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Influence of long-term mineral fertilization on metal contents and properties of soil samples taken from different locations in Hesse, Germany

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fertilizers over a long period of time will cause an accumulation of metals to high levels, thereby increasing risk to environmental and human health (Huang et al., 2004). Fertilizers and soil amendments can contain significant amounts of potentially hazardous trace elements of geologic or man-made origin. The risk of soil and environmental pollution through the application of these materials to agricultural lands has therefore raised some concern (Raven et al., 1997). Inorganic fertilizers contain elevated quantities of metals like Cd, Pb, As, and other trace elements of environmental relevance (Ajayi et al., 2012; Nicholson et al., 2003).

Metals can be toxic to humans and plants, therefore a long-term application of inorganic fertilizers, organic waste and pesticides to soils, requires a detailed risk assessment of heavy metal accumulation in agricultural lands (Papafilippaki et al., 2007). Metals required by plants include Mn, Fe, Cu, Zn, Mo, and, possibly, Ni. The phytotoxicity of such relatively common metals as Cd, Cu, Hg, and Ni is substantially greater than that of Pb and Zn (Raskin et al., 1994).

Cadmium concentrations in soils in many countries are increasing due to inadvertent additions in fertilizer, biosolids, soil amendments and additions from the atmosphere (McLaughlin et al., 1999). The management of phosphate (P) fertilizer application, both in the short- and long-term, can influence the potential accumulation of Cd in foods. The Cd added to agricultural systems in P fertilizers accumulates over time if application rates are in excess of Cd removal as an effect of long-term P fertilizer application (Grant, 2011). Nicholson et al. (2003) predicated that P fertilizers in particular, are an important source of metals, particularly for Zn, Cu and Cd, entering agricultural soils. Increase of Cu and Zn in soils is associated mainly with NPK fertilizers (Kabata-Pendias, 2011). High application rates of nitrogen fertilizer to agricultural soils resulted in increased accumulation of some heavy metals such as Cd and Pb in agricultural products (Zhou, 2003).

In this study we analyzed soils from long-term fertilizer experiments in Hesse, Germany. We investigated the long-term effects of different NPK fertilizer regimes on soil properties, and soil metal contents. Therefore, the aim of this paper was to answer the

following question: how do (1) soil organic carbon content and cation exchange capacity, (2) soil pH and (3) pseudo and mobile concentrations of metals differ between fertilized and control plots after 14 years of fertilizer application and after 8 years of fertilization termination in Hesse, Germany?

2 Materials and methods

2.1 Study area and fertilizer treatments

This study was carried out at four different locations 210, 260, 360, and 620 m, respectively, above sea level, in a Latin rectangle design, where a wide range of different fertilization regime has been undertaken since 1986 until 2002, in Hesse, Germany. The size of the field trial was 45 × 54 m and divided in 81 plots each of 30 m² size. Soil samples (0–30 cm) were taken in April, August, and November in three replications in the year 2000 (14 years after fertilization application) and in July 2011 from soil profiles (8 years after fertilization termination). *Lolium-Cynosuretum* was the cultivation crop during the study. Some basic physicochemical characteristics of the non-fertilized (control) plots in the studied areas are given in Table 1.

Calciumammoniumnitrate (CAN) (74 % NH₄NO₃, 26 % CaCO₃) as N fertilizer, hyper phosphate (26 % P₂O₅) and superphosphate (16–22 % P₂O₅) as P fertilizer, and magnesia kainit (11 % K₂O, 5 % MgO, 20 % Na, 4 % S) and K₂SO₄ as K fertilization, were used in various levels which refer to different application rates (Table 2).

N fertilization was used in the same proportions in middle of April, May, June, and July; P and K fertilizers were applied in the middle of April as hyper phosphate and magnesia kainit and in the middle of June as superphosphate and K₂SO₄.

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2.2 Analytical procedures

Soil samples were air dried, crushed, homogenized, and sieved through 2 mm screen for the analysis of physical and chemical characteristics and stored in clean HNO₃ treated plastic bottles.

5 The characteristics of soil such as soil pH (DIN ISO 10390, 2005), soil carbonate content (DIN ISO 10693, 1997), particle size distribution (DIN ISO 11277, 2002) were determined for the samples collected. Moreover, total contents of carbon (C_t) were examined on finely ground samples by a CNS analyzer with a Thermal Conductivity Detector (VarioEL III Elementar, Germany). Inorganic carbon (C_i) was calculated from
10 the carbonate content by multiplying the factor 0.1199. The amounts of organic C (C_{org}) resulted from the difference between C_t and C_i. Effective cation exchange capacity (CEC_{eff}) was examined with NH₄Cl extraction according to Trüby and Aldinger (1989) and exchangeable cations were determined with AAS (AA240FS, Varian, USA).

Pseudo total metal contents of soil samples were determined by modified USEPA
15 Method 3051A (USEPA, 1998) with microwave assisted extraction-aqua regia (MAE-AR). The modified program for MAE-AR (0.3 g of soil, 6 mL HCl and 2 mL HNO₃) is described in Öztan and Düring (2012). Mobile fraction of metals was measured after the extraction of soils with ammonium nitrate (NH₄NO₃) (20 g soil, 50 mL 1M NH₄NO₃, shaken for 120 min) (DIN 19730, 2009). Metal concentrations in soil extracts were de-
20 termined by inductively coupled plasma-optical emission spectrometer (ICP-OES; Agilent 720ES) which provides rapid multi-elemental analysis.

The reagents used were all of analytical-reagent grade certified for the impurities. Distilled and deionized water, purified with a Milli-Q plus system (Merck Millipore, Darmstadt, Germany), was used for the experiments. For quality assurance, two certified
25 reference materials (CRMs), “7001” (light sandy soil) and “7004” (loam), from Analytika Co. Ltd, Prague, Czech Republic, were analyzed repeatedly during analysis of the sample set. To designate the amount of possible cross contamination, blank values were determined within each sample series.

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2.3 Statistics

For data analysis SPSS 19.0 for Windows was used. Before testing for differences in analyte concentrations, each element across samples was tested for normality by examination of histograms and residual plots. All data were subjected to analysis of variance (ANOVA) to determine the main effect of each fertilizer variation on soil metal content and soil properties at the $p < 0.05$ probability level. The results of statistical analysis were shown in the figures by alphabetical letters. Variants with the same letters mean no significant differences.

3 Results and discussion

The results show that long-term N, P, and K fertilizer treatments evidently have considerable effects on soil properties. During the study, significant decrease in soil pH was detected due to fertilization (Table 3). Decrease in pH of the surface layer in the fertilizer might be attributed to the nitrification and acidification processes stimulated by continuous application of fertilizers as well as by H^+ released by roots (Liang et al., 2012). Where N fertilizer was applied the pH slightly decreased with respect to the initial value. Significant differences were determined between the control plots and N, P, NP, and NPK applications in soil pH level in April. However a decrease in soil pH was found when compared to control plots in soil August and November pH degrees, no significant differences were seen for N and P treatments in soil-August and for P application in soil-November pH level (Table 3). Continued use of ammonia-based fertilizers can induce soil acidity (Schwab et al., 1990). Results from the study were in agreement with the findings of Tsadilas et al. (2005) who reported that the application of ammonium fertilizer significantly decreased soil pH more than the nitrate treatments. Results from a pot fertilizer experiment which showed that application of NH_4Cl lowered soil pH from 4.51 to 4.07 was presented by Liu et al. (2007). The major mechanism of soil acidification by N fertilization is related to hydrogen ion (H^+) release through nitrification of

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NH_4^+ and subsequent leaching of NO_3^- . The most important acid forming reaction by fertilizers is microbial oxidation of ammoniacal fertilizers (Barak et al., 1997).

Conspicuous differences in soil organic carbon (C_{org}) content and effective cation exchange capacity (CEC) which was caused by fertilization treatments can be seen in Figs. 1 and 2. Significant increase in C_{org} content was determined in April, August and November soil samples. All fertilizer treatments show significant differences when compared to the control plots. In April and August soil samples, no significant differences were determined on C_{org} content between N and P treatments. Differences between N and P variations are evident in November C_{org} content. C_{org} level with P fertilizer was almost two times higher than C_{org} content in N treatment (Fig. 1). Addition of fertilizers to soil influences the chemical composition of soil solution. Increased plant biomass produced by fertilizers results in increased returns of organic material to the soil in the form of decaying roots, litter and crop residues (Haynes and Naidu, 1998). Our results are similar with the findings of Raun et al. (1998) and with the study of Halvorson et al. (1999) who demonstrated a rise in C_{org} with the applied increasing N ratios. Messiga and colleagues (2013) although found no significant affection for total carbon content in 0 to 5 cm soil layer by N application, they noted a tendency for greater total carbon concentration with increasing N application. Liu et al. (2005) reported that C_{org} in surface soil (0–15 cm) layer was 7.7 %, higher in mineral fertilizer treatment than without fertilizer application. Similar results obtained in a long-term experiment showed that low nitrogen rates poorly increased C_{org} content compared to the control (0.14 vs. 0.03 $\text{Mg ha}^{-1} \text{ year}^{-1}$), while the medium and high rates increased it by 0.45 and 0.49 $\text{Mg ha}^{-1} \text{ year}^{-1}$, respectively (Mazzoncini et al., 2011). After 14 years of fertilizer application, there was an increasing trend to C_{org} in April in NPK treatment (6.02 %) contrasting to the control soil content of 0.43 %. This is in accordance with the results of Xie et al. (2011) who found in a pot experimental study the highest C_{org} content was in the NPK (80 N mg kg^{-1} , 35 P mg kg^{-1} , and 60 K mg kg^{-1}) treatment, which was significantly higher than that in the NK (80 N mg kg^{-1} , and 60 K mg kg^{-1}) treatment. Gong and co-workers (2009) determined an increase in C by long-term applications of

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the accelerated acidification of agricultural soils (Bolan and Hedley, 2003). Acidification from soil organic matter accumulation or the direct effects of fertilizers on the soil chemistry can, however, be of significant importance (de Klein et al., 1997). Results from our study were in agreement with the findings of Cakmak et al. (2007) who also reported a negative correlation between soil pH and SOC content. Schwab et al. (1990) determined a decrease in soil pH and an increase in soil organic matter content due to applied high N rate.

In addition to effects of fertilizers on soil properties data were subjected to analysis of variance (ANOVA) for determination of the main effect of each fertilizer variation on soil pseudo total and mobile (NH_4NO_3 -extractable) metal (Cd, Cu, Mn, Pb, and Zn) contents (Figs. 4 and 5). Though total concentrations of metals in soil are used to characterize its base-line elemental composition (parent geological material), pseudo total metal concentration analysis allows to assess soil pollution and to ascertain heavy metal and other pollutant contents in the soil (Ure, 1990). Pseudo total metal contents in soils increased due to application of long-term chemical fertilizer treatments when compared to control plots (Fig. 4). Applications of N, P, NP, and NPK fertilizers increased soil April, August, and November Cd levels compared to non-fertilized soil, but no significant differences were found between P and NP fertilizer variations for pseudo total Cd concentrations (Fig. 4). Significant differences between P and NP applications were obtained for mobile Cd content (Fig. 5). Trace metal enrichment in soils through mineral fertilizer applications is well documented for Cd. Loganathan et al. (1997) reported that in pasture systems in New Zealand, ten years of P fertilizer application caused a marked increase in surface soil pseudo total Cd concentration. Identical findings reported Gray et al. (1999) who specified a significant increase in total Cd concentration in soils under pasture with application of superphosphate fertilizer over a period of 44 years in New Zealand.

No significant differences were determined between N, P, and NP fertilizer variations for pseudo total soil Cu concentrations in April and November, but August Cu level increased significantly with NP application (Fig. 4). Same effects can be seen by mobile

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(NH₄NO₃-extractable) Cu content (Fig. 5). Thomas et al. (2012) reported that total Zn, Cu and Cd concentrations show an increase in the soil concentration with an increase in levels of phosphate fertilizer.

Although NP fertilizer applications show significant differences to N and P treatments for soil-April Mn concentration, no significant differences were obtained by addition of NP when compared to N and P fertilization variations for August and November Mn levels (Fig. 4). Soil mobile Mn concentration increased due to fertilizer application. Significant differences were observed between NPK and NP treatments for April and August, but no significant differences were determined for November Mn concentrations (Fig. 5).

April Pb concentrations increased in N and P fertilizer variations, however no significant effects were found in Pb contents when compared to control plots (Fig. 4). No significant differences were noted on soil August and November Pb mobile concentrations among NP and NPK treatments (Fig. 5), whereas distinct significant differences were obtained for pseudo total contents (Fig. 4). A recent research showed that the amount of hot acid-extractable Pb in the 0 to 30 cm soil interval increased significantly ($p < 0.05$) in accordance with the amount of added P fertilizers (Cakmak et al., 2010). Our results agree with the findings of Atafar and co-workers (2010) who reported that total Cd, Pb, and As concentrations increased in the cultivated soils due to fertilizer application.

Addition of fertilizer at all rates increased soil Zn level when compared to control plots. Soil pseudo total Zn concentration increased from 63.1 mg kg⁻¹ in control treatment to 151.5 mg kg⁻¹, 242.5 mg kg⁻¹ and 185.4 mg kg⁻¹, respectively, in April, August, and November Zn concentrations in NPK application. The highest Zn concentration in soils in August recorded in NPK treatment was probably related to lower soil pH (4.64). Based on the results, Zn content in soils increased significantly by N and P applications, but no significant increase was determined between N and P treatments (Fig. 4). Significant differences between NP and P treatments were found in pseudo total soil

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August Zn concentration (Fig. 4), however, no significant differences were noted for mobile Zn level for the same fertilizer variations (Fig. 5).

Although many grassland fertilizer experiments have been performed worldwide, information about residual effects of fertilizer applications on grassland ecosystem functioning is still rare (Hejzman and Schnellberg, 2009). Fertilization is one of the major paths for metal input to agricultural soils, therefore monitoring of the long term impact of fertilization residual effects is necessary. On this account, in July 2011 mixed soil samples from soil profiles of the fertilized plots (NPK) in the study areas were taken and analyzed for pseudo total and mobile metal concentration and for soil properties such as pH, C_{org} and CEC. Soil pseudo and mobile metal contents (Cd, Cu, Mn, Pb, Zn) and soil pH, C_{org} and CEC of the soil profiles are summarized in Table 4 and detailed distribution of the metals in soil profiles are given in Appendix A.

After 8 years of finalization of the experimental fertilization design an appreciable decrease in pseudo total Cd, Cu, Mn, Pb, and Zn concentration in soil samples was determined. Cd concentration in the studied region was in the range of $0.06\text{--}0.19\text{ mg kg}^{-1}$ with mean value of $0.10 \pm 0.05\text{ mg kg}^{-1}$ in plots without fertilization (Fig. 4), and in NPK treated plots Cd level was between $0.64\text{--}0.78\text{ mg kg}^{-1}$ in August 2000, nevermore, in the soil samples taken from July 2011 Cd level was determined between 0.08 and 0.19 mg kg^{-1} with a mean value of 0.12 mg kg^{-1} (Table 4). The reason of decrease in soil Cd concentration could be due to its mobility and its uptake by plants. Soil Cd is mostly adsorbed at exchange sites and its plant availability is closely related to pH and soil organic matter content (Puschenreiter and Horak, 2000). Detected pseudo total Cd, Cu, Mn, Pb, and Zn concentrations in August 2000 were 0.69 mg kg^{-1} , 72.3 mg kg^{-1} , 2704.4 mg kg^{-1} , 72.8 mg kg^{-1} , and 242.5 mg kg^{-1} , respectively. A decrease of 82.6%, 54.2%, 48.5%, 74.4%, and 56.9%, respectively, in July 2011 soil samples in pseudo total Cd, Cu, Mn, Pb, and Zn contents was determined (Fig. 4 and Table 4). Determined mobile (NH_4NO_3 extractable) Cd, Cu, Mn, Pb, and Zn concentrations in August 2000 were 0.19 mg kg^{-1} , 1.97 mg kg^{-1} , 73.4 mg kg^{-1} , 2.01 mg kg^{-1} , and 6.67 mg kg^{-1} , respectively. In July 2011 soil samples in mobile Cd, Cu, Mn, Pb, and Zn contents,

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an evident decrease of 89.4 %, 94.4 %, 70.4 %, 99.0 %, and 92.5 %, respectively, was found (Fig. 5 and Table 4). The reason behind decrease in metal content could be due to absorption level which may modify the effect of soil pH. Greater proportion of soluble Cd, Cu, Mn, Pb, and Zn was taken up by plants and thus leaving lower amount in the soil. It should be noted that the pH value determines the solubility of metals in the soil and their availability for uptake by plants. Low soil pH may induce metals to be easily soluble and have ion exchange (Golía et al., 2008). In the soil samples taken in August 2000, significant increase in C_{org} and CEC content was determined in April, August and November due to N, P, NP, and NPK treatments when compared to non-fertilized plots (Figs. 1 and 2). Rasool and colleagues reported that, $N_{100}P_{50}K_{50}$ increased soil organic matter content by 21 % (Rasool et al., 2008). After finalization of the fertilization design an appreciable decrease in July 2011 soil C_{org} and CEC was detected (Table 4). Grassland suffered throughout centuries from nutrient depletion through removal of organic material as dung and forage, and subsequent transfer to agricultural land or high productive lowland grassland (Hejcman and Schnellberg, 2009).

4 Conclusions

Monitoring of the long term impact of fertilization is necessary as fertilization is one of the major paths for metal input to agricultural soils. Increase of metal contents in soil due to fertilization should be considered not only as a matter of soil contamination but also a potential risk for human health. Therefore, main objective of our study was to test the effects of different mineral fertilizer variations on soil metal contents of soils after 14 years of fertilizer application and after 8 years of fertilization termination. 14 years of mineral fertilization variations (N, P, NP, NPK) had a significant effect on pseudo total and mobile metal contents of soil (Cd, Cu, Mn, Pb, Zn), when compared to control plots. Apart from the metal contamination in soils, properties of soil such as pH, organic carbon and cation exchange capacity were also affected due to fertilizer treatments. During the study, a significant decrease on soil pH and significant increase

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in soil carbon content and cation exchange capacity was determined. All fertilizer treatments show significant increases in C_{org} content when compared to the control plots. This could be due to the fact that fertilizer application enhanced organic matter decomposition resulting in increased crop residues being returned to the soil. Moreover, positive linear relationships between C_{org} and CEC were determined. As is the case with C_{org} content, N, P, NP, and NPK treatments significantly increased soil CEC content in April, August and November soil samples when compared to control plots. In the soil samples taken from soil profiles for monitoring the residual effects of the fertilization 8 years after the cessation of the fertilizer application, a decrease in pseudo total and mobile Cd, Cu, Mn, Pb, and Zn concentration in soil samples was determined.

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Table 1. Basic physicochemical properties of soil samples (control plots) (mean values of 4 locations).

Properties			April	August	November
pH [CaCl ₂]	min		5.10	5.03	5.10
	max		5.57	5.82	5.45
	mean ± sd		5.32 ± 0.18	5.28 ± 0.32	5.27 ± 0.15
Clay [%]	min		24.1	30.0	27.8
	max		48.4	43.0	43.1
	mean ± sd		32.9 ± 7.9	34.9 ± 4.4	34.8 ± 4.5
C _{org} [%]	min		0.39	0.39	0.40
	max		0.52	0.51	0.49
	mean ± sd		0.43 ± 0.05	0.42 ± 0.03	0.42 ± 0.03
CEC _{eff.} [cmol _c kg ⁻¹]	min		9.21	9.23	9.22
	max		14.2	17.3	17.7
	mean ± sd		11.9 ± 1.83	13.6 ± 2.43	13.5 ± 2.51
Cd	min		0.059	0.057	0.056
	max		0.190	0.190	0.190
	mean ± sd		0.097 ± 0.04	0.093 ± 0.05	0.095 ± 0.04
Cu	min		8.92	9.81	8.99
	max		12.0	18.1	15.9
	mean ± sd		10.3 ± 0.73	11.4 ± 2.22	11.7 ± 2.08
Mn	min		538.9	532.2	533.6
	max		786.0	782.2	785.3
	mean ± sd		627.3 ± 98.6	624.4 ± 99.9	631.0 ± 97.5
Pb	min		8.01	8.12	8.12
	max		19.1	19.4	19.2
	mean ± sd		11.3 ± 4.62	11.4 ± 4.61	11.3 ± 4.54
Zn	min		41.1	42.0	41.0
	max		87.4	87.3	86.6
	mean ± sd		63.5 ± 17.9	63.3 ± 17.6	63.1 ± 17.9

CEC_{eff.}: Cation Exchange Capacity-effective, C_{org.}: Soil organic carbon.

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**Table 2.** Fertilization variations of the study sites.

Fertilizer Variations	kg ha ⁻¹		
	N	P ₂ O ₅	K ₂ O
Control	0	0	0
N	320	0	0
P	0	120	0
NP	320	120	0
NPK	320	120	160

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Table 3. Effect of fertilizers on soil pH (mean values of 4 locations).

Fertilizer	Value	pH (April)	pH (August)	pH (November)
Control	min	5.10	5.03	5.10
	max	5.57	5.82	5.45
	mean ± sd	5.32 ± 0.18 ^a	5.28 ± 0.32 ^a	5.27 ± 0.15 ^a
N	min	4.00	4.99	4.07
	max	5.07	5.16	5.18
	mean ± sd	4.76 ± 0.43 ^b	5.07 ± 0.05 ^{a,b}	4.78 ± 0.44 ^b
P	min	4.53	5.00	4.90
	max	5.16	5.26	5.07
	mean ± sd	4.96 ± 0.25 ^b	5.06 ± 0.07 ^{a,b}	5.00 ± 0.05 ^{a,b,c}
NP	min	4.52	4.66	4.49
	max	5.07	5.05	5.09
	mean ± sd	4.80 ± 0.24 ^b	4.88 ± 0.13 ^b	4.90 ± 0.23 ^b
NPK	min	4.20	4.19	4.35
	max	5.08	5.02	5.09
	mean ± sd	4.78 ± 0.33 ^b	4.64 ± 0.35 ^c	4.71 ± 0.32 ^b

Mean values followed by different superscript letters between fertilizer variations differ at 5% probability.

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Table 4. Soil pseudo and mobile metal contents (Cd, Cu, Mn, Pb, Zn) and soil pH, C_{org} and CEC of the soil profiles in July 2011 in the 4 study areas.

Properties		min	max	mean
pH [CaCl_2]		4.03	5.98	5.14
C_{org} [%]		0.31	4.58	2.09
$\text{CEC}_{\text{eff.}}$ [$\text{cmol}_c \text{kg}^{-1}$]		20.4	23.8	21.3
Aqua regia extractable metals mg kg^{-1}	Cd	0.08	0.19	0.12
	Cu	20.7	54.9	33.1
	Mn	993.0	1867.5	1390.2
	Pb	6.66	38.9	18.6
	Zn	87.1	146.3	104.5
NH_4NO_3 extractable metals mg kg^{-1}	Cd	0.00	0.08	0.02
	Cu	0.03	0.22	0.11
	Mn	1.20	48.2	21.7
	Pb	0.00	0.19	0.02
	Zn	0.01	1.45	0.50

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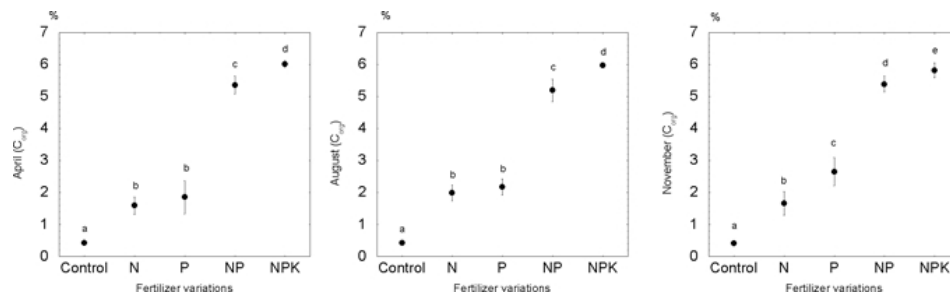


Figure 1. Effect of fertilizers on soil organic carbon content (mean values of 4 study areas–Error bars indicate standard deviation). Mean values followed by different superscript letters between fertilizer variations differ at 5% probability.

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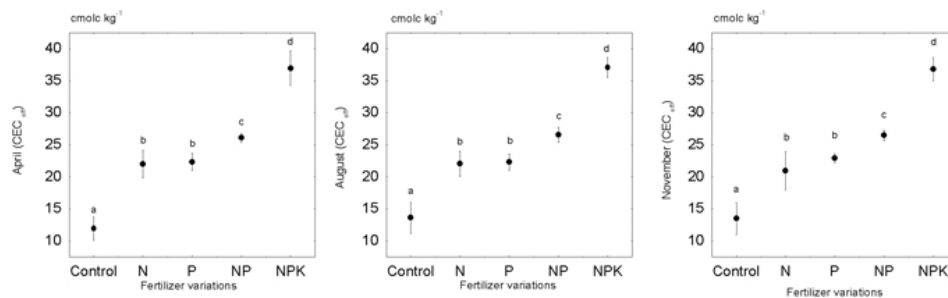


Figure 2. Effect of fertilizers on soil effective cation exchange capacity (mean values of 4 study areas—Error bars indicate standard deviation). (Mean values followed by different superscript letters between fertilizer variations differ at 5 % probability).

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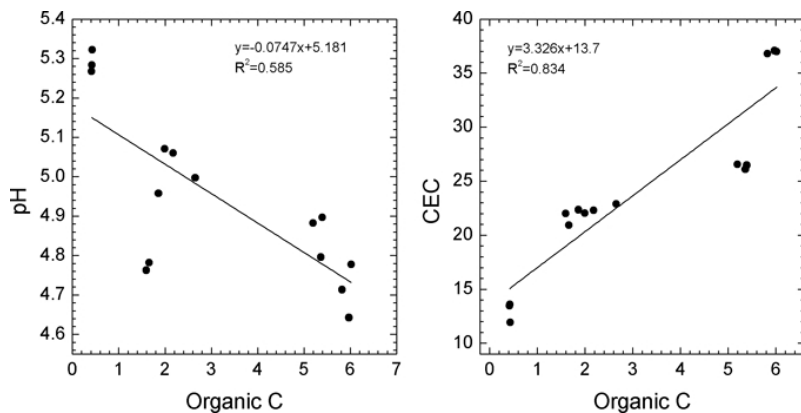


Figure 3. Relationship between organic C and pH and soil CEC_{eff} using mean values of all treatments (mean values of 4 study areas).

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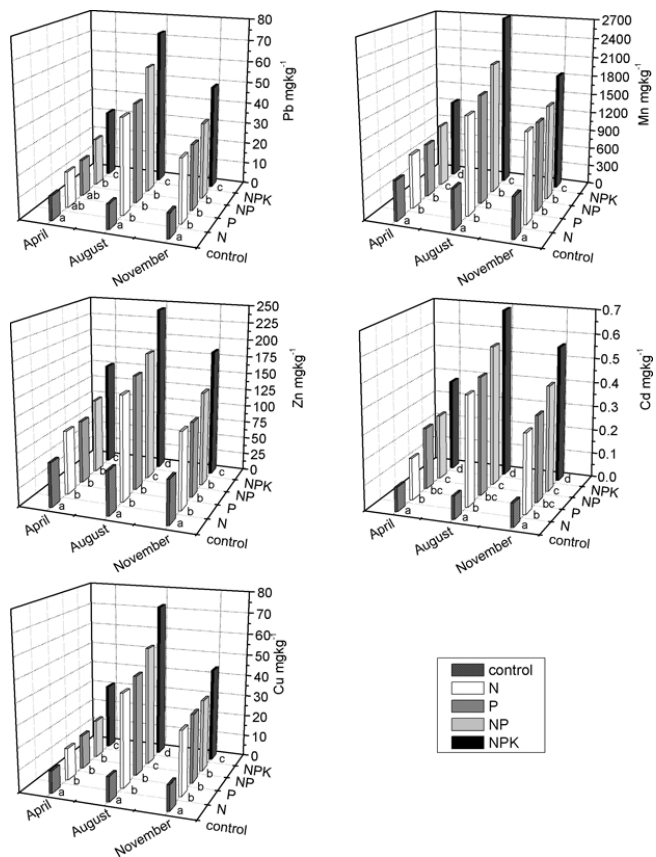


Figure 4. Effect of fertilizer treatments on pseudo total (aqua regia) metal contents in soil samples (Mean values followed by different superscript letters for each month between the fertilizer variations differ at 5 % probability).

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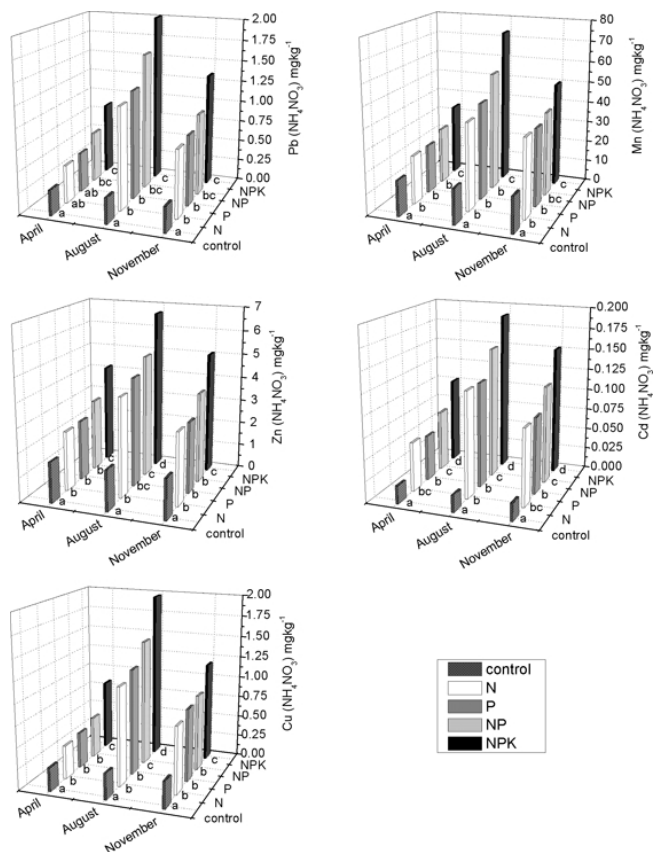


Figure 5. Effect of fertilizer treatments on mobile (NH_4NO_3) metal contents in soil samples (Mean values followed by different superscript letters for each month between the fertilizer variations differ at 5% probability).

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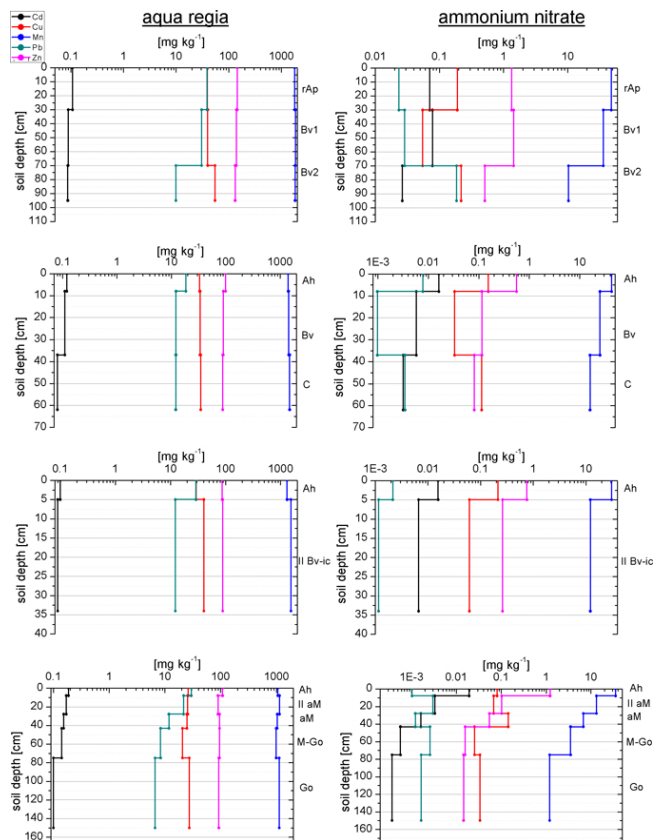


Figure A1. Soil pseudo and mobile metal contents (Cd, Cu, Mn, Pb, Zn) of the soil profiles in 2011.

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