¹ Potential short-term losses of N₂O and N₂ from high

² concentrations of biogas digestate depend mainly on soil

texture and moisture, not on NH₄⁺

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11 Abstract. Biogas digestate (BD) is increasingly used as organic fertiliser, but has a high potential for NH_3 12 losses. Its proposed injection into soils as a counter-measure has been suggested to promote the generation of N_2O_1 , leading to a potential trade-off. Furthermore, the effect of high nutrient concentrations on N_2 losses as they 13 14 may appear after injection of BD into soil has not yet been evaluated. Hence, we performed an incubation 15 experiment with soil cores in a helium-oxygen atmosphere to examine the influence of soil substrate (loamy sand, clayey silt), water-filled pore space (WFPS; 35, 55, 75%) and application rate (0, 17.6 and 35.2 mL BD per 16 17 soil core [250 cm³]) on the emissions of N_2O , N_2 and CO_2 after the usage of high loads of BD. To determine the 18 potential capacity for gaseous losses, we applied anaerobic conditions by purging with helium for the last 24 h of 19 incubation. Immediate N_2O and N_2 emissions as well as the $N_2/(N_2O+N_2)$ ratio depended on soil type and 20 increased with WFPS indicating a crucial role of soil gas diffusivity for the formation of nitrogenous gases in 21 agricultural soils. However, the emissions did not increase with the application rate of BD probably due to an 22 inhibitory effect of the high NH_4^+ content in BD. Our results suggest a larger potential for N_2O formation in the 23 fine-textured clayey silt compared to the coarse loamy sand after applying high concentrations of BD as 24 appearing after injection. However, the loamy sand showed basically a large potential for N_2 formation under 25 anaerobic headspace conditions. Nevertheless, our results show the need for further investigations on the 26 dynamics and the duration of the observed effects and their significance for field conditions.

27 1 Introduction

Nitrous oxide (N₂O) is a potent greenhouse gas (Myhre et al., 2013), with agriculture being the largest single 28 source of anthropogenic N_2O emissions, contributing about 4.1 Tg N_2O -N yr⁻¹ or 66% of total gross 29 30 anthropogenic emissions mainly as a result of mineral nitrogen (N) fertiliser and manure application (Davidson 31 and Kanter, 2014). The generation of nitrogen gas (N2) is of agronomic interest in terms of nutrient management, 32 since such gaseous losses may imply a significant loss of N from the soil/plant system (Friedl et al., 2016; Cameron et al., 2013). However, from an environmental stance, N₂ is innocuous and, thus, the preferred type of 33 34 gaseous N-loss from soil (Davidson et al., 2015). Further, emission of ammonia (NH_3) is of environmental concern, e.g., due to acid deposition or conversion to N_2O (Ferm, 1998; Mosier et al., 1998). In general, the 35 improvement of N use efficiency and thus the decrease of N losses in crop production are paramount in the 36 37 presence of challenges like food security, environmental degradation and climate change (Zhang et al., 2015).

38 In Germany, the increased demand for renewable energy sources like methane from biogas plants entails an 39 expanded amount of digestion residues (biogas digestate, BD) used as organic amendment in agriculture (Möller 40 and Müller, 2012). Digestion in biogas reactors increases pH and the proportion of ammonium (NH_4^+) and 41 narrows the C to N ratio due to the depletion of labile C fractions of the feedstock (Möller and Müller, 2012). 42 Compared to undigested amendments like animal slurry, this altered chemical properties may promote 43 biochemical reactions in the soil that are responsible for the formation of gaseous N species like N_2O , nitric 44 oxide (NO), N_2 and especially NH₃ (Nkoa, 2013). In general, the effect strengths of BD on gaseous N losses 45 from soil is still under debate (Möller, 2015).

46 Significant losses of N as NH₃ can occur within the first hours after manure application (Ouakernack et al., 47 2012). To reduce NH₃ losses, the application of BD by injection is recommended, but this measure can 48 simultaneously increase the potential for N₂O losses compared to surface-application (Wulf et al., 2002; Velthof 49 and Mosquera, 2011). On the one hand, high NH_4^+ concentrations in the injection band promote nitrification, 50 which is a significantly O₂ consuming process releasing N₂O (Christensen and Rowe, 1984). On the other hand, 51 increased amounts of C in the injection band also promote respiration and, thus, additionally deplete the O_2 52 supply (Dell et al., 2011). Altogether, the conditions during the initial phase after injection of BD foster 53 microsites favourable for microbial denitrification, which may promote also the formation of N_2 due to anaerobic 54 conditions (Köster et al., 2015; Webb et al., 2010).

55 There is a wealth of biotic and abiotic processes in soils that produce N_2O and N_2 , most of which are enhanced 56 by anaerobic or at least hypo-aerobic conditions (Butterbach-Bahl et al., 2013). Also the amounts and the relative 57 share of N_2 and N_2O in the overall gaseous N emissions depend – among other factors like the favoured 58 reduction of NO_3^- rather than N_2O as alternative electron acceptor – on the degree of O_2 restriction (Firestone 59 and Davidson, 1989). Soil physical and biotic factors (i.e. diffusion and consumption of O_2) as well as their interactions control the aerobic status of a soil. Diffusion of O_2 depends on the porosity of the soil substrate in 60 61 conjunction with water-filled pore space (WFPS), while O_2 is consumed by heterotrophic respiration and 62 nitrification which depend on mineral N content, carbon (C) availability as well as on temperature (Ball, 2013; 63 Uchida et al., 2008; Maag and Vinther, 1999). In general, fine textured soils with higher clay contents exhibit a 64 lower gas diffusivity compared to coarse textured soils, which result regularly in higher denitrification activity in the former with higher N_2O emission rates, but also a higher probability for the consecutive reduction to N_2 65 (Senbayram et al., 2014; Gu et al., 2013; Ball, 2013). Simultaneously, the supply of substrates for 66 67 microorganisms is determined by liquid diffusion rates in soil water and, thus, by WFPS (Blagodatsky and

Smith, 2012; Maag and Vinther, 1999). However, though high within injection bands, nutrient concentrations 68 and WFPS should theoretically increase further with the row spacing between the injection bands, if a given 69 amount of BD per area is assumed. We are not aware of studies addressing the effect of such high BD 70 71 concentrations. 72 Hence, there is a general lack of knowledge about effects of BD injection on gaseous N-losses and especially 73 about the effects of high BD concentrations and their interaction with O2 limiting factors like soil texture and 74 WFPS, as well as temperature and heterotrophic respiration. The indicated knowledge gaps are caused not the 75 least by methodological constrains with the direct determination of N_2 fluxes due to the high background level of N_2 in the atmosphere, while indirect applications like acetylene-based methods and ¹⁵N tracers are unfavourable 76 77 since the former implicates serious underestimations and the latter has rather high detection limits (Groffman et 78 al., 2006). 79 Thus, we applied the helium-oxygen (He-O₂) incubation technique (Butterbach-Bahl et al., 2002) in a laboratory 80 experiment to evaluate the effect of above suggested factors on the emission of N_2O and N_2 after soil amendment 81 with relatively high amounts of BD as they might occur after injection into soils. Simultaneously, CO₂ flux was

82 determined as an indicator for microbial O_2 consumption, O_2 diffusion and also for the degradability of organic

C applied with BD (Blagodatsky and Smith, 2012), but with the restriction that inorganic sources could not be differentiated. We hypothesised that (1) N_2O and N_2 emissions will increase with WFPS, (2) this gaseous N losses will also be affected by BD application rate, i.e. the concentration resulting from injection, and (3) the fine textured clayey silt will induce higher gaseous N losses than the coarse loamy sand.

87 2 Material and Methods

88 2.1 Selected soils, sampling of soil cores and biogas digestate

Two soils were selected and both were adjusted to three levels of WFPS and three quantities of BD (Table 1), resulting in 18 factor combinations with three repetitions each. Temperature was increased from 2 °C during the first two days to 15 °C for the last three days of the incubation. Intact soil cores (diameter 7.2 cm, height 6.1 cm, volume 250 cm³) were taken with sample rings in the range from 0–0.10 m depth from two sites with different textures, i.e. sandy loam and clayey silt. The sandy loam samples were gathered from a stagnic luvisol (IUSS Working Group WRB, 2006) located in Gülzow (North-East Germany) in the ground moraine of the Weichselian glacial period at 53° 48' 35" N and 12° 4' 20" E. The clayey silt samples were collected from a

- haplic luvisol located in Dornburg between the foothills and the lowlands of Central Germany at 51° 0' 8" N and
- 97 11° 39' 25" E (see Table 2 for more details on soil characteristics). After field sampling, the soil cores were dried
- 98 for 48 h at 40 °C to facilitate adjustment of WFPS.

Both sites have been cultivated with similar crop rotations used as feedstock for biogas production and have been amended with biogas digestate for the past nine years. The crop rotation on the sandy loam consisted of maize (*Zea mays* L.), rye (*Secale cereale* L.), sorghum (*Sorghum bicolor* (L.) MOENCH), winter triticale (× *Triticosecale* Wittmack), ryegrass (*Lolium perenne* L.) and winter wheat (*Triticum aestivum* L.). The only difference in the crop rotation on the clayey silt was the cultivation of sudangrass (*Sorghum × drummondii*) instead of sorghum.

The biogas digestate used for the incubation was obtained from a biogas plant at 'Gut Dalwitz', an organic farm in northeast Germany. The feedstock for the anaerobic fermentation in the plant consisted of 60 % maize, 20 % solid cattle manure, 10 % dry chicken manure and 10 % rye. The digestate was analysed by 'LUFA', Rostock, Germany and had a pH of 8.3, 2.91% organic C, 0.16% dissolved organic C (DOC), 0.54% N and 0.27% NH₄-N in undried material with a dry matter content of 9.4%.

110 2.2 Adjustment of WFPS and addition of N

111 For adjustment of WFPS, the dry and undisturbed soil cores were moistened dropwise. The respective quantities 112 of water were calculated based on the bulk density, an assumed particle density of 2.65 g cm⁻¹ and reduced by the expected moisture input from subsequent addition of BD. The soil cores were then mixed with BD and 113 114 finally repacked to reach nutrient concentrations comparable to that in injection bands. The mixing was done for 115 methodical reasons since the available space in the incubation vessels was limited and, hence, 'real' injection not feasible. However, injection bands have actually a thickness comparable to the sample rings we used (Markfoged 116 et al., 2011). The amounts of added BD were calculated with an assumed injection of 160 kg N ha⁻¹ into soil with 117 row spaces of 0.15 m (narrow injection bands with low BD concentration, LOBD) and 0.30 m (wide injection 118 119 bands with high BD concentration, HIBD), which are common ranges used by injection machinery and which 120 correspond to 17.6 and 25.3 mL BD, respectively, per sample ring. After this procedure, the soil cores were 121 sealed with plastic lids and stored immediately at 2 °C until the beginning of the incubation within a week.

123 The measurements of N₂, N₂O and CO₂ fluxes were applied following the He-O₂ method (Scholefield et al., 124 1997; Butterbach-Bahl et al., 2002). Six soil cores (i.e. the repetitions of two factor combinations at a time, Table 125 3) were placed simultaneously in special gas-tight incubation vessels inside a climate chamber. Analyses were conducted in the laboratory of the Institute for Landscape Biogeochemistry, Leibniz Centre for Agricultural 126 127 Landscape Research (ZALF), Müncheberg, Germany. Before flux measurements, the vessels were evacuated moderately (0.047 bar) and flushed with an artificial He/O₂ gas mixture (20.49 % O₂, 345.5 ppm CO₂, 359 ppb 128 129 N_2O_2 , 1863 ppb CH₄, 2.46 ppm N_2 , rest He) four times consecutively to remove ambient N_2 . Subsequently, the air temperature of the climate chamber was set to 2 °C and a continuous He/O₂ gas flow rate of 15 ml min⁻¹ was 130 131 applied to the vessel headspaces for 72 h to remove residues of N_2 from soil cores by diffusion, including a restricted N_2 production by decreased microbial activity. After this pre-incubation, during the following two 132 133 days, the headspace concentration of N_2O and CO_2 was measured once daily in the morning. To compensate for 134 the lower precision of the detector for N_2 in relation to the detector for N_2O and CO_2 (cf., Eickenscheidt et al., $\frac{2014}{100}$, N₂ concentrations were measured consecutively three times daily in the morning. Immediately after the 135 136 last measurement on the second day, the temperature was set to 15 °C and the measurements were continued for 137 another two days. Finally, the He/O₂ gas mixture was substituted by pure He and, following 24 h of 138 acclimatisation, gas measurements were carried out once again (Figure 1) to determine the generation of N_2O 139 and N_2 in a completely anaerobic soil matrix. The latter step is important to get a clue about the actual potential 140 for gaseous N losses after highly concentrated BD application. The settings of the chromatographs for gas 141 analyses are described in Eickenscheidt et al. (2014). Gas fluxes were calculated according to Eq. (1):

142
$$f = \frac{M \times p \times v \times dc}{R \times T \times A},$$
(1)

where *f* is the flux (N₂ and CO₂: mg m⁻² h⁻¹, N₂O: μ g m⁻² h⁻¹), *M* the molar mass in g mol⁻¹ (N₂: 28, CO₂: 44, N₂O: 44), *p* the air pressure (Pa), *v* the air flow (L h⁻¹), *R* the gas constant (8.31 J mol⁻¹ K⁻¹), *T* the temperature inside the chamber (K), *A* the area of the incubation vessel (m²), and *dc* the difference of gas concentrations (N₂ and CO₂: ppm, N₂O: ppb) between inlet and outlet of a vessel.

- 147 To enhance the tightness against atmospheric N_2 contamination, the lids of the incubation vessels were purged
- 148 permanently with helium. We obtained blank values by inserting aluminium blocks into the vessels before each
- 149 measurement cycle. Since this blank values were usually steady with means of 1.9 ($1\sigma = 0.9$) ppm N₂, 349.6 (1σ

150 = 11.4) ppb N₂O and 353.9 (1 σ = 13.5) ppm CO₂, we suggest that the vessels were tight. Derived from the blank 151 values, lowest detectable fluxes were on average 0.427 (1 σ = 0.271) mg N₂-N m⁻² h⁻¹, 3.6 (3.1) µg N₂O-N m⁻² h⁻¹ 152 and 0.918 (0.693) mg CO₂-C m⁻² h⁻¹. For flux estimation, the blank values were subtracted from the values 153 measured at the respective outlet. Estimated fluxes from the soil cores smaller than the respective blank fluxes of 154 each day were set to zero.

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156 2.4 Soil analyses after incubation

After incubation, the soil cores were stored at 2 °C until they were extracted with 0.1 M KCl solution (soil to extract ratio 1:4, standardised extraction method of the commissioned laboratory at Leibniz Centre for Agricultural Landscape Research e. V.) and analysed for NH_4^+ and nitrate (NO_3^-) by spectrophotometry according to DIN ISO 14256 with a continuous flow analyser 'CFA-SAN', Skalar Analytical B.V., the Netherlands and for DOC by combustion according to DIN ISO 10694 with an analyser 'RC 612', Leco Instruments GmbH, Germany.

163 2.5 Statistical analysis

All statistical analyses were done using R version 3.2.3 (R Core Team, 2015) with the data of the measuring days under He-O₂ atmosphere. Data from the vessels with the factor combination of 35% WFPS and LOBD with clayey silt were omitted due to technical reasons during sample preparation. For the final period of pure He headspace, some gas concentration data are missing due to logistical reasons. For the loamy sand, this affects all WFPS levels with LOBD (N₂ and N₂O), the treatment 75% WFPS with 320 kg N h⁻¹ (N₂O and CO₂) and for the clayey silt the treatment 35% WFPS without amendment (N₂O and CO₂).

170 To account for repeated measurement of vessels, linear mixed effect models were applied with package 171 'ImerTest' version 2.0-33 (Kuznetsova et al., 2016) for fluxes of each gas type. The three pseudo-replicated fluxes from the N_2 measurements of each vessel were averaged for each day to obtain the same number of 172 observations as for N₂O and CO₂ fluxes. The fixed structure of models included soil type, WFPS, amount of 173 174 digestate, temperature, NO_3 and DOC contents after incubation as well as the fluxes of N_2O (in the model for 175 N_2) and CO_2 (in the models for N_2 , N_2O and $N_2/[N_2+N_2O]$ product ratio). Soil NH_4^+ was omitted since it showed high autocorrelation with the amount of BD applied. The individual soil cores in the vessels were set as random 176 177 effect (nested within the week of incubation and with allowance for a variable slope of the effect each day) with 178 regard to lack of independence of consecutive measurements. The model responses for N2, N2O and CO2 where 179 log transformed ($\ln[value + 1]$) since gas fluxes from soils usually show lognormal distributions (Kaiser et al., 180 1998). The function 'step' was used for automatic backward selection of models based on AIC (Akaike's 'An 181 Information Criterion'). The skewness (γ) was calculated with R package 'moments' version 0.14 (Komsta and 182 Novomestky, 2015) to check residuals for normal distribution and $|\gamma| \le 2$ was assumed as appropriate (West et 183 al., 1995). For mixed effects models, p-values of the ANOVA (type 2) were calculated based on Satterthwaite's 184 approximation) 185 Cumulated gas fluxes were estimated with a bootstrap method using function 'auc.mc' of R package 'flux' 186 version 0.3-0 (Jurasinski et al., 2014) for the R statistical software version 3.2.3 (R Core Team, 2015). In short,

187 the fluxes for the period of aerobic headspace were cumulated in 100 iterations, while for each run 2 fluxes were

188 omitted randomly. Then, the resulting data were used to calculate means and standard deviations.

189 3 Results

190 **3.1** Soil NH_4^+ , NO_3^- and DOC contents

191 The calculated application of NH_4^+ -N from BD per kg soil approximated for the sandy loam 247.0 mg (LOBD) 192 and 494.0 mg (HIBD), and for the clayey silt 266.0 mg (LOBD) and 532.0 mg (HIBD). The NO₃⁻ content of BD 193 was negligible. After incubation, the recovered NH_4^+ -N contents increased with the level of amendment with BD 194 in both soils and were not affected by WFPS, with the exception of treatments of clayey silt with 35% WFPS 195 (Fig. 2). In the loamy sand, the mean amounts of NH_4^+ -N per kg soil ranged from 8.5 to 10.0 mg (no 196 amendment), from 170.4 to 185.6 mg (LOBD) and from 273.7 to 314.0 mg (HIBD). In the clayey silt, NH₄⁺-N 197 contents per kg soil reached only 1.8 to 8.8 mg (no amendment), 89.7 to 98.9 mg (LOBD) and 146.8 to 194.0 mg 198 (HIBD) and, thus, roughly half the amounts of the clayey silt. However, in contrast to the loamy sand, the clayey 199 silt showed also substantial NO₃⁻ contents between 25.7 (35% WFPS without amendment) and 49.8 mg NO₃⁻-N (kg soil)⁻¹ (55% WFPS with LOBD). Negligible amounts of NO₃⁻ were detected in the loamy sand after 200 incubation: except for a mean value of 2.4 mg NO₃⁻-N (kg soil)⁻¹ in the unamended treatment with 75% WFPS, 201 202 the values of all other treatments ranged between 0.2 and 0.5 mg.

The amounts of recovered DOC increased with the application rate of BD, but with different magnitudes for both soils. While mean values from 38.6 (55 % WFPS without amendment) to 500.1 mg DOC per kg soil (75 % WFPS, HIBD) were determined for the loamy sand after incubation, lower mean values from 18.9 (55 % WFPS without amendment) to 358.1 mg (35 % WFPS, HIBD) were found in the clayey silt, where the respective second highest values were considerably lower for both soils (loamy sand: 362.2 mg for 75 % WFPS with LOBD, clayey silt: 105.9 mg for 75 % WFPS with HIBD) (Table 4).

209 **3.2** CO₂ fluxes

210 CO_2 fluxes showed clear differences between the soils: under all combinations of temperature and oxygen, the 211 fluxes were always larger from the loamy sand compared with the clayey silt (Table A1). In general, the mean 212 fluxes from the loamy sand increased with the amount of digestate from 8.3 to 57.6 (aerobic atmosphere at 2°C), from 34.0 to 168.7 (aerobic at 15 °C) and from 11.2 to 87.9 mg CO₂-C m⁻² h⁻¹ (anaerobic at 15 °C), but showed 213 214 no obvious pattern with WFPS. Although the mean fluxes from the clayey silt were also always smallest in the 215 unamended treatments, there was no clear trend of fluxes with the amount of amendment. There was a slight trend of decreasing fluxes with increasing WFPS for the clayey silt. However, the predictive power of WFPS on 216 217 CO₂-C fluxes was minor since it was eliminated during stepwise regression fitting. By contrast, soil type, amount 218 of digestate, temperature as well as the DOC content after the incubation had significant (p < 0.01) effects (Table 219 5).

220 3.3 N₂O fluxes

221 The mean N₂O fluxes from the loamy sand at 2 °C under the He-O₂-atmosphere were virtually zero and, thus, negligible (Fig. 3, Day 2 in Table A2). This was similar at 15°C with the exception of 35% WFPS without 222 digestate (0.1 mg N₂O-N m⁻² h⁻¹, Fig. 3, Day 4 in Table A2). The clayey silt showed much larger fluxes than the 223 loamy sand: even at 2 °C, up to 1.5 mg N₂O-N m⁻² h⁻¹ were detected (55% WFPS with LOBD). After shifting the 224 temperature to 15 °C, the same factor combination had a mean flux of 6.2 mg N₂O-N m⁻² h⁻¹ and the other 225 treatments emitted in mean between 1.0 and 3.0 mg N₂O-N m⁻² h⁻¹ with the exception of incubations with 35% 226 WFPS, where fluxes were smaller. The sand showed weak N2O emissions, independent of temperature and 227 WFPS as well as the amount of BD application. In contrast, the emissions of the clayey silt increased with 228 229 temperature and were highest with intermediate WFPS and amount of BD, i.e. 55% and LOBD, respectively. 230 Surprisingly, at 15 °C, increasing the amount of BD up to HIBD did not increase the observed N₂O efflux; rather it decreased the efflux significantly (p < 0.05, Tuckey's HSD) at 55% and also, but not significantly, at 75% 231 232 WFPS (Fig. 3, Table A2). However, this effect was not noticed at 35% WFPS due to generally low emissions at 233 this moisture level.

According to the linear mixed model for N₂O fluxes in aerobic conditions, WFPS, amount of digestate, temperature, DOC content of soil after incubation and CO₂ fluxes had significant (p < 0.001) effects on N₂O flux (Table 5).

Under anaerobic headspace conditions, the overall highest mean N₂O flux was observed from the clayey silt at 35% WFPS and HIBD (11.7 mg N₂O-N m⁻² h⁻¹). The same soil showed a tendency of decreasing N₂O fluxes with increasing WFPS. Fluxes were largest with LOBD amendment. In the loamy sand, the pure He-atmosphere induced increasing mean N₂O fluxes (up to 1.3 mg N₂O-N m⁻² h⁻¹) with increasing WFPS (Fig. 3, Table A2). So, the anaerobic headspace induced a change only in the loamy sand by increasing emissions.

242 **3.4** N₂ fluxes

From the loamy sand, no N₂ fluxes were detected at 2 °C under He-O₂-atmosphere (Fig. 4, Day 2 in Table A3). 243 Under the same conditions, the clayey silt showed mean fluxes from zero (all incubations with 35% WFPS) to 244 1.4 mg N_2 m⁻² h⁻¹ (all incubations with 75% WFPS). After increasing the temperature to 15 °C, again, the sandy 245 loam released mostly negligible rates of N₂, except for 0.5 mg m⁻² h⁻¹ with 55% WFPS and 320 kg N ha⁻¹ (Fig. 4, 246 Day 4 in Table A3). In contrast, up to 3.8 mg N_2 m⁻² h⁻¹ (75% WFPS with LOBD) were detected in the clayey 247 silt. However, the clayey silt showed also no fluxes in all BD treatments with 35% WFPS. Put simply, 248 249 temperature had a small effect on N_2 emissions from the sandy loam, but WFPS and amount of BD showed no 250 consistent influence. In contrast, the clayey silt emitted clearly increasing emissions with increasing temperature, WFPS and also with the application of BD, where a raise from LOBD up to HIBD at 15 °C, however, resulted in 251 252 slightly, but not significantly (p > 0.05, Tuckey's HSD), decreased fluxes (Fig. 4, Table A3). The summary of the linear mixed model for N₂ fluxes under aerobic conditions revealed significant effects (p < 0.05) of soil type, 253 WFPS, the amount of digestate, temperature, DOC content after incubation and N_2O flux (Table 5). 254

255 After switching the atmosphere to pure He, the N_2 fluxes from the sandy loam increased more than 60-fold. In 256 contrast to aerobic conditions, all measured factor combinations showed mean fluxes from 3.3 (35% WFPS without N) to 35.1 mg N₂ m⁻² h⁻¹ (55% with 320 kg N ha⁻¹), where the fluxes from amended treatments were 257 258 always higher than fluxes from the unamended ones (Fig. 2, Day 5 in Table A3). For the clayey silt, compared with aerobic atmosphere, mean fluxes increased slightly to 1.9 mg N_2 m⁻² h⁻¹ in unamended treatments and more 259 remarkably to 9.3 mg N₂ m⁻² h⁻¹ in amended ones, still not reaching the amounts observed for the sandy loam. 260 261 This implies that the N_2 emissions were increased from both soils under anaerobic headspace conditions, but the 262 loamy sand exhibited a much more intense reaction.

263 **3.5** $N_2/(N_2 + N_2O-N)$ product ratio

264 No clear trend of the product ratio of $N_2/(N_2 + N_2O-N)$ was found for incubations of the loamy sand. However, there was a clear distinction of the ratios for this soil under aerobic and anaerobic atmospheres: while the ratios 265 were close to zero in the former, they were close to 1 in the latter (Fig. 5). In contrast, in the clayey silt the ratios 266 increased with WFPS and were affected by digestate amendment under both the aerobic and the anaerobic 267 atmospheres, where the highest ratios (up to 0.8) were found in treatments without digestate and at least 55% 268 269 WFPS. The digestate-amended treatments showed mostly ratios around or above 0.5, with exception of the 35% 270 WFPS treatments, which had ratios close to zero. According to the linear mixed model, the product ratio under 271 aerobic conditions was affected significantly (p < 0.01) by soil type and the amount of digestate (Table 5).

272 **4 Discussion**

4.1 Increased BD application rate did not increase N₂O and N₂ losses probably due to inhibitory effect of high NH₄⁺ concentrations

275 The overall N₂O fluxes corresponded well with those from other studies with similar incubation conditions and application rates of BD in terms of N ha⁻¹ (Severin et al., 2015; Senbayram et al., 2012; Köster et al., 2015). 276 277 However, the latter studies assumed a distribution of BD into soil by a cultivator, which implies a smaller 278 concentration of BD compared to its occurrence in injection slits. Although we observed differences in N₂O 279 emissions between soils, soil type was not confirmed as a significant effect. Nevertheless, WFPS and 280 temperature, which are well known controllers of N₂O generation (Maag and Vinther, 1999), showed significant 281 influences. Both are physical (by gas diffusion) and biological (by increased metabolic activity and consequently 282 increased O_2 consumption by respiration) drivers for O_2 availability, respectively (Maag and Vinther, 1999; Ball, 2013). Accordingly, the CO_2 flux (resulting from respiration of O_2) generally increased with temperature and 283 284 was also identified as significant by regression selection.

The mean N_2 fluxes of up to 0.5 (loamy sand) and 3.8 mg N m⁻² h⁻¹ (clayey silt) at 15° C (Fig. 5, Table A3) were considerably smaller than the mean fluxes of up to 13.0 mg m⁻² h⁻¹ observed by Köster et al. (2015) during the first five days of their incubation. Although the amount of BD in terms of applied N (250 kg ha⁻¹) was comparable, Köster et al. (2015) used a higher WFPS of 90%, which may have increased the generation of N₂. In contrast to N₂O emission rates, the observed N₂ fluxes depended not only on WFPS, but also on soil type (Table 5), most likely due to the direct influence of soil structure on diffusivity and, thus, the supply with O_2 (Balaine et al. 2016; Butterbach-Bahl et al. 2013). N₂O flux showed also a significant effect during regression selection for N₂. N₂O is the direct precursor of N₂ in denitrification and, hence, the flux of the latter depends on the availability of the former. However, temperature showed no significant effect.

The $N_2/(N_2+N_2O)$ ratios were significantly determined only by soil type and WFPS: while no clear trend was observable for the loamy sand, there was a pronounced effect in the clayey silt (Fig 4). We attribute the lack of a trend in the loamy sand to generally adverse conditions for the formation of N₂O and N₂. Contrary, the influence of WFPS apparently mirrored favourable conditions in the clayey silt (Table 5). Simultaneously, with increasing WFPS, the reduction of N₂O accelerates as an alternative electron acceptor under reduced O₂ supply (Tiedje, 1988). Accordingly, no or rather small fluxes of the investigated gaseous N species were generally found in our presumably well aerated treatments with 35% WFPS.

301 In our study, one treatment (clayey silt, 55% WFPS, LOBD) showed exceptionally large mean N₂O fluxes of up to 7.1 mg N m⁻² h⁻¹ (Fig. 3, Table A2). This could be evidence that injection of such commonly applied amounts 302 of BD-N (i.e., 160 kg N ha⁻¹) may favour much larger losses of N₂O compared to an even distribution of BD in 303 304 soils due to larger substrate concentration in injection slits. However, with higher amendments (i.e. HIBD), we observed surprisingly partially significant (p < 0.05, Tuckey's HSD) reductions of N₂O and a decreasing 305 tendency of N₂ emissions (Table A2, Table A3). In line with this, the amount of BD showed a significant effect 306 307 during the regression selection on N_2O , but not on N_2 fluxes (Table 5). A coherent reason for the rather smaller 308 emissions of highly amended HIBD treatments might be the inhibitory effect of NH_3 on nitrification. 309 Accordingly, Anthonisen et al. (1976) found an inhibition by concentrations from 0.1 to 150 mg NH₃ L⁻¹. The application rate in the treatments with HIBD amounted to approximately 500 mg NH_4^+ -N (kg soil)⁻¹ (Fig. 3) 310 which correspond to 25.8 mg NH₃-N (kg soil)⁻¹ at 15 °C if we use the pH of the BD and assume that all 311 extractable NH_4^+ -N was in solution (Emerson et al., 1975). Hence, we consider this inhibitory effect as the 312 313 reason for the missing increase of N₂O and N₂. Additionally, the amount of NH₄⁺ fixed as NH₃ by soil organic 314 matter increases with pH and, moreover, this fixed NH₃ is not readily extractable by the KCl method we have 315 applied (Kissel et al., 2008). This is consistent with the observation of generally higher N_2O and N_2 fluxes from 316 the clayey silt since clay increases the sorption capacity of soils for NH_4^+ and may, thus, reduce the inhibitory 317 effect on nitrification (Kissel et al., 2008). However, since we mixed the BD with the soil, we would expect a lower NH₃ fixation in tubular injection slits *in situ*, resulting in probably lower N₂O and N₂ fluxes from clayey 318 319 soils.

- An increasing application of BD tended also to decrease the $N_2/(N_2+N_2O)$ ratio, but this effect was also not significant (p > 0.05, Tuckey's HSD). In general, nitrite (NO_2^-) and NO_3^- are preferably reduced compared to
- N_2O during denitrification sequence since the energy yield of each reduction step decreases from NO₃⁻ to N₂O
- 323 (Koike and Hattori, 1975). Additionally, the reaction rate of reduction is higher for NO_3^- and NO_2^- than for $N_2O_3^-$,
- 324 which results in an accumulation of N_2O , if NO_3^- or NO_2^- concentration is not limited (Betlach and Tiedje,
- 325 1981). Hence, increasing application rates of BD increase the availability NO_2^- and NO_3^- from NH_4^+ oxidation
- 326 which, consequently, decreases N₂O reduction. However, in field situations, sooner or later an important fraction
- 327 of this NH_4^+ will be nitrified and can lead to further N_2O and N_2 emissions if the WFPS is at sufficient levels.
- 328 The inhibitory effect is in line with the strong influence of NO_3^- content of the soils after incubation (Table 5).

329 Since NO₂⁻ oxidising bacteria are less resilient against high concentrations of NH₃ than NH₃ oxidising bacteria

- 330 (Anthonisen et al., 1976), the accumulation of NO₂⁻ is likely. This NO₂⁻ protonates then partly to the toxic and
- 331 unstable HNO₂, which drives biological and chemical production of NO and N₂O for detoxification (Venterea et
- al., 2015). Hence, we suggest a dominant role of denitrification, i.e., NO_2^- reduction, in the generation of N_2O
- 333 during our experiment. Nevertheless, Indeed, coupled nitrification-denitrification and bacterial denitrification
- have been found to dominate the production of N_2O directly after application of BD (Köster et al., 2011;
- 335 Senbayram et al., 2009). However, N₂O-N losses were clearly larger than N₂ losses under aerobic headspace in
- the clayey silt, indicating that much of the N gas loss was driven by processes other than canonical
- denitrification. Under the above mentioned conditions, NO-N losses may exceed N₂O losses (Venterea et al.,
- 338 2015), calling for taking account of NO measurements in future studies.
- Notably, in contrast to the clayey silt, no or negligible concentrations of NO_3^- were found in all treatments with loamy sand. Although we have not determined NO_2^- , it was certainly a substantial source for reduction by nitrifier denitrification in this soil, especially during the anaerobic headspace conditions at the end of the incubation. Actually, high NH_4^+ loads in conjunction with alkaline conditions are typical for BD (Möller and Müller, 2012), which favour NO_2^- accumulation and may be the reason for the relatively small NO_3^- recovery in both soils (van Cleemput and Samater, 1995).

345 **4.2** Different effects of soil diffusivity on N₂O and N₂ fluxes

- Apparently, the tested factors affected the N_2O and N_2 fluxes from both soils in a different way. A specific soil
- 347 characteristic that exhibits such a fundamental control on biogeochemical processes such as denitrification is the
- diffusivity for O₂ (Ball, 2013; Letey et al., 1980; Parkin and Tiedje, 1984), which is a main soil characteristic

349 responsible for the appearance of anaerobic microsites. In general, diffusivity integrates the soil porosity, i.e., pore continuity and size as well as WFPS, which control both soil N₂O and N₂ emissions (Balaine et al., 2016; 350 351 Letey et al., 1980; Ball, 2013). Soils with a coarser texture like the loamy sand have a higher proportion of 352 macro-pores and thus a higher gas diffusion compared with fine textured soils like the clayey silt we used 353 (Groffman and Tiedje, 1991). This lets us expect conditions that are more favourable for N_2O and N_2 generation 354 in the latter due to relatively poor diffusion characteristics and, thus, a smaller O_2 supply. Actually, although we incubated the soils at comparable levels of WFPS and BD amendments, the apparent lower diffusivity led to 355 larger N₂O and N₂ production in the treatments with the clayey silt in relation to the loamy sand. 356

357 The role of the distinct diffusivities of both soils is corroborated by our observations of the gas fluxes in anaerobic headspace. With switching the $He-O_2$ atmosphere in the headspace to pure He, the denitrification 358 359 potential can be tested because anaerobicity eliminates respiration processes that use O_2 as electron acceptor 360 (Parkin and Tiedje, 1984). We acknowledge e.g. DNRA and anammox as possible additional sources of N₂O and N₂, respectively, under such conditions, but we were not able to quantify their contribution. The anaerobic 361 362 headspace induced a considerable increase of N₂O fluxes in the loamy sand, but not in the clayey silt. 363 Concurrently, the N₂ fluxes increased in both soils, but pronounced, i.e. more than 60-fold, in the sandy loam. 364 These observed changes resulting from oxygen deprivation imply that, during the previous aerobic conditions, the diffusivity of the sandy loam was too high to allow for a sufficient establishment of anaerobic microsites, 365 366 while the clayey silt ensured a moderate diffusional constraint to maintain hypo-aerobic conditions. However, 367 the large production rates indicate that also the loamy sand harboured the necessary microbial community able to generate N₂ as soon as the atmospheric conditions become favourable. In general, only N₂O fluxes from 368 369 treatments with negligible fluxes during the previous aerobic period increased under anaerobic conditions. This 370 included all treatments with loamy sand and the highly amended clayey silt with 35% WFPS (Fig. 3, Table A2). 371 At the same time, there was a reduction of N_2O fluxes in all other clayey silt treatments. However, when we take 372 a closer look at the simultaneous changes of N_2 fluxes after atmosphere change, virtually all of the respective 373 treatments showed increased rates. Hence, there was an enhanced reduction of N2O to N2, which is reflected in 374 the increased $N_2/(N_2 + N_2O)$ ratio (Fig. 5) and points to intensified reduction of N_2O due to the lack of oxygen 375 (Parkin and Tiedje, 1984). The much larger N₂ fluxes from the loamy sand compared to the clayey silt might 376 have been caused additionally by small NO_3^- availability (Fig. 2) and a high availability of C (Table 4), which 377 promoted the reduction of N_2O to N_2 (Senbayram et al., 2012). Alternatively, the much smaller increase of N_2 378 fluxes from the clayey silt could have resulted from depleted mineral N stocks (NO₃⁻ and NH₄⁺) due to the

379 previous gaseous N losses during the course of incubation. However, the cumulated fluxes of both N_2 and N_2O amounted to a maximum absolute loss of 9.4 ($1\sigma = 0.3$) mg N per kg soil in the clayey silt with LOBD and 55% 380 381 WFPS, which was roughly 3.5% of the calculated NH_4^+ -N applied with BD (Fig. 2). Thus, we found no evidence 382 for any shortage of substrate in the clayey silt during the subsequent anaerobic headspace conditions. On the 383 other hand, the $N_2/(N_2+N_2O)$ ratios increased only slightly (Fig. 5) and, in contrast to the loamy sand, there were 384 still significant N₂O fluxes in the clayey silt (Fig. 3), which point to still sufficient stocks of NO_3^- in the latter 385 (Senbayram et al., 2012). In fact, the NO₃ stock was greater in the clayey silt than in loamy sand after incubation 386 (Fig. 2). Thus, we suggest that the gas fluxes were unaffected by the change to anaerobic headspace in the clayey 387 silt due to already low O₂ concentrations as a result of poor diffusivity. In conclusion, distinct gas diffusivities 388 of both soils can be proposed as the main reason for the differing N_2O and N_2 fluxes.

389 In interaction with soil diffusivity, also respiration affects the aerobicity of a soil matrix by concurrent 390 consumption and formation of O₂ and CO₂, respectively. Similarly to N₂O and N₂ generation by denitrification, 391 respiration depends on the microbial availability of carbon as well. Although anaerobic digestion reduces readily 392 degradable organic matter in BD, a 'labile' fraction usually remains, but the biodegradability of the respective 393 residual organic carbon is variable, depending on the origin of BD (Askri et al., 2015). However, DOC could be 394 used as an approximate indicator for microbial availability of carbon, though not all DOC might be readily 395 degradable (Cook and Allan, 1992). Generally, the DOC contents after our incubation increased with application 396 rate of BD (Table 4), but the DOC contents were always smaller in the clayey silt both in the not amended and 397 especially in the amended treatments. This might reflect a stronger sorption of C and thus a lower availability for 398 respiration in the clayey silt compared to the loamy sand (Kaiser and Guggenberger, 2000). If we compare the DOC recoveries with the cumulated flux rates of CO_2 over the period of aerobic headspace, we find a good 399 400 regression fit ($R^2 = 0.91$, p < 0.001) for both soils (Fig. 6) indicating a sufficient availability of C from BD for 401 respiration and, thus, implicitly also for denitrification (Reddy et al., 1982). Moreover, as increased DOC 402 enhanced respiration (Table A1), it consequently affected O₂ consumption and, thus, also the emergence of 403 anaerobic microsites (Azam et al., 2002). Accordingly, there is also a good correlation between cumulated CO_2 and N₂O + N₂ fluxes for the same period from the clayev silt ($R^2 = 0.93$, p = 0.001), when the treatments with 35 404 405 % WFPS (which showed virtually no N emissions) are omitted (Fig. 7). However, there was no such a correlation for the loamy sand. This confirms the interactive effect of diffusivity (induced by both the soils and 406 WFPS) and C availability on the emissions of N_2O and N_2 . Although CO₂ fluxes were mostly higher in the 407 408 treatments with 320 kg compared to LOBD, this behaviour was not generally reflected in the separate emissions

of N_2O and N_2 which might be a result of the inhibitory effect of high NH_4^+ loads on nitrification (see chapter 4.1). However, the $N_2/(N_2O+N_2)$ ratios implied a tendency of N_2O reduction due to a shortage of alternative electron acceptors like O_2 in the highly amended treatments. Additionally, increasing temperature influenced indirectly the aerobic status of the soils due to increased microbial activity and, hence, respiration (Maag and Vinther, 1999).

414 **4.3** No indications for BD induced short-term priming effect

415 We further checked for a short-term priming effect after amendment with BD as suggested recently by Coban et 416 al. (2015). After balancing cumulated net CO2-C-fluxes (difference between amended and unamended 417 treatments) against the calculated DOC-C application with BD for the period of aerobic headspace, we found no 418 evidence for a short-term priming effect. In the loamy sand with LOBD, between 76% (35% WFPS) and 103% (75% WFPS) of the DOC-C had been respired (data not shown). In the respective treatments with 320 kg N ha⁻¹, 419 the CO₂-C losses ranged from 47% (35% WFPS) to 76% (75% WFPS). By contrast, only between 11% (320 kg 420 421 N ha⁻¹) and 42% (LOBD) has been respired in the clayey silt (both at 55% WFPS). However, if one would 422 consider the period after BD application for a longer time than we would, the light loamy sand may be 423 vulnerable for C losses after BD application than the cloggy clayey silt.

424 **5 Limitations**

425 Our study does not allow for conclusions about the long-term dynamics of N_2 and N_2O emissions after BD 426 application in concentrations similar to injection, since we have data for at most two days of static conditions in 427 terms of temperature and headspace aerobicity. However, it should be emphasised that our aim was to estimate 428 the effect of differing soil environmental conditions on gaseous N₂ losses. Generally, the initial phase, i.e., the 429 first week after fertilizer application, is crucial for N_2O emissions (Dobbie et al., 1999) and most probably also 430 for N_2 because the same processes are involved. Köster et al. (2011; 2015) and Senbayram et al. (2009) observed in incubation experiments N_2O peaks within the first and third day, which indicate a rather immediate reaction 431 also for N₂ at least in vitro. Nevertheless, the former studies recorded a second plateau of N₂O emission 432 433 consistently after around two weeks, though, at very high WFPS. At a lower WFPS of 65%, Senbayram et al. (2009) measured only one peak within two days without a repeated increase later, regardless the amount of 434 435 applied BD. Thus, we assume a single peak shortly after application holds also true for our incubation. Moreover, on the one hand, we observed no changes of N_2O in the clayey silt under anaerobic headspace, which 436

437 suggest no further increase would have awaited if we had extended the incubation period with aerobic 438 headspace. The increased N₂ emissions on the last day showed the potential, which would have arisen if the soil 439 cores had been completely anaerobic. The latter has, however, no implications for mineral soils since such 440 conditions are unlikely to occur *in situ*. On the other hand, the extremely increased N_2 emissions from the loamy 441 sand on the last day verify that this soil permitted abundantly oxygen diffusion, which let us assume no 442 appearances of possible second emission increases in the former aerobic headspace. We assume also the 443 measurements after only 24 hours of anaerobicity in the headspace as representative for the emission potential 444 since microbes associated with the production of N₂O and N₂ in soils are able to react fast to changing 445 environmental conditions by utilising existing enzymes within minutes or by de novo synthesis within 4-8hours (Rudaz et al., 1991). Wang et al. (2011; 2013) showed in similar studies to ours that the emission of N_2 446 447 and N₂O peaked within less than 24 hours after switching their headspace from aerobic to anaerobic conditions, 448 which emphasise our study design as appropriate.

449 6 Conclusions

450 As hypothesised, N₂O and N₂ emissions as well as the N₂/(N₂O+N₂) ratio increased with WFPS, most probably 451 due to restricted supply of O_2 . Contrary to our second hypothesis, the gaseous losses of N_2O and N_2 did not 452 increase with the application rate of BD. This indicates an inhibitory effect of high NH_3 and NH_4^+ 453 concentrations, respectively, on nitrification, which are found typically in biogas digestates (BD). However, the 454 $N_2/(N_2O+N_2)$ ratio tended to decrease with application rate as supposed, probably due to a copious supply with 455 NO₂⁻ and NO₃⁻ from oxidised BD-NH₄⁺. Confirming our third hypothesis, the fine textured clayey silt induced 456 larger gaseous N losses and a higher $N_2/(N_2O+N_2)$ ratio than the coarse loamy sand by the apparent distinct 457 diffusivities of both soils. Overall, there was a larger potential for formation of N₂O in the fine-textured clayey 458 silt compared to the coarse loamy sand after applying high concentrations of BD as they may appear after 459 injection. However, the loamy sand showed a large potential for N₂ formation under anaerobic headspace 460 conditions. Nevertheless, further investigations are needed in regarding the dynamics and the duration of the 461 observed effects and their reliability for field conditions.

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 for sustainable development, NATURE, 528, 51–59, 2015.

647 Table 1: The examined factors soil texture, water-filled pore space (WFPS), and amount (i.e., concentration) of

nitrogen (N) applied with biogas digestate (BD) with their respective levels applied in the present study, resulting in
 18 treatments with three replicates each. The temperature was manipulated consecutively during the incubation.

Factor [n]		Levels	
Soil texture [2]	loamy s	and	clayey silt
WFPS (%) [3]	35	55	75
BD-N (kg ha ⁻¹) [3]	0	160	320
Temperature (°C) [2]	2		15

Table 2: Characteristics of both soils. Texture and mean values with standard deviations (in brackets) for carbon (C, n = 9), nitrogen (N, n = 9), pH (n = 3), bulk density (BD, n = 3) and mineral N (NO₃⁻ and NH₄⁺, n = 3) of both soils in 0–10 cm depth after field sampling.

Texture	$C (mg g^{-1})^a$	N $(mg g^{-1})^{a}$	pH^b	Bulk density $(g \text{ cm}^{-3})^{c}$	$NO_3^{-1} (mg kg^{-1})^d$	$NH_4^+ (mg kg^{-1})^d$
Loamy sand	6.99 (0.29)	0.67 (0.05)	7.2 (0.1)	1.4 (0.0)	1.0 (0.2)	0.6 (0.3)
Clayey silt	10.77 (0.28)	1.19 (0.06)	7.2 (0.0)	1.5 (0.0)	1.8 (0.2)	0.3 (0.2)

^a measured with analyser "Truspec CNS", Leco Instruments GmbH, Germany, performed according to ISO 10694 ("elemental analysis") for C and according to ISO

654 13878 ("elemental analysis") for N

^b measured in H₂O with TitraMaster85, Radiometer Analytical SAS, France, performed according to VDLUFA Methodenbuch, Vol. 1, chap. 5.1.1

656 ^c measured on 250 cm³ soil cores

657 d measured with analyser "CFA-SAN", Skalar Analytical B.V., Netherlands, performed according ISO 14256 the to

- Table 3: Chronological order of the incubated factor combinations. Two different factor combinations with their
- respective repetitions (n = 3) were placed together for each weekly incubation course (cf. Fig. 1). The factors were combined by (1) soil (loamy sand: LS, clayey silt: CS), (2) amount (kg) of applied N from digestate per ha and (3) WFPS (%).

Week	Factor combination 1	Factor combination 2
1	LS - 0 N - 35%	LS - 0 N - 55%
2	LS - 0 N - 75%	LS - 160 N - 35%
3	LS - 160 N - 55%	LS - 160 N - 75%
4	LS - 320 N - 35%	LS - 320 N - 55%
5	LS - 320 N - 75%	CS - 0 N - 35%
6	CS - 0 N - 55%	CS - 0 N - 75%
7	CS - 160 N - 35%	CS - 160 N - 55%
8	CS - 160 N - 75%	CS - 320 N - 35%
9	CS - 320 N - 55%	CS - 320 N - 75%

663Table 4: Mean recovered DOC values from soils after incubation with standard deviations in brackets for the664respective treatments differing in amount of applied biogas digestate (BD) and water-filled pore space (WFPS).

kg digestate-N ha ⁻¹	WFPS (%)	mg DOC	(kg soil) ⁻¹
	(,,,)	Loamy sand	Clayey silt
	35	41.4 (2.7)	18.9 (1.1)
0	55	38.6 (3.1)	19.8 (1.4)
	75	43.7 (1.4)	19.0 (1.8)
	35	197.4 (20.7)	n.a.
160	55	190.5 (19.3)	68.3 (12.7)
	75	362.2 (40.0)	63.2 (9.6)
	35	316.8 (25.3)	358.1 (26.3)
320	55	312.5 (14.3)	94.8 (13.6)
	75	500.1 (33.4)	105.9 (14.8)

665 n.a.: data not available

Table 5: ANOVA table (type 2, p-values calculated based on Satterthwaite's approximation) of the linear mixed effects models for estimated fluxes of N₂, N₂O, N₂/(N₂+N₂O)

product ratio and CO_2 in aerobic He-O₂ atmosphere. Soil type, water-filled pore space (WFPS), amount of digestate, temperature, NO₃⁻ and DOC content of soil after incubation

as well as fluxes of N₂O and CO₂ were set as possible independent variables. The individual soil rings were set as random effect (nested within the respective week and with the

669 allowance for varying slopes for each day of measurements). The random effect was always significant.

					Fixed effects			
Response	Soil type	WFPS	Digestate amount	Temperature	NO ₃ ⁻ post	DOC post	N ₂ O flux	CO ₂ flux
N ₂	0.026	< 0.001	0.008	0.037	Ť	0.001	< 0.001	Ť
N ₂ O	Ť	< 0.001	< 0.001	< 0.001	Ť	< 0.001	*	< 0.001
N ₂ /(N ₂ +N ₂ O)	0.005	0.004	Ť	Ť	Ť	Ť	*	Ť
CO_2	< 0.001	Ť	< 0.001	< 0.001	Ť	0.007	Ť	*

670

671 [†] Variable eliminated during stepwise regression selection

672 * Variable was not included into original regression

- Figure 1: Course of incubation and gas measurements with respect to atmosphere and temperature of the headspace

- after two days of pre-incubation at 2 °C in He/O₂ gas mixture. Gas concentrations of the headspace were determined on five consecutive days, i.e. Monday to Friday in the morning. After the first two measurement days, the headspace temperature was increased from 2 to 15 °C. Additionally, after the fourth measurement day, the aerobic Helium/oxygen gas mixture in the headspace was replaced by a pure Helium atmosphere.

679 Fig. 2: Ammonium and nitrate contents from loamy sand and clayey silt after incubation with different water-filled

680 pore spaces (WFPS, %) and amounts of digestate (0 mL per sample ring: 'No BD', 17.6 mL: 'Low BD' and 35.2 mL:

681 'High BD'). Error bars denote standard deviations. In general, the ammonium content increased with digestate

application with lower amounts detected in the clayey silt. Nitrate was found almost exclusively in the latter soil. For
 comparison, calculated amounts of ammonium applied with biogas digestate are shown by triangles. One treatment
 (*) was omitted from all analyses due to technical reasons.

- 686 Fig. 3: Mean N₂O fluxes (mg N m⁻² h⁻¹) from a loamy sand and a clayey silt incubated under different water-filled pore
- 687 spaces (WFPS, %) with different amounts of digestate (0 mL per sample ring: 'No BD', 17.6 mL: 'Low BD' and 35.2
- 688 mL: 'High BD'). The first till the fourth day of the incubation were measured in an aerobic He-O₂ headspace (with
- 689 two days at 2 °C followed by another two days at 15 °C) while on the fifth day measurements where conducted in an anaerobic headspace with pure He (at 15 °C). Error bars show standard deviations; if bars are not visible, they are
- 691 smaller than the symbols of the means. Under aerobic atmosphere, N₂O fluxes from loamy sand were negligible, while
- fluxes from clayey silt showed an increase with temperature, especially with higher WFPS and intermediate amounts
- 693 of digestate. Under anaerobic atmosphere, mean fluxes from loamy sand increased slightly, but significantly (Tukey's
- HSD, p < 0.05). The fluxes from clayey silt showed no significant differences (Tukey's HSD, p < 0.05) compared to the
- day before, with the exception of 35% WFPS, where mean flux increased strongly in the treatment with 320 kg
- 696 **digestate-N ha**⁻¹.

- 697 Fig. 4: Mean N₂ fluxes (mg m⁻² h⁻¹) from a loamy sand and a clayey silt incubated under different water-filled pore
- 698 spaces (WFPS, %) with different amounts of digestate (0 mL per sample ring: 'No BD', 17.6 mL: 'Low BD' and 35.2
- 699 mL: 'High BD'). The first till the fourth day of the incubation were measured in an aerobic He-O₂ headspace (with
- 700 two days at 2 °C followed by another two days at 15 °C) while on the fifth day measurements where conducted in an
- anaerobic headspace with pure He (at 15 $^{\circ}$ C). Error bars show standard deviations; if bars are not visible, they are smaller than the symbols of the means. The dotted horizontal lines depict the average blank value; single flux rates
- rol and respective lank value were set zero.
 Inder aerobic atmosphere, N₂ fluxes from loamy sand were zero
- 704 or rather negligible, while fluxes from clayey silt show a distinct increase with WFPS and higher fluxes at 15 °C.
- 705 Under anaerobic atmosphere, mean fluxes from loamy sand increased by orders of magnitude, while the fluxes from
- 706 clayey silt increased as well, but more gently compared to the sand.

- Fig. 5: Mean $N_2/(N_2 + N_2O-N)$ product ratio from a loamy sand and a clayey silt incubated under different water-filled
- 708 pore spaces (WFPS, %) with different amounts of digestate (0 mL per sample ring: 'No BD', 17.6 mL: 'Low BD' and
- 709 35.2 mL: 'High BD'). The first till the fourth day of the incubation were measured in an aerobic He-O₂ headspace
- 710 (with two days at 2 °C followed by another two days at 15 °C) while on the fifth day measurements where conducted
- 711 in an anaerobic headspace with pure He (at 15 °C). Error bars show standard deviations; if bars are not visible, they
- are smaller than the symbols of the means. For the loamy sand, there was a clear distinction of the ratios between aerobic and anaerobic atmospheres: while the ratios tended to 0 in the former, they tended to 1 in the latter,
- 714 irrespectively of temperature or amount of digestate. For the clayey silt, ratios increased with WFPS and were highest
- 715 from the unamended treatments under both the aerobic and the anaerobic atmospheres.

- Fig. 6: Regression between DOC recoveries (mg per 100 g soil) after the incubation and the respective cumulated CO_2 emissions (g C m⁻²) during the period of aerobic headspace with their standard deviations and confidence interval (95%). If error bars are not visible, they are smaller than the symbols of the means. Both soils showed increasing 717

- 719 emissions with increasing soil DOC contents as well a good regression fit ($R^2 = 0.91$, p < 0.001).

- 721 722 723 724 Fig. 7: Regression between cumulated CO₂ emissions (g C m⁻²) and the respective cumulated N₂O + N₂ emissions (g N m²) from the clayey silt with WFPS > 35 % during the period of aerobic headspace with their standard deviations
- and confidence interval (95%). If error bars are not visible, they are smaller than the symbols of the means. The proportional increase of CO₂ and the N gas species shows a good regression fit of $R^2 = 0.93$, (p = 0.001).

days of the respective incubation cycle. Different letters after fluxes indicate significant differences (Tukey's HSD, p < 0.05) within each soil and measuring day. Zeros as last digits were omitted.

D	A	T	WFPS	1	mg CO ₂	CO_2 -C m ⁻² h ⁻¹	
Day	Atmosphere	Temperature (*C)	(%)	kg in na	Loamy sand	Clayey silt	
1	He-O ₂	2	35	0	6.8 ± 2.4 cd	0 ± 0 c	
1	He-O ₂	2	35	160	$22 \pm 3.5 \text{ bcd}$	NA	
1	He-O ₂	2	35	320	23.3 ± 9.3 bc	22.8 ± 2.8 ab	
1	He-O ₂	2	55	0	$6 \pm 0.7 \ d$	$4.6 \pm 7.9 \text{ bc}$	
1	He-O ₂	2	55	160	34.4 ± 3.1 b	34.5 ± 11.6 a	
1	He-O ₂	2	55	320	$28\pm3.2\ b$	15.9 ± 3.4 abc	
1	He-O ₂	2	75	0	$9.4 \pm 1.4 \ cd$	0 ± 0 c	
1	He-O ₂	2	75	160	$37.5\pm 6\ b$	15.5 ± 12.1 abc	
1	He-O ₂	2	75	320	68.3 ± 12.1 a	$24.5\pm2.7~a$	
2	He-O ₂	2	35	0	9.8 ± 3.5 c	1.3 ± 1.4 b	
2	He-O ₂	2	35	160	23 ± 3.9 bc	NA	
2	He-O ₂	2	35	320	$30.9\pm2.2~b$	$22.2\pm2.4~a$	
2	He-O ₂	2	55	0	$8.7\pm1.5\ c$	$0.6 \pm 1 \text{ b}$	
2	He-O ₂	2	55	160	$33.4\pm0.9~b$	27.6 ± 12.3 a	
2	He-O ₂	2	55	320	$35.9\pm2.7~b$	14.4 ± 1.9 ab	
2	He-O ₂	2	75	0	$8.3\pm1.5\ c$	$0\pm 0 \; b$	
2	He-O ₂	2	75	160	$31.9\pm3\ b$	13 ± 9.3 ab	
2	He-O ₂	2	75	320	$57.6 \pm 14.8 \ a$	18.3 ± 4 a	
3	He-O ₂	15	35	0	$42.5 \pm 4.5 \text{ c}$	$6.7\pm0.7~b$	
3	He-O ₂	15	35	160	$114.3\pm12.2~b$	NA	
3	He-O ₂	15	35	320	$149.5\pm9.4\ b$	$130.9\pm105~a$	
3	He-O ₂	15	55	0	$41.3 \pm 3.5 \text{ c}$	$3.2\pm0.4\ b$	
3	He-O ₂	15	55	160	$108.7\pm10.1~b$	$57.8 \pm 12.2 \text{ bc}$	
3	He-O ₂	15	55	320	$162.1\pm9.6\ b$	$26.8\pm0.7\ bc$	
3	He-O ₂	15	75	0	$44.1 \pm 9.8 \ c$	$3.2\pm0.7\;b$	
3	He-O ₂	15	75	160	$150.4\pm19~\text{b}$	$26.4\pm11.8\ bc$	
3	He-O ₂	15	75	320	249.7 ± 53.5 a	$35.3 \pm 6 \text{ bc}$	
4	He-O ₂	15	35	0	$48.7\pm6\ c$	$15.1 \pm 4.9 \text{ cd}$	
4	He-O ₂	15	35	160	$114.3\pm6.4\ b$	NA	
4	He-O ₂	15	35	320	156.9 ± 15.4 a	65.7 ± 2.2 a	
4	He-O ₂	15	55	0	48 ± 3.4 c	$4.2\pm0.2\;d$	
4	He-O ₂	15	55	160	$109 \pm 14.4 \text{ b}$	51.2 ± 15.1 ab	
4	He-O ₂	15	55	320	177.7 ± 7.5 a	$26.6 \pm 2.3 \text{ cd}$	

Table A1: Mean CO₂-C fluxes with standard deviations in mg m⁻² h⁻¹ from the loamy sand and the clayey silt, treated with different water-filled pore spaces (WFPS, %), amounts of digestate (kg N ha⁻¹) as well as different temperature regimes ($^{\circ}$ C) under aerobic (He-O₂) and anaerobic (He) atmosphere. Column 'Day' denotes the consecutive measuring 725 726 727

4	He-O ₂	15	75	0	34 ± 7.8 c	6.7 ± 4 d
4	He-O ₂	15	75	160	$168.7\pm0.4~a$	$22.1 \pm 14.8 \text{ cd}$
4	He-O ₂	15	75	320	166.3 ± 23.1 a	$34.1 \pm 5.7 \text{ bc}$
5	He	15	35	0	$11.2\pm0.6~d$	NA
5	He	15	35	160	$54.8\pm9.3\ c$	NA
5	He	15	35	320	$149.3 \pm 3.9 \text{ a}$	$45.8 \pm 2.1 \text{ a}$
5	He	15	55	0	$13.6\pm1.9~d$	$3.4\pm0.6\;c$
5	He	15	55	160	$55.2 \pm 4.4 \text{ bc}$	32 ± 11.4 ab
5	He	15	55	320	$164.5 \pm 3.5 \text{ a}$	$15.2 \pm 10.7 \text{ bc}$
5	He	15	75	0	$20.9\pm2.3~d$	$3.6\pm0.1\ c$
5	He	15	75	160	$75\pm7.3\ b$	$20.6\pm8.5\ bc$
5	He	15	75	320	NA	26.1 ± 2.6 ab

Table A2: Mean N₂O-N fluxes with standard deviations in mg m⁻² h⁻¹ from the loamy sand and the clayey silt, treated with different water-filled pore spaces (WFPS, %), amounts of digestate (kg N ha⁻¹) as well as different temperature regimes (°C) under aerobic (He-O₂) and anaerobic (He) atmosphere. Column 'Day' denotes the consecutive measuring 731 732 733

734 735

days of the respective incubation cycle. Different letters after fluxes indicate significant differences (Tukey's HSD, p < 0.05) within each soil and measuring day. Zeros as last digits were omitted.

Davi	A tracemberro	Temperature		ko N ha ⁻¹	mg N ₂ O-N m ⁻² h ⁻¹		
Day	Aunosphere	(°C)	WFF3 (%)	kg in na	Loamy sand	Clayey silt	
1	He-O ₂	2	35	0	0 ± 0	0 ± 0 c	
1	He-O ₂	2	35	160	0 ± 0	NA	
1	He-O ₂	2	35	320	0 ± 0	$0\pm 0\ c$	
1	He-O ₂	2	55	0	0 ± 0	$0.3\pm0.1\ c$	
1	He-O ₂	2	55	160	0 ± 0	1.7 ± 0.4 a	
1	He-O ₂	2	55	320	0 ± 0	$1.1\pm0.1\ b$	
1	He-O ₂	2	75	0	0 ± 0	$0.4\pm0.1\ c$	
1	He-O ₂	2	75	160	0 ± 0	$1\pm0.1~b$	
1	He-O ₂	2	75	320	0 ± 0	$1\pm0.2\;b$	
2	He-O ₂	2	35	0	0 ± 0	$0\pm 0 \; d$	
2	He-O ₂	2	35	160	0 ± 0	NA	
2	He-O ₂	2	35	320	0 ± 0	$0\pm 0\ cd$	
2	He-O ₂	2	55	0	0 ± 0	$0.3\pm0.1\ bc$	
2	He-O ₂	2	55	160	0 ± 0	$1.5\pm0.6\;a$	
2	He-O ₂	2	55	320	0 ± 0	1.2 ± 0.2 a	
2	He-O ₂	2	75	0	0 ± 0	$0.4 \pm 0.1 \ bc$	
2	He-O ₂	2	75	160	0 ± 0	$1 \pm 0.1 \text{ ab}$	
2	He-O ₂	2	75	320	0 ± 0	1.1 ± 0.2 a	
3	He-O ₂	15	35	0	$0 \pm 0 \ cd$	$0\pm 0\ c$	
3	He-O ₂	15	35	160	0 ± 0 abc	NA	
3	He-O ₂	15	35	320	$0 \pm 0 ab$	$0\pm 0\ c$	
3	He-O ₂	15	55	0	$0 \pm 0 \ bcd$	$0.8\pm0.2\ c$	
3	He-O ₂	15	55	160	$0\pm 0 \ bcd$	$7.1\pm0.9~a$	
3	He-O ₂	15	55	320	0 ± 0 a	$3.5\pm0.7\ b$	
3	He-O ₂	15	75	0	$0 \pm 0 ab$	$0.8\pm0.2\ c$	
3	He-O ₂	15	75	160	$0\pm 0 \; d$	$3.2\pm0.7\ b$	
3	He-O ₂	15	75	320	0 ± 0 cd	$3 \pm 0.9 \text{ b}$	
4	He-O ₂	15	35	0	$0\pm 0 \; b$	$0\pm 0\ c$	
4	He-O ₂	15	35	160	$0\pm 0 ab$	NA	
4	He-O ₂	15	35	320	$0\pm 0 ab$	$0.1\pm0.1\ c$	
4	He-O ₂	15	55	0	$0\pm 0\;b$	1 ± 0.2 bc	
4	He-O ₂	15	55	160	0.1 ± 0.1 a	6.2 ± 1.1 a	
4	He-O ₂	15	55	320	0 ± 0 ab	$3\pm0.8~b$	

4	He-O ₂	15	75	0	0 ± 0 ab	1.1 ± 0.3 bc
4	He-O ₂	15	75	160	$0\pm 0 \; b$	$2.6\pm1\ b$
4	He-O ₂	15	75	320	$0\pm 0 \; b$	$2.2\pm0.9\ b$
5	He	15	35	0	0.1 ± 0	NA
5	He	15	35	160	NA	NA
5	He	15	35	320	0.9 ± 0.1	11.7 ± 2 a
5	He	15	55	0	0.1 ± 0	$0.1 \pm 0 c$
5	He	15	55	160	NA	$5 \pm 1 b$
5	He	15	55	320	1.2 ± 0.7	$1.4\pm0.8\;c$
5	He	15	75	0	1.3 ± 1.6	$0.1 \pm 0 c$
5	He	15	75	160	NA	$1.7 \pm 0.3 \text{ c}$
5	He	15	75	320	NA	1 ± 0.3 c

Table A3: Mean N₂ fluxes with standard deviations in mg m⁻² h⁻¹ from the loamy sand and the clayey silt, treated with different water-filled pore spaces (WFPS, %), amounts of digestate (kg N ha⁻¹) as well as different temperature regimes ($^{\circ}$ C) under aerobic (He-O₂) and anaerobic (He) atmosphere. Column 'Day' denotes the consecutive measuring

737 738 739 740 741 days of the respective incubation cycle. Different letters after fluxes indicate significant differences (Tukey's HSD, p <

741	0.05) within	each soil and	l measuring	day. Zeros	as last digits	were omitted.
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	A tracer here	Temperature		ha N ha ⁻¹	mg N ₂	$m^{-2} h^{-1}$
	Atmosphere	(°C)	WFP3 (%)	kg in na	Loamy sand	Clayey silt
1	He-O ₂	2	35	0	0 ± 0	$0\pm 0\ bc$
1	He-O ₂	2	35	160	0 ± 0	NA
1	He-O ₂	2	35	320	0 ± 0	$0.1\pm0.1\ bc$
1	He-O ₂	2	55	0	0 ± 0	1.5 ± 0.3 a
1	He-O ₂	2	55	160	0 ± 0	1.5 ± 0.3 a
1	He-O ₂	2	55	320	0 ± 0	1.5 ± 0 a
1	He-O ₂	2	75	0	0 ± 0	1.2 ± 1.1 a
1	He-O ₂	2	75	160	0 ± 0	1.5 ± 0.2 a
1	He-O ₂	2	75	320	0 ± 0	$1.7 \pm 0.4 \ a$
2	He-O ₂	2	35	0	0 ± 0	$0\pm 0\ c$
2	He-O ₂	2	35	160	0 ± 0	NA
2	He-O ₂	2	35	320	0 ± 0	$0\pm 0\ c$
2	He-O ₂	2	55	0	0 ± 0	1.3 ± 0.1 a
2	He-O ₂	2	55	160	0 ± 0	$0.8\pm0.5\ b$
2	He-O ₂	2	55	320	0 ± 0	1.3 ± 0.3 a
2	He-O ₂	2	75	0	0 ± 0	$1.4 \pm 0.4 \ a$
2	He-O ₂	2	75	160	0 ± 0	1.4 ± 0.5 a
2	He-O ₂	2	75	320	0 ± 0	$1.4 \pm 0.1 \ a$
3	He-O ₂	15	35	0	0 ± 0 b	$0 \pm 0 e$
3	He-O ₂	15	35	160	$0\pm 0 \; b$	NA
3	He-O ₂	15	35	320	0.1 ± 0.1 ab	$0 \pm 0 e$
3	He-O ₂	15	55	0	$0\pm 0 b$	$1.8 \pm 0.3 \text{ cd}$
3	He-O ₂	15	55	160	$0\pm 0 b$	2.3 ± 0.4 bc
3	He-O ₂	15	55	320	$0\pm 0 b$	2.5 ± 0.2 ab
3	He-O ₂	15	75	0	$0.2 \pm 0.3 a$	$1.5\pm0.2\;d$
3	He-O ₂	15	75	160	$0\pm 0 b$	$3 \pm 0.9 a$
3	He-O ₂	15	75	320	$0\pm 0 b$	2.6 ± 0.8 ab
4	He-O ₂	15	35	0	$0\pm 0 b$	$0 \pm 0 c$
4	He-O ₂	15	35	160	$0.1\pm0.2\;b$	NA
4	He-O ₂	15	35	320	$0\pm 0 \; b$	$0\pm 0\ c$
4	He-O ₂	15	55	0	$0.1\pm0.1\;b$	$1.5\pm0.2\;b$
4	He-O ₂	15	55	160	$0\pm 0 \; b$	$2.9\pm0.6\;a$
4	He-O ₂	15	55	320	0.5 ± 0.4 a	2.8 ± 0.7 a

4	He-O ₂	15	75	0	$0\pm 0 b$	1.3 ± 0.2 bc
4	He-O ₂	15	75	160	$0\pm 0\ b$	$3.8\pm1.6\;a$
4	He-O ₂	15	75	320	$0\pm 0~b$	3.3 ± 0.8 a
5	Не	15	35	0	$3.3 \pm 0.4 \text{ d}$	$0 \pm 0 c$
5	He	15	35	160	NA	NA
5	He	15	35	320	$22.9\pm5.7~b$	$1.8\pm0.1\ c$
5	He	15	55	0	$6 \pm 2.2 \ cd$	1.8 ± 0.2
5	He	15	55	160	NA	9.5 ± 2.7 a
5	He	15	55	320	$35.1\pm8.6~a$	$5.1 \pm 1.8 \text{ bc}$
5	He	15	75	0	$9.2\pm0.4\ c$	$1.9\pm0.1\ c$
5	He	15	75	160	NA	4.8 ± 1.6 bc
5	He	15	75	320	$26.8\pm1.1\ b$	$6.7\pm0.8\;b$