of the observed effects and their significance for field conditions.

27 1 Introduction

28 Nitrous oxide (N₂O) is a potent greenhouse gas (Myhre et al., 2013), with agriculture being the largest single
29 source of anthropogenic N₂O emissions, contributing about 4.1 Tg N₂O-N yr⁻¹ or 66% of total gross
30 anthropogenic emissions mainly as a result of mineral nitrogen (N) fertiliser and manure application (Davidson
31 and Kanter, 2014). The generation of nitrogen gas (N₂) is of agronomic interest in terms of nutrient management,
32 since such gaseous losses may imply a significant loss of N from the soil/plant system (Friedl et al., 2016;
33 Cameron et al., 2013). However, from an environmental stance, N₂ is innocuous and, thus, the preferred type of
34 gaseous N-loss from soil (Davidson et al., 2015). Further, emission of ammonia (NH₃) is of environmental
35 concern, e.g., due to acid deposition or conversion to N₂O (Ferm, 1998; Mosier et al., 1998). In general, the
36 improvement of N use efficiency and thus the decrease of N losses in crop production are paramount in the
37 presence of challenges like food security, environmental degradation and climate change (Zhang et al., 2015).

38 In Germany, the increased demand for renewable energy sources like methane from biogas plants entails an
39 expanded amount of digestion residues (biogas digestate, BD) used as organic amendment in agriculture (Möller
40 and Müller, 2012). Digestion in biogas reactors increases pH and the proportion of ammonium (NH_4^+) and
41 narrows the C to N ratio due to the depletion of labile C fractions of the feedstock (Möller and Müller, 2012).
42 Compared to undigested amendments like **animal** slurry, this altered chemical properties may promote
43 biochemical reactions in the soil that are responsible for the formation of gaseous N species like N_2O , **nitric**
44 **oxide (NO)**, N_2 and especially NH_3 (Nkoa, 2013). In general, the **effect strengths** of BD on gaseous N losses
45 from soil is still under debate (Möller, 2015).

46 Significant losses of N as NH_3 can occur within the first hours after manure application (Quakernack et al.,
47 2012). To reduce NH_3 losses, the application of BD by injection is recommended, **but** this measure can
48 simultaneously increase the potential for N_2O losses compared to surface-application (Wulf et al., 2002; Velthof
49 and Mosquera, 2011). **On the one hand, high NH_4^+ concentrations in the injection band promote nitrification,**
50 **which is a significantly O_2 consuming process releasing N_2O (Christensen and Rowe, 1984). On the other hand,**
51 **increased amounts of C in the injection band also promote respiration and, thus, additionally deplete the O_2**
52 **supply (Dell et al., 2011). Altogether, the** conditions during the initial phase after injection of BD foster
53 microsites favourable for microbial denitrification, which may promote also the formation of N_2 due to anaerobic
54 conditions (Köster et al., 2015; Webb et al., 2010).

55 There is a wealth of biotic and abiotic processes in soils that produce N_2O and N_2 , most of which are enhanced
56 by anaerobic or at least hypo-aerobic conditions (Butterbach-Bahl et al., 2013). Also the amounts and the relative
57 share of N_2 and N_2O in the overall gaseous N emissions depend – among other factors like the favoured
58 reduction of NO_3^- rather than N_2O as alternative electron acceptor – on the degree of O_2 restriction (Firestone
59 and Davidson, 1989). Soil physical and biotic factors (i.e. diffusion and consumption of O_2) as well as their
60 interactions control the aerobic status of a soil. Diffusion of O_2 depends on the porosity of the soil substrate in
61 conjunction with water-filled pore space (WFPS), while O_2 is consumed by heterotrophic respiration **and**
62 **nitrification** which depend on mineral N content, carbon (C) availability as well as on temperature (Ball, 2013;
63 Uchida et al., 2008; Maag and Vinther, 1999). **In general, fine textured soils with higher clay contents exhibit a**
64 **lower gas diffusivity compared to coarse textured soils, which result regularly in higher denitrification activity in**
65 **the former with higher N_2O emission rates, but also a higher probability for the consecutive reduction to N_2**
66 **(Senbayram et al., 2014; Gu et al., 2013; Ball, 2013).** Simultaneously, the supply of substrates for
67 microorganisms is determined by liquid diffusion rates in soil water and, thus, by WFPS (Blagodatsky and

68 Smith, 2012; Maag and Vinther, 1999). However, though high within injection bands, nutrient concentrations
69 and WFPS should theoretically increase further with the row spacing between the injection bands, if a given
70 amount of BD per area is assumed. We are not aware of studies addressing the effect of such high BD
71 concentrations.

72 Hence, there is a general lack of knowledge about effects of BD injection on gaseous N-losses and especially
73 about the effects of high BD concentrations and their interaction with O₂ limiting factors like soil texture and
74 WFPS, as well as temperature and heterotrophic respiration. The indicated knowledge gaps are caused not the
75 least by methodological constraints with the direct determination of N₂ fluxes due to the high background level of
76 N₂ in the atmosphere, while indirect applications like acetylene-based methods and ¹⁵N tracers are unfavourable
77 since the former implicates serious underestimations and the latter has rather high detection limits (Groffman et
78 al., 2006).

79 Thus, we applied the helium-oxygen (He-O₂) incubation technique (Butterbach-Bahl et al., 2002) in a laboratory
80 experiment to evaluate the effect of above suggested factors on the emission of N₂O and N₂ after soil amendment
81 with relatively high amounts of BD as they might occur after injection into soils. Simultaneously, CO₂ flux was
82 determined as an indicator for microbial O₂ consumption, O₂ diffusion and also for the degradability of organic
83 C applied with BD (Blagodatsky and Smith, 2012), but with the restriction that inorganic sources could not be
84 differentiated. We hypothesised that (1) N₂O and N₂ emissions will increase with WFPS, (2) this gaseous N
85 losses will also be affected by BD application rate, i.e. the concentration resulting from injection, and (3) the fine
86 textured clayey silt will induce higher gaseous N losses than the coarse loamy sand.

87 2 Material and Methods

88 2.1 Selected soils, sampling of soil cores and biogas digestate

89 Two soils were selected and both were adjusted to three levels of WFPS and three quantities of BD (Table 1),
90 resulting in 18 factor combinations with three repetitions each. Temperature was increased from 2 °C during the
91 first two days to 15 °C for the last three days of the incubation. Intact soil cores (diameter 7.2 cm, height 6.1 cm,
92 volume 250 cm³) were taken with sample rings in the range from 0–0.10 m depth from two sites with different
93 textures, i.e. sandy loam and clayey silt. The sandy loam samples were gathered from a stagnic luvisol (IUSS
94 Working Group WRB, 2006) located in Gülzow (North-East Germany) in the ground moraine of the
95 Weichselian glacial period at 53° 48' 35" N and 12° 4' 20" E. The clayey silt samples were collected from a

96 haplic luvisol located in Dornburg between the foothills and the lowlands of Central Germany at 51° 0' 8" N and
97 11° 39' 25" E (see Table 2 for more details on soil characteristics). After field sampling, the soil cores were dried
98 for 48 h at 40 °C to facilitate adjustment of WFPS.

99 Both sites have been cultivated with similar crop rotations used as feedstock for biogas production and have been
100 amended with biogas digestate for the past nine years. The crop rotation on the sandy loam consisted of maize
101 (*Zea mays* L.), rye (*Secale cereale* L.), sorghum (*Sorghum bicolor* (L.) MOENCH), winter triticale (×
102 *Triticosecale* Wittmack), ryegrass (*Lolium perenne* L.) and winter wheat (*Triticum aestivum* L.). The only
103 difference in the crop rotation on the clayey silt was the cultivation of sudangrass (*Sorghum × drummondii*)
104 instead of sorghum.

105 The biogas digestate used for the incubation was obtained from a biogas plant at 'Gut Dalwitz', an organic farm
106 in northeast Germany. The feedstock for the anaerobic fermentation in the plant consisted of 60 % maize, 20 %
107 solid cattle manure, 10 % dry chicken manure and 10 % rye. The digestate was analysed by 'LUFA', Rostock,
108 Germany and had a pH of 8.3, 2.91% organic C, 0.16% dissolved organic C (DOC), 0.54% N and 0.27% NH₄-N
109 in undried material with a dry matter content of 9.4%.

110 2.2 Adjustment of WFPS and addition of N

111 For adjustment of WFPS, the dry and undisturbed soil cores were moistened dropwise. The respective quantities
112 of water were calculated based on the bulk density, an assumed particle density of 2.65 g cm⁻¹ and reduced by
113 the expected moisture input from subsequent addition of BD. The soil cores were then mixed with BD and
114 finally repacked to reach nutrient concentrations comparable to that in injection bands. The mixing was done for
115 methodical reasons since the available space in the incubation vessels was limited and, hence, 'real' injection not
116 feasible. However, injection bands have actually a thickness comparable to the sample rings we used (Markfoged
117 et al., 2011). The amounts of added BD were calculated with an assumed injection of 160 kg N ha⁻¹ into soil with
118 row spaces of 0.15 m (narrow injection bands with low BD concentration, LOBD) and 0.30 m (wide injection
119 bands with high BD concentration, HIBD), which are common ranges used by injection machinery and which
120 correspond to 17.6 and 25.3 mL BD, respectively, per sample ring. After this procedure, the soil cores were
121 sealed with plastic lids and stored immediately at 2 °C until the beginning of the incubation within a week.

122 2.3 Determination of gas fluxes

123 The measurements of N₂, N₂O and CO₂ fluxes were applied following the He-O₂ method (Scholefield et al.,
124 1997; Butterbach-Bahl et al., 2002). Six soil cores (i.e. the repetitions of two factor combinations at a time, Table
125 3) were placed simultaneously in special gas-tight incubation vessels inside a climate chamber. Analyses were
126 conducted in the laboratory of the Institute for Landscape Biogeochemistry, Leibniz Centre for Agricultural
127 Landscape Research (ZALF), Müncheberg, Germany. Before flux measurements, the vessels were evacuated
128 moderately (0.047 bar) and flushed with an artificial He/O₂ gas mixture (20.49 % O₂, 345.5 ppm CO₂, 359 ppb
129 N₂O, 1863 ppb CH₄, 2.46 ppm N₂, rest He) four times consecutively to remove ambient N₂. Subsequently, the air
130 temperature of the climate chamber was set to 2 °C and a continuous He/O₂ gas flow rate of 15 ml min⁻¹ was
131 applied to the vessel headspaces for 72 h to remove residues of N₂ from soil cores **by diffusion, including a**
132 **restricted N₂ production by decreased microbial activity**. After this pre-incubation, during the following two
133 days, the headspace concentration of N₂O and CO₂ was measured once daily in the morning. To compensate for
134 **the lower precision of the detector for N₂ in relation to the detector for N₂O and CO₂ (cf., Eickenscheidt et al.,**
135 **2014)**, N₂ concentrations were measured consecutively three times daily in the morning. Immediately after the
136 last measurement on the second day, the temperature was set to 15 °C and the measurements were continued for
137 another two days. Finally, the He/O₂ gas mixture was substituted by pure He and, following 24 h of
138 acclimatisation, gas measurements were carried out once again (Figure 1) to determine the **generation of N₂O**
139 **and N₂ in a completely anaerobic soil matrix. The latter step is important to get a clue about the actual potential**
140 **for gaseous N losses after highly concentrated BD application**. The settings of the chromatographs for gas
141 analyses are described in Eickenscheidt et al. (2014). Gas fluxes were calculated according to Eq. (1):

$$142 \quad f = \frac{M \times p \times v \times dc}{R \times T \times A}, \quad (1)$$

143 where f is the flux (N₂ and CO₂: mg m⁻² h⁻¹, N₂O: µg m⁻² h⁻¹), M the molar mass in g mol⁻¹ (N₂: 28, CO₂: 44,
144 N₂O: 44), p the air pressure (Pa), v the air flow (L h⁻¹), R the gas constant (8.31 J mol⁻¹ K⁻¹), T the temperature
145 inside the chamber (K), A the area of the incubation vessel (m²), and dc the difference of gas concentrations (N₂
146 and CO₂: ppm, N₂O: ppb) **between inlet and outlet of a vessel**.

147 **To enhance the tightness against atmospheric N₂ contamination, the lids of the incubation vessels were purged**
148 **permanently with helium. We obtained blank values by inserting aluminium blocks into the vessels before each**
149 **measurement cycle. Since this blank values were usually steady with means of 1.9 (1σ = 0.9) ppm N₂, 349.6 (1σ**

150 = 11.4) ppb N₂O and 353.9 (1σ = 13.5) ppm CO₂, we suggest that the vessels were tight. Derived from the blank
151 values, lowest detectable fluxes were on average 0.427 (1σ = 0.271) mg N₂-N m⁻² h⁻¹, 3.6 (3.1) μg N₂O-N m⁻² h⁻¹
152 and 0.918 (0.693) mg CO₂-C m⁻² h⁻¹. For flux estimation, the blank values were subtracted from the values
153 measured at the respective outlet. Estimated fluxes from the soil cores smaller than the respective blank fluxes of
154 each day were set to zero.

155

156 2.4 Soil analyses after incubation

157 After incubation, the soil cores were stored at 2 °C until they were extracted with 0.1 M KCl solution (soil to
158 extract ratio 1:4, standardised extraction method of the commissioned laboratory at Leibniz Centre for
159 Agricultural Landscape Research e. V.) and analysed for NH₄⁺ and nitrate (NO₃⁻) by spectrophotometry
160 according to DIN ISO 14256 with a continuous flow analyser 'CFA-SAN', Skalar Analytical B.V., the
161 Netherlands and for DOC by combustion according to DIN ISO 10694 with an analyser 'RC 612', Leco
162 Instruments GmbH, Germany.

163 2.5 Statistical analysis

164 All statistical analyses were done using R version 3.2.3 (R Core Team, 2015) with the data of the measuring
165 days under He-O₂ atmosphere. Data from the vessels with the factor combination of 35% WFPS and LOBD with
166 clayey silt were omitted due to technical reasons during sample preparation. For the final period of pure He
167 headspace, some gas concentration data are missing due to logistical reasons. For the loamy sand, this affects all
168 WFPS levels with LOBD (N₂ and N₂O), the treatment 75% WFPS with 320 kg N h⁻¹ (N₂O and CO₂) and for the
169 clayey silt the treatment 35% WFPS without amendment (N₂O and CO₂).

170 To account for repeated measurement of vessels, linear mixed effect models were applied with package
171 'lmerTest' version 2.0-33 (Kuznetsova et al., 2016) for fluxes of each gas type. The three pseudo-replicated
172 fluxes from the N₂ measurements of each vessel were averaged for each day to obtain the same number of
173 observations as for N₂O and CO₂ fluxes. The fixed structure of models included soil type, WFPS, amount of
174 digestate, temperature, NO₃⁻ and DOC contents after incubation as well as the fluxes of N₂O (in the model for
175 N₂) and CO₂ (in the models for N₂, N₂O and N₂/[N₂+N₂O] product ratio). Soil NH₄⁺ was omitted since it showed
176 high autocorrelation with the amount of BD applied. The individual soil cores in the vessels were set as random
177 effect (nested within the week of incubation and with allowance for a variable slope of the effect each day) with

178 regard to lack of independence of consecutive measurements. The model responses for N₂, N₂O and CO₂ were
179 log transformed (ln[value + 1]) since gas fluxes from soils usually show lognormal distributions (Kaiser et al.,
180 1998). The function ‘step’ was used for automatic backward selection of models based on AIC (Akaike’s ‘An
181 Information Criterion’). The skewness (γ) was calculated with R package ‘moments’ version 0.14 (Komsta and
182 Novomestky, 2015) to check residuals for normal distribution and $|\gamma| \leq 2$ was assumed as appropriate (West et
183 al., 1995). For mixed effects models, *p*-values of the ANOVA (type 2) were calculated based on Satterthwaite's
184 approximation)
185 Cumulated gas fluxes were estimated with a bootstrap method using function ‘auc.mc’ of R package ‘flux’
186 version 0.3-0 (Jurasinski et al., 2014) for the R statistical software version 3.2.3 (R Core Team, 2015). In short,
187 the fluxes for the period of aerobic headspace were cumulated in 100 iterations, while for each run 2 fluxes were
188 omitted randomly. Then, the resulting data were used to calculate means and standard deviations.

189 3 Results

190 3.1 Soil NH₄⁺, NO₃⁻ and DOC contents

191 The calculated application of NH₄⁺-N from BD per kg soil approximated for the sandy loam 247.0 mg (LOBD)
192 and 494.0 mg (HIBD), and for the clayey silt 266.0 mg (LOBD) and 532.0 mg (HIBD). The NO₃⁻ content of BD
193 was negligible. After incubation, the recovered NH₄⁺-N contents increased with the level of amendment with BD
194 in both soils and were not affected by WFPS, with the exception of treatments of clayey silt with 35% WFPS
195 (Fig. 2). In the loamy sand, the mean amounts of NH₄⁺-N per kg soil ranged from 8.5 to 10.0 mg (no
196 amendment), from 170.4 to 185.6 mg (LOBD) and from 273.7 to 314.0 mg (HIBD). In the clayey silt, NH₄⁺-N
197 contents per kg soil reached only 1.8 to 8.8 mg (no amendment), 89.7 to 98.9 mg (LOBD) and 146.8 to 194.0 mg
198 (HIBD) and, thus, roughly half the amounts of the clayey silt. However, in contrast to the loamy sand, the clayey
199 silt showed also substantial NO₃⁻ contents between 25.7 (35% WFPS without amendment) and 49.8 mg NO₃⁻-N
200 (kg soil)⁻¹ (55% WFPS with LOBD). Negligible amounts of NO₃⁻ were detected in the loamy sand after
201 incubation: except for a mean value of 2.4 mg NO₃⁻-N (kg soil)⁻¹ in the unamended treatment with 75% WFPS,
202 the values of all other treatments ranged between 0.2 and 0.5 mg.

203 The amounts of recovered DOC increased with the application rate of BD, but with different magnitudes for both
204 soils. While mean values from 38.6 (55 % WFPS without amendment) to 500.1 mg DOC per kg soil (75 %
205 WFPS, HIBD) were determined for the loamy sand after incubation, lower mean values from 18.9 (55 % WFPS

206 without amendment) to 358.1 mg (35 % WFPS, HIBD) were found in the clayey silt, where the respective
207 second highest values were considerably lower for both soils (loamy sand: 362.2 mg for 75 % WFPS with
208 LOBD, clayey silt: 105.9 mg for 75 % WFPS with HIBD) (Table 4).

209 3.2 CO₂ fluxes

210 CO₂ fluxes showed clear differences between the soils: under all combinations of temperature and oxygen, the
211 fluxes were always larger from the loamy sand compared with the clayey silt (Table A1). In general, the mean
212 fluxes from the loamy sand increased with the amount of digestate from 8.3 to 57.6 (aerobic atmosphere at 2°C),
213 from 34.0 to 168.7 (aerobic at 15 °C) and from 11.2 to 87.9 mg CO₂-C m⁻² h⁻¹ (anaerobic at 15°C), but showed
214 no obvious pattern with WFPS. Although the mean fluxes from the clayey silt were also always smallest in the
215 unamended treatments, there was no clear trend of fluxes with the amount of amendment. There was a slight
216 trend of decreasing fluxes with increasing WFPS for the clayey silt. However, the predictive power of WFPS on
217 CO₂-C fluxes was minor since it was eliminated during stepwise regression fitting. By contrast, soil type, amount
218 of digestate, temperature as well as the DOC content after the incubation had significant ($p < 0.01$) effects (Table
219 5).

220 3.3 N₂O fluxes

221 The mean N₂O fluxes from the loamy sand at 2 °C under the He-O₂-atmosphere were virtually zero and, thus,
222 negligible (Fig. 3, Day 2 in Table A2). This was similar at 15°C with the exception of 35% WFPS without
223 digestate (0.1 mg N₂O-N m⁻² h⁻¹, Fig. 3, Day 4 in Table A2). The clayey silt showed much larger fluxes than the
224 loamy sand: even at 2 °C, up to 1.5 mg N₂O-N m⁻² h⁻¹ were detected (55% WFPS with LOBD). After shifting the
225 temperature to 15 °C, the same factor combination had a mean flux of 6.2 mg N₂O-N m⁻² h⁻¹ and the other
226 treatments emitted in mean between 1.0 and 3.0 mg N₂O-N m⁻² h⁻¹ with the exception of incubations with 35%
227 WFPS, where fluxes were smaller. The sand showed weak N₂O emissions, independent of temperature and
228 WFPS as well as the amount of BD application. In contrast, the emissions of the clayey silt increased with
229 temperature and were highest with intermediate WFPS and amount of BD, i.e. 55% and LOBD, respectively.
230 Surprisingly, at 15 °C, increasing the amount of BD up to HIBD did not increase the observed N₂O efflux; rather
231 it decreased the efflux significantly ($p < 0.05$, Tuckey's HSD) at 55% and also, but not significantly, at 75%
232 WFPS (Fig. 3, Table A2). However, this effect was not noticed at 35% WFPS due to generally low emissions at
233 this moisture level.

234 According to the linear mixed model for N₂O fluxes in aerobic conditions, WFPS, amount of digestate,
235 temperature, DOC content of soil after incubation and CO₂ fluxes had significant ($p < 0.001$) effects on N₂O flux
236 (Table 5).

237 Under anaerobic headspace conditions, the overall highest mean N₂O flux was observed from the clayey silt at
238 35% WFPS and HIBD (11.7 mg N₂O-N m⁻² h⁻¹). The same soil showed a tendency of decreasing N₂O fluxes
239 with increasing WFPS. Fluxes were largest with LOBD amendment. In the loamy sand, the pure He-atmosphere
240 induced increasing mean N₂O fluxes (up to 1.3 mg N₂O-N m⁻² h⁻¹) with increasing WFPS (Fig. 3, Table A2). So,
241 the anaerobic headspace induced a change only in the loamy sand by increasing emissions.

242 3.4 N₂ fluxes

243 From the loamy sand, no N₂ fluxes were detected at 2 °C under He-O₂-atmosphere (Fig. 4, Day 2 in Table A3).
244 Under the same conditions, the clayey silt showed mean fluxes from zero (all incubations with 35% WFPS) to
245 1.4 mg N₂ m⁻² h⁻¹ (all incubations with 75% WFPS). After increasing the temperature to 15 °C, again, the sandy
246 loam released mostly negligible **rates** of N₂, except for 0.5 mg m⁻² h⁻¹ with 55% WFPS and 320 kg N ha⁻¹ (Fig. 4,
247 Day 4 in Table A3). In contrast, up to 3.8 mg N₂ m⁻² h⁻¹ (75% WFPS with LOBD) were detected in the clayey
248 silt. However, **the clayey silt showed** also no fluxes in all **BD treatments** with 35% WFPS. Put simply,
249 temperature had a small effect on N₂ emissions from the sandy loam, but WFPS and amount of BD showed no
250 consistent influence. In contrast, the clayey silt **emitted** clearly increasing emissions with increasing temperature,
251 WFPS and also with the application of BD, where a raise from LOBD up to HIBD at 15 °C, however, resulted in
252 slightly, but not significantly ($p > 0.05$, Tuckey's HSD), decreased fluxes (Fig. 4, Table A3). The summary of
253 the linear mixed model for N₂ fluxes under aerobic conditions **revealed significant effects ($p < 0.05$) of soil type,**
254 **WFPS, the amount of digestate, temperature, DOC content after incubation and N₂O flux** (Table 5).

255 After switching the atmosphere to pure He, the N₂ fluxes from the sandy loam increased more than 60-fold. In
256 contrast to aerobic conditions, all measured factor combinations showed mean fluxes from 3.3 (35% WFPS
257 without N) to 35.1 mg N₂ m⁻² h⁻¹ (55% with 320 kg N ha⁻¹), where the fluxes from amended treatments were
258 always higher than fluxes from the unamended ones (Fig. 2, Day 5 in Table A3). For the clayey silt, compared
259 with aerobic atmosphere, mean fluxes increased slightly to 1.9 mg N₂ m⁻² h⁻¹ in unamended treatments and more
260 remarkably to 9.3 mg N₂ m⁻² h⁻¹ in amended ones, still not reaching the amounts observed for the sandy loam.
261 This implies that the N₂ emissions were increased from both soils under anaerobic headspace conditions, but the
262 loamy sand exhibited a much more intense reaction.

263 3.5 $N_2/(N_2 + N_2O-N)$ product ratio

264 No clear trend of the product ratio of $N_2/(N_2 + N_2O-N)$ was found for incubations of the loamy sand. However,
265 there was a clear distinction of the ratios for this soil under aerobic and anaerobic atmospheres: while the ratios
266 were close to zero in the former, they were close to 1 in the latter (Fig. 5). In contrast, in the clayey silt the ratios
267 increased with WFPS and were affected by digestate amendment under both the aerobic and the anaerobic
268 atmospheres, where the highest ratios (up to 0.8) were found in treatments without digestate and at least 55%
269 WFPS. The digestate-amended treatments showed mostly ratios around or above 0.5, with exception of the 35%
270 WFPS treatments, which had ratios close to zero. According to the linear mixed model, the product ratio under
271 aerobic conditions was affected significantly ($p < 0.01$) by soil type and the amount of digestate (Table 5).

272 4 Discussion

273 4.1 Increased BD application rate did not increase N_2O and N_2 losses probably due to inhibitory effect 274 of high NH_4^+ concentrations

275 The overall N_2O fluxes corresponded well with those from other studies with similar incubation conditions and
276 application rates of BD in terms of $N\ ha^{-1}$ (Severin et al., 2015; Senbayram et al., 2012; Köster et al., 2015).
277 However, the latter studies assumed a distribution of BD into soil by a cultivator, which implies a smaller
278 concentration of BD compared to its occurrence in injection slits. Although we observed differences in N_2O
279 emissions between soils, soil type was not confirmed as a significant effect. Nevertheless, WFPS and
280 temperature, which are well known controllers of N_2O generation (Maag and Vinther, 1999), showed significant
281 influences. Both are physical (by gas diffusion) and biological (by increased metabolic activity and consequently
282 increased O_2 consumption by respiration) drivers for O_2 availability, respectively (Maag and Vinther, 1999; Ball,
283 2013). Accordingly, the CO_2 flux (resulting from respiration of O_2) generally increased with temperature and
284 was also identified as significant by regression selection.

285 The mean N_2 fluxes of up to 0.5 (loamy sand) and 3.8 $mg\ N\ m^{-2}\ h^{-1}$ (clayey silt) at 15° C (Fig. 5, Table A3) were
286 considerably smaller than the mean fluxes of up to 13.0 $mg\ m^{-2}\ h^{-1}$ observed by Köster et al. (2015) during the
287 first five days of their incubation. Although the amount of BD in terms of applied N (250 $kg\ ha^{-1}$) was
288 comparable, Köster et al. (2015) used a higher WFPS of 90%, which may have increased the generation of N_2 . In
289 contrast to N_2O emission rates, the observed N_2 fluxes depended not only on WFPS, but also on soil type (Table

290 5), most likely due to the direct influence of soil structure on diffusivity and, thus, the supply with O₂ (Balaine et
291 al. 2016; Butterbach-Bahl et al. 2013). N₂O flux showed also a significant effect during regression selection for
292 N₂. N₂O is the direct precursor of N₂ in **denitrification** and, hence, the flux of the latter depends on the
293 availability of the former. However, temperature showed no significant effect.

294 The N₂/(N₂+N₂O) ratios were significantly determined only by soil type and WFPS: while no clear trend was
295 observable for the loamy sand, there was a pronounced effect in the clayey silt (Fig 4). We attribute the lack of a
296 trend in the loamy sand to generally adverse conditions for the formation of N₂O and N₂. Contrary, the influence
297 of WFPS apparently mirrored favourable conditions in the clayey silt (Table 5). Simultaneously, with increasing
298 WFPS, the reduction of N₂O accelerates as an alternative electron acceptor under reduced O₂ supply (Tiedje,
299 1988). **Accordingly**, no or rather small fluxes of the investigated gaseous N species were generally found in our
300 **presumably well aerated** treatments with 35% WFPS.

301 In our study, one treatment (clayey silt, 55% WFPS, LOBD) showed exceptionally large mean N₂O fluxes of up
302 to 7.1 mg N m⁻² h⁻¹ (Fig. 3, Table A2). This could be evidence that injection of such **commonly applied amounts**
303 **of BD-N (i.e., 160 kg N ha⁻¹)** may favour much larger losses of N₂O compared to an even distribution of BD in
304 soils due to larger substrate concentration in injection slits. However, with higher amendments (i.e. HIBD), we
305 observed surprisingly partially significant (*p* < 0.05, Tuckey's HSD) reductions of N₂O and a decreasing
306 tendency of N₂ emissions (Table A2, Table A3). In line with this, the amount of BD showed a significant effect
307 during the regression selection on N₂O, but not on N₂ fluxes (Table 5). A coherent reason for the rather smaller
308 emissions of **highly** amended HIBD treatments might be the inhibitory effect of NH₃ on nitrification.
309 **Accordingly**, Anthonisen et al. (1976) found an inhibition by concentrations from 0.1 to 150 mg NH₃ L⁻¹. The
310 application rate in the treatments with HIBD amounted to approximately 500 mg NH₄⁺-N (kg soil)⁻¹ (Fig. 3)
311 **which correspond to 25.8 mg NH₃-N (kg soil)⁻¹ at 15 °C if we use the pH of the BD and assume that all**
312 **extractable NH₄⁺-N was in solution (Emerson et al., 1975). Hence**, we consider this inhibitory effect as **the**
313 **reason** for the missing increase of N₂O and N₂. Additionally, the amount of NH₄⁺ fixed as NH₃ by soil organic
314 matter increases with pH and, moreover, this fixed NH₃ is not readily extractable by the KCl method we have
315 applied (Kissel et al., 2008). This is consistent with the observation of generally higher N₂O and N₂ fluxes from
316 the clayey silt since clay increases the sorption capacity of soils for NH₄⁺ and may, thus, reduce the inhibitory
317 effect on nitrification (Kissel et al., 2008). **However, since we mixed the BD with the soil, we would expect a**
318 **lower NH₃ fixation in tubular injection slits *in situ*, resulting in probably lower N₂O and N₂ fluxes from clayey**
319 **soils.**

320 An increasing application of BD tended also to decrease the $N_2/(N_2+N_2O)$ ratio, but this effect was also not
321 significant ($p > 0.05$, Tuckey's HSD). In general, nitrite (NO_2^-) and NO_3^- are preferably reduced compared to
322 N_2O during denitrification sequence since the energy yield of each reduction step decreases from NO_3^- to N_2O
323 (Koike and Hattori, 1975). Additionally, the reaction rate of reduction is higher for NO_3^- and NO_2^- than for N_2O ,
324 which results in an accumulation of N_2O , if NO_3^- or NO_2^- concentration is not limited (Betlach and Tiedje,
325 1981). Hence, increasing application rates of BD increase the availability NO_2^- and NO_3^- from NH_4^+ oxidation
326 which, consequently, decreases N_2O reduction. However, in field situations, sooner or later an important fraction
327 of this NH_4^+ will be nitrified and can lead to further N_2O and N_2 emissions if the WFPS is at sufficient levels.
328 The inhibitory effect is in line with the strong influence of NO_3^- content of the soils after incubation (Table 5).
329 Since NO_2^- oxidising bacteria are less resilient against high concentrations of NH_3 than NH_3 oxidising bacteria
330 (Anthonisen et al., 1976), the accumulation of NO_2^- is likely. This NO_2^- protonates then partly to the toxic and
331 unstable HNO_2 , which drives biological and chemical production of NO and N_2O for detoxification (Venterea et
332 al., 2015). Hence, we suggest a dominant role of denitrification, i.e., NO_2^- reduction, in the generation of N_2O
333 during our experiment. Nevertheless, Indeed, coupled nitrification-denitrification and bacterial denitrification
334 have been found to dominate the production of N_2O directly after application of BD (Köster et al., 2011;
335 Senbayram et al., 2009). However, N_2O -N losses were clearly larger than N_2 losses under aerobic headspace in
336 the clayey silt, indicating that much of the N gas loss was driven by processes other than canonical
337 denitrification. Under the above mentioned conditions, NO -N losses may exceed N_2O losses (Venterea et al.,
338 2015), calling for taking account of NO measurements in future studies.
339 Notably, in contrast to the clayey silt, no or negligible concentrations of NO_3^- were found in all treatments with
340 loamy sand. Although we have not determined NO_2^- , it was certainly a substantial source for reduction by
341 nitrifier denitrification in this soil, especially during the anaerobic headspace conditions at the end of the
342 incubation. Actually, high NH_4^+ loads in conjunction with alkaline conditions are typical for BD (Möller and
343 Müller, 2012), which favour NO_2^- accumulation and may be the reason for the relatively small NO_3^- recovery in
344 both soils (van Cleemput and Samater, 1995).

345 4.2 Different effects of soil diffusivity on N_2O and N_2 fluxes

346 Apparently, the tested factors affected the N_2O and N_2 fluxes from both soils in a different way. A specific soil
347 characteristic that exhibits such a fundamental control on biogeochemical processes such as denitrification is the
348 diffusivity for O_2 (Ball, 2013; Letey et al., 1980; Parkin and Tiedje, 1984), which is a main soil characteristic

349 responsible for the appearance of anaerobic microsites. In general, diffusivity integrates the soil porosity, i.e.,
350 pore continuity and size as well as WFPS, which control both soil N₂O and N₂ emissions (Balaine et al., 2016;
351 Letey et al., 1980; Ball, 2013). Soils with a coarser texture like the loamy sand have a higher proportion of
352 macro-pores and thus a higher gas diffusion compared with fine textured soils like the clayey silt we used
353 (Groffman and Tiedje, 1991). This lets us expect conditions that are more favourable for N₂O and N₂ generation
354 in the latter due to **relatively poor** diffusion characteristics and, thus, a smaller O₂ supply. Actually, although we
355 incubated the soils at comparable levels of WFPS and BD amendments, the apparent lower diffusivity led to
356 larger N₂O and N₂ production in the treatments with the clayey silt in relation to the loamy sand.

357 The role of the distinct diffusivities of both soils is corroborated by our observations of the gas fluxes in
358 anaerobic headspace. With switching the He-O₂ atmosphere in the headspace to pure He, the denitrification
359 potential can be tested because anaerobicity eliminates respiration processes that use O₂ as electron acceptor
360 (Parkin and Tiedje, 1984). We acknowledge e.g. DNRA and anammox as possible additional sources of N₂O and
361 N₂, respectively, under such conditions, but we were not able to quantify their contribution. The anaerobic
362 headspace induced a considerable increase of N₂O fluxes in the loamy sand, but not in the clayey silt.
363 Concurrently, the N₂ fluxes increased in both soils, but pronounced, i.e. more than 60-fold, in the sandy loam.
364 These observed changes resulting from oxygen deprivation imply that, during the previous aerobic conditions,
365 the diffusivity of the sandy loam was too high to allow for a **sufficient** establishment of anaerobic microsites,
366 while the clayey silt **ensured a moderate diffusional constraint** to maintain hypo-aerobic conditions. However,
367 the large production rates indicate that also the loamy sand harboured the necessary microbial community able to
368 generate N₂ as soon as the atmospheric conditions become favourable. In general, only N₂O fluxes from
369 treatments with negligible fluxes during the previous aerobic period increased under anaerobic conditions. This
370 included all treatments with loamy sand and the highly amended clayey silt with 35% WFPS (Fig. 3, Table A2).
371 At the same time, there was a reduction of N₂O fluxes in all other clayey silt treatments. However, when we take
372 a closer look at the simultaneous changes of N₂ fluxes after atmosphere change, virtually all of the respective
373 treatments showed increased rates. Hence, there was an enhanced reduction of N₂O to N₂, which is reflected in
374 the increased N₂/(N₂ + N₂O) ratio (Fig. 5) and points to intensified reduction of N₂O due to the lack of oxygen
375 (Parkin and Tiedje, 1984). The much larger N₂ fluxes from the loamy sand compared to the clayey silt might
376 have been caused additionally by small NO₃⁻ availability (Fig. 2) and a high availability of C (Table 4), which
377 promoted the reduction of N₂O to N₂ (**Senbayram et al., 2012**). Alternatively, the much smaller increase of N₂
378 fluxes from the clayey silt could have resulted from depleted mineral N stocks (NO₃⁻ and NH₄⁺) due to the

379 previous gaseous N losses during the course of incubation. However, the cumulated fluxes of both N₂ and N₂O
380 amounted to a maximum absolute loss of 9.4 (1σ = 0.3) mg N per kg soil in the clayey silt with LOBD and 55%
381 WFPS, which was roughly 3.5% of the calculated NH₄⁺-N applied with BD (Fig. 2). Thus, we found no evidence
382 for any shortage of substrate in the clayey silt during the subsequent anaerobic headspace conditions. On the
383 other hand, the N₂/(N₂+N₂O) ratios increased only slightly (Fig. 5) and, in contrast to the loamy sand, there were
384 still significant N₂O fluxes in the clayey silt (Fig. 3), which point to still **sufficient** stocks of NO₃⁻ in the latter
385 (Senbayram et al., 2012). In fact, the NO₃⁻ stock was greater in the clayey silt than in loamy sand after incubation
386 (Fig. 2). Thus, we suggest that the gas fluxes were unaffected by the change to anaerobic headspace in the clayey
387 silt due to already low O₂ concentrations as a result of poor diffusivity. In conclusion, distinct gas diffusivities
388 of both soils can be **proposed** as the main reason for the differing N₂O and N₂ fluxes.

389 In interaction with soil diffusivity, also respiration affects the aerobicity of a soil matrix by concurrent
390 consumption and formation of O₂ and CO₂, respectively. Similarly to N₂O and N₂ generation by **denitrification**,
391 respiration depends on the microbial availability of carbon as well. Although anaerobic digestion reduces readily
392 degradable organic matter in BD, a 'labile' fraction usually remains, but the biodegradability of the respective
393 residual organic carbon is variable, depending on the origin of BD (Askri et al., 2015). However, DOC could be
394 used as an approximate indicator for microbial availability of carbon, though not all DOC might be readily
395 degradable (Cook and Allan, 1992). Generally, the DOC contents after our incubation increased with application
396 rate of BD (Table 4), but the DOC contents were always smaller in the clayey silt both in the not amended and
397 especially in the amended treatments. This might reflect a stronger sorption of C and thus a lower availability for
398 respiration in the clayey silt compared to the loamy sand (Kaiser and Guggenberger, 2000). If we compare the
399 DOC recoveries with the cumulated flux rates of CO₂ over the **period of aerobic headspace**, we find a good
400 regression fit ($R^2 = 0.91$, $p < 0.001$) for both soils (Fig. 6) indicating a sufficient availability of C from BD for
401 respiration and, thus, implicitly also for denitrification (Reddy et al., 1982). Moreover, as increased DOC
402 enhanced respiration (Table A1), it consequently affected O₂ consumption and, thus, also the emergence of
403 anaerobic microsites (Azam et al., 2002). **Accordingly, there is also a good correlation between cumulated CO₂**
404 **and N₂O + N₂ fluxes for the same period from the clayey silt ($R^2 = 0.93$, $p = 0.001$), when the treatments with 35**
405 **% WFPS (which showed virtually no N emissions) are omitted (Fig. 7). However, there was no such a**
406 **correlation for the loamy sand. This confirms the interactive effect of diffusivity (induced by both the soils and**
407 **WFPS) and C availability on the emissions of N₂O and N₂.** Although CO₂ fluxes were mostly higher in the
408 treatments with 320 kg compared to LOBD, this behaviour was not generally reflected in the **separate** emissions

409 of N₂O and N₂ which might be a result of the inhibitory effect of high NH₄⁺ loads on nitrification (see chapter
410 4.1). However, the N₂/(N₂O+N₂) ratios implied a tendency of N₂O reduction due to a shortage of alternative
411 electron acceptors like O₂ in the highly amended treatments. Additionally, **increasing** temperature influenced
412 indirectly the aerobic status of the soils due to increased microbial activity and, hence, respiration (Maag and
413 Vinther, 1999).

414 **4.3 No indications for BD induced short-term priming effect**

415 We further checked for a short-term priming effect after amendment with BD as suggested recently by Coban et
416 al. (2015). After balancing cumulated net CO₂-C-fluxes (difference between amended and unamended
417 treatments) against the calculated DOC-C application with BD for the period of aerobic headspace, we found no
418 evidence for a short-term priming effect. In the loamy sand with LOBD, between 76% (35% WFPS) and 103%
419 (75% WFPS) of the DOC-C had been respired (data not shown). In the respective treatments with 320 kg N ha⁻¹,
420 the CO₂-C losses ranged from 47% (35% WFPS) to 76% (75% WFPS). By contrast, only between 11% (320 kg
421 N ha⁻¹) and 42% (LOBD) has been respired in the clayey silt (both at 55% WFPS). However, if one would
422 consider the period after BD application for a longer time than we would, the light loamy sand may be
423 vulnerable for C losses after BD application than the **cloggy** clayey silt.

424 **5 Limitations**

425 Our study does not allow for conclusions about the long-term dynamics of N₂ and N₂O emissions after BD
426 application in concentrations similar to injection, since we have data for at most two days of static conditions in
427 terms of temperature and headspace aerobicity. However, it should be emphasised that our aim was to estimate
428 the effect of differing soil environmental conditions on gaseous N₂ losses. Generally, the initial phase, i.e., the
429 first week after fertilizer application, is crucial for N₂O emissions (Dobbie et al., 1999) and most probably also
430 for N₂ because the same processes are involved. Köster et al. (2011; 2015) and Senbayram et al. (2009) observed
431 in incubation experiments N₂O peaks within the first and third day, which indicate a rather immediate reaction
432 also for N₂ at least *in vitro*. Nevertheless, the former studies recorded a second plateau of N₂O emission
433 consistently after around two weeks, though, at very high WFPS. At a lower WFPS of 65%, Senbayram et al.
434 (2009) measured only one peak within two days without a repeated increase later, regardless the amount of
435 applied BD. Thus, we assume a single peak shortly after application holds also true for our incubation.
436 Moreover, on the one hand, we observed no changes of N₂O in the clayey silt under anaerobic headspace, which

437 suggest no further increase would have awaited if we had extended the incubation period with aerobic
438 headspace. The increased N₂ emissions on the last day showed the potential, which would have arisen if the soil
439 cores had been completely anaerobic. The latter has, however, no implications for mineral soils since such
440 conditions are unlikely to occur *in situ*. On the other hand, the extremely increased N₂ emissions from the loamy
441 sand on the last day verify that this soil permitted abundantly oxygen diffusion, which let us assume no
442 appearances of possible second emission increases in the former aerobic headspace. We assume also the
443 measurements after only 24 hours of anaerobicity in the headspace as representative for the emission potential
444 since microbes associated with the production of N₂O and N₂ in soils are able to react fast to changing
445 environmental conditions by utilising existing enzymes within minutes or by *de novo* synthesis within 4 – 8
446 hours (Rudaz et al., 1991). Wang et al. (2011; 2013) showed in similar studies to ours that the emission of N₂
447 and N₂O peaked within less than 24 hours after switching their headspace from aerobic to anaerobic conditions,
448 which emphasise our study design as appropriate.

449 **6 Conclusions**

450 As hypothesised, N₂O and N₂ emissions as well as the N₂/(N₂O+N₂) ratio increased with WFPS, most probably
451 due to restricted supply of O₂. Contrary to our second hypothesis, the gaseous losses of N₂O and N₂ did not
452 increase with the application rate of BD. This indicates an inhibitory effect of high NH₃ and NH₄⁺
453 concentrations, respectively, on nitrification, which are found typically in biogas digestates (BD). However, the
454 N₂/(N₂O+N₂) ratio tended to decrease with application rate as supposed, probably due to a copious supply with
455 NO₂⁻ and NO₃⁻ from oxidised BD-NH₄⁺. Confirming our third hypothesis, the fine textured clayey silt induced
456 **larger** gaseous N losses and a higher N₂/(N₂O+N₂) ratio than the coarse loamy sand by the apparent distinct
457 diffusivities of both soils. Overall, there was a larger potential for formation of N₂O in the fine-textured clayey
458 silt compared to the coarse loamy sand **after applying high concentrations of BD as they may appear after**
459 injection. However, the loamy sand showed a large potential for N₂ formation under anaerobic headspace
460 conditions. Nevertheless, further investigations are needed in regarding **the dynamics and** the duration of the
461 observed effects and their reliability for field conditions.

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470 **References**

- 471 Anthonisen, A. C., Loehr, R. C., Prakasam, T. B. S., and Srinath, E. G.: Inhibition of Nitrification by Ammonia
472 and Nitrous Acid, *Journal (Water Pollution Control Federation)*, 48, 835–852, 1976.
- 473 Askri, A., Laville, P., Trémier, A., and Houot, S.: Influence of Origin and Post-treatment on Greenhouse Gas
474 Emissions After Anaerobic Digestate Application to Soil: Waste and Biomass Valorization, *Waste Biomass*
475 *Valor*, 1–14, doi:10.1007/s12649-015-9452-6, 2015.
- 476 Azam, F., Müller, C., Weiske, A., Benckiser, G., and Ottow, J.: Nitrification and denitrification as sources of
477 atmospheric nitrous oxide – role of oxidizable carbon and applied nitrogen, *Biology and Fertility of Soils*,
478 35, 54–61, doi:10.1007/s00374-001-0441-5, 2002.
- 479 Balaine, N., Clough, T. J., Beare, M. H., Thomas, S. M., and Meenken, E. D.: Soil Gas Diffusivity Controls N₂O
480 and N₂ Emissions and their Ratio, *Soil Science Society of America Journal*, 80, 529–540,
481 doi:10.2136/sssaj2015.09.0350, 2016.
- 482 Ball, B. C.: Soil structure and greenhouse gas emissions: a synthesis of 20 years of experimentation, *Eur J Soil*
483 *Sci*, 64, 357–373, doi:10.1111/ejss.12013, 2013.
- 484 Betlach, M. R. and Tiedje, J. M.: Kinetic Explanation for Accumulation of Nitrite, Nitric Oxide, and Nitrous
485 Oxide During Bacterial Denitrification, *Applied and Environmental Microbiology*, 42, 1074–1084, 1981.
- 486 Blagodatsky, S. and Smith, P.: Soil physics meets soil biology: Towards better mechanistic prediction of
487 greenhouse gas emissions from soil, *Soil Biology and Biochemistry*, 47, 78–92,
488 doi:10.1016/j.soilbio.2011.12.015, 2012.
- 489 Butterbach-Bahl, K., Baggs, E. M., Dannenmann, M., Kiese, R., and Zechmeister-Boltenstern, S.: Nitrous oxide
490 emissions from soils: how well do we understand the processes and their controls?, *Philosophical*
491 *Transactions of the Royal Society of London B: Biological Sciences*, 368, doi:10.1098/rstb.2013.0122,
492 2013.
- 493 Butterbach-Bahl, K., Willibald, G., and Papen, H.: Soil core method for direct simultaneous determination of N₂
494 and N₂O emissions from forest soils, *Plant and Soil*, 240, 105–116, doi:10.1023/A:1015870518723, 2002.
- 495 Cameron, K. C., Di, H. J., and Moir, J. L.: Nitrogen losses from the soil/plant system: a review, *Ann Appl Biol*,
496 162, 145–173, doi:10.1111/aab.12014, 2013.
- 497 Christensen, J. P. and Rowe, G. T.: Nitrification and oxygen consumption in northwest Atlantic deep-sea
498 sediments, *Journal of Marine Research*, 42, 1099–1116, doi:10.1357/002224084788520828, 1984.

499 Coban, H., Miltner, A., Elling, F. J., Hinrichs, K. U., and Kästner, M.: The contribution of biogas residues to soil
500 organic matter formation and CO₂ emissions in an arable soil, *Soil Biology and Biochemistry*, 86, 108–115,
501 doi:10.1016/j.soilbio.2015.03.023, 2015.

502 Cook, B. D. and Allan, D. L.: Dissolved organic carbon in old field soils: Total amounts as a measure of
503 available resources for soil mineralization, *Soil Biology and Biochemistry*, 24, 585–594, doi:10.1016/0038-
504 0717(92)90084-B, 1992.

505 Davidson, E. A. and Kanter, D.: Inventories and scenarios of nitrous oxide emissions, *Environmental Research*
506 *Letters*, 9, 105012, 2014.

507 Davidson, E. A., Suddick, E. C., Rice, C. W., and Prokopy, L. S.: More Food, Low Pollution (Mo Fo Lo Po): A
508 Grand Challenge for the 21st Century, *Journal of Environmental Quality*, 44, doi:10.2134/jeq2015.02.0078,
509 2015.

510 Dell, C. J., Meisinger, J. J., and Beegle, D. B.: Subsurface Application of Manures Slurries for Conservation
511 Tillage and Pasture Soils and Their Impact on the Nitrogen Balance, *Journal of Environmental Quality*, 40,
512 doi:10.2134/jeq2010.0069, 2011.

513 Dobbie, K. E., McTaggart, I. P., and Smith, K. A.: Nitrous oxide emissions from intensive agricultural systems:
514 Variations between crops and seasons, key driving variables, and mean emission factors, *J. Geophys. Res.*,
515 104, 26891–26899, doi:10.1029/1999JD900378, 1999.

516 Eickenscheidt, T., Heinichen, J., Augustin, J., Freibauer, A., and Drösler, M.: Nitrogen mineralization and
517 gaseous nitrogen losses from waterlogged and drained organic soils in a black alder (*Alnus glutinosa* (L.)
518 Gaertn.) forest, *Biogeosciences*, 11, 2961–2976, doi:10.5194/bg-11-2961-2014, 2014.

519 Emerson, K., Russo, R. C., Lund, R. E., and Thurston, R. V.: Aqueous Ammonia Equilibrium Calculations:
520 Effect of pH and Temperature, *Journal of the Fisheries Research Board of Canada*, 32, 2379–2383,
521 doi:10.1139/f75-274, 1975.

522 Ferm, M.: Atmospheric ammonia and ammonium transport in Europe and critical loads: a review, *Nutrient*
523 *Cycling in Agroecosystems*, 51, 5–17, doi:10.1023/A:1009780030477, 1998.

524 Firestone, M. K. and Davidson, E. A.: Microbiological basis of NO and N₂O production and consumption in
525 soil, in: *Exchange of Trace Gases Between Terrestrial Ecosystems and the Atmosphere*, Andreae, M. a. S. D.
526 (Ed.), Wiley, Chichester, 7–21, 1989.

527 Friedl, J., Scheer, C., Rowlings, D. W., McIntosh, H. V., Strazzabosco, A., Warner, D. I., and Grace, P. R.:
528 Denitrification losses from an intensively managed sub-tropical pasture – Impact of soil moisture on the
529 partitioning of N₂ and N₂O emissions, *Soil Biology and Biochemistry*, 92, 58–66,
530 doi:10.1016/j.soilbio.2015.09.016, 2016.

531 Groffman, P. M., Altabet, M. A., Böhlke, J. K., Butterbach-Bahl, K., David, M. B., Firestone, M. K., Giblin, A.
532 E., Kana, T. M., Nielsen, L. P., and Voytek, M. A.: Methods for measuring denitrification: diverse
533 approaches to a difficult problem, *Ecological Applications*, 16, 2091–2122, doi:10.1890/1051-
534 0761(2006)016[2091:MFMDDA]2.0.CO;2, 2006.

535 Groffman, P. M. and Tiedje, J. M.: Relationships between denitrification, CO₂ production and air-filled porosity
536 in soils of different texture and drainage, *Soil Biology and Biochemistry*, 23, 299–302, doi:10.1016/0038-
537 0717(91)90067-T, 1991.

538 Gu, J., Nicoullaud, B., Rochette, P., Gossel, A., Hénault, C., Cellier, P., and Richard, G.: A regional experiment
539 suggests that soil texture is a major control of N₂O emissions from tile-drained winter wheat fields during
540 the fertilization period, *Soil Biology and Biochemistry*, 60, 134–141, doi:10.1016/j.soilbio.2013.01.029,
541 2013.

542 Jurasinski, G., Koebsch, F., and Hagemann, U.: flux: Flux rate calculation from dynamic closed chamber
543 measurements, 2014.

544 Kaiser, E. A., Kohrs, K., Kucke, M., Schnug, E., Heinemeyer, O., and Munch, J. C.: Nitrous oxide release from
545 arable soil: Importance of N-fertilization, crops and temporal variation, *Soil Biology and Biochemistry*, 30,
546 1553–1563, doi:10.1016/S0038-0717(98)00036-4, 1998.

547 Kaiser, K. and Guggenberger, G.: The role of DOM sorption to mineral surfaces in the preservation of organic
548 matter in soils, *Organic Geochemistry*, 31, 711–725, doi:10.1016/S0146-6380(00)00046-2, 2000.

549 Kissel, D. E., Cabrera, M. L., and Paramasivam, S.: Ammonium, Ammonia, and Urea Reactions in Soils, in:
550 Nitrogen in Agricultural Systems, *Agronomy Monographs*, American Society of Agronomy, Crop Science
551 Society of America, Soil Science Society of America, Madison, WI, 101–155, 2008.

552 Koike, I. and Hattori, A.: Energy Yield of Denitrification: An Estimate from Growth Yield in Continuous
553 Cultures of *Pseudomonas denitrificans* under Nitrate-, Nitrite- and Nitrous Oxide-limited Conditions,
554 *Microbiology*, 88, 11–19, doi:10.1099/00221287-88-1-11, 1975.

555 Komsta, L. and Novomestky, F.: moments: Moments, cumulants, skewness, kurtosis and related tests, 2015.

556 Köster, J. R., Cárdenas, L., Senbayram, M., Bol, R., Well, R., Butler, M., Mühling, K. H., and Dittert, K.: Rapid
557 shift from denitrification to nitrification in soil after biogas residue application as indicated by nitrous oxide
558 isotopomers, *Soil Biology and Biochemistry*, 43, 1671–1677, doi:10.1016/j.soilbio.2011.04.004, 2011.

559 Köster, J. R., Cárdenas, L. M., Bol, R., Lewicka-Szczebak, D., Senbayram, M., Well, R., Giesemann, A., and
560 Dittert, K.: Anaerobic digestates lower N₂O emissions compared to cattle slurry by affecting rate and
561 product stoichiometry of denitrification – An N₂O isotopomer case study, *Soil Biology and Biochemistry*,
562 84, 65–74, doi:10.1016/j.soilbio.2015.01.021, 2015.

563 Kuznetsova, A., Brockhoff, P. B., and Christensen, R. H. B.: lmerTest: Tests in Linear Mixed Effects Models,
564 2016.

565 Letey, J., Jury, W. A., Hadas, A., and Valoras, N.: Gas Diffusion as a Factor in Laboratory Incubation Studies on
566 Denitrification1, *Journal of Environmental Quality*, 9, doi:10.2134/jeq1980.00472425000900020012x,
567 1980.

568 Maag, M. and Vinther, F. P.: Effect of temperature and water on gaseous emissions from soils treated with
569 animal slurry, *Soil Science Society of America Journal*, 63, 858–865, 1999.

570 Markfoged, R., Nielsen, L. P., Nyord, T., Ottosen, L. D. M., and Revsbech, N. P.: Transient N₂O accumulation
571 and emission caused by O₂ depletion in soil after liquid manure injection, *Eur J Soil Sci*, 62, 541–550,
572 doi:10.1111/j.1365-2389.2010.01345.x, 2011.

573 Möller, K.: Effects of anaerobic digestion on soil carbon and nitrogen turnover, N emissions, and soil biological
574 activity. A review, *Agronomy for Sustainable Development*, 1–21, doi:10.1007/s13593-015-0284-3, 2015.

575 Möller, K. and Müller, T.: Effects of anaerobic digestion on digestate nutrient availability and crop growth: A
576 review, *Eng. Life Sci.*, 12, 242–257, doi:10.1002/elsc.201100085, 2012.

577 Mosier, A., Kroeze, C., Nevison, C., Oenema, O., Seitzinger, S., and van Cleemput, O.: Closing the global
578 N₂O budget: nitrous oxide emissions through the agricultural nitrogen cycle - OECD/IPCC/IEA phase II
579 development of IPCC guidelines for national greenhouse gas inventory methodology, *Nutrient Cycling in*
580 *Agroecosystems*, 52, 225–248, doi:10.1023/A:1009740530221, 1998.

581 Myhre, G., Shindell, D., Bréon, F. M., Collins, W., Fuglestedt, J., Huang, J., Koch, D., Lamarque, J. F., Lee,
582 D., and Mendoza, B.: Anthropogenic and natural radiative forcing, in: *Climate Change 2013: The Physical*
583 *Science Basis: Contribution of Working Group I to the Fifth Assessment Report of the Intergovernmental*
584 *Panel on Climate Change*, Stocker, T. F., Qin, D., Plattner, G. K., Tignor, M., Allen, S. K., Boschung, J.,
585 Nauels, A., Xia, Y., Bex, V., and Midgley, P. M. (Eds.), Cambridge University Press, Cambridge, United
586 Kingdom, New York, NY, USA, 659–740, 2013.

587 Nkoa, R.: Agricultural benefits and environmental risks of soil fertilization with anaerobic digestates: a review,
588 *Agron. Sustain. Dev.*, 1–20, doi:10.1007/s13593-013-0196-z, 2013.

589 Parkin, T. B. and Tiedje, J. M.: Application of a soil core method to investigate the effect of oxygen
590 concentration on denitrification, *Soil Biology and Biochemistry*, 16, 331–334, doi:10.1016/0038-
591 0717(84)90027-0, 1984.

592 Quakernack, R., Pacholski, A., Techow, A., Herrmann, A., Taube, F., and Kage, H.: Ammonia volatilization and
593 yield response of energy crops after fertilization with biogas residues in a coastal marsh of Northern
594 Germany, *Agriculture, Ecosystems & Environment*, 160, 66–74, doi:10.1016/j.agee.2011.05.030, 2012.

595 R Core Team: R: A Language and Environment for Statistical Computing, Vienna, Austria: [https://www.R-](https://www.R-project.org/)
596 [project.org/](https://www.R-project.org/), 2015.

597 Reddy, K. R., Rao, P. S. C., and Jessup, R. E.: The Effect of Carbon Mineralization on Denitrification Kinetics
598 in Mineral and Organic Soils¹, *Soil Science Society of America Journal*, 46, 62–68,
599 doi:10.2136/sssaj1982.03615995004600010011x, 1982.

600 Rudaz, A. O., Davidson, E. A., and Firestone, M. K.: Sources of nitrous oxide production following wetting of
601 dry soil, *FEMS Microbiology Letters*, 85, 117–124, doi:10.1111/j.1574-6968.1991.tb04703.x-i1, 1991.

602 Scholefield, D., Hawkins, J. M. B., and Jackson, S. M.: Development of a helium atmosphere soil incubation
603 technique for direct measurement of nitrous oxide and dinitrogen fluxes during denitrification, *Soil Biology*
604 *and Biochemistry*, 29, 1345–1352, doi:10.1016/S0038-0717(97)00021-7, 1997.

605 Senbayram, M., Chen, R., Budai, A., Bakken, L., and Dittert, K.: N₂O emission and the N₂O/(N₂O + N₂)
606 product ratio of denitrification as controlled by available carbon substrates and nitrate concentrations,
607 *Agriculture, Ecosystems & Environment*, 147, 4–12, doi:10.1016/j.agee.2011.06.022, 2012.

608 Senbayram, M., Chen, R., Mühling, K. H., and Dittert, K.: Contribution of nitrification and denitrification to
609 nitrous oxide emissions from soils after application of biogas waste and other fertilizers, *Rapid Commun.*
610 *Mass Spectrom.*, 23, 2489–2498, doi:10.1002/rcm.4067, 2009.

611 Senbayram, M., Chen, R., Wienforth, B., Herrmann, A., Kage, H., Mühling, K. H., and Dittert, K.: Emission of
612 N₂O from Biogas Crop Production Systems in Northern Germany, *BioEnergy Research*, 1–14,
613 doi:10.1007/s12155-014-9456-2, 2014.

614 Severin, M., Fuss, R., Well, R., Garlipp, F., and van den Weghe, H.: Soil, slurry and application effects on
615 greenhouse gas emissions, *Plant, Soil and Environment*, 61, 344–351, 2015.

616 Tiedje, J. M.: Ecology of denitrification and dissimilatory nitrate reduction to ammonium, in: *Biology of*
617 *anaerobic microorganisms*, Zehnder, A. J. B. (Ed.), John Wiley and Sons Inc, New York, 179–244, 1988.

618 Uchida, Y., Clough, T. J., Kelliher, F. M., and Sherlock, R. R.: Effects of aggregate size, soil compaction, and
619 bovine urine on N₂O emissions from a pasture soil, *Soil Biology and Biochemistry*, 40, 924–931,
620 doi:10.1016/j.soilbio.2007.11.007, 2008.

621 van Cleemput, O. and Samater, A. H.: Nitrite in soils: accumulation and role in the formation of gaseous N
622 compounds, *Fertilizer research*, 45, 81–89, doi:10.1007/BF00749884, 1995.

623 Velthof, G. L. and Mosquera, J.: The impact of slurry application technique on nitrous oxide emission from
624 agricultural soils, *Agriculture, Ecosystems & Environment*, 140, 298–308, doi:10.1016/j.agee.2010.12.017,
625 2011.

626 Venterea, R. T., Clough, T. J., Coulter, J. A., Breuillin-Sessoms, F., Wang, P., and Sadowsky, M. J.: Ammonium
627 sorption and ammonia inhibition of nitrite-oxidizing bacteria explain contrasting soil N₂O production,
628 *Scientific Reports*, 5, 12153 EP -, 2015.

629 Wang, R., Feng, Q., Liao, T., Zheng, X., Butterbach-Bahl, K., Zhang, W., and Jin, C.: Effects of nitrate
630 concentration on the denitrification potential of a calcic cambisol and its fractions of N₂, N₂O and NO, *Plant*
631 *and Soil*, 363, 175–189, doi:10.1007/s11104-012-1264-x, 2013.

632 Wang, R., Willibald, G., Feng, Q., Zheng, X., Liao, T., Brüggemann, N., and Butterbach-Bahl, K.: Measurement
633 of N₂, N₂O, NO, and CO₂ Emissions from Soil with the Gas-Flow-Soil-Core Technique, *Environ. Sci.*
634 *Technol.*, 45, 6066–6072, doi:10.1021/es1036578, 2011.

635 Webb, J., Pain, B., Bittman, S., and Morgan, J.: The impacts of manure application methods on emissions of
636 ammonia, nitrous oxide and on crop response—A review, *Special section Harvested perennial grasslands:*
637 *Ecological models for farming's perennial future*, 137, 39–46, doi:10.1016/j.agee.2010.01.001, 2010.

638 West, S. G., Finch, J. F., and Curran, P. J.: Structural equation models with nonnormal variables: Problems and
639 remedies, in: *Structural equation modeling: Concepts, issues, and applications*, Hoyle, R. H. (Ed.), Sage,
640 Thousand Oaks, 56–75, 1995.

641 Wulf, S., Maeting, M., and Clemens, J.: Application technique and slurry co-fermentation effects on ammonia,
642 nitrous oxide, and methane emissions after spreading: II. Greenhouse gas emissions, *Journal of*
643 *Environmental Quality*, 31, 1795–1801, 2002.

644 Zhang, X., Davidson, E. A., Mauzerall, D. L., Searchinger, T. D., Dumas, P., and Shen, Y.: Managing nitrogen
645 for sustainable development, *NATURE*, 528, 51–59, 2015.

646

647 **Table 1: The examined factors soil texture, water-filled pore space (WFPS), and amount (i.e., concentration) of**
 648 **nitrogen (N) applied with biogas digestate (BD) with their respective levels applied in the present study, resulting in**
 649 **18 treatments with three replicates each. The temperature was manipulated consecutively during the incubation.**

Factor [<i>n</i>]	Levels		
Soil texture [2]	loamy sand	clayey silt	
WFPS (%) [3]	35	55	75
BD-N (kg ha ⁻¹) [3]	0	160	320
Temperature (°C) [2]	2	15	

650

651 **Table 2: Characteristics of both soils. Texture and mean values with standard deviations (in brackets) for carbon (C, $n = 9$), nitrogen (N, $n = 9$), pH ($n = 3$), bulk density (BD, $n =$**
 652 **3) and mineral N (NO_3^- and NH_4^+ , $n = 3$) of both soils in 0–10 cm depth after field sampling.**

Texture	C (mg g^{-1}) ^a	N (mg g^{-1}) ^a	pH ^b	Bulk density (g cm^{-3}) ^c	NO_3^- (mg kg^{-1}) ^d	NH_4^+ (mg kg^{-1}) ^d
Loamy sand	6.99 (0.29)	0.67 (0.05)	7.2 (0.1)	1.4 (0.0)	1.0 (0.2)	0.6 (0.3)
Clayey silt	10.77 (0.28)	1.19 (0.06)	7.2 (0.0)	1.5 (0.0)	1.8 (0.2)	0.3 (0.2)

653 ^a measured with analyser “Truspec CNS”, Leco Instruments GmbH, Germany, performed according to ISO 10694 (“elemental analysis”) for C and according to ISO

654 13878 (“elemental analysis”) for N

655 ^b measured in H_2O with TitraMaster85, Radiometer Analytical SAS, France, performed according to VDLUFA Methodenbuch, Vol. 1, chap. 5.1.1

656 ^c measured on 250 cm^3 soil cores

657 ^d measured with analyser “CFA-SAN”, Skalar Analytical B.V., the Netherlands, performed according to ISO 14256

658 **Table 3: Chronological order of the incubated factor combinations. Two different factor combinations with their**
 659 **respective repetitions ($n = 3$) were placed together for each weekly incubation course (cf. Fig. 1). The factors were**
 660 **combined by (1) soil (loamy sand: LS, clayey silt: CS), (2) amount (kg) of applied N from digestate per ha and (3)**
 661 **WFPS (%).**

Week	Factor combination 1	Factor combination 2
1	LS - 0 N - 35%	LS - 0 N - 55%
2	LS - 0 N - 75%	LS - 160 N - 35%
3	LS - 160 N - 55%	LS - 160 N - 75%
4	LS - 320 N - 35%	LS - 320 N - 55%
5	LS - 320 N - 75%	CS - 0 N - 35%
6	CS - 0 N - 55%	CS - 0 N - 75%
7	CS - 160 N - 35%	CS - 160 N - 55%
8	CS - 160 N - 75%	CS - 320 N - 35%
9	CS - 320 N - 55%	CS - 320 N - 75%

662

663 **Table 4: Mean recovered DOC values from soils after incubation with standard deviations in brackets for the**
 664 **respective treatments differing in amount of applied biogas digestate (BD) and water-filled pore space (WFPS).**

kg digestate-N ha ⁻¹	WFPS (%)	mg DOC (kg soil) ⁻¹	
		Loamy sand	Clayey silt
0	35	41.4 (2.7)	18.9 (1.1)
	55	38.6 (3.1)	19.8 (1.4)
	75	43.7 (1.4)	19.0 (1.8)
160	35	197.4 (20.7)	n.a.
	55	190.5 (19.3)	68.3 (12.7)
	75	362.2 (40.0)	63.2 (9.6)
320	35	316.8 (25.3)	358.1 (26.3)
	55	312.5 (14.3)	94.8 (13.6)
	75	500.1 (33.4)	105.9 (14.8)

665 n.a.: data not available

666 **Table 5: ANOVA table (type 2, *p*-values calculated based on Satterthwaite's approximation) of the linear mixed effects models for estimated fluxes of N₂, N₂O, N₂/(N₂+N₂O)**
 667 **product ratio and CO₂ in aerobic He-O₂ atmosphere. Soil type, water-filled pore space (WFPS), amount of digestate, temperature, NO₃⁻ and DOC content of soil after incubation**
 668 **as well as fluxes of N₂O and CO₂ were set as possible independent variables. The individual soil rings were set as random effect (nested within the respective week and with the**
 669 **allowance for varying slopes for each day of measurements). The random effect was always significant.**

670

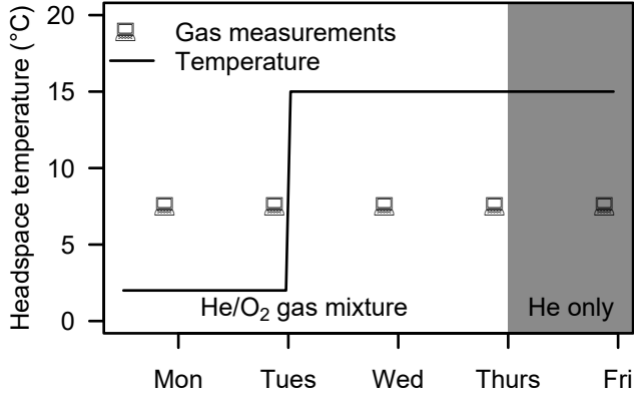
Response	Fixed effects							
	Soil type	WFPS	Digestate amount	Temperature	NO ₃ ⁻ post	DOC post	N ₂ O flux	CO ₂ flux
N ₂	0.026	< 0.001	0.008	0.037	†	0.001	< 0.001	†
N ₂ O	†	< 0.001	< 0.001	< 0.001	†	< 0.001	*	< 0.001
N ₂ /(N ₂ +N ₂ O)	0.005	0.004	†	†	†	†	*	†
CO ₂	<0.001	†	<0.001	<0.001	†	0.007	†	*

671 † Variable eliminated during stepwise regression selection

672 * Variable was not included into original regression

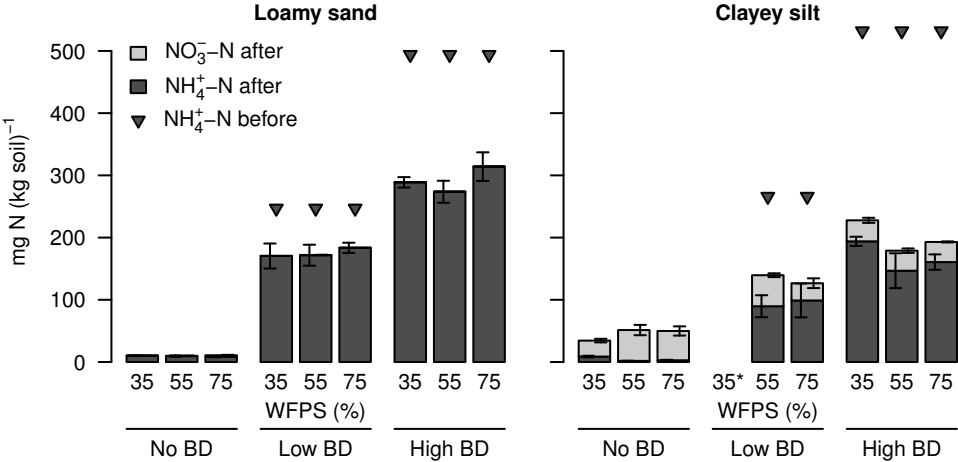
673 **Figure 1: Course of incubation and gas measurements with respect to atmosphere and temperature of the headspace**
674 **after two days of pre-incubation at 2 °C in He/O₂ gas mixture. Gas concentrations of the headspace were determined**
675 **on five consecutive days, i.e. Monday to Friday in the morning. After the first two measurement days, the headspace**
676 **temperature was increased from 2 to 15 °C. Additionally, after the fourth measurement day, the aerobic**
677 **Helium/oxygen gas mixture in the headspace was replaced by a pure Helium atmosphere.**

678

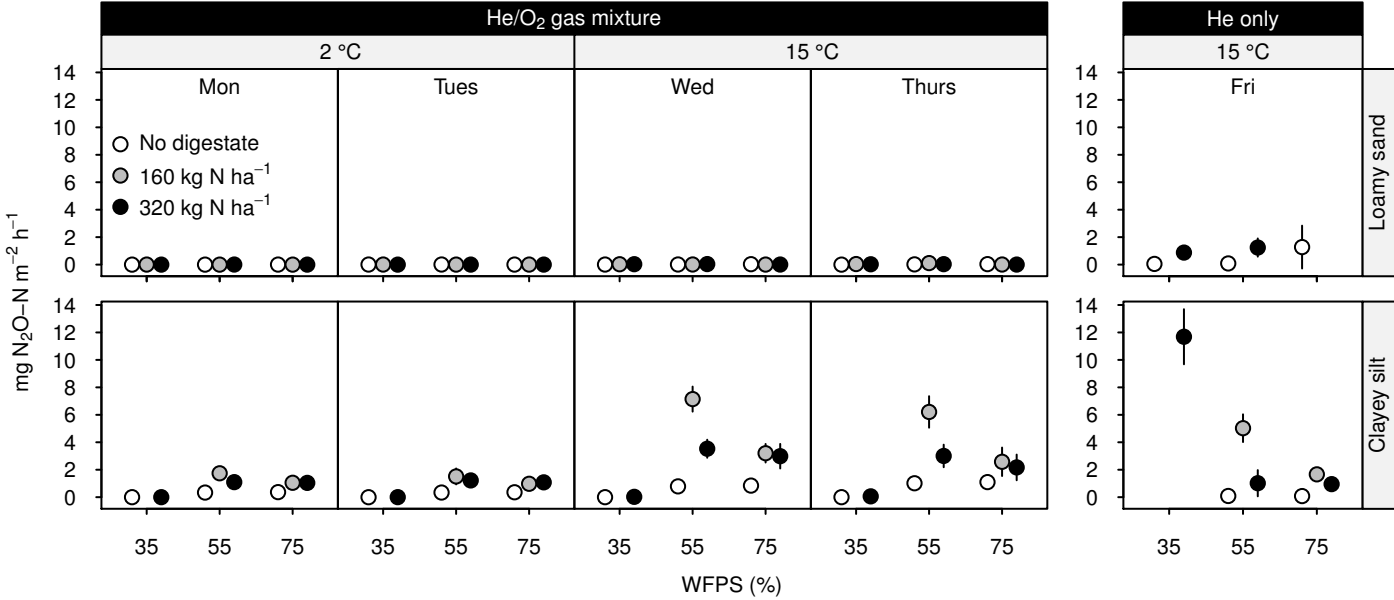


679 **Fig. 2: Ammonium and nitrate contents from loamy sand and clayey silt after incubation with different water-filled**
680 **pore spaces (WFPS, %) and amounts of digestate (0 mL per sample ring: 'No BD', 17.6 mL: 'Low BD' and 35.2 mL:**
681 **'High BD'). Error bars denote standard deviations. In general, the ammonium content increased with digestate**
682 **application with lower amounts detected in the clayey silt. Nitrate was found almost exclusively in the latter soil. For**
683 **comparison, calculated amounts of ammonium applied with biogas digestate are shown by triangles. One treatment**
684 **(*) was omitted from all analyses due to technical reasons.**

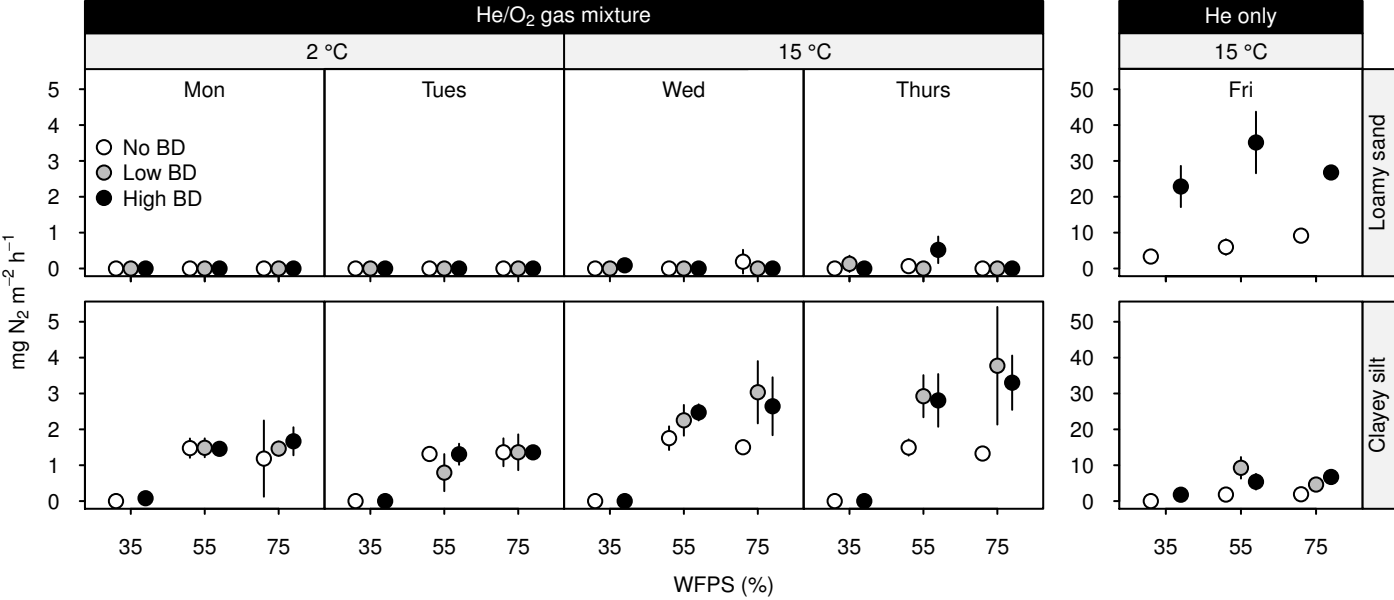
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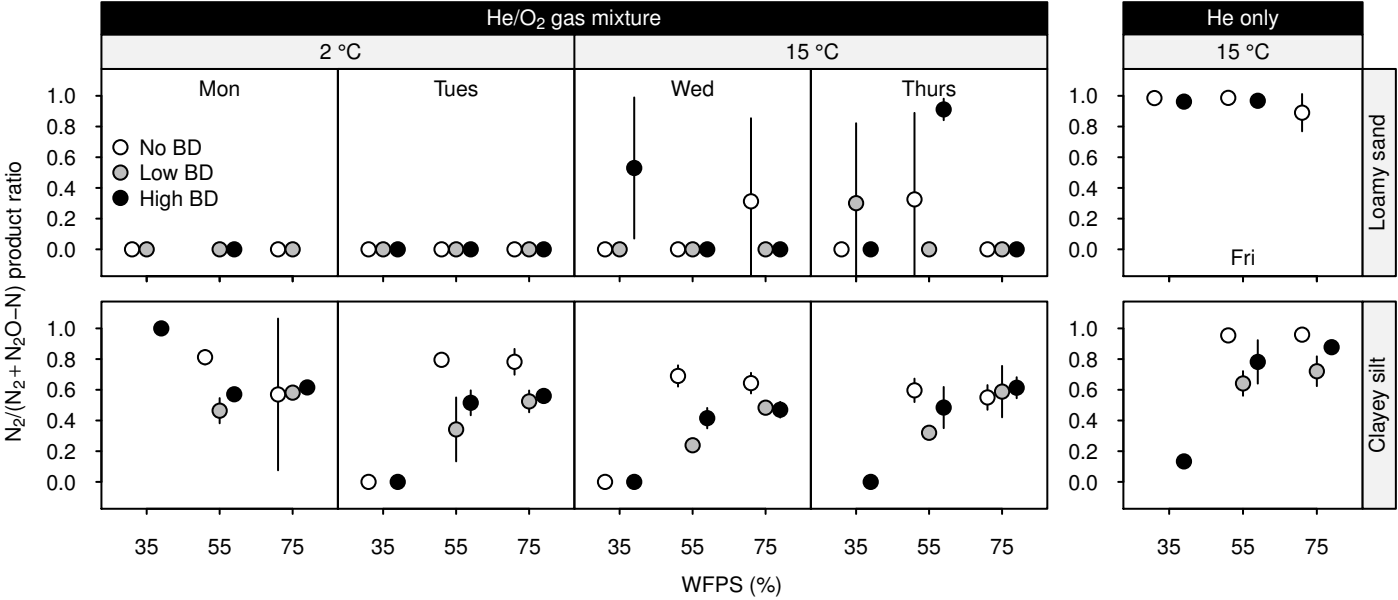
686 Fig. 3: Mean N₂O fluxes (mg N m⁻² h⁻¹) from a loamy sand and a clayey silt incubated under different water-filled pore
687 spaces (WFPS, %) with different amounts of digestate (0 mL per sample ring: 'No BD', 17.6 mL: 'Low BD' and 35.2
688 mL: 'High BD'). The first till the fourth day of the incubation were measured in an aerobic He-O₂ headspace (with
689 two days at 2 °C followed by another two days at 15 °C) while on the fifth day measurements were conducted in an
690 anaerobic headspace with pure He (at 15 °C). Error bars show standard deviations; if bars are not visible, they are
691 smaller than the symbols of the means. Under aerobic atmosphere, N₂O fluxes from loamy sand were negligible, while
692 fluxes from clayey silt showed an increase with temperature, especially with higher WFPS and intermediate amounts
693 of digestate. Under anaerobic atmosphere, mean fluxes from loamy sand increased slightly, but significantly (Tukey's
694 HSD, *p* < 0.05). The fluxes from clayey silt showed no significant differences (Tukey's HSD, *p* < 0.05) compared to the
695 day before, with the exception of 35% WFPS, where mean flux increased strongly in the treatment with 320 kg
696 digestate-N ha⁻¹.



697 Fig. 4: Mean N₂ fluxes (mg m⁻² h⁻¹) from a loamy sand and a clayey silt incubated under different water-filled pore
698 spaces (WFPS, %) with different amounts of digestate (0 mL per sample ring: 'No BD', 17.6 mL: 'Low BD' and 35.2
699 mL: 'High BD'). The first till the fourth day of the incubation were measured in an aerobic He-O₂ headspace (with
700 two days at 2 °C followed by another two days at 15 °C) while on the fifth day measurements were conducted in an
701 anaerobic headspace with pure He (at 15 °C). Error bars show standard deviations; if bars are not visible, they are
702 smaller than the symbols of the means. The dotted horizontal lines depict the average blank value; single flux rates
703 lower than the respective blank value were set zero. Under aerobic atmosphere, N₂ fluxes from loamy sand were zero
704 or rather negligible, while fluxes from clayey silt show a distinct increase with WFPS and higher fluxes at 15 °C.
705 Under anaerobic atmosphere, mean fluxes from loamy sand increased by orders of magnitude, while the fluxes from
706 clayey silt increased as well, but more gently compared to the sand.

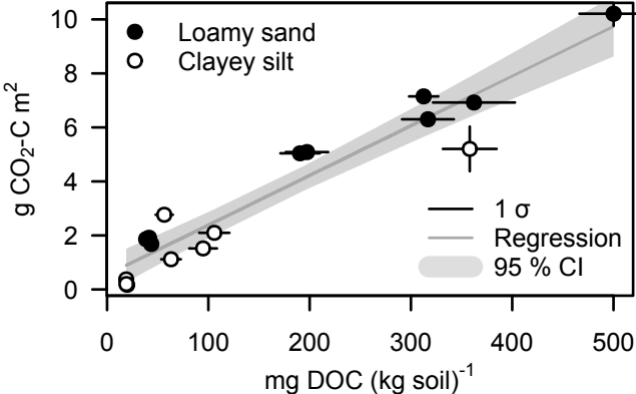


707 **Fig. 5: Mean $N_2/(N_2 + N_2O-N)$ product ratio from a loamy sand and a clayey silt incubated under different water-filled**
708 **pore spaces (WFPS, %) with different amounts of digestate (0 mL per sample ring: 'No BD', 17.6 mL: 'Low BD' and**
709 **35.2 mL: 'High BD'). The first till the fourth day of the incubation were measured in an aerobic He-O₂ headspace**
710 **(with two days at 2 °C followed by another two days at 15 °C) while on the fifth day measurements were conducted**
711 **in an anaerobic headspace with pure He (at 15 °C). Error bars show standard deviations; if bars are not visible, they**
712 **are smaller than the symbols of the means. For the loamy sand, there was a clear distinction of the ratios between**
713 **aerobic and anaerobic atmospheres: while the ratios tended to 0 in the former, they tended to 1 in the latter,**
714 **irrespectively of temperature or amount of digestate. For the clayey silt, ratios increased with WFPS and were highest**
715 **from the unamended treatments under both the aerobic and the anaerobic atmospheres.**

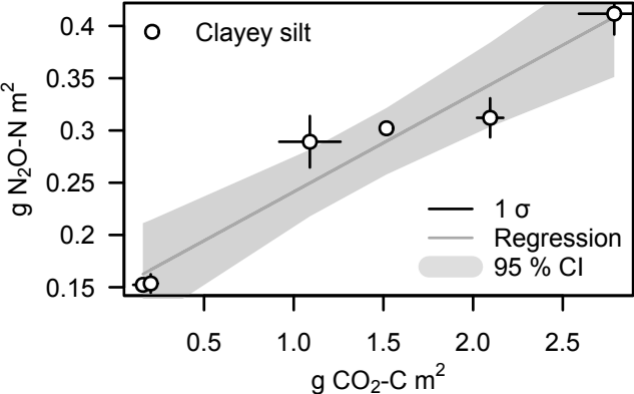


716 **Fig. 6: Regression between DOC recoveries (mg per 100 g soil) after the incubation and the respective cumulated CO₂**
717 **emissions (g C m⁻²) during the period of aerobic headspace with their standard deviations and confidence interval**
718 **(95%). If error bars are not visible, they are smaller than the symbols of the means. Both soils showed increasing**
719 **emissions with increasing soil DOC contents as well a good regression fit ($R^2 = 0.91, p < 0.001$).**

720



721 **Fig. 7: Regression between cumulated CO₂ emissions (g C m⁻²) and the respective cumulated N₂O + N₂ emissions (g N**
722 **m⁻²) from the clayey silt with WFPS > 35 % during the period of aerobic headspace with their standard deviations**
723 **and confidence interval (95%). If error bars are not visible, they are smaller than the symbols of the means. The**
724 **proportional increase of CO₂ and the N gas species shows a good regression fit of $R^2 = 0.93$, ($p = 0.001$).**



725 **Table A1: Mean CO₂-C fluxes with standard deviations in mg m⁻² h⁻¹ from the loamy sand and the clayey silt, treated**
726 **with different water-filled pore spaces (WFPS, %), amounts of digestate (kg N ha⁻¹) as well as different temperature**
727 **regimes (°C) under aerobic (He-O₂) and anaerobic (He) atmosphere. Column 'Day' denotes the consecutive measuring**
728 **days of the respective incubation cycle. Different letters after fluxes indicate significant differences (Tukey's HSD, *p* <**
729 **0.05) within each soil and measuring day. Zeros as last digits were omitted.**

Day	Atmosphere	Temperature (°C)	WFPS (%)	kg N ha ⁻¹	mg CO ₂ -C m ⁻² h ⁻¹	
					Loamy sand	Clayey silt
1	He-O ₂	2	35	0	6.8 ± 2.4 cd	0 ± 0 c
1	He-O ₂	2	35	160	22 ± 3.5 bcd	NA
1	He-O ₂	2	35	320	23.3 ± 9.3 bc	22.8 ± 2.8 ab
1	He-O ₂	2	55	0	6 ± 0.7 d	4.6 ± 7.9 bc
1	He-O ₂	2	55	160	34.4 ± 3.1 b	34.5 ± 11.6 a
1	He-O ₂	2	55	320	28 ± 3.2 b	15.9 ± 3.4 abc
1	He-O ₂	2	75	0	9.4 ± 1.4 cd	0 ± 0 c
1	He-O ₂	2	75	160	37.5 ± 6 b	15.5 ± 12.1 abc
1	He-O ₂	2	75	320	68.3 ± 12.1 a	24.5 ± 2.7 a
2	He-O ₂	2	35	0	9.8 ± 3.5 c	1.3 ± 1.4 b
2	He-O ₂	2	35	160	23 ± 3.9 bc	NA
2	He-O ₂	2	35	320	30.9 ± 2.2 b	22.2 ± 2.4 a
2	He-O ₂	2	55	0	8.7 ± 1.5 c	0.6 ± 1 b
2	He-O ₂	2	55	160	33.4 ± 0.9 b	27.6 ± 12.3 a
2	He-O ₂	2	55	320	35.9 ± 2.7 b	14.4 ± 1.9 ab
2	He-O ₂	2	75	0	8.3 ± 1.5 c	0 ± 0 b
2	He-O ₂	2	75	160	31.9 ± 3 b	13 ± 9.3 ab
2	He-O ₂	2	75	320	57.6 ± 14.8 a	18.3 ± 4 a
3	He-O ₂	15	35	0	42.5 ± 4.5 c	6.7 ± 0.7 b
3	He-O ₂	15	35	160	114.3 ± 12.2 b	NA
3	He-O ₂	15	35	320	149.5 ± 9.4 b	130.9 ± 105 a
3	He-O ₂	15	55	0	41.3 ± 3.5 c	3.2 ± 0.4 b
3	He-O ₂	15	55	160	108.7 ± 10.1 b	57.8 ± 12.2 bc
3	He-O ₂	15	55	320	162.1 ± 9.6 b	26.8 ± 0.7 bc
3	He-O ₂	15	75	0	44.1 ± 9.8 c	3.2 ± 0.7 b
3	He-O ₂	15	75	160	150.4 ± 19 b	26.4 ± 11.8 bc
3	He-O ₂	15	75	320	249.7 ± 53.5 a	35.3 ± 6 bc
4	He-O ₂	15	35	0	48.7 ± 6 c	15.1 ± 4.9 cd
4	He-O ₂	15	35	160	114.3 ± 6.4 b	NA
4	He-O ₂	15	35	320	156.9 ± 15.4 a	65.7 ± 2.2 a
4	He-O ₂	15	55	0	48 ± 3.4 c	4.2 ± 0.2 d
4	He-O ₂	15	55	160	109 ± 14.4 b	51.2 ± 15.1 ab
4	He-O ₂	15	55	320	177.7 ± 7.5 a	26.6 ± 2.3 cd

4	He-O ₂	15	75	0	34 ± 7.8 c	6.7 ± 4 d
4	He-O ₂	15	75	160	168.7 ± 0.4 a	22.1 ± 14.8 cd
4	He-O ₂	15	75	320	166.3 ± 23.1 a	34.1 ± 5.7 bc
5	He	15	35	0	11.2 ± 0.6 d	NA
5	He	15	35	160	54.8 ± 9.3 c	NA
5	He	15	35	320	149.3 ± 3.9 a	45.8 ± 2.1 a
5	He	15	55	0	13.6 ± 1.9 d	3.4 ± 0.6 c
5	He	15	55	160	55.2 ± 4.4 bc	32 ± 11.4 ab
5	He	15	55	320	164.5 ± 3.5 a	15.2 ± 10.7 bc
5	He	15	75	0	20.9 ± 2.3 d	3.6 ± 0.1 c
5	He	15	75	160	75 ± 7.3 b	20.6 ± 8.5 bc
5	He	15	75	320	NA	26.1 ± 2.6 ab

730

731 **Table A2: Mean N₂O-N fluxes with standard deviations in mg m⁻² h⁻¹ from the loamy sand and the clayey silt, treated**
732 **with different water-filled pore spaces (WFPS, %), amounts of digestate (kg N ha⁻¹) as well as different temperature**
733 **regimes (°C) under aerobic (He-O₂) and anaerobic (He) atmosphere. Column 'Day' denotes the consecutive measuring**
734 **days of the respective incubation cycle. Different letters after fluxes indicate significant differences (Tukey's HSD, *p* <**
735 **0.05) within each soil and measuring day. Zeros as last digits were omitted.**

Day	Atmosphere	Temperature (°C)	WFPS (%)	kg N ha ⁻¹	mg N ₂ O-N m ⁻² h ⁻¹	
					Loamy sand	Clayey silt
1	He-O ₂	2	35	0	0 ± 0	0 ± 0 c
1	He-O ₂	2	35	160	0 ± 0	NA
1	He-O ₂	2	35	320	0 ± 0	0 ± 0 c
1	He-O ₂	2	55	0	0 ± 0	0.3 ± 0.1 c
1	He-O ₂	2	55	160	0 ± 0	1.7 ± 0.4 a
1	He-O ₂	2	55	320	0 ± 0	1.1 ± 0.1 b
1	He-O ₂	2	75	0	0 ± 0	0.4 ± 0.1 c
1	He-O ₂	2	75	160	0 ± 0	1 ± 0.1 b
1	He-O ₂	2	75	320	0 ± 0	1 ± 0.2 b
2	He-O ₂	2	35	0	0 ± 0	0 ± 0 d
2	He-O ₂	2	35	160	0 ± 0	NA
2	He-O ₂	2	35	320	0 ± 0	0 ± 0 cd
2	He-O ₂	2	55	0	0 ± 0	0.3 ± 0.1 bc
2	He-O ₂	2	55	160	0 ± 0	1.5 ± 0.6 a
2	He-O ₂	2	55	320	0 ± 0	1.2 ± 0.2 a
2	He-O ₂	2	75	0	0 ± 0	0.4 ± 0.1 bc
2	He-O ₂	2	75	160	0 ± 0	1 ± 0.1 ab
2	He-O ₂	2	75	320	0 ± 0	1.1 ± 0.2 a
3	He-O ₂	15	35	0	0 ± 0 cd	0 ± 0 c
3	He-O ₂	15	35	160	0 ± 0 abc	NA
3	He-O ₂	15	35	320	0 ± 0 ab	0 ± 0 c
3	He-O ₂	15	55	0	0 ± 0 bcd	0.8 ± 0.2 c
3	He-O ₂	15	55	160	0 ± 0 bcd	7.1 ± 0.9 a
3	He-O ₂	15	55	320	0 ± 0 a	3.5 ± 0.7 b
3	He-O ₂	15	75	0	0 ± 0 ab	0.8 ± 0.2 c
3	He-O ₂	15	75	160	0 ± 0 d	3.2 ± 0.7 b
3	He-O ₂	15	75	320	0 ± 0 cd	3 ± 0.9 b
4	He-O ₂	15	35	0	0 ± 0 b	0 ± 0 c
4	He-O ₂	15	35	160	0 ± 0 ab	NA
4	He-O ₂	15	35	320	0 ± 0 ab	0.1 ± 0.1 c
4	He-O ₂	15	55	0	0 ± 0 b	1 ± 0.2 bc
4	He-O ₂	15	55	160	0.1 ± 0.1 a	6.2 ± 1.1 a
4	He-O ₂	15	55	320	0 ± 0 ab	3 ± 0.8 b

4	He-O ₂	15	75	0	0 ± 0 ab	1.1 ± 0.3 bc
4	He-O ₂	15	75	160	0 ± 0 b	2.6 ± 1 b
4	He-O ₂	15	75	320	0 ± 0 b	2.2 ± 0.9 b
5	He	15	35	0	0.1 ± 0	NA
5	He	15	35	160	NA	NA
5	He	15	35	320	0.9 ± 0.1	11.7 ± 2 a
5	He	15	55	0	0.1 ± 0	0.1 ± 0 c
5	He	15	55	160	NA	5 ± 1 b
5	He	15	55	320	1.2 ± 0.7	1.4 ± 0.8 c
5	He	15	75	0	1.3 ± 1.6	0.1 ± 0 c
5	He	15	75	160	NA	1.7 ± 0.3 c
5	He	15	75	320	NA	1 ± 0.3 c

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737 **Table A3: Mean N₂ fluxes with standard deviations in mg m⁻² h⁻¹ from the loamy sand and the clayey silt, treated with**
738 **different water-filled pore spaces (WFPS, %), amounts of digestate (kg N ha⁻¹) as well as different temperature**
739 **regimes (°C) under aerobic (He-O₂) and anaerobic (He) atmosphere. Column 'Day' denotes the consecutive measuring**
740 **days of the respective incubation cycle. Different letters after fluxes indicate significant differences (Tukey's HSD, *p* <**
741 **0.05) within each soil and measuring day. Zeros as last digits were omitted.**

	Atmosphere	Temperature (°C)	WFPS (%)	kg N ha ⁻¹	mg N ₂ m ⁻² h ⁻¹	
					Loamy sand	Clayey silt
1	He-O ₂	2	35	0	0 ± 0	0 ± 0 bc
1	He-O ₂	2	35	160	0 ± 0	NA
1	He-O ₂	2	35	320	0 ± 0	0.1 ± 0.1 bc
1	He-O ₂	2	55	0	0 ± 0	1.5 ± 0.3 a
1	He-O ₂	2	55	160	0 ± 0	1.5 ± 0.3 a
1	He-O ₂	2	55	320	0 ± 0	1.5 ± 0 a
1	He-O ₂	2	75	0	0 ± 0	1.2 ± 1.1 a
1	He-O ₂	2	75	160	0 ± 0	1.5 ± 0.2 a
1	He-O ₂	2	75	320	0 ± 0	1.7 ± 0.4 a
2	He-O ₂	2	35	0	0 ± 0	0 ± 0 c
2	He-O ₂	2	35	160	0 ± 0	NA
2	He-O ₂	2	35	320	0 ± 0	0 ± 0 c
2	He-O ₂	2	55	0	0 ± 0	1.3 ± 0.1 a
2	He-O ₂	2	55	160	0 ± 0	0.8 ± 0.5 b
2	He-O ₂	2	55	320	0 ± 0	1.3 ± 0.3 a
2	He-O ₂	2	75	0	0 ± 0	1.4 ± 0.4 a
2	He-O ₂	2	75	160	0 ± 0	1.4 ± 0.5 a
2	He-O ₂	2	75	320	0 ± 0	1.4 ± 0.1 a
3	He-O ₂	15	35	0	0 ± 0 b	0 ± 0 e
3	He-O ₂	15	35	160	0 ± 0 b	NA
3	He-O ₂	15	35	320	0.1 ± 0.1 ab	0 ± 0 e
3	He-O ₂	15	55	0	0 ± 0 b	1.8 ± 0.3 cd
3	He-O ₂	15	55	160	0 ± 0 b	2.3 ± 0.4 bc
3	He-O ₂	15	55	320	0 ± 0 b	2.5 ± 0.2 ab
3	He-O ₂	15	75	0	0.2 ± 0.3 a	1.5 ± 0.2 d
3	He-O ₂	15	75	160	0 ± 0 b	3 ± 0.9 a
3	He-O ₂	15	75	320	0 ± 0 b	2.6 ± 0.8 ab
4	He-O ₂	15	35	0	0 ± 0 b	0 ± 0 c
4	He-O ₂	15	35	160	0.1 ± 0.2 b	NA
4	He-O ₂	15	35	320	0 ± 0 b	0 ± 0 c
4	He-O ₂	15	55	0	0.1 ± 0.1 b	1.5 ± 0.2 b
4	He-O ₂	15	55	160	0 ± 0 b	2.9 ± 0.6 a
4	He-O ₂	15	55	320	0.5 ± 0.4 a	2.8 ± 0.7 a

4	He-O ₂	15	75	0	0 ± 0 b	1.3 ± 0.2 bc
4	He-O ₂	15	75	160	0 ± 0 b	3.8 ± 1.6 a
4	He-O ₂	15	75	320	0 ± 0 b	3.3 ± 0.8 a
5	He	15	35	0	3.3 ± 0.4 d	0 ± 0 c
5	He	15	35	160	NA	NA
5	He	15	35	320	22.9 ± 5.7 b	1.8 ± 0.1 c
5	He	15	55	0	6 ± 2.2 cd	1.8 ± 0.2
5	He	15	55	160	NA	9.5 ± 2.7 a
5	He	15	55	320	35.1 ± 8.6 a	5.1 ± 1.8 bc
5	He	15	75	0	9.2 ± 0.4 c	1.9 ± 0.1 c
5	He	15	75	160	NA	4.8 ± 1.6 bc
5	He	15	75	320	26.8 ± 1.1 b	6.7 ± 0.8 b

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